

TOTAL SYNTHESSES OF (±)-CHANOCLAVINE I AND (±)-DIHYDROSETOCLAVINE

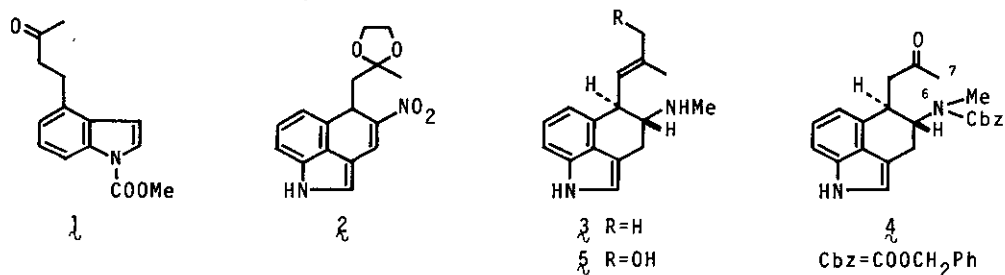
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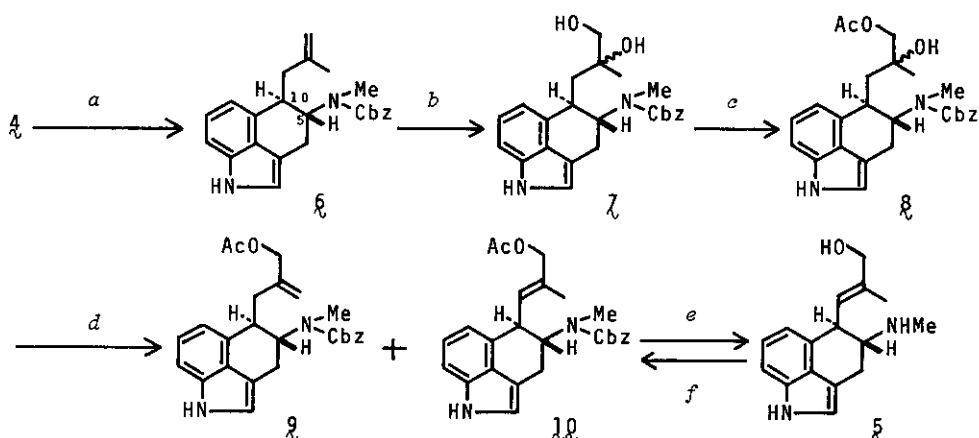
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Abstract: Syntheses of the title ergot alkaloids $\mathbf{1}$ and $\mathbf{10}$ were achieved from the common intermediate $\mathbf{2}$, obtained by a series of reactions including our synthetic method of 4-alkylindoles.

In the previous three papers,^{1,2,3} we reported (i) a synthetic method of functionalized 4-alkylindoles such as $\mathbf{1}$, (ii) its transformation into a tricyclic indole derivative $\mathbf{2}$, which is expected to be a common intermediate for the synthesis of ergot alkaloids, and (iii) the first synthesis of 6,7-secoagroclavine ($\mathbf{3}$) from $\mathbf{2}$ by way of a N-protected ketone derivative $\mathbf{4}$. $\mathbf{4}$ is an important compound for the synthesis of 6,7-secoergoline type of alkaloids and this time, a synthesis of (±)-chanoclavine I⁴ ($\mathbf{5}$) was carried out as shown in Chart 1.^{5,6}



Owing to the insoluble character of chanoclavine I in most organic solvents, identification [TLC, ¹H NMR (CDCl₃), and IR (CHCl₃)] was performed at the stage of the compound $\mathbf{10}$. Natural $\mathbf{5}$ was treated with ClCOOCH₂Ph in the presence of Et₃N and the resulting diacyl derivative was partially hydrolyzed⁷ to an N-benzyloxy-carbonyl alcohol, which was acetylated to afford $\mathbf{10}$ of the natural origin. Both natural and synthetic $\mathbf{10}$'s were treated with warm diluted alkali,⁷ followed by cleavage of the N-protecting group, and the recovered natural $\mathbf{5}$ was identical with chanoclavine I [mixed mp, IR (KBr)]. Synthetic $\mathbf{5}$ ⁸ exhibited the same MS pattern as natural $\mathbf{5}$, thus completing a total synthesis of (±)- $\mathbf{5}$.



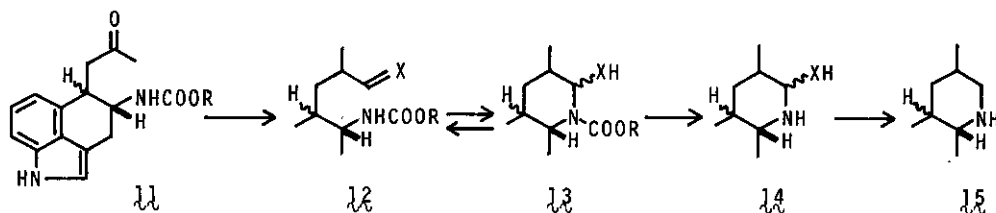
a $\text{Ph}_3\text{P}=\text{CH}_2$, THF, 0° , 46% (78%[†]). *b* OsO_4 , $\text{Et}_2\text{O}-\text{Py}$, $0^\circ \rightarrow \text{rt}$, 70% (80%[†]). *c* Ac_2O , Py, quant. *d* $p\text{-TsOH}$, PhH, reflux. 9: 22%, 10: 25%. *e* (i) 2% KOH in $t\text{-BuOH}-\text{H}_2\text{O}$ (3:1), $55-60^\circ$, (ii) Na, liq. NH_3 -THF. *ca.* 80% yield for both synthetic and natural compounds. *f* (i) $\text{ClCOOCH}_2\text{Ph}$, CH_2Cl_2 -Py, Et_3N , (ii) 3.5% KOH in $t\text{-BuOH}-\text{H}_2\text{O}$ (3:1), $55-60^\circ$, (iii) Ac_2O , Py.

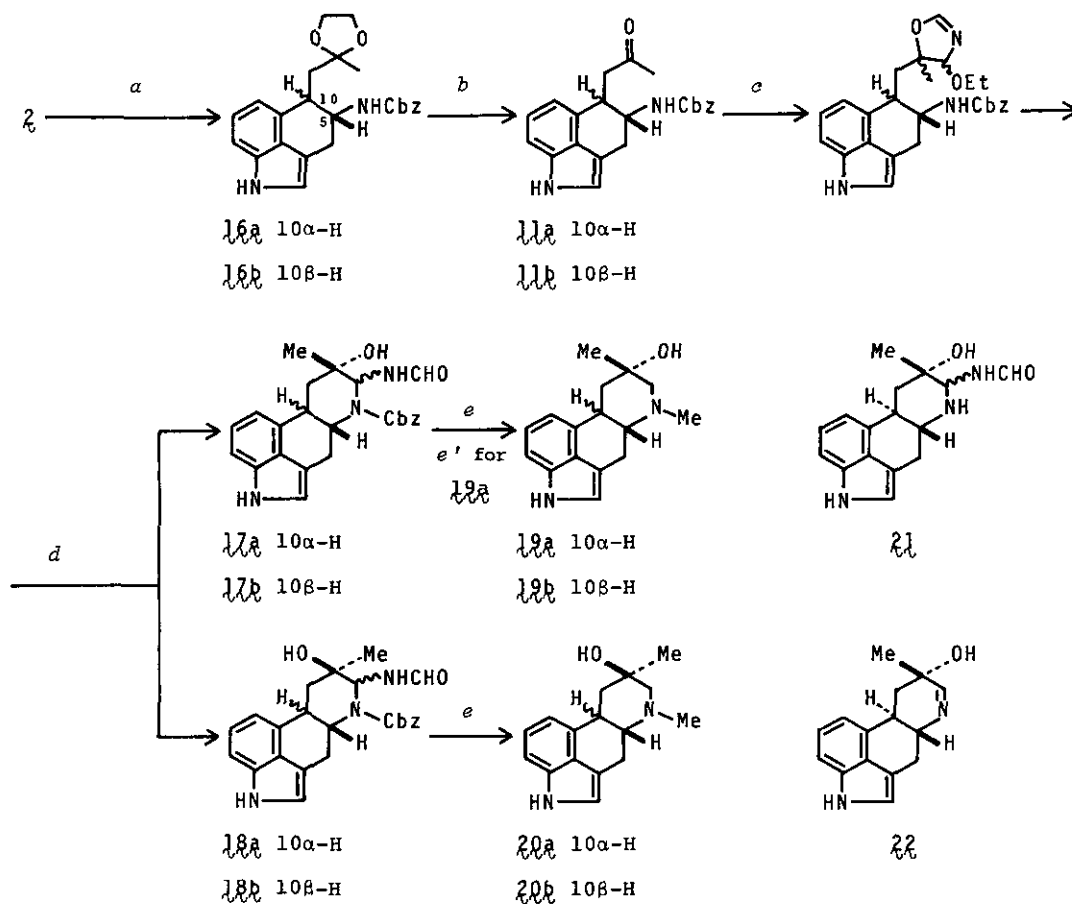
[†] Yield calculated on the basis of converted starting material.

Chart 1

Construction of the tetracyclic ergoline skeleton was next attempted by assuming an intramolecular cyclization from 12 to 13, if one could achieve the introduction of an aldehyde equivalent into the ketone group of 11. When R equaled to the benzyl group, removal of the N-protecting group from 13 by the catalytic hydrogenation would produce 14 at first and then end up in the formation of a stable D ring as 15, whereas, in the case of $\text{R}=\text{Me}$, any reaction on 13 might involve the participation of an equilibrium form 12 to afford ring-opened derivatives as by-products. Based on this consideration, 11a⁵ and 11b⁵ were synthesized from 2 (Chart 2) and submitted to the one carbon elongation reaction producing an aldehyde function.⁹

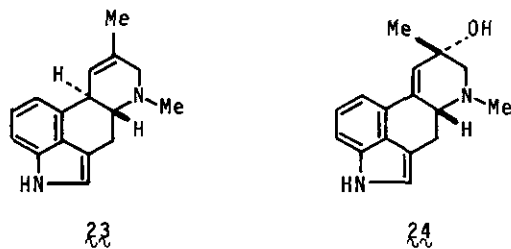
A satisfactory result was obtained by the condensation of tosylmethylisocyanide (TosMIC) with 11b using TlOEt as a base¹⁰ and subsequent treatment with $p\text{-TsOH}$ in





a (i) LiAlH_4 , THF, reflux, (ii) $\text{ClCOOCH}_2\text{Ph}$, CH_2Cl_2 , Et_3N ; **16a**: 28%, **16b**: 33%. *b* Me_2CO , *p*-TsOH, rt; **11a**: 85%, **11b**: 85%. *c* *p*-TsCH₂NC, TiOEt , EtOH-DME (4:1), rt. *d* *p*-TsOH, $\text{DME-H}_2\text{O}$ (6:1), rt. *e* H_2 , 10% Pd-C, $\text{CH}_2\text{O-H}_2\text{O}$, MeOH. *e'* (i) H_2 , 10% Pd-C, MeOH, (ii) 5% KOH in $\text{MeOH-H}_2\text{O}$ (14:1), reflux, (iii) H_2 , 10% Pd-C, $\text{CH}_2\text{O-H}_2\text{O}$, MeOH.

Chart 2



DME-H₂O afforded **17b** [MS *m/e*: 419 (M⁺), 401 (M⁺-H₂O), 356 (M⁺-H₂O-NH₂CHO); ¹H NMR (CDCl₃, 60°) ¹¹ δ: 1.45 (≡C-Me), 7.93 (>N-CHO)] and **18b** [MS *m/e*: 419 (M⁺), 401 (M⁺-H₂O); ¹H NMR (CDCl₃, 60°) δ: 1.20 (≡C-Me), 8.35 (>N-CHO)], which were hydrogenated over 10% Pd-C in the presence of CH₂O. Formation of (±)-**19b**⁸ [mp 88-91°, MS *m/e*: 256 (M⁺), ¹H NMR (CDCl₃) δ: 1.34 (s, Me), 2.19 (br. s, OH), 2.39 (s, N-Me), 6.88 (br. s, H-2), 7.96 (br., indole NH)] and (±)-**20b**⁸ [mp 217-219°, MS *m/e*: 256 (M⁺), ¹H NMR (CDCl₃-CD₃OD) δ: 1.19 (s, Me), 2.57 (s, N-Me)] was observed in 33% and 20% yields, respectively, from **11b**. The structure of **19b** was confirmed by comparison with a hydrogenation product of setoclavine (*vide infra*).

The same series of reaction were applied to **11a**. A mixture of the tetracyclic derivatives **17a** and **18a** was formed analogously, but the catalytic hydrogenation required the prolonged reaction time, and yet **21** [¹H NMR (DMSO-d₆) δ: 1.11 (≡C-Me), 8.12 and 8.15 (-NHCHO), 10.58 (indole NH)] was isolated in addition to (±)-dihydroisotococlavine^{8,12} (**20a**) [mp 232-236°, MS *m/e*: 256 (M⁺), ¹H NMR (CDCl₃-CD₃OD) δ: 1.50 (s, Me), 2.43 (s, N-Me)] in 16% yield from **11a**. **21** was once treated with 5% KOH in MeOH-H₂O (formation of **22**), followed by the catalytic hydrogenation in the presence of CH₂O. (±)-Dihydrosetoclavine⁸ (**19a**) [mp 252-256°, MS *m/e*: 256 (M⁺), ¹H NMR (DMSO-d₆) δ: 1.18 (s, Me), 2.37 (s, N-Me), 10.63 (br., indole NH)] was obtained in 12% yield from **11a**.

In order to confirm the structures of synthetic **19a** and **19b**, preparation of the authentic samples was carried out from agroclavine (**23**). **23** was converted to setoclavine¹³ (**24**) according to the procedure described in the literature^{4a,14} and the catalytic hydrogenation¹⁵ of **24** yielded dihydrosetoclavine^{13,16} (**19a**) (43%) and **19b** (7%). Identification of (±)-**19a** with dihydrosetoclavine was performed by TLC (10% MeOH-CHCl₃, R_f=0.2), MS, ¹H NMR (CDCl₃-CD₃OD, DMSO-d₆), and ¹³C NMR (DMSO-d₆) spectra.

Acknowledgement — Authors' thanks are due to Dr. S. Ohmomo of the University of Tsukuba for his exceptional cooperation in providing us with precious samples of chanoclavine I and agroclavine. A part of this work was supported by Grant-in-Aid for Special Project Research from the Ministry of Education, Science and Culture, which is gratefully acknowledged.

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8. Satisfactory results of high resolution mass spectra were obtained for these
compounds.
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ful results, see Ref. 3.
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13. The material exhibited the same IR spectrum as reported in the literature.^{4a}
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Received, 8th December, 1980