

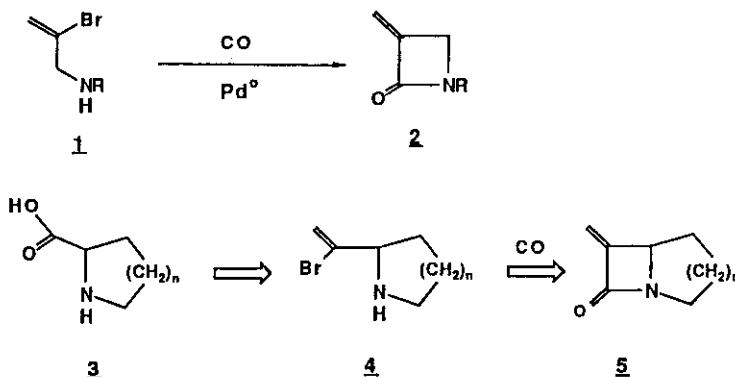
RING CONSTRUCTION OF BICYCLIC- $\beta$ -LACTAM BY USE OF PALLADIUM CATALYZED CARBOXYLATION

Miwako Mori,\* Yukako Higuchi, Katsuji Kagechika, and Masakatsu Shibasaki\*

Faculty of Pharmaceutical Sciences, Hokkaido University, Sapporo, 060 Japan

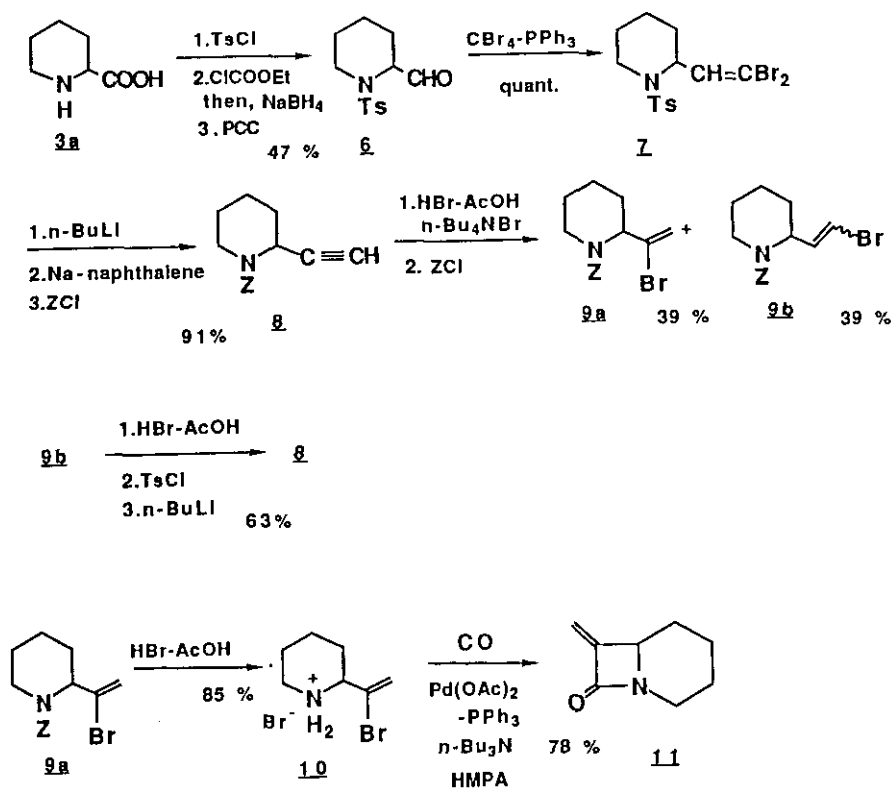
**Abstract**---Palladium catalyzed carbonylation into vinyl halide 10 afforded carbacepham 11 in good yield. The introduction of methoxycarbonyl group at C-4 position of carbacepham was achieved by conversion of methoxy group introduced by anodic oxidation in MeCN-MeOH to carboxyl group.

The search for  $\beta$ -lactam antibiotics possessing enhanced activity and resistance to  $\beta$ -lactamase has generated strong interest in methods of preparing the carbacephem and carbapenem skeletons. We have already reported the new synthetic method of  $\alpha$ -methylene- $\beta$ -lactams by use of palladium catalyzed carbonylation into 2-bromoallylamine derivatives.<sup>1</sup> This procedure prompted us to develop a new synthetic method of bicyclic  $\beta$ -lactam 5 from vinyl halide 4.



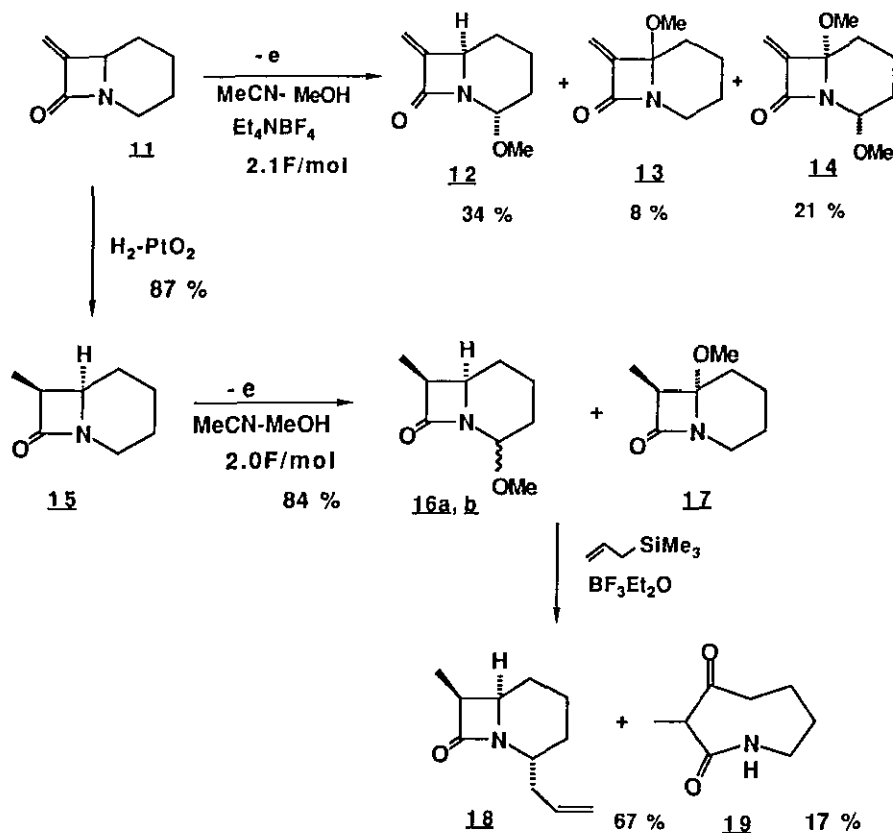
In order to prepare vinyl halide 4, the attempt to convert the carboxyl group of cyclic amino acid 3 such as proline (n=1) or pipercolinic acid (n=2), into vinyl halide was made. Pipercolinic acid 3a was converted to aldehyde 6 by usual method, which was treated with CBr<sub>4</sub>-PPh<sub>3</sub> to afford vinyl dibromide 7. Treatment of 7 with excess n-BuLi<sup>2</sup> was followed by conversion of protecting group

from tosyl group to benzyloxycarbonyl group<sup>3</sup>. Addition of HBr to compound **8**<sup>4</sup> followed by protection of amino group with ZCl provided vinyl bromides, **9a** and **9b** in a ratio of 1 to 1. However, the latter vinyl halide **9b** could easily give back to acetylene **8**. Removal of the protecting group of **9a** with HBr-AcOH afforded the desired vinyl bromide hydrogen bromide **10**, which was successfully converted to bicyclic  $\beta$ -lactam **11** by palladium catalyzed carbonylation. Namely, a solution of vinyl halide **10**, Pd(OAc)<sub>2</sub>(2 mol %), PPh<sub>3</sub>(4 mol %) and n-Bu<sub>3</sub>N(2.5 eq) in hexamethylphosphoric triamide(HMPA) was heated at 100°C for 4 h under carbon monoxide(1 atm) to give  $\beta$ -lactam **11** in 78 % yield.

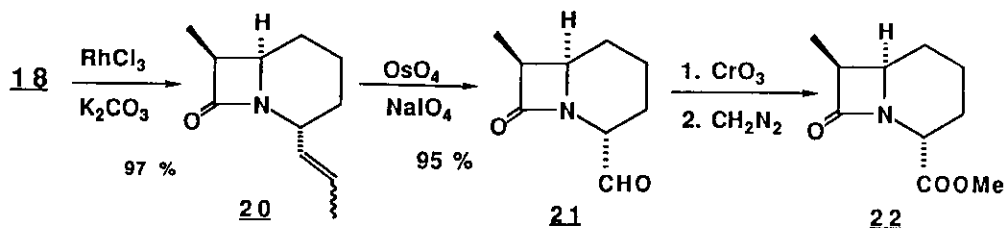


In order to introduce the carboxyl group at C-4 position of carbacepham **11**, the anodic oxidation should be a suitable method because the methoxy group at the  $\alpha$ -position of lactam<sup>5</sup> introduced by the anodic oxidation could be replaced by carbon nucleophile.<sup>6</sup> Thus, the electrochemical oxidation to  $\beta$ -lactam **11** was carried out in an undivided cell using platinum plates as electrode in MeCN-MeOH (9:1) containing Et<sub>4</sub>NBF<sub>4</sub> as supporting electrolyte. After 2.1 F/mol of electricity was passed

through the solution, methoxylated compounds **12**, **13** and **14** were obtained in 34 %, 8 %, and 21 % yields, respectively.<sup>7</sup> Since the allylic position should be easy to oxidize for the electrolysis,<sup>5b</sup>  $\alpha$ -methylene- $\beta$ -lactam **11** was hydrogenated with  $\text{PtO}_2$  to give compound **15** as a single product. The methyl group of compound **15** should be oriented to the  $\beta$ -position because the catalyst might approach from the less hindered site. When 2.0 F/mol of electricity was passed through the MeCN-MeOH (9:1) solution of compound **15**, inseparable mixture of methoxylated compounds **16** and **17** was obtained in 84 % yield. The nmr spectrum indicated that the ratio of **16** to **17** was 7 to 1. Treatment of the mixture of **16** and **17** with allylsilane in the presence of  $\text{BF}_3\text{Et}_2\text{O}$ <sup>8</sup> gave compound **18** in 67 % yield along with compound **19** (17 % yield).<sup>9</sup> The latter compound **19** should be obtained from compound **17** by treatment with  $\text{BF}_3\text{Et}_2\text{O}$  in the presence of a small amount of water.



Compound **18** was treated with  $\text{RhCl}_3$  in the presence of  $\text{K}_2\text{CO}_3$  in EtOH followed by treatment with  $\text{OsO}_4$  and  $\text{NaIO}_4$  to give aldehyde **21** in good yield. Oxidation of compound **21** with  $\text{CrO}_3$  provided carboxylic acid, which was converted into methyl ester **22**<sup>10</sup> by treatment with  $\text{CH}_2\text{N}_2$ .



These results suggested that palladium catalyzed carbonylation into vinyl halide 10 afforded bicyclic  $\beta$ -lactam 11 in good yield. In order to introduce the carboxyl group at C-4 position of carbacepham skeleton, introduction of the methoxy group to the  $\alpha$ -position of lactam by anodic oxidation was a good procedure because carbon nucleophile could be introduced to the methoxylated position. If proline was used for this reaction, carbapenam skeleton would be formed.

Further studies are in progress.

#### REFERENCES AND NOTES

1. M. Mori, K. Chiba, M. Okita, and Y. Ban, *Chem. Comm.*, 1979, 698. M. Mori, K. Chiba, M. Okita, I. Kayo, and Y. Ban, *Tetrahedron*, **41**, 1985, 375.
2. E. J. Corey and P. L. Fuchs, *Tetrahedron Lett.*, 1972, 3769.
3. The deprotection of the tosyl group with Na-naphthalene would accompany the debromination of vinyl halide.
4. J. Coussean, *Synthesis*, 1980, 805.
5. a) M. Okita, T. Wakamatsu, and Y. Ban, *Chem. Comm.*, 1979, 749. b) M. Okita, M. Mori, T. Wakamatsu, and Y. Ban, *Heterocycles*, **23**, 1985, 247.
6. T. Shono, Y. Matsumura, and K. Tsubata, *J. Am. Chem. Soc.*, **103**, 1981, 1172.
7. Compounds 12 and 13 were inseparable mixture, but the nmr spectrum indicated that compound 12 was a single isomer. Presumably, methoxy group should attack from the less hindered site of the acyl iminium cation generated by electrolysis.
8. G. A. Kraus and K. Neuenschwander, *Chem. Comm.*, 1982, 134.
9. From the nmr spectrum of compounds 16 and 17, methoxylated compound 16 was a mixture of two isomers and the ratio of  $\alpha$ -(16a) to  $\beta$ -methoxylated compound(16b) was 4 to 1. However, compound 18 was obtained as a single isomer.
10. Compound 22:  $\text{ir } \nu_{\text{max}}(\text{CHCl}_3)$  1730  $\text{cm}^{-1}$ ;  $\text{ms m/e}$  197( $\text{M}^+$ ), 169( $\text{M}^+ - \text{CO}$ ), 138( $\text{M}^+ - \text{COOMe}$ ), 110, 82, 68, 55, high resolution mass spectrum Calcd for  $\text{C}_{10}\text{H}_{15}\text{NO}_3$  196.1069, found 196.1061;  $\text{nmr } \delta(\text{CDCl}_3)$  1.19(d,  $J=6\text{Hz}$ , 3 H), 1.4-2.2(m, 6 H), 3.4(m, 1 H), 3.74(s, 3 H, OMe), 3.8(m, 1 H), 4.56(bd,  $J=7\text{ Hz}$ , 1 H).

Received, 23rd January, 1989