

HETEROCYCLES, Vol. 76, No. 2, 2008, pp. 1069 - 1074. © The Japan Institute of Heterocyclic Chemistry
Received, 8th May, 2008, Accepted, 12th June, 2008, Published online, 19th June, 2008. COM-08-S(N)113

STEREOSELECTIVE SYNTHESIS OF MAITOTOXIN GHI-RING SYSTEM HAVING A 1,2-DIOL SIDE CHAIN

Masanori Nagatomo and Tadashi Nakata*

Department of Chemistry, Faculty of Science, Tokyo University of Science, 1-3
Kagurazaka, Shinjuku-ku, Tokyo 162-8601, Japan

E-mail address: nakata@rs.kagu.tus.ac.jp

Abstract – The stereoselective synthesis of maitotoxin GHI-ring system having a 1,2-diol side chain was accomplished via successive SmI_2 -induced reductive cyclizations of β -alkoxyacrylate–aldehyde and optically active (*E*)- β -alkoxyvinylsulfoxide–aldehyde, dihydroxylation of α,β -unsaturated δ -lactone, and Sharpless asymmetric dihydroxylation of the side chain.

Maitotoxin (MTX; **1**, Figure 1), one of the toxins implicated in ciguatera, was first discovered from the viscera of coral reef fish¹ and later isolated from the epiphytic dinoflagellate *Gambierdiscus toxicus*.² MTX is the most toxic and largest natural product (molecular weight 3422) known to date, except for biopolymers.³ The full structure of MTX, including a partial stereochemical assignment, was established by the Murata-Yasumoto group.⁴ The relative stereochemistry of the remaining acyclic parts and the absolute structure of MTX were determined independently by the Tachibana⁵ and Kishi⁶ groups.⁷ The giant structure of MTX contains 32 fused ether rings, 28 hydroxy groups, 21 methyl groups, 2 sulfates, and 98 chiral centers. The unusual complex structure and potent toxicity have attracted the attention of both chemists and biologists. Partial syntheses of MTX have been reported by Tachibana⁵, Kishi,⁶ Nicolaou,⁸ and our groups⁹ so far. We have already reported the stereoselective synthesis of the BCDE-ring,^{9a} GHI-ring,^{9c} WXYZA'-ring,^{9f} and C'D'E'F'-ring having a side chain.^{9d,e} Our previous synthesis of the GHI-ring was accomplished starting from the G-ring.^{9c} We now report an alternative and efficient synthesis of the GHI-ring system having a 1,2-diol side chain starting from the I-ring.

This paper is dedicated to Prof. Dr. Ryoji Noyori on the occasion of his 70th birthday.

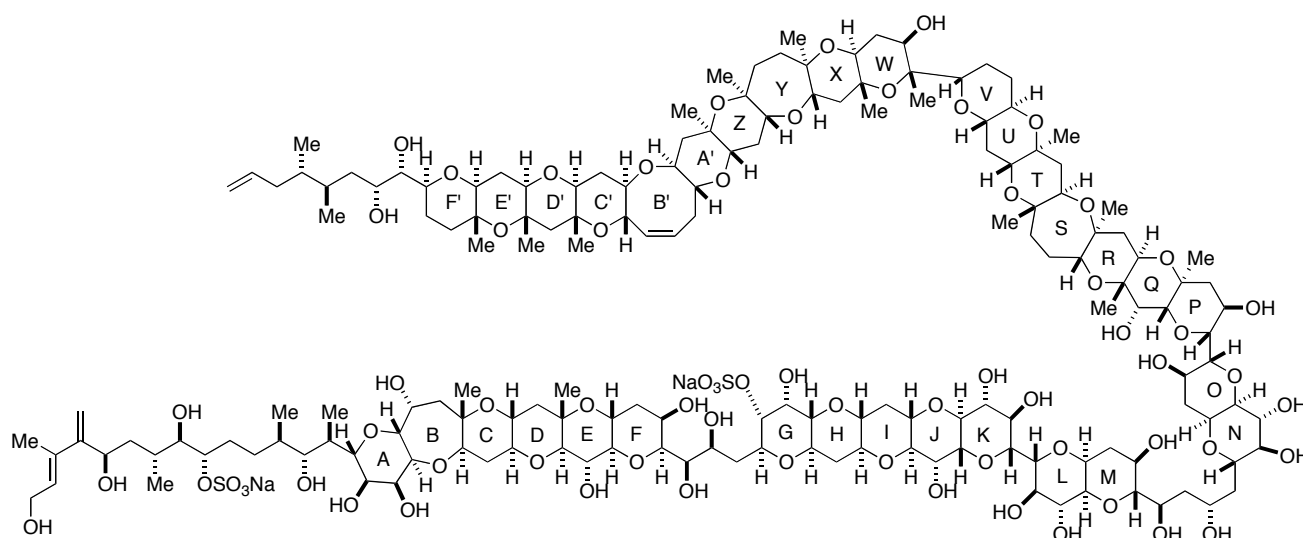
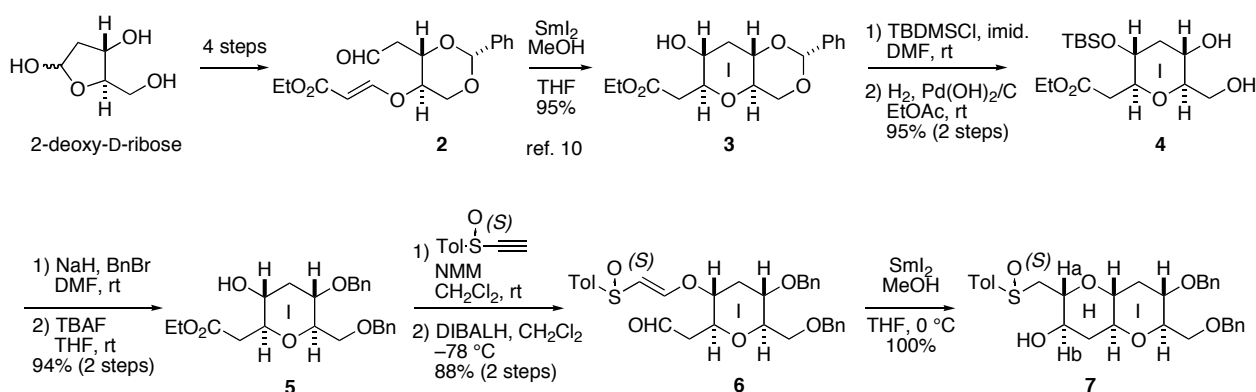


Figure 1. Structure of maitotoxin (1).

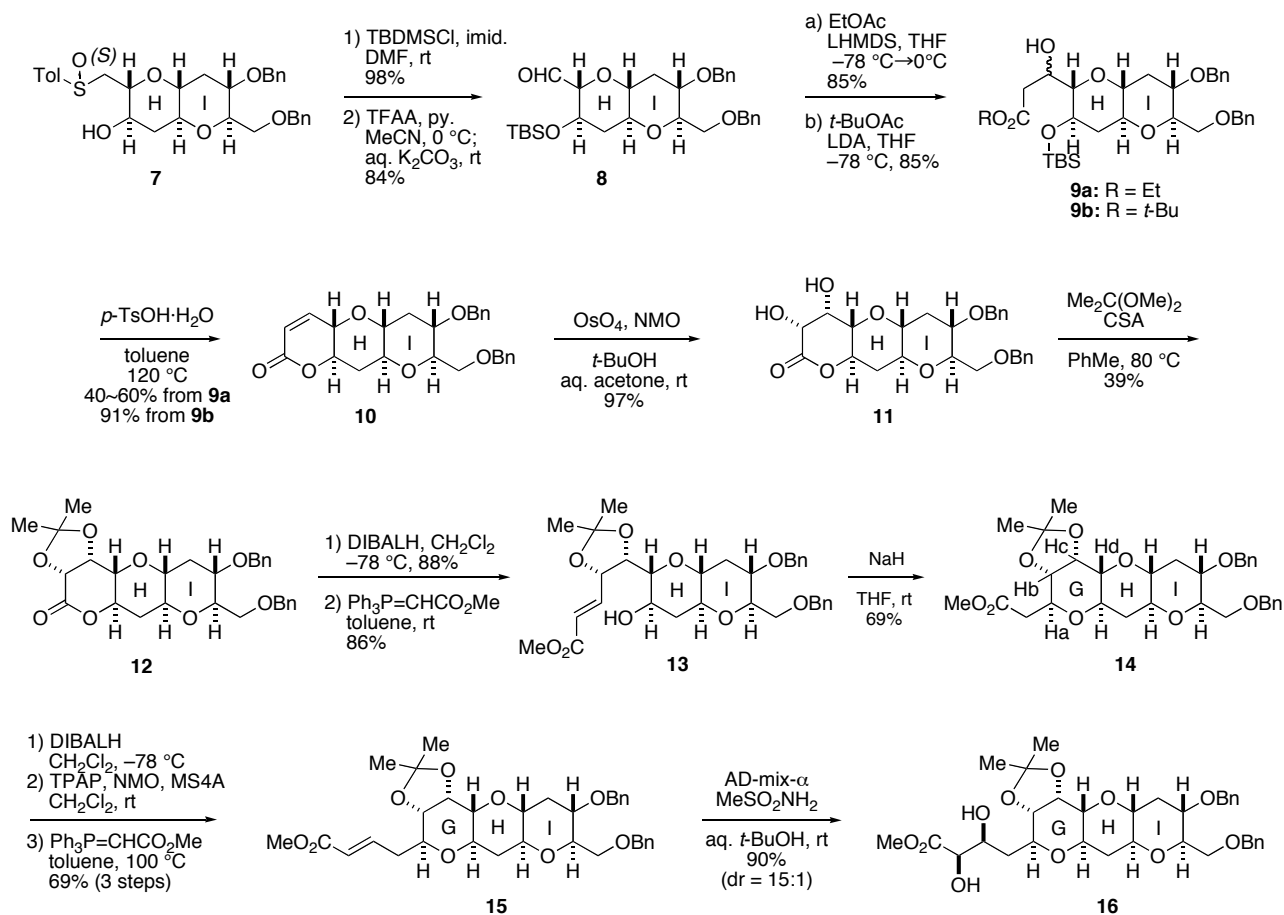
The synthesis of the GHI-ring started with the known compound **3**,¹⁰ corresponding to the I-ring (Scheme 1). The tetrahydropyran I-ring **3** was stereoselectively synthesized by SmI_2 -induced cyclization¹¹ of β -alkoxyacrylate–aldehyde **2**, prepared from 2-deoxy-D-ribose. Protection of alcohol **3** as the TBDMS ether followed by removal of the benzylidene group by hydrogenolysis afforded diol **4** in 95% yield (two steps), which was benzylated followed by desilylation to give dibenzyl ether **5** in 94% yield (two steps). The H-ring was then constructed based on our recently developed SmI_2 -induced cyclization of β -alkoxyvinylsulfoxide–aldehyde.¹² Treatment of **5** with (*S*)-ethynyl-*p*-tolylsulfoxide in the presence of *N*-methylmorpholine (NMM) effected hetero-Michael reaction to give (*E*)- β -alkoxyvinylsulfoxide, which was reduced with DIBALH to give aldehyde **6** in 88% yield (two steps). Reductive cyclization of (*E*)- β -alkoxyvinylsulfoxide–aldehyde **6** with SmI_2 in the presence of MeOH in THF constructed *syn-trans*-tetrahydropyran to give the HI-ring **7** as a single product, quantitatively. The stereochemistry



Scheme 1. Synthesis of the HI-ring **7**.

was confirmed by coupling constant ($J_{a,b} = 10.8$ Hz) between Ha and Hb of **7**.

Next, construction of the G-ring and the side chain having β -diol was investigated (Scheme 2). After protection of **7** with TBDMSCl (98%), Pummerer rearrangement by treatment with TFAA in pyridine followed by aqueous K_2CO_3 afforded aldehyde **8** in 84% yield. After aldol reaction of **8** with EtOAc and LHMDS (85%), treatment of the resulting hydroxy ester **9a** with *p*-TsOH·H₂O in toluene at reflux effected desilylation, lactonization and dehydration in one pot to give α,β -unsaturated δ -lactone **10** in 40~60% yield. The yield of **10** was improved to 91% by using *t*-butyl ester **9b**, which was obtained by aldol reaction of **8** with *t*-BuOAc and LDA (85%). Treatment of the α,β -unsaturated δ -lactone **10** with OsO_4 in the presence of *N*-methylmorpholine *N*-oxide (NMO) exclusively afforded the desired α -diol **11** in 97% yield. After protection of the diol **11** as the acetonide **12** (39%),¹³ DIBALH reduction followed by Wittig reaction using $Ph_3P=CHCO_2Me$ afforded α,β -unsaturated ester **13** in 86% yield. Upon treatment of **13** with NaH in THF, an intramolecular hetero-Michael addition¹⁴ took place stereoselectively to give the GHI-ring **14** in 69% yield. The stereochemistry of the G-ring was confirmed by coupling constants ($J_{a,b} = 9.2$ Hz, $J_{b,c} = J_{c,d} = 3.6$ Hz). Reduction of **14** with DIBALH followed by tetra-*n*-propylammonium perruthenate (TPAP)-NMO oxidation¹⁵ gave an aldehyde, which was treated with $Ph_3P=CHCO_2Me$ to



Scheme 2. Synthesis of the GHI-ring **16** having a 1,2-diol side chain.

give α,β -unsaturated ester **15** in 69% yield (three steps). The Sharpless asymmetric dihydroxylation¹⁶ of **15** furnished β -diol **16** (dr = 15:1) in 90% yield.¹⁷ The stereostructure of **16** was confirmed by NOE measurement of the corresponding diacetate **17** (Figure 2),¹⁸ which was prepared by acetylation of **16**.

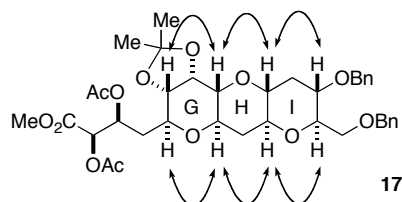


Figure 2. Observed NOE of the acetate **17**.

In summary, MTX GHI-ring system **16** having a 1,2-diol side chain was stereoselectively synthesized starting from 2-deoxy-D-ribose via successive SmI_2 -induced cyclizations, dihydroxylation of α,β -unsaturated δ -lactone, intramolecular hetero-Michael addition, and Sharpless asymmetric dihydroxylation.¹⁹

ACKNOWLEDGEMENTS

This work was financially supported in part by the NOVARTIS Foundation (Japan) for the Promotion of Science and by a Grant-in-Aid for Scientific Research on Priority Areas from the Ministry of Education, Culture, Sports, Science and Technology of Japan.

REFERENCES AND NOTES

1. T. Yasumoto, R. Bagnis, and J. P. Vernoux, *Bull. Jpn. Soc. Sci. Fish.*, 1976, **42**, 359.
2. T. Yasumoto, I. Nakajima, R. Bagnis, and R. Adachi, *Bull. Jpn. Soc. Sci. Fish.*, 1977, **43**, 1021.
3. For reviews on marine polycyclic ethers, see: (a) T. Yasumoto and M. Murata, *Chem. Rev.*, 1993, **93**, 1897. (b) M. Murata and T. Yasumoto, *Nat. Prod. Rep.*, 2000, **17**, 293. (c) T. Yasumoto, *Chem. Rec.*, 2001, **1**, 228.
4. (a) M. Murata, T. Iwashita, A. Yokoyama, M. Sasaki, and T. Yasumoto, *J. Am. Chem. Soc.*, 1992, **114**, 6594. (b) M. Murata, H. Naoki, T. Iwashita, S. Matsunaga, M. Sasaki, A. Yokoyama, and T. Yasumoto, *J. Am. Chem. Soc.*, 1993, **115**, 2060. (c) M. Murata, H. Naoki, S. Matsunaga, M. Satake, and T. Yasumoto, *J. Am. Chem. Soc.*, 1994, **116**, 7098. (d) M. Satake, S. Ishida, T. Yasumoto, M. Murata, H. Utsumi, and T. Hinomoto, *J. Am. Chem. Soc.*, 1995, **117**, 7019.
5. (a) M. Sasaki, N. Matsumori, T. Maruyama, T. Nonomura, M. Murata, K. Tachibana, and T. Yasumoto, *Angew. Chem., Int. Ed. Engl.*, 1996, **35**, 1672. (b) T. Nonomura, M. Sasaki, N.

- Matsumori, M. Murata, K. Tachibana, and T. Yasumoto, *Angew. Chem., Int. Ed. Engl.*, 1996, **35**, 1675. (c) M. Sasaki, T. Nonomura, M. Murata, and K. Tachibana, *Tetrahedron Lett.*, 1994, **35**, 5023. (d) M. Sasaki, T. Nonomura, M. Murata, and K. Tachibana, *Tetrahedron Lett.*, 1995, **36**, 9007. (e) M. Sasaki, N. Matsumori, M. Murata, and K. Tachibana, *Tetrahedron Lett.*, 1995, **36**, 9011.
6. (a) L. R. Cook, H. Oinuma, M. A. Semones, and Y. Kishi, *J. Am. Chem. Soc.*, 1997, **119**, 7928. (b) W. Zheng, J. A. DeMattei, J.-P. Wu, J. J.-W. Duan, L. R. Cook, H. Oinuma, and Y. Kishi, *J. Am. Chem. Soc.*, 1996, **118**, 7946.
7. The stereochemistry at the J/K ring junction was recently questioned, but the originally assigned structure was supported through synthesis of the GHIJK- and GHIJKLMNO-ring system by Nicolaou *et al.*^{8b,c} (a) A. R. Gallimore and J. B. Spencer, *Angew. Chem. Int. Ed.*, 2006, **45**, 4406. (b) K. C. Nicolaou and M. O. Frederick, *Angew. Chem. Int. Ed.*, 2007, **46**, 5278.
8. (a) K. C. Nicolaou, M. H. D. Postema, E. W. Yue, and A. Nadin, *J. Am. Chem. Soc.*, 1996, **118**, 10335. (b) K. C. Nicolaou, K. P. Cole, M. O. Frederick, R. J. Aversa, and R. M. Denton, *Angew. Chem. Int. Ed.*, 2007, **46**, 8875. After the submission of this paper, the synthesis of the GHIJKLMNO-ring has been reported. (c) K. C. Nicolaou, M. O. Frederick, A. C. B. Burtoloso, R. M. Denton, F. Rivas, K. P. Cole, R. J. Aversa, R. Gibe, T. Umezawa, and T. Suzuki, *J. Am. Chem. Soc.*, 2008, **130**, 7466.
9. (a) T. Nakata, S. Nomura, and H. Matsukura, *Chem. Pharm. Bull.*, 1996, **44**, 627. (b) K. Nagasawa, N. Hori, R. Shiba, and T. Nakata, *Heterocycles*, 1997, **44**, 105. (c) M. Satoh, M. Mori, and T. Nakata, *Heterocycles*, 2007, **74**, 259. (d) Y. Sakamoto, G. Matsuo, H. Matsukura, and T. Nakata, *Org. Lett.*, 2001, **3**, 2749. (e) M. Motita, S. Ishiyama, H. Koshino, and T. Nakata, *Org. Lett.*, 2008, **10**, 1675. (f) M. Morita, T. Haketa, H. Koshino, and T. Nakata, *Org. Lett.*, 2008, **10**, 1679. (g) M. Satoh, H. Koshino, and T. Nakata, *Org. Lett.*, 2008, **10**, 1683.
10. G. Matsuo, K. Kawamura, N. Hori, H. Matsukura, and T. Nakata, *J. Am. Chem. Soc.*, 2004, **126**, 14374.
11. (a) N. Hori, H. Matsukura, G. Matsuo, and T. Nakata, *Tetrahedron Lett.*, 1999, **40**, 2811. (b) N. Hori, H. Matsukura, and T. Nakata, *Org. Lett.*, 1999, **1**, 1099. (c) G. Matsuo, N. Hori, and T. Nakata, *Tetrahedron Lett.*, 1999, **40**, 8859. (d) K. Suzuki, H. Matsukura, G. Matsuo, H. Koshino, and T. Nakata, *Tetrahedron Lett.*, 2002, **43**, 8653. (e) G. Matsuo, H. Kadohama, and T. Nakata, *Chem. Lett.*, 2002, 148. (f) N. Hori, G. Matsuo, H. Matsukura, and T. Nakata, *Tetrahedron*, 2002, **58**, 1853.
12. (a) T. Kimura, M. Hagiwara, and T. Nakata, *Tetrahedron Lett.*, 2007, **48**, 9171. See, also: (b) J. H. Jung, Y. W. Kim, M. A. Kim, S. Y. Choi, Y. K. Chung, T.-R. Kim, S. Shin, and E. Lee, *Org. Lett.*, 2007, **9**, 3225.
13. The yield was not optimized yet. The reaction was accompanied by decomposition.

14. For several examples of the synthesis of cyclic ethers via intramolecular herero-Michael reaction, see: (a) K. C. Nicolaou, C.-K. Hwang, and M. E. Duggan, *J. Am. Chem. Soc.*, 1989, **111**, 6682. (b) J. M. Betancort, V. S. Martín, J. M. Pardón, J. M. Palazón, M. A. Ramírez, and M. A. Soler, *J. Org. Chem.*, 1997, **62**, 4570. (c) H. Fuwa, N. Kaimuna, K. Tachibana, and M. Sasaki, *J. Am. Chem. Soc.*, 2002, **124**, 14983.
15. (a) W. P. Griffith and S. V. Ley, *Aldrichimica Acta*, 1990, **23**, 13. (b) S. V. Ley, J. Norman, W. P. Griffith, and S. P. Marsden, *Synthesis*, 1994, 639.
16. (a) S. G. Hentges and K. B. Sharpless, *J. Am. Chem. Soc.*, 1980, **102**, 4263. (b) H. C. Kolb, M. S. VanNieuwenhze, and K. B. Sharpless, *Chem. Rev.*, 1994, **94**, 2483.
17. The stereochemistry of the β -diol in **16** was assigned according to the empirical rule for Sharpless asymmetric dihydroxylation.^{16a}
18. Data for **17**: IR (neat) 2926, 2855, 1749, 1455, 1373, 1218, 1097, 1057, 867, 742, 700, 602, 508 cm^{-1} ; ^1H NMR (600 MHz, CDCl_3) δ 7.53-7.28 (m, 8H), 7.21-7.19 (m, 2H), 5.56 (ddd, $J = 7.9, 4.9, 2.6$ Hz, 1H), 5.16 (d, $J = 2.6$ Hz, 1H), 4.62 (d, $J = 12.5$ Hz, 1H), 4.55 (d, $J = 11.3$ Hz, 2H), 4.49 (br dd, $J = 4.2, 3.8$ Hz, 1H), 4.40 (d, $J = 11.3$ Hz, 1H), 3.82 (dd, $J = 9.3, 4.2$ Hz, 1H), 3.75 (dd, $J = 13.2, 11.0$ Hz, 1H), 3.73 (s, 3H), 3.66 (dd, $J = 11.0, 4.5$ Hz, 1H), 3.56 (ddd, $J = 11.3, 10.2, 4.4$ Hz, 1H), 3.51 (ddd, $J = 11.0, 10.3, 4.3$ Hz, 1H), 3.41 (m, 1H), 3.32 (ddd, $J = 9.6, 9.3, 2.5$ Hz, 1H), 3.28 (dd, $J = 9.1, 4.2$ Hz, 1H), 3.16 (ddd, $J = 11.0, 9.1, 4.2$ Hz, 1H), 3.12 (ddd, $J = 11.3, 9.1, 4.0$ Hz, 1H), 2.57 (ddd, $J = 11.3, 4.4, 4.0$ Hz, 1H), 2.46 (ddd, $J = 11.0, 4.3, 4.0$ Hz, 1H), 2.19 (s, 3H), 2.06 (ddd, $J = 14.3, 7.9, 2.5$ Hz, 1H), 2.04 (s, 3H), 1.70 (ddd, $J = 14.3, 9.6, 4.9$ Hz, 1H), 1.64 (ddd, $J = 11.3, 11.3, 11.3$ Hz, 1H), 1.53 (s, 3H), 1.44 (ddd, $J = 11.0, 11.0, 11.0$ Hz, 1H), 1.39 (s, 3H); ^{13}C NMR (150 MHz, CDCl_3) δ 170.1, 169.8, 167.9, 138.2, 138.0, 128.4 (2C), 128.3 (2C), 127.9 (2C), 127.7 (2C), 127.6 (2C), 111.4, 80.4, 77.6, 77.1, 75.9, 74.7, 73.5, 73.3, 73.1, 72.3, 71.0, 69.9, 69.1, 69.0, 52.6, 35.1, 35.0, 33.6, 29.7, 28.3, 26.4, 20.7, 20.5; HRMS (ESI) calcd for $\text{C}_{38}\text{H}_{48}\text{O}_{13}\text{Na}$ ($\text{M}+\text{Na}^+$) 735.2987, found 735.2958.
19. The previous synthesis of the GHI-ring was accomplished in 31 steps from methyl α -D-glucopyranoside.^{9c} The present synthesis of the GHI-ring, including the construction of 1,2-diol side chain, was efficiently accomplished in 25 steps from 2-deoxy-D-ribose.