

REACTIVITY OF OXYTRYPTAMINES

CONVERSION TO 3-(*o*-AMINOPHENYL)-2-PYRROLIDONE AND KYNURENINE

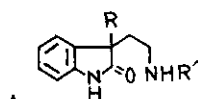
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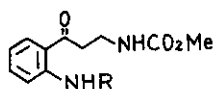
Oxytryptamine hydrochloride **1a** HCl was converted to 3-hydroxytryptamine **1c** (14%) together with kynurettamine derivative **2a** (3%) and 1-ethoxy-benzo(d)-(1,3)-oxazine-3-one **3b** (6%) when an ethanolic solution of **1a** HCl was basified with 10% NaOH and the mixture was stirred for 1 hr at room temperature in an open air followed by treatment with methyl chloroformate. No detectable amount of **4** was found in the reaction mixture. Instead, **1a** was recovered in 43% yield as **1b**. The similar reaction of **1a** in methanol gave **2a** (2%), **3a** (3%), **1c** (7%), and **1b** (40%). The similar treatment of **1a** HCl, under an oxygen atmosphere, for 15 min led to a 28% yield of **3b**, along with **2a** (13%) and **2b** (2%). The prolonged treatment of **1a** for 60 min under the similar condition and work-up yielded **2a** as the main product in 31% yield, accompanied with **2b** (3%). In contrast, when the reaction mixture was stirred for 60 min in nitrogen or argon atmosphere followed by the similar work-up, **1a** was recovered as **1b** in 71% yield. The Rose Bengal-sensitized photooxygenation of **1b** in MeOH in the presence of MeONa provided **1d** in 33% yield which was converted to **3a** (54%) and **2a** (9%) when treated with 10% NaOH for 30 min.

These results suggest the instability of **1a** might be associated with its susceptibility to air oxidation under alkaline conditions. A possible pathway to account for the unexpected formation of **2,3**, and **1c** is discussed.

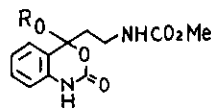
On the other hand, the intramolecular N,N'-transacylation of **1a** to **4** was achieved by refluxing **1a** with 10% NaOH in MeOH under argon for 11 hr and a 30% yield of **4**, mp 120.5-121.5°, was obtained. 3-(*o*-Aminophenyl)-2-pyrrolidone **4** was readily reverted to **1a** in an acidic media such as AcOH-MeOH or HCl-MeOH.



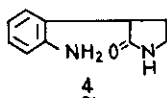
- 1**
 a. R = R' = H
 b. R = H, R' = CO₂Me
 c. R = OH, R' = CO₂Me
 d. R = OH, R' = CO₂Me



- 2**
 a. R = H
 b. R = CO₂Me



- 3**
 a. R = Me
 b. R = Et



4