

A GENERAL SYNTHESIS OF INDOLO[2,3-a]QUINOLIZINE FROM A SYMMETRICAL STARTING MATERIAL, CIS- Δ^4 -TETRAHYDROPHTHALIC ACID ANHYDRIDE

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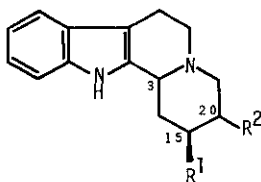
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Abstract — A general synthesis of indolo[2,3-a]quinolizine derivatives (12) and (13) has been achieved by a condensation of tryptamine with the aldehyde (11) derived from a symmetrical starting material, cis- Δ^4 -tetrahydrophthalic acid anhydride (2).

In the field of alkaloid chemistry, the indoloquinolizine skeleton (1) is a common structure to a number of indole alkaloids. Change in the geometry of the hydrogen atoms attached to certain vital centers (position 3, 20) completely alters the properties of this family such as corynantheine,¹ hirsuteine,² corynantheidine,³ and spiciociliatine.⁴



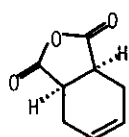
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Recently we have shown the anhydride (2) to be an excellent synthon for the construction of the pentacyclic skeleton of yohimbine family.⁵ In our continuous efforts for the synthesis of natural products and its related compounds⁶ using a symmetrical starting material, we intrigued a synthesis of the corynanthe-type indole alkaloids. Here we wish to report a synthesis of the indolo[2,3-a]quinolizine derivatives which

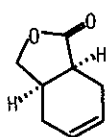
could be potential intermediates leading to indole alkaloids such as dehydrocorynantheol and corynantheine.

The diacetal (5), prepared in 55.7 % yield via dialdehyde (4) from the known γ -butyrolactone (3)⁷ derived from cis- Δ^4 -tetrahydrophthalic acid anhydride (2) was heated with potassium cyanide in dimethyl sulfoxide at 190°C to give the cyanated acid (6). Without further purification, the crude carboxylic acid (6) was treated with an ethereal diazomethane to produce the trans orientated methyl ester (7)⁸ [ir (CHCl₃) 2240, 1720 cm⁻¹; δ 3.70 (3H, s, OCH₃), 3.70 - 4.07 (8H, m, 2 x CH<_O-CH₂), 4.90 (1H, t, J = 4 Hz, CH<_O), 4.92 (1H, t, J = 4 Hz, CH<_O); mass m/e 299 (M⁺)] in 74 % yield from the γ -butyrolactone (5). The conformation of λ was determined by a comparison with an authentic sample derived from λ via known compound (8).⁹ The reduction of this ester (7) with 4 equivalents of diisobutylaluminum hydride in toluene at -60°C afforded, in 80.2 % yield, the alcohol (9) [ir (CHCl₃) 3600 - 3200, 2250 cm⁻¹; δ 2.63 - 3.0 (1H, br s, OH, exchanged with D₂O), 3.50 - 3.77 (2H, br s, CH₂OH), 3.88, 3.93 (each 4H, each s, 2 x CH<_O-CH₂), 4.9 (2H, t, J = 4 Hz, 2 x CH<_O)] which on treatment with p-toluenesulfonyl chloride in pyridine at 0°C produced the tosylate (10) in 95 % yield [ir (CHCl₃) 2250, 1180 cm⁻¹; δ 2.45 (3H, s, ArCH₃), 3.67 - 3.97 (8H, m, 2 x CH<_O-CH₂), 3.97 - 4.23 (2H, m, CH₂OSO₂), 4.80 (2H, t, J = 4 Hz, 2 x CH<_O), 7.30 (2H, d, J = 8 Hz, ArH), 7.73 (2H, d, J = 8 Hz, ArH); mass m/e 425 (M⁺)]. The conversion of nitrile to aldehyde was also done by use of diisobutylaluminum hydride. Thus, the reduction of the nitrile (10) with 6 equivalents of diisobutylaluminum hydride followed by a treatment of the mixture with saturated ammonium chloride solution produced the desired aldehyde (11) in 82.4 % yield [ir (CHCl₃) 2730, 1725, 1180 cm⁻¹; δ 2.43 (3H, s, ArCH₃), 3.82, 3.85 (each 4H, each s, 2 x CH<_O-CH₂), 3.90 - 4.23 (2H, m, CH₂OSO₂), 4.63 - 5.0 (2H, m, 2 x CH<_O), 7.28 (2H, d, J = 8 Hz, ArH), 7.25 (2H, d, J = 8 Hz, ArH), 9.60 (1H, br s, CHO)]. Since this aldehyde is unstable, it was used directly in the next step. Thus, heating the aldehyde (11) with tryptamine in acetic acid at 85°C for 1 hr afforded a separable mixture of indoloquinolizines, 3S,15R,20R-diethylenedioxyethylindoloquinolizine (12) in 27.3 % yield [ir (CHCl₃) 3460 cm⁻¹; δ 3.85, 3.89 (each 4H, each s, 2 x CH<_O-CH₂), 4.67 - 5.14 (2H, m, 2 x CH<_O), 6.80 - 7.48 (4H, m, ArH), 7.80 - 8.12 (1H, br s, NH, exchanged with D₂O); mass m/e 398.2188 (M⁺)] and 3R,15R,20R-diethylenedioxyethylindoloquinolizine (13) in 28.5 % yield [ir (CHCl₃) 3460 cm⁻¹; δ 3.80, 3.85 (each 4 H, each s, 2 x CH<_O-CH₂), 4.67 - 5.08 (2H, m, 2 x CH<_O), 6.80 - 7.48 (4H, m, ArH), 7.88 - 8.08 (1H, NH, exchanged with D₂O); mass m/e 398.2186 (M⁺)], whose configuration at C₃ is not clear at this stage.

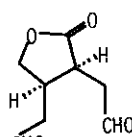
In order to certify the configuration at C₃ of these indoloquinolizines, the compound (13) was dehydrogenated with mercuric acetate¹⁰ to give the iminium base (14) which on reduction with sodium borohydride produced 11 in 80.8 % yield. On the other hand, treatment of 12 under the same conditions resulted in recovered starting material. These conversions indicated that the compounds (12) and (13) were the C(3)-epimers of each other.



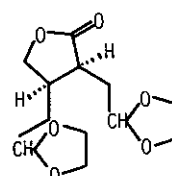
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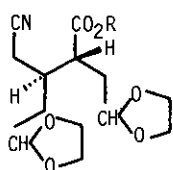
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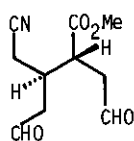
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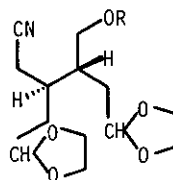
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6 : R=H

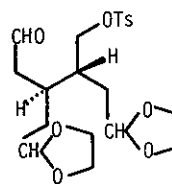
7 : R=CH₃

8

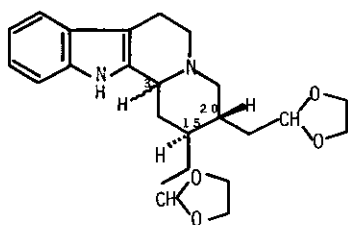


9 : R=H

10 : R=Ts

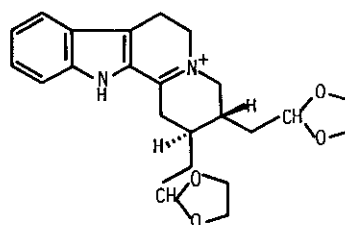


11



12 : 3αH

13 : 3βH



14

Thus, we achieved a general synthesis of the indolo[2,3-a]quinolizine derivatives by a condensation of tryptamine with the aldehyde (11) derived from a symmetrical starting material, *cis*- Δ^4 -tetrahydrophthalic acid anhydride (2) and these compounds would be potential intermediates leading to corynan, dihydrocorynantheol and corynantheidol. According to this methodology, a synthesis of corynanthe-type indole alkaloids is under investigation in our laboratory.

REFERENCES AND NOTES

1. E. E. van Tamelen, P. E. Alderich, and T. J. Katz, J. Amer. Chem. Soc., 1957, 79, 6426, and references are cited therein.
2. S. Sakai and N. Shinma, Chem. and Pharm. Bull. (Japan), 1978, 26, 2596.
3. M. F. Bartlett, R. Sklar, W. I. Taylor, E. Schlitter, R. L. S. Amai, P. Beak, N. V. Bringi, and E. Wenkert, J. Amer. Chem. Soc., 1962, 84, 622.
4. W. F. Trager, C. M. Lee, and A. H. Berckett, Tetrahedron, 1967, 23, 365, 375.
5. T. Suzuki, A. Tomino, K. Unno, and T. Kametani, Heterocycles, 1979, 13, 301.
6. T. Suzuki, S. Kagaya, A. Tomino, K. Unno, and T. Kametani, J. Chem. Soc. Perkin I, 1980, 2801; T. Suzuki, A. Tomino, K. Unno, and T. Kametani, Chem. and Pharm. Bull. (Japan), 1981, 29, 76; T. Kametani, T. Suzuki, S. Kamada, and K. Unno, J. Chem. Soc. Perkin I, in press; T. Kametani, T. Suzuki, A. Tomino, S. Kamada, and K. Unno, Heterocycles, 1981, 16, 905; T. Kametani, T. Suzuki, E. Sato, and K. Unno, J. Chem. Soc., Chem. Comm., in press.
7. B. Belleau and J. Puranen, Canad. J. Chem., 1965, 43, 2551.
8. This compound would be formed as a result of the epimerization at C₂₀ during the introduction of the cyano group into the γ -butyrolactone (4); see reference 7) and T. Suzuki, A. Tomino, K. Unno, and T. Kametani, Heterocycles, 1980, 14, 439; Chem. and Pharm. Bull. (Japan), in press.
9. T. Kametani, T. Suzuki, S. Kamada, and K. Unno, Tetrahedron Lett., to be submitted.
10. H. T. Openshaw and N. Whittaker, J. Chem. Soc., 1963, 1461; N. J. Dastoor, A. A. Gorman, and J. Schmid, Helv. Chim. Acta, 1967, 50, 213; S. Takano, S. Hatakeyama, and K. Ogasawara, Tetrahedron Lett., 1978, 2519; S. Takano, K. Masuda, and K. Ogasawara, J. Chem. Soc., Chem. Comm., 1980, 887.

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