

PRENYLATED ISOFLAVONES FROM *DERRIS SCANDENS*

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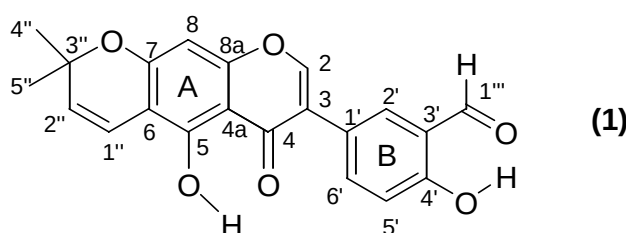
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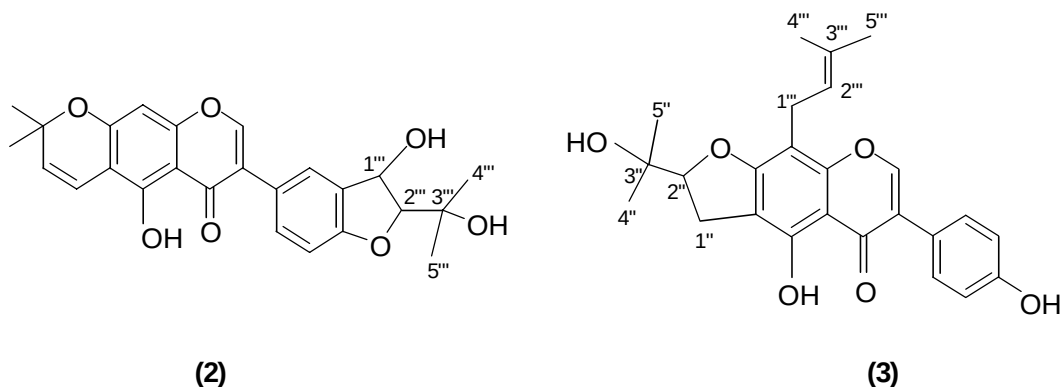
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Abstract- 3'-Formylalpinumisoflavone (**1**), and 2-(1-hydroxy-1-methylethyl)-3-hydroxy-2,3-dihydrofuranoalpinumisoflavone (**2**) were isolated from the stem of *Derris scandens* along with five known isoflavones. The two new compounds were characterised by extensive use of high resolution NMR spectroscopy. Senegalensin (**3**) was reported here for the first time in *D. scandens*

D. scandens (Leguminosae) stem has been widely used as Thai traditional medicine for relieving muscular pain and diuretic.¹ Hypotensive,² immunostimulating activities¹ and smooth muscle stimulant³ were also evaluated. Investigations of the plant from various sources obtained numerous isoflavones.⁴⁻⁸ However, rotenoids were isolated from our earlier investigation of the chemical constituents of the *Derris* species.⁹

We recently examined *D. scandens* stem collected from Chantaburi Province, Thailand during 1997 and reported here two new prenylated isoflavone, 3'-formylalpinumisoflavone (**1**), 2-(1-hydroxy-1-methylethyl)-3-hydroxy-2,3-dihydrofuranoalpinumisoflavone (**2**) along with revised chemical shifts senegalensin (**3**). Structures of the known compounds were principally deduced from NMR spectroscopic data and by comparison with those found in *Lupinus albus* roots¹¹ and previous reports.^{7,8,10} They were identified as lupalbigenin,⁸ lupinisoflavone G,⁸ lupinisol A⁸ and 5,7,4'-trihydroxy-6,8-diprenylisoflavone.⁸





3'-Formylalpinumisoflavone (1) was a colorless solid (mp 205-208°C) with molecular formula $C_{21}H_{16}O_6$ determined by HREIMS ($m/z = 364.0945$). The 1H NMR spectrum revealed a characteristic singlet signal (δ 7.84, H-2) of isoflavone,¹² 2,2-dimethylpyrano ring (δ 1.48, s, 6H, 2xMe; two olefinic protons at δ 5.64 d, $J = 10$ Hz and 6.73 d, $J = 10$ Hz) associated with a singlet H-8 at δ 6.36. Chemical shifts of H-2' (δ 7.82, d, $J = 1.2$ Hz), H-6' (δ 7.66, dd, $J = 8.8, 1.2$ Hz) and H-5' (δ 7.08, d, $J = 8.8$ Hz) were deduced as three aromatic protons having *m*-, *o/m*- and *o*-coupling of B-ring. Moreover, the location of two hydroxy protons at C-5 and C-4' which nearby carbonyl groups suggested magnetically anisotropic down field shifted signals at δ 12.97 and δ 11.10 respectively, indicating intramolecular hydrogen bondings in both cases. The IR spectrum displayed strong absorption at 1650 cm^{-1} which indicated the presence of α,β -unsaturated carbonyl group along with downward shifted absorption at 3200 cm^{-1} due to the chelated hydroxy group. An intense MS ion at m/z 349 belongs to ion $[M-15]^+$ which was typically found in pyranoisoflavone due to loss of methyl group.¹⁰ The ^{13}C NMR spectrum indicated the presence of 15 distinct carbon resonances of the isoflavone moiety (δ 95.0-161.6) including a carbonyl carbon (C-4) at δ 180.4. Carbon resonances of 2,2-dimethylpyrano ring were also apparent from olefinic ^{13}C NMR signals (δ 115.3, C-1'' and δ 128.4, C-2'') associated with a quaternary carbon (δ 78.2, C-3'') and two methyl carbons (δ 28.3, C-4'', 5''). The location of the ring was assigned at C-6 and C-7 by HMBC experiment which revealed correlations of olefinic H-1'' (δ 6.73) with C-5 (δ 156.8), C-6 (δ 105.8) and C-7 (δ 159.8) (Figure 1). Olefinic and aromatic carbons of the isoflavone moiety contained a total of 14 resonances according to the HMQC and DEPT spectra. These were 5 tertiary carbons (δ 95.0, C-8; δ 118.1, C-5', δ 137.2, C-6'; δ 134.2, C-2' δ 152.6, C-2) and 9 quaternary carbons (δ 122.2, C-1'; δ 107.5, C-4a; δ 156.8, C-5; δ 105.8, C-6; δ 159.8, C-7; 157.2, C-8a; 122.6, C-3; δ 120.5, C-3'; δ 161.6, C-4'). The placement of the formyl group at C-3' was confirmed by the HMBC spectrum, in which the formyl proton H-1''' (δ 9.96) showed long range correlation with C-2' (δ 134.2) and C-4' (δ 161.6) in agreement with correlation

of H-2' (δ 7.82) with intense signal of the formyl carbonyl C-1''' (δ 196.5) (Figure 1). Therefore, the structure of **1** was deduced as 3'-formylalpinumisoflavone.

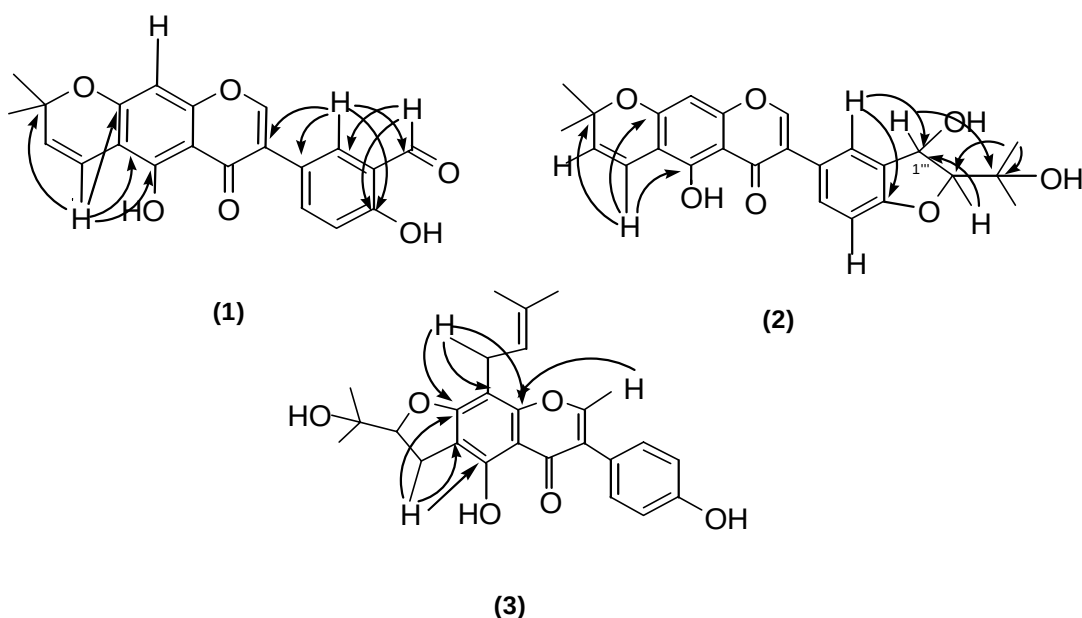


Figure 1. Important H-C correlations observed in HMBC spectra of (1), (2) and (3)

2-(1-Hydroxy-1-methylethyl)-3-hydroxy-2,3-dihydrofuranoalpinumisoflavone (2) was found as colorless gum and showed the location of 2,2-dimethylpyranoisoflavone moiety in ^1H NMR, ^{13}C NMR, DEPT and HMQC spectra like those of compound **(1)** thus, only B-ring substitution remain to be established. Three aromatic protons of the B-ring assembled the same pattern *m*-, *o/m* and *o*-coupling as compound **(1)** with coupling constants 8.3 (*ortho*) and 1.9 (*meta*) Hz in ^1H NMR spectrum. Chemical shifts of H-1''' (δ 5.42, br d, J = 4.6 Hz), H-2''' (δ 4.34, d, J = 4.8 Hz), HO-1''' (δ 2.50, br s) and H-4''' ,5''' (δ 1.31, 1.37, s, 6H) were deduced as dihydrofuran ring with 2-(1-hydroxy-1-methylethyl) and 3-hydroxy substituted groups. The position of H-1''' and H-2''' signals were confirmed from the COSY spectrum in which they demonstrated vicinal couplings of the H-1''' to both H-2''' and HO-1''' protons along with 3-bonds correlation of H-1''' with C-3''' carbon in HMBC spectrum. As evidenced from the 3-bond correlation between aromatic H-2' (δ 7.54) proton of the B-ring and the C-1''' (δ 73.4) of dihydrofuran ring indicated the location of the dihydrofuran ring at 3' and 4' positions of the B-ring (Figure 1).

Senegalensin (3) showed most of the ^1H and ^{13}C signals very similar to previous report.¹³ Careful examination of the COSY, HMQC and HMBC spectra led to some revision of the assignments as showed in the EXPERIMENTAL.

EXPERIMENTAL

General : Mps were uncorrected. Analytical thin-layer separation was carried out on Merck pre-coated silica gel plates (F-254; layer thickness 0.25 mm). Silica gel 60, particle size (0.063-0.260 mm) and less than 0.063 mm were used in column chromatography. UV spectra were taken on a JASCO Uvidex-650 spectrophotometer. IR spectra were recorded on a JASCO FT-IR 700 spectrophotometer. ^1H and ^{13}C NMR spectra were recorded on a Bruker Avance 400 spectrometer at 400 or 100 MHz. EIMS, HREIMS and HRFABMS spectra were determined on a Finnigan MAT 8200 mass spectrometer.

Plant material: The white stems of *D. scandens* were collected from Chantaburi Province, Thailand during summer of 1997, the plant was identified by comparison with the herbarium specimen kept at the Forest Herbarium, Royal Forest Department, Bangkok, Thailand. A voucher specimen (MCDS/1997) is kept at the Department of Chemistry, Faculty of Science, Ramkhamhaeng University.

Extraction and Isolation: Dried and pulverized white stems (1.6 kg) were extracted successively with n-hexane, CHCl_3 and MeOH at refluxing temperature in a Soxhlet extractor (5 days each). After evaporation of the solvents under reduced pressure, the n-hexane (42 g), CHCl_3 (35 g) and MeOH (40 g) extracts were obtained. The extracts were each chromatographed on silica gel column eluting with a gradient of n-hexane, CHCl_3 and finally MeOH giving combined 3'-formylalpinumisoflavone (**1**) (120 mg), 2-(1-hydroxy-1-methylethyl)-3-hydroxy-2,3-dihydrofuranoalpinumisoflavone (**2**) (32 mg) and senegalensin (**3**) (56 mg).

3'-Formylalpinumisoflavone (1): pale yellow solid, mp 205-208°C; IR (CHCl_3 solution) ν_{max} : 3200 (chelated OH), 1654 (conjugated C=O), 1589, 1488, 1463, 1266 cm^{-1} ; UV λ_{max} (MeOH) [$\log \epsilon$]: 289 (2.2) nm; ^1H NMR (CDCl_3 , 400 MHz) δ 1.48 (6H, s, H-4'', 5''), 5.64 (1H, d, $J=10$ Hz, H-2''), 6.36 (1H, s, H-8), 6.73 (1H, d, $J=10$ Hz, H-1''), 7.08 (1H, d, $J=8.8$ Hz, H-5'), 7.66 (1H, dd, $J=8.8, 1.2$ Hz, H-6'), 7.82 (1H, d, $J=1.2$ Hz, H-2'), 7.84 (1H, s, H-2), 9.96 (1H, s, formyl H-1'''), 11.10 (1H, s, HO-4') and 12.97 (1H, s, HO-5); ^{13}C NMR (CDCl_3 , 100 MHz): δ 28.3, 28.3 (C-4'', 5''), 78.2 (C-3''), 95.0 (C-8), 105.8 (C-6), 107.5 (C-4a), 115.3 (C-1''), 118.1 (C-5'), 122.2* (C-1'), 122.6* (C-3), 120.5 (C-3'), 128.4 (C-2''), 134.2 (C-2'), 137.2 (C-6'), 152.6 (C-2), 156.8 (C-5), 157.2 (C-8a), 159.8 (C-7), 161.6 (C-4'), 180.4 (C-4), 196.5 (C-1''') *interchangeable assignments; EIMS m/z (rel.int.) 364 [M] $^+$ (17), 349 [$\text{M}-15$] $^+$ (100), 350 (20), 321 (2), 203 (3), 174 (3), 61 (2.4); HREIMS m/z 364.0945 [M] $^+$ (calcd for $\text{C}_{21}\text{H}_{16}\text{O}_6$, 364.0947).

2-(1-Hydroxy-1-methylethyl)-3-hydroxy-2,3-dihydrofuranoalpinumisoflavone (2) colorless gum; IR (CHCl_3 solution) ν_{max} : 3574 (free OH), 2968, 2909, 1653 (conjugated C=O), 1616, 1583, 1494, 1456, 1249 cm^{-1} ; UV λ_{max} (CHCl_3) [$\log \epsilon$]: 285 (6.6) nm; ^1H NMR (CDCl_3 , 400 MHz): δ 1.31, 1.37 (6H, s, H-4''', 5'''), 1.47 (6H, s, H-4'', 5''), 2.50 (1H, br s, HO-1'''), 5.42 (1H, br d, $J=4.6$ Hz, H-1'''), 4.34 (1H, d,

$J=4.8$ Hz, H-2'''), 5.63 (1H, d, $J=10$ Hz, H-2''), 6.34 (1H, s, H-8), 6.72 (1H, d, $J=10$ Hz, H-1''), 6.93 (1H, d, $J=8.3$ Hz, H-5'), 7.37 (1H, dd, $J=8.3, 1.9$ Hz, H-6'), 7.54 (1H, d, $J=1.9$ Hz, H-2'), 7.83 (1H, s, H-2), 13.08 (1H, s, HO-5); ^{13}C NMR(CDCl_3 , 100 MHz): δ 24.5, 25.7 (C-4''', 5'''), 28.3, 28.3 (C-4'', 5''), 71.3 (C-3'''), 73.4 (C-1'''), 94.9 (C-8), 97.3 (C-2'''), 105.5 $\bullet$ (C-6), 105.6 $\bullet$ (C-4a), 110.3 (C-5'), 115.4 (C-1''), 123.5* (C-3), 123.6*(C-1'), 126.1 (C-2'), 128.2 (C-2''), 129.1 (C-3'), 131.3 (C-6'), 152.6 (C-2), 156.0 (C-5), 156.8 (C-8a), 159.6 (C-7), 160 (C-4'), 180.8 (C-4) $\bullet$ * interchangable assignments; EIMS m/z (rel.int.): 400 (8), 385 (27), 360 (17), 346 (21), 345 (100), 203 (25); HRFABMS m/z : 437.1604 $[\text{M}+\text{H}]^+$ (calcd for $\text{C}_{25}\text{H}_{24}\text{O}_7$, 437.1600).

Senegalensin (3): yellow amorphous solid, IR (KBr) ν_{max} : 3357 (free OH), 1651 (conjugated C=O), 1576, 1515, 1215 cm^{-1} ; UV λ_{max} (MeOH) [$\log \epsilon$]: 273 (3.9) nm; ^1H -NMR (CDCl_3 , 400 MHz): δ 1.21, 1.31 (6H, s, H-4'', 5''), δ 1.67, 1.77 (6H, s, H-4''', 5'''), δ 3.11, 3.25 (2H, dd, $J_{\text{gem}}=15.7, J_{\text{vic}}=9.4, 7.8$ Hz, H-1''), δ 3.37 (2H, d, $J=6.9$ Hz, H-1'''), δ 4.77 (1H, dd, $J=9.4, 7.8$ Hz, H-2''), δ 5.20 (1H, t-like, $J=6.9$, H-2'''), δ 6.83 (2H, d, $J=8.5$ Hz, H-3', 5'), δ 7.38 (2H, d, $J=8.5$ Hz, H-2', 6'), δ 7.84 (1H, s, H-2), δ 12.95 (1H, s, HO-5); ^{13}C NMR ($\text{CDCl}_3+\text{CD}_3\text{OD}$, 100 MHz): δ 17.7, 25.6 (C-4''', 5'''), 21.9 (C-1'''), 24.2, 25.0 (C-4'', 5''), 27.0 (C-1''), 71.7 (C-3'''), 91.2 (C-2''), 102.5 (C-8), 106.7 (C-4a), 108.5 (C-6), 110.6 (C-3', 5'), 121.6 (C-2'''), 122.2 (C-1'), 123.8 (C-3), 130.5 (C-2', 6'), 132.3 (C-3'''), 152.5 (C-2), 153.4 (C-5), 154.0 (C-8a), 157.0 (C-4'), 164.2 (C-7), 181.1 (C-4); EIMS m/z (rel.int.): 422 $[\text{M}]^+$ (100), 420 (35), 407 $[\text{M}-\text{Me}]^+$ (21), 405 $[\text{M}-\text{OH}]^+$ (24), 389 $[\text{M}-\text{H}_2\text{O}-\text{Me}]^+$ (34), 363 $[\text{M}-59]^+$ (19), 349 (71), 335 (35); HRFABMS m/z : 423.1808 $[\text{M}+\text{H}]^+$ (calcd for $\text{C}_{25}\text{H}_{27}\text{O}_6$, 423.1808).

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REFERENCES

1. B. Sriwanthana and P. Chavalittumrong, *J. Ethnopharmacology*, 2001, 125 and references cited.
2. C. Jansakul, A. Srichanbarn, and A. Saelee, *J. Sci. Soc. Thailand*, 1997, **23**, 323.
3. M. Mokkahasmit, W. Ngarmwathana, K. Sawasdimongkol, and U. Permiphaphat, *J. Med. Ass. Thailand*, 1971, **54**, 490.
4. C. P. Falshaw, R. A. Harmer, W. D. Ollis, R. E. Wheeler, V. R. Lalitha, and N. V. Subba Rao, *J. Chem. Soc. (C)*, 1969, 374.
5. A. Pelter and P. Stainton, *J. Chem. Soc. (C)*, 1966, 701.

6. A. P. Johnson, A. Pelter, and P. Stainton, *J. Chem. Soc. (C)*, 1966, 192.
7. M. N. Rao, G. L. D. Krupadanam, and G. Srimannarayana, *Phytochemistry*, 1994, **37**, 267 and references cited.
8. T. Sekine, M. Inagaki, F. Ikegami, Y. Fujii, and N. Ruangrunsi, *Phytochemistry*, 1999, **52**, 87.
9. N. Thasana, M. Chuankamnerdkarn, and S. Ruchirawat, *Heterocycles*, 2001, **55**, 1121.
10. A. K. Singhal, R. P. Sharma, G. Thyagarajan, W. Herz, and S. V. Govindan, *Phytochemistry*, 1980, **19**, 929.
11. S. Tahara, S. Orihara, J. L. Ingham, and J. Mizutani, *Phytochemistry*, 1989, **28**, 901.
12. K. R. Markham and T. J. Mabry, "The Flavanoids" ed. by J. B. Harborne, T. J. Mabry, and H. Mabry, Chapman & Hall, London, 1975, p. 62.
13. J. Wandji, E. Augustin, Z. Nkengfack, T. Fomum, R. Ubillas, K. B. Killday, and M. S. Tempesta, *J. Nat. Prod.*, 1990, **53**, 1425.