

FACILE SYNTHESIS OF PYRAZOLO[3,4-b]PYRIDINES

Abdulla Babagi, Ali El-Shekeil*, Mohammed Hassan, and

Sayed Shiba

Chemistry Department, Faculty of Science, Sana'a University

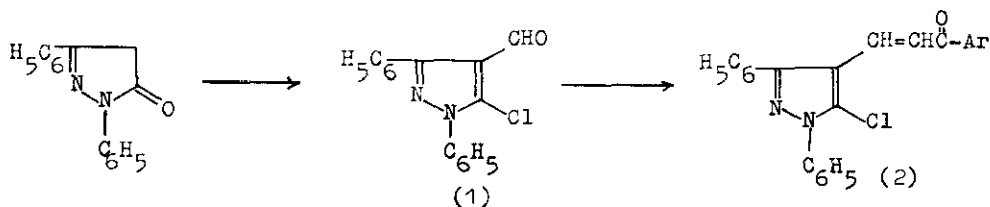
Sana'a, Yemen

Abstract — The base-catalyzed Michael addition of cyclohexanone, 1,3-diphenyl-2-pyrazolin-5-one, ethyl acetoacetate, and diethyl malonate on 1-aryl-3-(5-chloro-1,3-diphenyl-1H-4-pyrazolyl)-prop-2-en-1-ones (2) in the presence of ammonium acetate at 130°C gave the corresponding pyrazolo[3,4-b]pyridine derivatives (3-6). Structures were confirmed by microanalysis, ir, ^1H -nmr, and ^{13}C -nmr.

Pharmacological studies of pyrazolones as respiratory and cardio-vascular agents¹ fungicides, herbicides, insecticides^{2,3} and antibacterial compounds⁴ have been recently reported. Also, the antianxiety activities have been investigated^{5,6} and the carcinogenicity of some pyrazolones has been established.⁷

As an extension of our previous work⁸, we report here the facile synthesis of some new pyrazolo[3,4-b]pyridines from available chemicals by simple experimental procedures in high yields.

The Vilsmeier-Haack formylation of 1,3-diphenyl-2-pyrazolin-5-one was reported earlier^{9,10} to give 5-chloro-1,3-diphenyl-1H-pyrazole-4-carboxaldehyde (1). The key-starting 1-aryl-3-(5-chloro-1,3-diphenyl-1H-4-pyrazolyl)prop-2-en-1-ones (2 a,b) were prepared as a major product of Knoevenagel condensation of aldehyde (1) with acetophenone or *p*-methylacetophenone in alcoholic KOH.



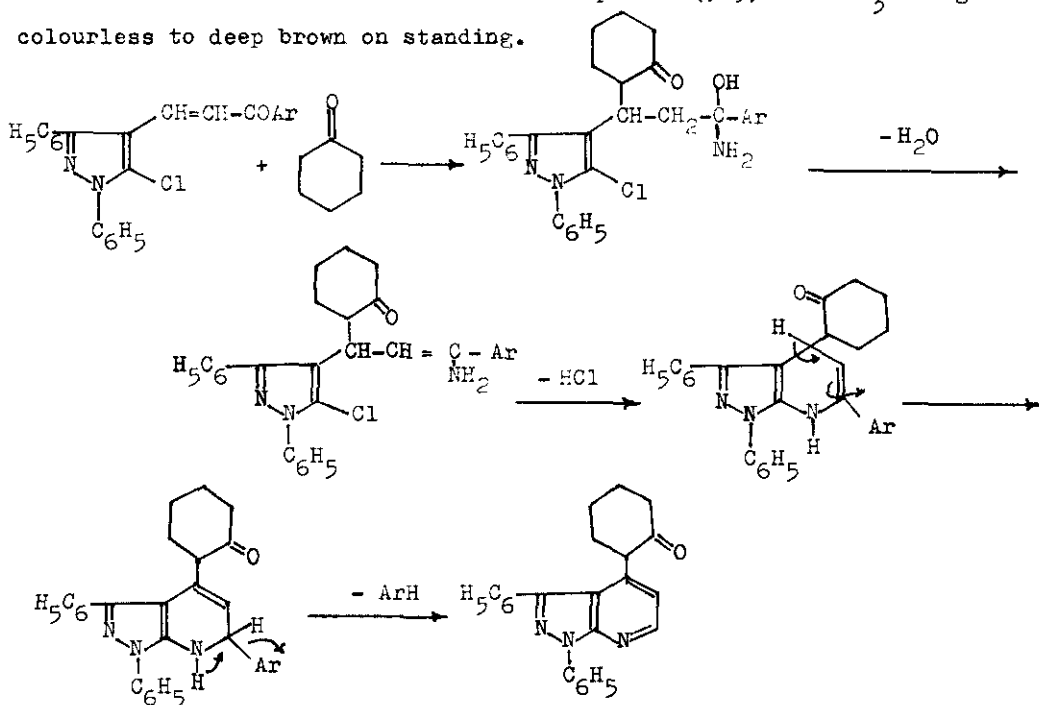
a) Ar = C_6H_5 , b) Ar = $\text{C}_6\text{H}_4-\text{CH}_3-\text{p}$

Treatment of (2 a,b) with cyclohexanone or 1,3-diphenyl-2-pyrazolin-5-one in the presence of ammonium acetate at 130°C underwent Michael addition, cyclization, and dearylation to give 1,3-diphenyl-4-(6-oxocyclohexyl)-1H-pyrazolo [3,4-b] -pyridine (3) or 4,7-dihydro-1,3-diphenyl-4-(4,5-dihydro-1,3-diphenyl-5-oxo-1H-4-pyrazolyl)-1H-pyrazolo[3,4-b] pyridine (4), respectively. The dearylation¹¹ is confirmed by ¹H-nmr, ¹³C-nmr and elemental analysis, and mechanism is proposed.

The base-catalyzed Michael addition of ethyl acetoacetate on (2) followed by cyclization, under the same conditions, yielded ethyl 1,3-diphenyl-6-methyl-4-phenacyl-1H-pyrazolo [3,4-b]pridine-5-carboxylate derivatives (5 a,b), while diethyl malonate reacted with (2) under the same conditions¹² to give ethyl 1,3-diphenyl-6-oxo-4-phenacyl-4,5,6,7-tetrahydro-1H-pyrazolo[3,4-b]pyridine -5-carboxylates (6 a,b).

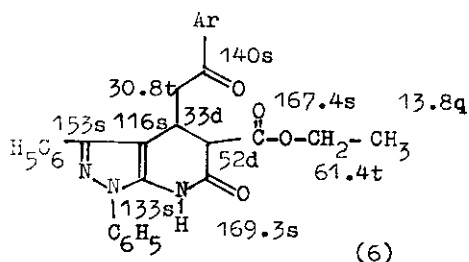
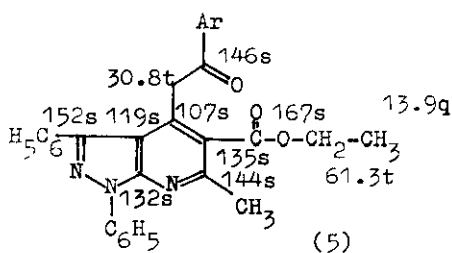
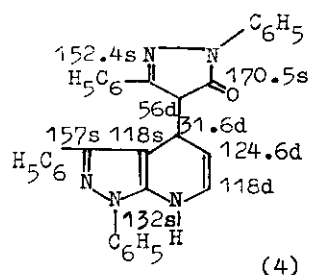
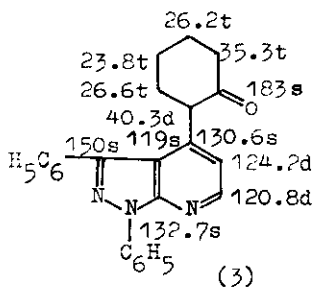
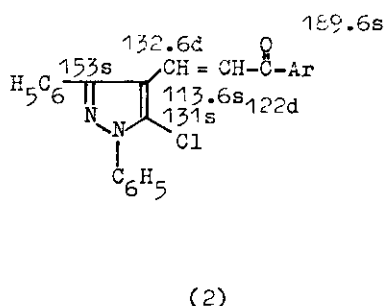
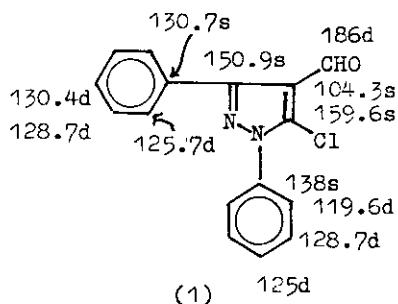
The structures of compounds (1-6) have been established by elemental analysis, ir and ¹H-nmr spectral data^{13,14}, which are listed in Table 1 together with yields, melting points, colour of crystals, and solvents of recrystallization.

The ¹³C-nmr chemical shifts (δ ppm) and off-resonance data¹⁵⁻¹⁸ are written on the structures below. The solution of compounds (3-5) in CDCl₃ changed from colourless to deep brown on standing.



EXPERIMENTAL

Melting points reported are uncorrected. Ir spectra were determined in KBr on Pye Unicam SP 200 G spectrophotometer. The ^1H -nmr spectra were recorded on a Varian T-60 spectrometer. The ^{13}C -nmr spectra were measured on a Jeol FX 90 Q Fourier Transform instrument operating at 22.50 MHz, 8192 data points were collected and a sweep width of 5000 Hz, i.e. digital resolution of 0.6 Hz (0.03 ppm). Pulse intervals of 5 seconds or more were used to achieve reasonable signal-to-noise ratios especially for the signal of carbons of long relaxation times e.g. C=O, C=N. The broad band proton noise decoupled and off-resonance spectra were recorded. In all nmr experiments the internal standard was TMS and the solvent was CDCl_3 . All chemical shifts are in ppm downfield from TMS.



5-Chloro-1,3-diphenyl-1H-pyrazole-4-carboxaldehyde (1) - A solution of 1,3-diphenyl-5-pyrazolone (47.5 gm, 0.2 mol) in DMF (100 ml) was cooled to 0°C in an ice - acetone bath. Then phosphoryl chloride (55 ml, 0.6 mol) was added dropwise at such as a rate as to maintain the temperature between 10-20°C. The reaction mixture was heated on steam-bath after complete addition for 90 min. The mixture was poured onto ice-water (21). The resulting mixture was allowed to stand overnight at 25°C. The solid product thus obtained was filtered, washed with water, dried and crystallized from pet. ether (80-100°C) to give (1) as colourless crystals in 60% yield.

1-Aryl-3-(5-chloro-1,3-diphenyl-1H-4-pyrazolyl)prop-2-en-1-ones (2 a,b) - A mixture of aldehyde (1) (7.8 g., 0.01 mol) and acetophenone/or p-methylacetophenone (0.02 mol) in ethanolic KOH (2 g. in 100 ml) was stirred at 0°C for 2 h. The solid product which separated during stirring was filtered off, washed with water, dried and crystallized from benzene-pet. ether (80-100°C) mixture to give (2) as colourless crystals in 80-86% yield.

General procedure of base-catalyzed cycloaddition of active methylene compounds on (2) : Formation of pyrazolo[3,4-b]pyridines (3-6): A mixture of propenones (2)(0.01 mol), active methylene compounds (0.02 mol), namely, cyclohexanone, 1,3-diphenyl-2-pyrazolin-5-one, ethyl acetoacetate, or diethyl malonate, and dry ammonium acetate (0.04 mol) was heated in an oil-bath at 130°C for 3 h. The reaction mixture was cooled, triturated with hot water (50 ml), filtered, washed with water, dried and crystallized from a suitable solvent to give (3-6) respectively.

Table 1- Physical data of compounds 1-6 :

Compd.	Mp(°C) (colour)	Solvent of crystn. (yield %)	Mol. formula	Analysis %			Ir spectra (cm ⁻¹)
				Calc./	(found)		
1*	110 (colourless)	B + P (60)	C ₁₆ H ₁₁ N ₂ OCl	68.0 (67.7)	3.9 4.0	9.9 9.9	1590(C=N, C=C), 1670 (C=O).
2a**	150 (colourless)	B + P (80)	C ₂₄ H ₁₇ N ₂ OCl	74.9 (74.7)	4.4 4.4	7.3 7.3	1590(C=N, C=C), 1660 (C=O).
2b**	142 (colourless)	B + P (86)	C ₂₅ H ₁₉ N ₂ OCl	75.3 (75.4)	4.8 4.7	7.0 6.7	1595(C=N, C=C), 1680 (C=O).
3*	170 (pale yellow)	B + P (60)	C ₂₄ H ₂₁ N ₃ O	78.5 (78.8)	5.7 5.9	11.4 11.2	1595(C=N, C=C), 1660 (C=O).
4**	220 (colourless)	B (66)	C ₃₃ H ₂₅ N ₅ O	78.1 (78.2)	5.0 4.7	13.8 13.6	1590(C=N, C=C), 1705 (C=O), 3300-3600(NH).
5a**	165 (pale yellow)	B + P (72)	C ₃₀ H ₂₅ N ₃ O ₃	75.8 (75.9)	5.3 5.5	8.8 9.1	1595(C=N, C=C), 1660 (C=O), 1720(C=O ester).
5b†	172 (pale yellow)	B + P (75)	C ₃₁ H ₂₇ N ₃ O ₃	76.1 (75.9)	5.5 5.7	8.6 8.9	1595(C=N, C=C), 1655 (C=O), 1720(C=O ester).
6a***	185 (colourless)	B + P (85)	C ₂₉ H ₂₅ N ₃ O ₄	72.6 (72.7)	5.3 5.1	8.8 8.6	1595(C=N, C=C), 1680 (C=O), 1740(C=O ester), 3230(NH).
6b***	220 (colourless)	B + P (87)	C ₃₀ H ₂₇ N ₃ O ₄	73.0 (73.2)	5.5 5.3	8.5 8.8	1590(C=N, C=C), 1670 (C=O), 1740(C=O ester), 3210(NH).

B = benzene; P = pet. ether 80-100°C

- * 1, ¹H-nmr : 7.2-7.9 ppm (m, 10H, Ar-H), and 10.0 (s, 1H, COH).
- ** 2a, ¹H-nmr : 7.2-8.3 ppm (m, Ar-H and CH = CH), while 2b gives additional 2.4(s, 3H, Ar-CH₃).
- + 2, ¹H-nmr : 0.8-2.4 (m, 9H, cyclohexyl-H), and 6.8-8.0(m, 12H, Ar-H, and olefinic C₅-H & C₆-H).
- ++ 4, ¹H-nmr : 4.0(d, J=5Hz, 1H, C₄-H), 5.7(d, J=5Hz, 1H, C₄-CH), 6.9-7.7 (m, 22H, Ar-H and olefinic C₅-H and C₆-H) and 8.2 (broad s, 1H, NH).
- † 5a, ¹H-nmr : 1.0(t, J=7Hz, 3H, ester CH₂-CH₃), 2.15(s, 3H, C₆-CH₃), 2.7(s, 2H, C₄-CH₂), 4.15(q, J=7Hz, ester CH₂-CH₃), and 7.2-8.0 (m, 15H, Ar-H), 5b gives also 2.4(s, 3H, Ar-CH₃).
- *** 6a, ¹H-nmr : Similar to 5a but with no peak at 2.15 but with s, 1H, NH at 8.3 ppm, but 6b gives additional signal at 2.4 (s, 3H, Ar-CH₃).

REFERENCES

1. R. Benigini, Minerva Med., 1941, 1, 436.
2. G. Tacconi, G. Gatti, G. Desimoni, and V. Messori, J. Prakt. Chem., 1980, 322, 831.
3. H. S. El-Kashef, M. A. Abdalla, B. E. Bayoumi, and A. A. M. El-Timawy, J. Chem. Tech. Biotechnol., 1983, 294.
4. N. S. Habib and G. G. Tawil, Sci. Pharm., 1981, 49, 42.
5. A. F. Heald and R. A. Wildonger, US Patent, 4, 364, 948 (Cl. 424 - 256; A 61 K 31/44); Chem. Abstr., 1983, 98, 89358.
6. J. Haeufel and E. Breitmaier, Angew. Chem., 1974, 13, 604.
7. US Patent, Gov. Rep. Announce Index (U.S.), 1979, 79, 79, Chem. Abstr., 1979, 90, 198641.
8. A. G. El-Shekeil, M. A. Hassan, A. S. Babaqi, and S. A. Shiba, Acta Chim. Sinica, 1985, 43, 1215.
9. S. B. Barnela, R. S. Pandit, and S. Seshadri, Indian J. Chem., 1976, 14, 665.
10. W. Dynek, B. Janik, and R. Zimon, Acta Polon. Pharm., 1963, 20, 9.
11. A. S. Babaqi, A. G. El-Shekeil, M. A. Hassan, and S. A. Shiba, Acta Chim. Hungarica, 1988 (ACCEPTED).
12. A. Sammour, A. Raouf, M. El-Kasaby, M. Abdalla, and M. A. Hassan, Acta Chim. Hungarica, 1974, 83, 209.
13. L. J. Bellamy "The Infrared Spectra of Complex Molecules", Methuen & Co. Ltd. London, 1968.
14. R. M. Silverstein and G. C. Bassler "Spectrophotometric Identification of Organic Compounds", John Wiley and Sons Inc., New York, 1981.
15. S. N. Ege, A. D. Adams, E. J. Gess, K. S. Ragone, B. J. Kober, M. B. Lampert, P. Umrigar, D. C. Lankin, and G. w. Griffin, J. Chem. Soc., Perkin Trans. I, 1983, 325.
16. F. W. Wehrli and J. Wirthlin "Interpretation of Carbon - ^{13}C NMR Spectra", Heyden & Son, London 1976.
17. E. Kleinpeter and R. Borsdorf " ^{13}C -NMR-Spektroskopie in der Organischen Chemie", Akademie - Verlage, Berlin, 1981.
18. J. T. Clerc, E. Pretsch, and S. Sternhell " ^{13}C -Kernresonanz Spektroskopie", Akadem Verlagsgesellschaft, Frankfurt, 1973.

Received, 25th January, 1988