

**STEREOCONTROLLED SYNTHESIS OF A HIGHLY  
FUNCTIONALIZED 1,7-DIOXASPIRO[4.4]NONANE DERIVATIVE  
RELATED TO ANTIBIOTIC PSEUROTINS<sup>‡</sup>**

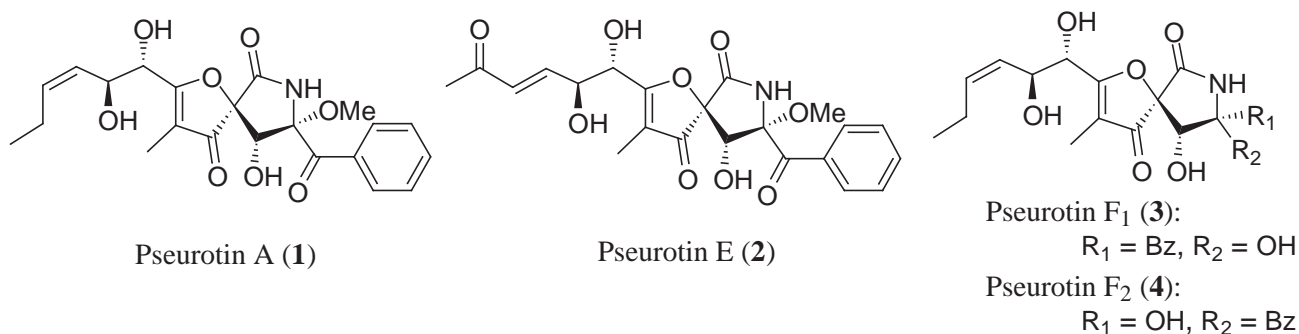
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**Abstract** — For a total synthesis of pseurotins, unusual heterospirocyclic antibiotics, a key intermediate (**5**), having a highly functionalized 1,7-dioxaspiro-[4.4]nonane structure, has been synthesized from diacetone D-glucose.

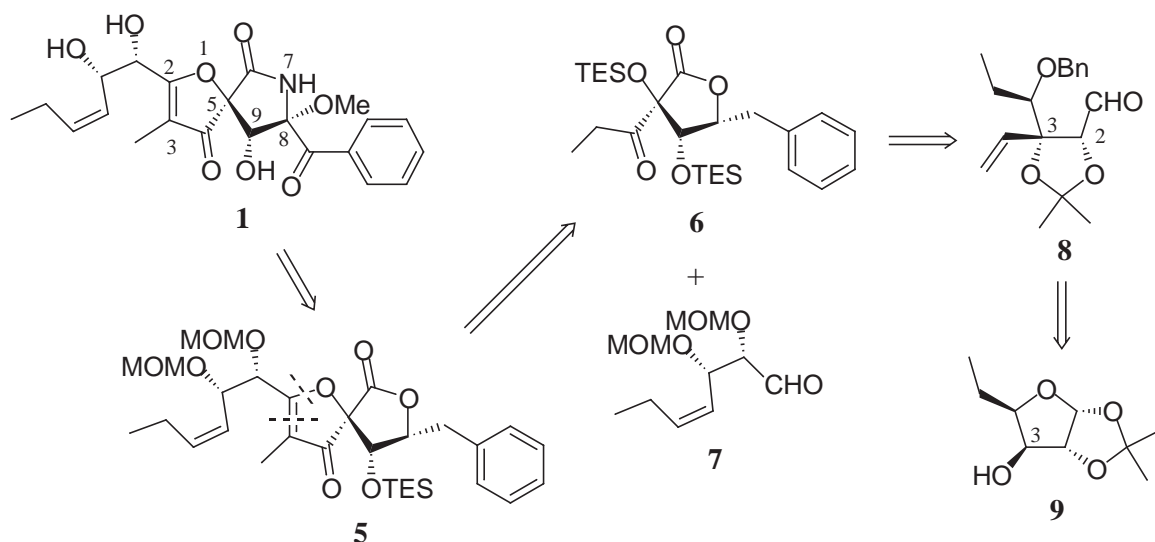
Pseurotins A–E (A **1**, E **2**) are a class of secondary microbial metabolites, which were isolated from the cultures of *Pseudeurotium ovalis* STOLK by Tamm *et al.*<sup>1</sup> Pseurotins F<sub>1</sub>/F<sub>2</sub> (**3**, **4**) were also found from *Aspergillus fumigatus* DSM 6598.<sup>2</sup> The structure of pseurotin A (**1**) including its absolute stereochemistry was determined by spectral analysis, chemical properties, and finally by X-Ray crystal analysis of its dibromo derivative.<sup>1b</sup> This antibiotic possesses a novel highly functionalized 1-oxa-7-azaspiro[4.4]nonane skeleton with five stereogenic centers, and exhibits a strong neurite formation activity to PC-12 pheochromocytoma cells.<sup>3</sup> Synthetic approaches toward the pseurotins have been reported by Tamm *et al.*<sup>4</sup> We have accomplished total synthesis of several  $\gamma$ -lactam natural products.<sup>5</sup> Herein, we report a synthetic approach toward this structurally as well as biologically intriguing pseurotin



**Figure 1.** Structures of pseurotins A, E, and F

<sup>‡</sup> This paper is dedicated to Professor Albert I. Meyers with respect and admiration on the occasion of his seventieth birthday.

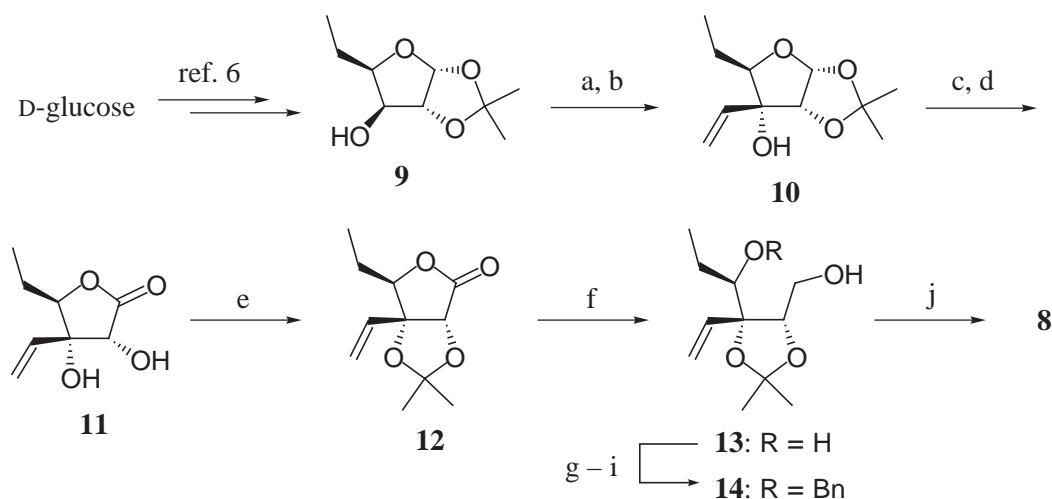
A (**1**). In this study, we synthesized a functionalized heterospirocyclic derivative (**5**), which possesses a 1,7-dioxaspiro[4.4]nonane core framework including four stereogenic centers required for pseurotins synthesis. Our retrosynthetic analysis is outlined in Scheme 1. We envisioned that the target (**1**) would be obtained from the highly functionalized heterospirocyclic derivative (**5**) by oxidations of C-8 and the benzylic carbon and transformation of the  $\gamma$ -lactone to a  $\gamma$ -lactam. Construction of the 3(*2H*)-furanone structure in **5** would arise from the aldol reaction of a keto  $\gamma$ -lactone (**6**) with an aldehyde (**7**) followed by ring closure. The  $\gamma$ -lactone (**6**) could be obtained from an aldehyde (**8**), of which two stereogenic centers (C-2 and C-3) are corresponding to C-9 and C-5 in **1**, respectively. The intermediate (**8**) could be prepared from the known 5,6-dideoxy-1,2-*O*-isopropylidene- $\alpha$ -D-xylo-hexofuranose (**9**)<sup>6</sup> via stereoselective vinyl Grignard addition to a carbonyl group at C-3 derived from **9**.



Scheme 1

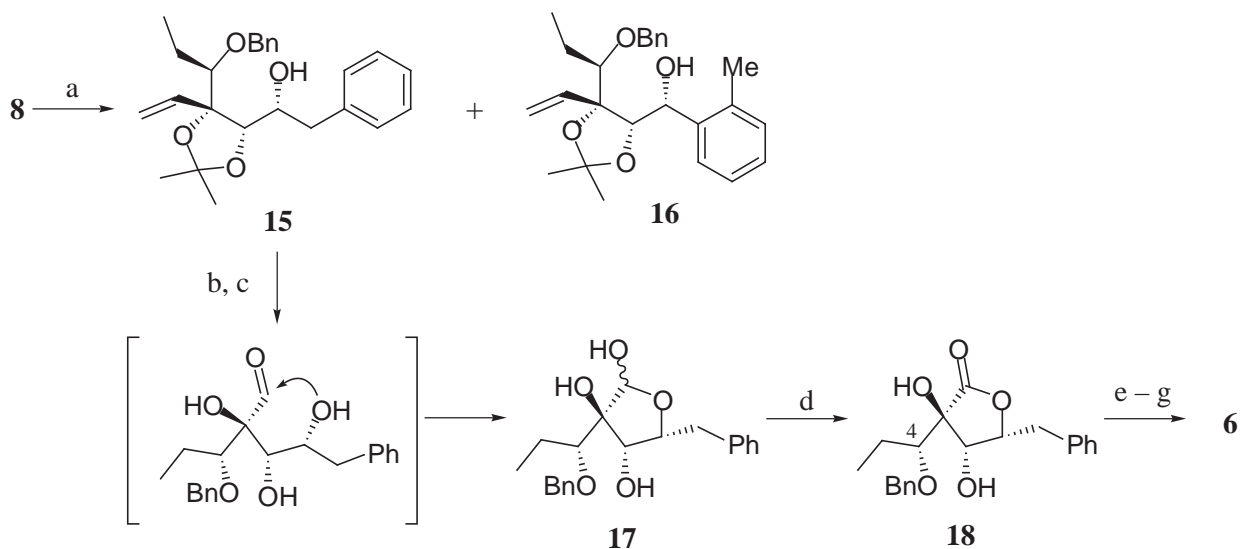
The synthesis of **8** began with **9**, which was prepared from diacetone D-glucose in the known five steps.<sup>6</sup> Oxidation of **9** with pyridinium chlorochromate (PCC) followed by the vinyl Grignard addition provided the adduct (**10**) in 83% yield as a single isomer (Scheme 2).<sup>7</sup> The nucleophile attacked exclusively from the convex face of the bicyclic structure of the C-3 keto derivative. Acidic hydrolysis of **10** and chemoselective oxidation of the resultant hemiacetal with *N*-iodosuccinimide (NIS) in the presence of *n*-Bu<sub>4</sub>N<sup>8</sup> gave  $\gamma$ -lactone (**11**). The *cis* diol in **11** was protected as an acetonide to provide bicyclic  $\gamma$ -lactone (**12**). Reduction of **12** with LiAlH<sub>4</sub> provided a ring-opened diol (**13**), which was converted into a primary alcohol derivative (**14**) by a protection–deprotection sequence. Dess–Martin oxidation<sup>9</sup> of **14** produced quantitatively the aldehyde (**8**). The introduction of a benzyl moiety into **8** was performed by a Cu(I)-mediated Grignard reaction to afford the adduct (**15**) in 89% yield as a single diastereomer along with a small amount of *ortho*-tolyl adduct (**16**) (Scheme 3). When the reaction was conducted in the absence of CuBr·Me<sub>2</sub>S, **16** was predominantly produced.<sup>10,11</sup> Ozonolysis of **15** and successive acidic hydrolysis of the acetonide group formed a five-membered hemiacetal, giving **17**. Chemoselective

oxidation of the  $\gamma$ -lactol (**17**) with NIS in the presence of *n*-Bu<sub>4</sub>NI provided  $\gamma$ -lactone (**18**). The two hydroxy groups in **18** were protected as triethylsilyl (TES) ethers, and deprotection of the O-4 benzyl group by hydrogenolysis followed by Dess–Martin oxidation provided the keto lactone (**6**), the substrate for the aldol reaction.



*Reagents and conditions:* (a) PCC, MS4A, CH<sub>2</sub>Cl<sub>2</sub>; (b) vinylMgBr, THF, -18 °C (83% for 2 steps); (c) 80% aqueous AcOH, 80 °C; (d) NIS, *n*-Bu<sub>4</sub>NI, CH<sub>2</sub>Cl<sub>2</sub> (95% for 2 steps); (e) Me<sub>2</sub>C(OMe)<sub>2</sub>, CSA, acetone, reduced pressure (*ca.* 300 hPa), 40 °C (79%); (f) LiAlH<sub>4</sub>, THF, 0 °C (91%); (g) TrCl, DMAP, pyr, reflux; (h) BnBr, NaH, DMF; (i) CSA, MeOH, EtOAc (78% for 3 steps); (j) Dess–Martin periodinane, CH<sub>2</sub>Cl<sub>2</sub> (99%).

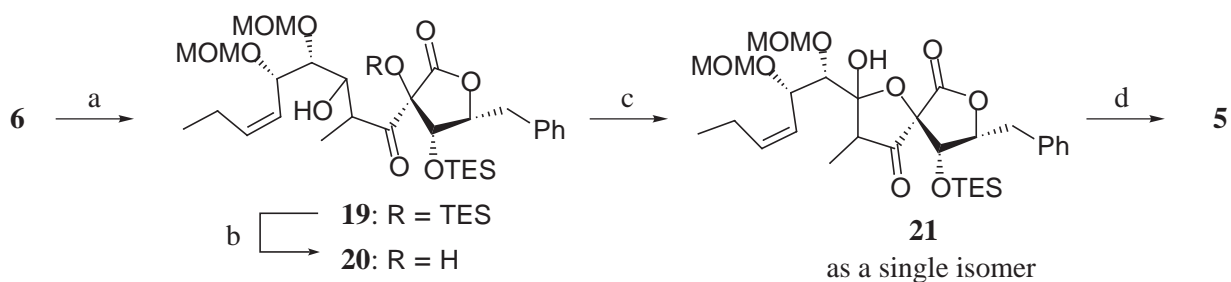
**Scheme 2**



*Reagents and conditions:* (a) BnMgCl, CuBr·Me<sub>2</sub>S, THF, Me<sub>2</sub>S, 0 °C (89% for **15** and 3% for **16**); (b) O<sub>3</sub>, CH<sub>2</sub>Cl<sub>2</sub>, -78 °C; Me<sub>2</sub>S; (c) 60% aqueous TFA; (d) NIS, *n*-Bu<sub>4</sub>NI, CH<sub>2</sub>Cl<sub>2</sub> (77% from **15**); (e) TESOTf, pyr, 50 °C (100%); (f) H<sub>2</sub>, 10% Pd on C, EtOAc (93%); (g) Dess–Martin periodinane, CH<sub>2</sub>Cl<sub>2</sub> (100%).

**Scheme 3**

We then performed the construction of the 3(2*H*)-furanone structure in **5** (Scheme 4). The aldol reaction of **6** with 4 molar equiv of **7**,<sup>12</sup> using 1 molar equiv of potassium bis(trimethylsilyl)amide (KHMDS) in THF at  $-78\text{ }^{\circ}\text{C}$ , proceeded smoothly to produce the aldol (**19**) with a high level of diastereoselectivity. The newly introduced stereochemistries were not determined. Treatment of **19** with hydrogen fluoride–pyridine complex in pyridine<sup>13</sup> cleaved selectively the TES ether of the tertiary alcohol to afford **20**. Oxidation of **20** followed by dehydration of the resultant spiro-ketal (**21**) gave the desired heterospirocyclic compound (**5**),<sup>14</sup> which would be a key intermediate for the pseurotins synthesis.



*Reagents and conditions:* (a) KHMDS, THF,  $-78\text{ }^{\circ}\text{C}$ ; **7**; (b) HF·pyr, pyr, THF; (c) Dess–Martin periodinane,  $\text{CH}_2\text{Cl}_2$  (49% from **6**); (d)  $\text{SOCl}_2$ , pyr,  $0\text{ }^{\circ}\text{C}$  (97%).

**Scheme 4**

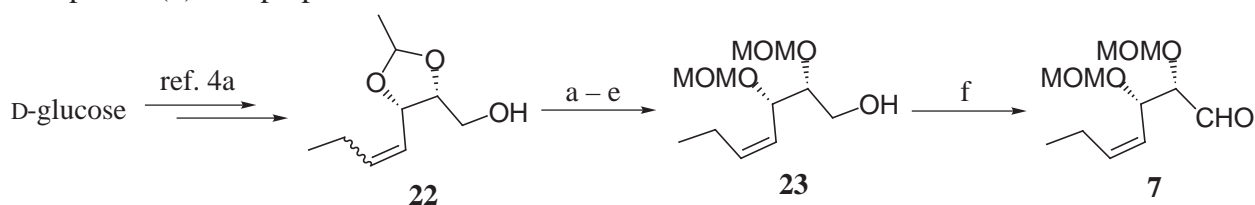
In conclusion, the stereoselective synthesis of the highly functionalized 1,7-dioxaspiro[4.4]nonane (**5**) from D-glucose has been accomplished. The total synthesis of pseurotins is in progress.

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- All new compounds were fully characterized by spectroscopic means [ $^1\text{H}$  (300 MHz in  $\text{CDCl}_3$ ) and

$^{13}\text{C}$  (75 MHz in  $\text{CDCl}_3$ ) NMR, IR] and gave satisfactory HRMS. Yields refer to homogeneous samples purified by chromatography on silica gel.

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10. The reaction of benzylmagnesium chloride with aldehydes was reported to produce the *ortho*-tolyl adducts *via* Mg(II)-mediated six-membered transition states.<sup>11</sup> We considered that the addition of the Cu(I) salt suppressed the formation of the six-membered transition state and the expected 1,2-addition occurred.
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12. Compound (**7**) was prepared as follows.



*Reagents and conditions:* (a) MPMCl, NaH, DMF; (b) Amberlite IR-120 ( $\text{H}^+$ ), MeOH (91% for 2 steps); (c) MOMCl, *i*-Pr<sub>2</sub>NEt,  $\text{CH}_2\text{Cl}_2$ ; (d) DDQ,  $\text{CH}_2\text{Cl}_2$ ,  $\text{H}_2\text{O}$ ; (e) separation of the geometrical isomers on silica gel (EtOAc/hexane, 1:4) (**23**: 88% for 2 steps, *E*-isomer: 7% for 2 steps); (f)  $(\text{COCl})_2$ , DMSO,  $\text{CH}_2\text{Cl}_2$ ;  $\text{Et}_3\text{N}$ ,  $-78\text{ }^\circ\text{C}$  (89%).

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14. Compound (**5**) was obtained as white crystals: mp 56.0–56.3  $^\circ\text{C}$ ; TLC  $R_f$  0.20 (EtOAc/hexane, 1:5);  $[\alpha]_D^{21} +23.0\text{ }^\circ$  ( $c$  2.09,  $\text{CHCl}_3$ ); IR (neat) 2960, 2880, 1790, 1715, 1640, 1150, 1040  $\text{cm}^{-1}$ ;  $^1\text{H}$  NMR (300 MHz,  $\text{CDCl}_3$ )  $\delta$  0.51–0.69 (m, 6H,  $\text{Si}(\text{CH}_2\text{CH}_3)_3$ ), 0.94 (t, 9H,  $J = 8.1$  Hz,  $\text{Si}(\text{CH}_2\text{CH}_3)_3$ ), 1.01 (t, 3H,  $J = 7.6$  Hz,  $\text{CH}_3$  of the side chain at C-2), 1.79 (s, 3H,  $\text{CH}_3$  at C-3), 2.09–2.27 (m, 2H, H-5, 5' of the side chain at C-2), 3.19 (dd, 1H,  $J = 2.7, 15.4$  Hz,  $\text{CHHPh}$ ), 3.31, 3.39 (2s, each 3H,  $\text{OCH}_3 \times 2$ ), 3.65 (dd, 1H,  $J = 11.0, 15.4$  Hz,  $\text{CHHPh}$ ), 4.57–4.68 (m, 5H, H-9,  $\text{OCH}_2\text{O} \times 2$ ), 4.71–4.78 (m, 1H, H-2 of the side chain at C-2), 4.83 (ddd, 1H,  $J = 2.7, 7.3, 11.0$  Hz, H-8), 5.00 (d, 1H,  $J = 7.3$  Hz, H-1 of the side chain at C-2), 5.35 (dd, 1H,  $J = 9.5, 11.0$  Hz, H-3 of the side chain at C-2), 5.78 (dt, 1H,  $J = 11.0, 7.3$  Hz, H-4 of the side chain at C-2), 7.21–7.35 (m, 5H,  $\text{C}_6\text{H}_5$ );  $^{13}\text{C}$  NMR (75 MHz,  $\text{CDCl}_3$ )  $\delta$  4.5  $\times$  3, 5.7, 6.6  $\times$  3, 14.0, 21.2, 36.3, 55.8, 56.0, 71.1, 72.7, 74.2, 82.7, 89.1, 94.5, 95.1, 114.9, 125.5, 126.6, 128.5  $\times$  2, 129.3  $\times$  2, 137.6, 138.6, 166.3, 183.4, 195.4; HRMS calcd for  $\text{C}_{30}\text{H}_{43}\text{O}_8\text{Si}$  ( $\text{M}^+ - \text{OMe}$ )  $m/z$  559.2727, found 559.2722.