

Although 2 are extremely stable not always in solid state at room temperature in the open air but also in ordinary solvent (CHCl_3 , THF, EtOH, benzene, etc) on reflux, sudden change of the diazo structures of 2 into 3 can be observed at the temperatures above ca. 100°C . Thus, when 1 mM of 2a-e¹ were allowed to be heated in 6 ml of refluxing toluene, yellow color of the solution disappeared and almost colorless crystals of pure 3a-e precipitated within a short period (< 30 min), which were isolated nearly quantitatively ($> 85\%$) simply by filtration of the reaction mixture on cool. Mass spectral and microanalyses data³ showed that these new crystals are isomers of the starting 2a-e. $^1\text{H-NMR}$ spectra were consistent with the assignment of the benzopyranopyrazole structures of 3. In the NMR spectrum of 3a in DMSO-d_6 , for instance, the characteristic diazomethyl proton (δ 5.90) and C^3 proton (δ 6.56) singlet signals of 2a disappeared and a new singlet signal attributable to $\text{C}^1\text{-H}$ (δ 8.63) and a broad singlet NH signal (δ 14.56, exchangeable with D_2O) appeared instead in addition to aromatic proton signals [δ 7.28-7.40 (3H, m, $\text{C}^{6-8}\text{-H}$), 7.94 (1H, d, $\text{C}^9\text{-H}$)]. Furthermore, N-methyl derivative of 3a (needles from isoPrOH, mp $163\text{-}164^\circ\text{C}$) obtained by treatment with diazomethane was identical with the sample prepared by the different route as reported.⁴ Some physical and spectral data of 3a-e are depicted in Table I. All the crystals of 3a-e do not melt below 270°C but tend to be sublimed at higher temperatures. 3c and 3e showed strong fluorescence (emission maximum in EtOH: 402 nm for 3c, 510 nm for 3e).

Table I. Benzopyrano[3,4-c]pyrazol-4(3H)-ones 3

Compound	Yield (%)	Appearance Recryst. Solvent	MS m/e [M ⁺]	IR $\nu_{\text{max}}^{\text{KBr}}$ cm ⁻¹		$^1\text{H-NMR}$ δ (DMSO-d_6)	
				NH	CO	$\text{C}^1\text{-H}$	NH
<u>3a</u>	85	needles (dioxane)	186	3282	1726	8.63	14.56
<u>3b</u>	91	leaves (dioxane)	200	3298	1727	8.56	14.50
<u>3c</u>	90	pale yellow needles (THF)	216	3284	1723	8.52	14.45
<u>3d</u>	95	pale yellow prisms (dioxane)	244	3278	1727 1757	8.71	14.52
<u>3e</u>	86	pale yellow prisms (CH_3CN)	257	3290	1724	8.46	14.24

Since the above transformations of 2 proceed rapidly at 100°C or above in dioxane, DMF, BuOH or pyridine as well as in toluene and also at 120-130°C without solvent, but proceed slowly at 90°C and never below 70°C regardless of the solvent used, the isomerization is supposed to be thermo-dependent 1,5-electrocyclic type of ring closure reaction which affords formal 1,3-dipolar cycloaddition product 3' followed by hydrogen migration leading to 3. As the final product we assigned the N³-H structure 3 rather than the tautomeric N²-H structure 3'' on the assumption that 3 is more stable. Possibilities of the other tautomeric structures can be eliminated from the NMR and IR spectral data. In spite of a number of intramolecular cyclization of diazoalkenes using tosylhydrazone precursors indirectly,⁵ those of the isolated pure diazo compounds of heterocyclic system such as 2 are apparently unknown.

4-Diazomethylcoumarins 2 are reported¹ to be easily obtained by three steps from the corresponding 4-methylcoumarins 1, and the isomerizations of 2 are very facile to afford 3 in high yields. Therefore, the present set of transformations starting from 1 through stable diazo intermediates 2 appears to provide a convenient route to benzopyrano[3,4-c]pyrazol-4(3H)-ones 3. Investigations on similar cyclizations of analogous stable aryl diazomethanes are in progress.

REFERENCES AND NOTES

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