

REGIOSELECTIVE ALLYLATION TO A TETRAHYDROISOQUINOLINE[§]

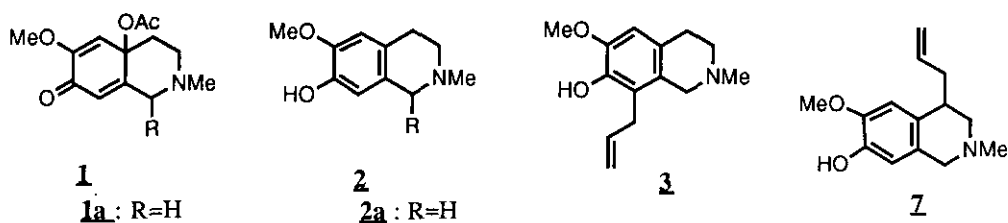
Hiroshi Hara, Akiko Seto, and Osamu Hoshino*

Faculty of Pharmaceutical Sciences, Science University of Tokyo, Shinjuku-ku,
Tokyo 162, Japan

Abstract--- The *p*-quinol acetate (**1a**) reacted with allyltrimethylsilane in dichloromethane in the presence of acid ($\text{BF}_3 \cdot \text{Et}_2\text{O}$ or CF_3COOH) to give 8-allylcorypalline (**3**), while in acetonitrile to give 4-allylcorypalline (**7**), regioselectively. Plausible pathway on formation of **3** and **7** is described.

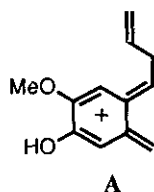
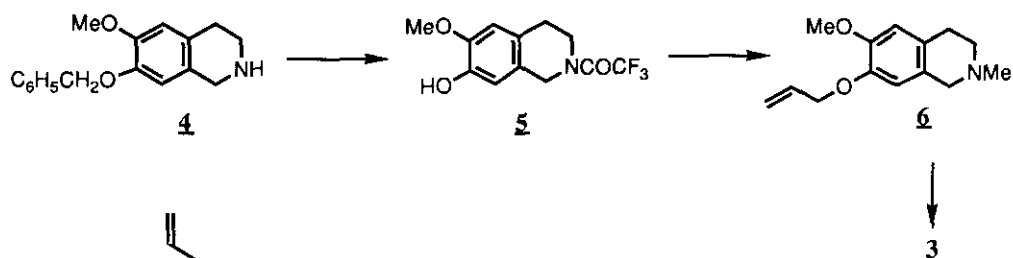
The *p*-quinol acetates (**1**), derived from tetrahydroisoquinolin-7-ol (**2**), are highly reactive to nucleophiles in the presence of acid¹ and are versatile compounds for synthesis of isoquinoline alkaloids. We have already reported syntheses of various isoquinoline alkaloids and their derivatives by use of the *p*-quinol acetates (**1**) as key compounds.² Especially, application of this nucleophilic reactions to C-C bond formation serves for synthesis of tetracyclic alkaloids^{2a} by intramolecular cyclization and for intermolecular construction of biaryls.^{2b}

At present silicon reagents should be indispensable for organic synthesis.³ Among them, allylsilanes are particularly valuable for C-C bond formation.⁴ For instance, allylsilanes react with *p*-quinones to give allyl-substi-

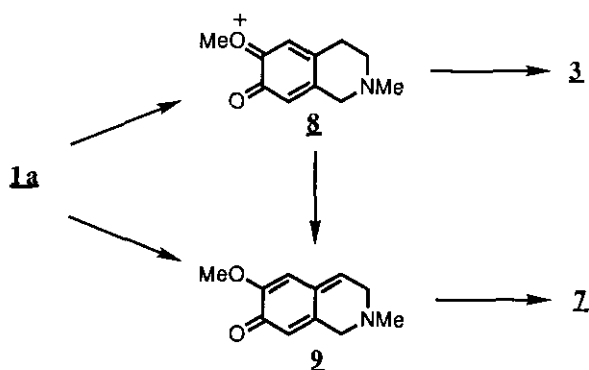


[§]Dedicated to Professor Emeritus Masatomo Hamana on the occasion of his 75th birthday.

Scheme 1



Scheme 2



Table

Solvent ^a	Acid	Reaction temp.	Time (h)	Product (%) ^b	
				3	7
CH ₂ Cl ₂	BF ₃ Et ₂ O	r.t. ^c	2	22.4	-----
CH ₂ Cl ₂	BF ₃ Et ₂ O	0°C	2	12.7	-----
CH ₂ Cl ₂	CF ₃ COOH	r.t. ^c	2	6.5	-----
MeCN	BF ₃ Et ₂ O	r.t. ^c	2	-----	10.2
MeCN	BF ₃ Et ₂ O	0°C	3	-----	20.0
MeCN	CF ₃ COOH	r.t. ^c	3	-----	7.5

^a 10 ml of solvent was used.

^b Yield from corypalline (2a) (100 mg).

^c Room temperature.

tuted hydroquinones⁵ and *p*-quinone methides bearing an allylsilyl moiety react intramolecularly to afford cyclized products.⁶ Those results suggested that the *p*-quinol acetate (**1a**) in the presence of acid could react with allylsilane. Here we wish to report a regioselective allylation⁷ to a 1,2,3,4-tetrahydroisoquinoline, corypaline (**2a**), by use of allyltrimethylsilane as a nucleophile.

The *p*-quinol acetate (**1a**), prepared from corypalline (**2a**) by the similar reaction as reported previously,² was not purified but was dissolved in dichloromethane containing allyltrimethylsilane (1.5 eq.).

Then, boron trifluoride etherate ($\text{BF}_3 \cdot \text{Et}_2\text{O}$) (1.5 eq.) was added to the stirred mixture at room temperature and stirring was continued for 1.5 h. Work-up as usual gave an oily product, which was purified by preparative tlc to give rise to 8-allylcorypalline (**3**),⁸ mp 112-113°C, in 22.4% yield. Other reaction conditions⁹ decreased yield of the product. The results are shown in Table.

Structure of the product (**3**) was determined as follows (Scheme 1). *N*-Trifluoroacetylation of a tetrahydroisoquinoline (**4**)¹⁰ followed by debenzylation gave *N*-trifluoroacetylcorypalline (**5**).⁸ By successive reactions (*O*-allylation, *N*-deprotection and *N*-methylation) the phenol (**5**) was transformed to *O*-allylcorypalline (**6**),⁸ the Claisen rearrangement (reflux in *N,N*-dimethylaniline) of which gave rise to an allylphenol (**3**) (29%) being identical with the product derived from the *p*-quinol acetate (**1a**).

On the other hand, when acetonitrile was used as the more polar solvent instead of dichloromethane, direction of the substitution was dramatically changed (see Table). Namely, similar reaction of **1a** in acetonitrile in the presence of $\text{BF}_3 \cdot \text{Et}_2\text{O}$ at 0°C followed by the similar purification as above produced 4-allylcorypalline (**7**),⁸ mp 74-75°C, in 20% yield, structure of which was established spectroscopically. ¹H-Nmr spectrum of **7** showed two aromatic protons (δ 6.50 and 6.65, each singlet) and methylene protons on 1-position (δ 3.31 and 3.53, each doublet, $J=12.9$ Hz). Those spectral data suggested an allyl group to be introduced 3 or 4 position. However, introduction to the 3-position was excluded because a characteristic fragment ion (**A**) (m/z 190) due to 1,2,3,4-tetrahydroisoquinoline bearing an allyl substituent at 4-position appeared in the mass spectrum.

As for formation of two allyl-substituted corypallines (**3** and **7**), plausible pathway could be proposed as shown in Scheme 2. Namely, *o*-quinonoid cation (**8**) was generated in dichloromethane solution, which was similar to that in C-C bond formation reported previously,² while in acetonitrile solution *p*-quinone methide (**9**) was formed *via* **8** or directly from **1a**. Then both intermediates reacted with allyl anion to give 8-(**3**) or 4-allylcorypalline (**7**), respectively. It was noticed that the allylation reaction gave a sole product being dependent on solvent.

In order to investigate scope and limitation of this regioselectivity, reaction of other substrates with organo-silicon reagents is now in progress.

ACKNOWLEDGEMENT

The authors are indebted to Miss N. Sawabe and Mrs. F. Hasegawa, this faculty, for ^1H -nmr and mass spectral measurements and to Sankyo Co., Ltd. for elemental analysis.

REFERENCES AND NOTES

1. B. Umezawa and O. Hoshino, *Heterocycles*, 1975, **3**, 1005.
2. a) H. Hara, O. Hoshino, and B. Umezawa, *Chem. Pharm. Bull.*, 1976, **24**, 253; H. Hara, O. Hoshino, B. Umezawa, and Y. Iitaka, *J. Chem. Soc., Perkin Trans. 1*, **1979**, 2657.
b) H. Hara, O. Hoshino, and B. Umezawa, *Chem. Pharm. Bull.*, 1983, **31**, 730; H. Hara, M. Murakata, O. Hoshino, B. Umezawa, and Y. Iitaka, *Chem. Pharm. Bull.*, 1988, **36**, 1627.
c) H. Hara, O. Hoshino, and B. Umezawa, *Chem. Pharm. Bull.*, 1981, **29**, 51.
3. "The Chemistry of Organic Silicon Compounds", Parts 1 and 2, ed. by S. Patai and Z. Rappoport, John Wiley & Sons, Chichester, 1989; E. W. Colvin, "Silicon Reagents in Organic Synthesis", Academic Press, London, 1988; *Idem*, "Silicon in Organic Synthesis", Butterworth, London, 1981.
4. a) I. Fleming, J. Dunogues, and R. Smithers, *Organic Reactions*, ed. by A. S. Kende, Vol. 37, John Wiley & Sons, New York, Chapter 2, 1989.
b) G. Majetich, "Organic Synthesis, Theory and Application", ed. by T. Hudlicky, Jai Press, Greenwich Vol. 1, 1989.
c) A. Hosomi, *Acc. Chem. Res.*, 1988, **21**, 200.
5. A. Hosomi and H. Sakurai, *Tetrahedron Lett.*, **1977**, 4041.
6. S. R. Angle and K. D. Turnbull, *J. Am. Chem. Soc.*, 1989, **111**, 1136.
7. Recently, a synthesis of 1-allyl-2-methoxycarbonyltetrahydroisoquinoline by using allyltrimethylsilane has been reported; T. Naota, T. Nakato, and S.-I. Murahashi, *Tetrahedron Lett.*, 1990, **31**, 7475.
8. All new compounds gave satisfactory ^1H -nmr and high resolution mass spectral or microanalytical data.
9. Titanium tetrachloride as acid was not used because of producing 8-chlorocorypalline.^{2c}
10. H. Hara, O. Hoshino, and B. Umezawa, *Chem. Pharm. Bull.*, 1985, **33**, 2705.

Received, 19th September, 1991