

**(S)-1,3-DIMETHYL-3-HYDROXY-5-METHOXYOXINDOLE: BY-PRODUCT
IN THE SYNTHESIS OF PHYSOSTIGMINE**

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Abstract- (S)-1,3-Dimethyl-3-hydroxy-5-methoxyoxindole (**4**) was obtained as a by-product in the asymmetric alkylation of oxindole (**1**) with methyl chloroacetate in the presence of *N*-[(4-trifluoromethyl)benzyl]cinchonium bromide. Its structure was established by spectral data. Optically active (**4**) of (S)-configuration was obtained from oxindole (**1**) on air oxidation in the presence of the chiral catalyst *N*-[(4-trifluoromethyl)benzyl]cinchonium bromide.

INTRODUCTION

Asymmetric alkylation of oxindole (**1**) with chloroacetonitrile in the presence of the chiral catalyst *N*-[(4-trifluoromethyl)benzyl]cinchonium bromide (CPTC) afforded a useful intermediate for preparing natural physostigmine.¹ This reaction, when executed with methyl chloroacetate and catalyzed by CPTC in the presence of sodium hydroxide, afforded as a neutral compound the ester (**2**), and as the material soluble in NaOH the acid (**3**) in a total yield of 85%. However, the optical purity of these materials was low (66% ee for **2** and 31.4% ee for **3**). It was found that a polar (tlc) by-product was obtained in this reaction, which was separated by chromatography, and shown on the basis of spectral data and its *m*s of 207 (*M*⁺) to be the oxindole (**4**) with a OH-group at C-3. The compound was optically active, and showed when compared with optically pure material an ee% of the (S)-enantiomer of 14 %. Optically pure **4** was obtained from **1** when treated in benzene and NaOH in the presence of CPTC. The material obtained after workup and chromatography on silica gel afforded **4** of mp 130-131 °C and [α]_D -14.1 ° (c 1.5, EtOH), and after repeated crystallization from ether optically pure **4** (14%) as slightly yellow crystals of mp 116-117 °C and [α]_D -104 ° (c 1.5, EtOH). ¹H-Nmr analysis of this material in using chiral

with 1.5% HCl (12.5 mlx3), washed with brine, dried over Na₂SO₄. After evaporation of solvent, the residue was chromatographed (silica gel, CH₂Cl₂/EtOAc = 14:1) to give product (2) as a gum 50 mg (16.9%): $[\alpha]_D -13.2^\circ$ (c 0.4, EtOH), ee = 66.3%.³ Ms and ¹H-nmr are identical with those of racemic material.⁴ The tlc of the ether extract (silica gel, CH₂Cl₂/MeOH = 9 : 1) showed the presence of a polar impurity of ms 207 (M⁺) which was found to be identical with compound (4) described below.

The aqueous layer obtained from extraction of reaction mixture with ether was acidified with 10% HCl, then extracted by CHCl₃ (25 mlx3). The organic layer was washed with brine, and dried over Na₂SO₄. Evaporation of solvent gave compound (3) as a powder (212 mg, 68.1%): $[\alpha]_D -15.3^\circ$ (c 1, EtOH), ee = 31.4%;³ Elms (m/z): 249 (M⁺); ¹H-Nmr (CDCl₃): δ 1.43 (s, 3H, C3-CH₃), 2.84 (m, 2H, -CH₂-), 3.17 (s, 3H, N-CH₃), 6.67-6.90 (m, 3H, Ar-H), 10.35 (br, 1H, -COOH).

1,3-Dimethyl-3-hydroxy-5-methoxyoxindole (4). To a 25 ml flask was added benzene (12 ml), oxindole (1) (95.5 mg, 0.5 mmol) and CPTC (34.9 mg, 0.075 mmol), then 2 ml of 50% NaOH solution were added. The mixture after standing for 12 h was extracted with ether (10 mlx3). The combined extracts were washed with brine, and dried over MgSO₄. After evaporation of solvent and chromatography (silica gel, petroleum ether : EtOH = 3 : 1), oxindole (4) was obtained as crystals (62 mg, 50%): mp 130-131 °C; $[\alpha]_D -14.1^\circ$ (c 1.5, EtOH), ee = 13%; Elms (m/z): 207 (M⁺, 100%), 190 (M⁺-OH, 35.43%); ¹H-Nmr (CDCl₃): δ 1.60 (s, 3H, C3-CH₃), 3.18 (s, 3H, N-CH₃), 3.80 (s, 3H, O-CH₃), 6.67-7.06 (m, 3H, Ar-H). Anal. Calcd for C₁₁H₁₃NO₃: C 63.71, H 6.28, N 6.76; Found: C 63.64, H 6.17, N 6.58.

Purification and CD-spectrum of 3(S)-1,3-dimethyl-3-hydroxy-5-methoxyoxindole (4). Compound (4) (955 mg) was dissolved in ether (80 ml) and refluxed until complete dissolution. The solution was cooled to room temperature and left standing in the refrigerator overnight. Recrystallization (2 X) gave pure (4) (65 mg, 13.6%): mp 116-117 °C; $[\alpha]_D -104^\circ$ (c 1.5, EtOH), $[\alpha]_D -85.3^\circ$ (c 0.4, CHCl₃); cd spectrum in MeOH: $\Delta\epsilon$ 220 nm, +24.8; $\Delta\epsilon$ 260 nm, -9; ¹H-nmr by using Eu(hfc)₃ showed ee $\geq$ 95%.

Determination of the optical purity of 4. 200 MHz ¹H-Nmr (CDCl₃): addition of Eu(hfc)₃ showed δ 1.60 (s, 3H, C3-CH₃) as two single peaks, with an ee $>$ 95% for the pure material and an ee = 13% for the original material. The $[\alpha]_D$ values measured for these compounds agree with nmr analysis.

The absolute configuration of optically pure hydroxyoxindole (4), 3(S)-3-Benzoyloxy-1,3-dimethyl-5-methoxyoxindole (5). 50 mg (0.242 mmol) of optically pure 4 and 36 mg (0.255 mmol) of benzoyl chloride were added with CCl₄ (10 drops) and dry pyridine (10 drops). The mixture was allowed to stand at room temperature for 12 h. The reaction mixture was dissolved in ether, washed with 10% NaOH, 10% HCl, brine, and then dried over MgSO₄. Evaporation of solvent gave compound (5) as a crystal (68 mg, 90%): mp 75-77 °C; $[\alpha]_D -39.3^\circ$ (c 0.15, CHCl₃); Elms (m/z): 311 (M⁺); ¹H-nmr (CDCl₃): δ 1.50 (s, 3H, C3-CH₃), 3.55 (s, 3H, N-CH₃), 3.70 (s, 3H, O-CH₃), 6.67-7.20 (m, 8H, Ar-H); cd spectrum in MeOH: $\Delta\epsilon$ 220 nm, +24.8; $\Delta\epsilon$ 260 nm, -11.3.

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