

HETEROCYCLES. XIV.¹ EFFICIENT STEREOCONTROLLED SYNTHESIS OF
RACEMIC FLAVONOIDS

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Abstract — Stereocontrolled synthesis of the flavonoids
(2, 4 and 5) is described. Racemic taxifolin (dihydro-
quercetin) (14) is synthesized by application of this method.

Hydroxyflavanonols and flavan-3,4-diols naturally occur in a variety of hardwoods and barks. The majority of these compounds were stereochemically established, and several racemic flavonoids were synthesized. For example, (2R*,3R*)-flavanonol (2) was prepared by alkaline hydrogen peroxide oxidation of 2'-hydroxy-chalcone (1), but the yield was very poor owing to its facile autoxidation to flavonol (3) in the alkaline medium.² (2R*,3S*,4R*)-flavan-3,4-diol (4) was obtained on hydrogenation of 2, and the 4S*-isomer (5) was prepared from 2 through the oxime and the corresponding amine.³ Our aim is to synthesize naturally occurring chiral flavonoids. We now report, as a model study, efficient stereocontrolled synthesis of the flavonoids (2, 4 and 5) and (2R*,3R*)-taxifolin (dihydroquercetin).

Treatment of 1 with methoxymethyl chloride in an alkaline medium gave the ether (6) (85%). Alkaline hydrogen peroxide oxidation of 6 yielded the epoxide (7) (92%), in which the $\alpha R^*, \beta S^*$ -configuration was deduced by $J_{\alpha, \beta}$ 2 Hz observed in the ¹H NMR spectrum.⁴ Treatment of 7 with trifluoroacetic acid provided 2 (92%) (2R*,3R*; $J_{2,3}$ 12.5 Hz). On reduction with sodium borohydride, 2 gave 4 (84%) (2R*,3S*,4R*; $J_{2,3}$ 10 Hz and $J_{3,4}$ 8.5 Hz) as a sole product. Reduction of 7 with sodium borohydride gave the erythro-epoxy alcohol (8) (68%) and the threo-isomer (9) (10%). The configurations of these compounds were deduced on the basis of those of the compounds obtained by the next reactions. On treatment with trifluoroacetic acid or hydrochloric acid, 8 and 9 gave 5 (71%) and 4 (71%), respectively. The 2R*,3S*,4S*-configuration in 5 was deduced by $J_{2,3}$ 9 Hz and $J_{3,4}$ 3.5 Hz. It is clear that the exclusively stereocontrolled

formations of the chroman rings in the compounds (2, 4 and 5) occurred via intramolecular S_N2 reactions.

(2R*,3R*)-Taxifolin (14) was prepared by the same process as employed for the synthesis of 2 in 92% yield. This is the first total synthesis of 14.

On reduction with sodium borohydride at -30°C , the tetramethyl ether (15), derived from 14, gave two isomeric flavan-3,4-diols (16 and 17) in 15 and 67% yields, respectively. The same reduction at refluxing temperature provided 16 and 17 in 33 and 49% yields, respectively.

EXPERIMENTAL SECTION

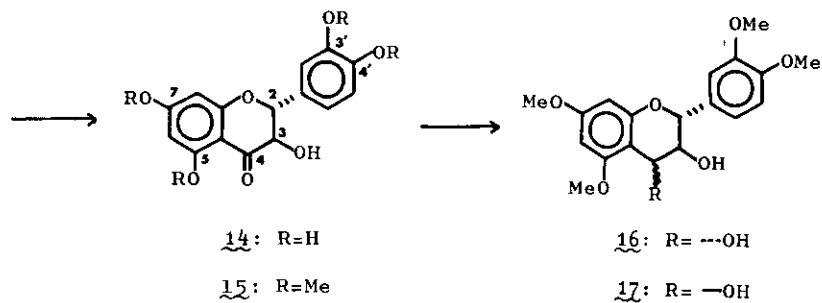
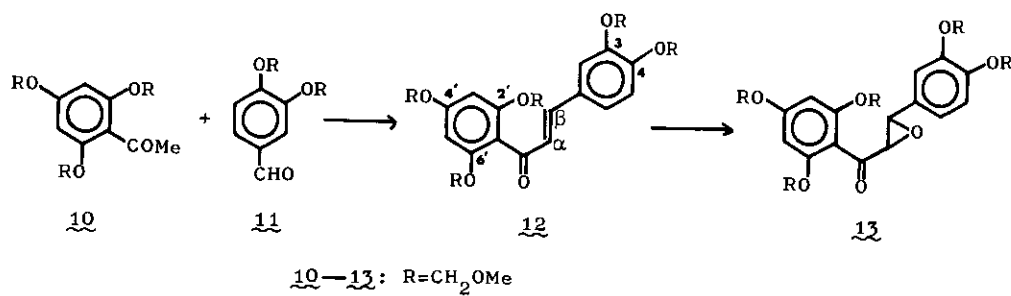
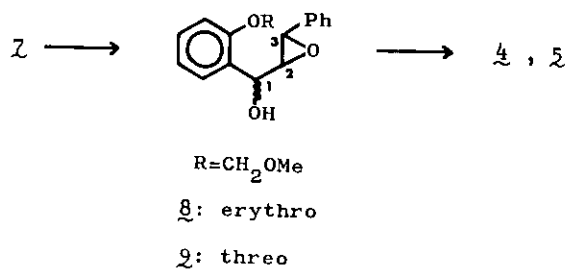
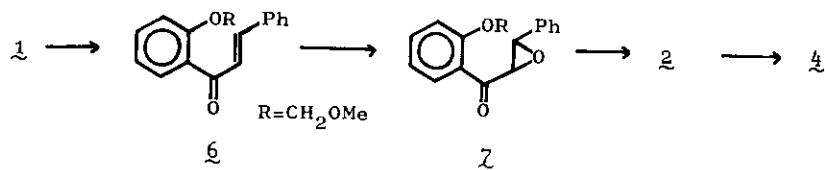
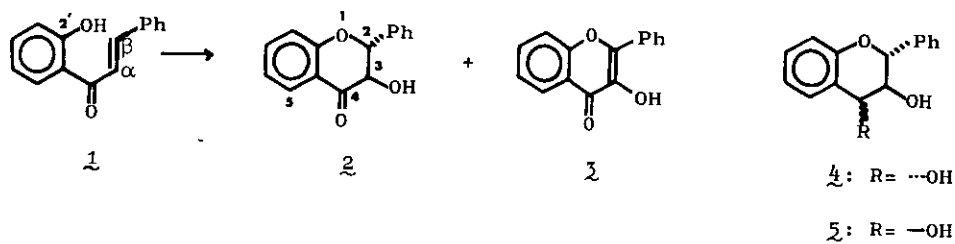
Melting points are uncorrected. Spectral data were recorded on the following spectrometers: IR — Hitachi 260-30; ^1H NMR — Varian EM-390 (90 MHz); MS — JEOL JMS DX-300.

2'-Methoxymethoxychalcone (6)

A mixture of 2'-hydroxychalcone (1)⁵ (220 mg), 1N NaOH (10 ml) and CH_2Cl_2 (10 ml) was stirred at room temperature for 10 min. $\text{N}(\text{Bu})_4\text{Cl}$ (30 mg) and then a solution of $\text{CH}_3\text{OCH}_2\text{Cl}$ (0.2 ml) in CH_2Cl_2 (0.5 ml) were added, and the mixture was stirred at room temperature for 6 h. The organic phase was washed with H_2O and dried over Na_2SO_4 , then concentrated in vacuo. The residue was purified by prep. TLC (Al_2O_3 , C_6H_6) to yield 6 (230 mg, 85%) as a colorless oil, R_f 0.60. IR $\nu_{\text{max}}^{\text{film}}$ cm^{-1} : 1650 (CO). ^1H NMR (CDCl_3) δ : 7.68–6.68 (11H, m, aromatic and vinylic H's), 5.68 (2H, s, OCH_2O), 3.42 (3H, s, OMe). MS Calcd for $\text{C}_{17}\text{H}_{16}\text{O}_3$: M, 268.110. Found m/z : M^+ , 268.109.

(αR *, βS *)- α,β -Epoxy-2'-methoxymethoxychalcone (7)

30% H_2O_2 (0.2 ml) and 1N NaOH (0.2 ml) were added to a solution of 6 (170 mg) in MeOH (5 ml), and the mixture was stirred at room temperature for 5 h. The reaction mixture was taken up in CHCl_3 , and the organic phase was washed with 5% aq. KI and then 5% aq. $\text{Na}_2\text{S}_2\text{O}_3$. Work-up gave an oil, which was purified by prep. TLC (Al_2O_3 , CHCl_3) to yield 7 (165 mg, 92%) as colorless needles of mp $85\text{--}90^\circ\text{C}$ (MeOH), R_f 0.70. IR $\nu_{\text{max}}^{\text{CHCl}_3}$ cm^{-1} : 1678 (CO). ^1H NMR (C_6D_6) δ : 7.94 (1H, dd, J 7.5 and 2 Hz, 6'-H), 7.23–6.73 (8H, m, aromatic H's), 4.65, 4.44 (each 1H, d, J 7 Hz, OCH_2O), 4.10, 3.96 (each 1H, d, J 2 Hz, α - and β -H's), 2.78 (3H, s, OMe).



MS Calcd for $C_{17}H_{16}O_4$: M, 284.105. Found m/z: M^+ , 284.105.

(2R*,3R*)-Flavanonol (2)

30% CF_3COOH (0.2 ml) was added to a solution of **7** (110 mg) in MeOH (0.3 ml), and the mixture was stirred at room temperature for 30 min, then taken up in $CHCl_3$. The organic phase was washed with 5% aq. $NaHCO_3$ and H_2O , then dried over Na_2SO_4 . Work-up gave an oil, which was crystallized from MeOH to yield **2** (76 mg, 82%) as colorless crystals of mp 172–175°C (lit.²; mp 174–177°C). Additional **2** (9.5 mg, 10%) was obtained from the mother liquor by prep. TLC (silica gel, $CHCl_3$), Rf 0.44. IR ν_{max}^{KBr} cm^{-1} : 3475 (OH), 1690 (CO). 1H NMR ($CDCl_3$) δ : 7.76 (1H, dd, J 8 and 1.5 Hz, 5-H), 7.73–6.96 (8H, m, aromatic H's), 5.13 (1H, d, J 12.5 Hz, 2-H), 4.56 (1H, dd, J 12.5 and 2 Hz, 3-H),⁶ 3.64 (1H, d, J 2 Hz, 3-OH).⁷ MS Calcd for $C_{15}H_{12}O_3$: M, 240.080. Found m/z: M^+ , 240.080.

(1S*,2S*,3S*)-2,3-Epoxy-1-2'-methoxymethoxyphenyl-3-phenylpropanol (erythro) (8)
and The 1R*-Isomer (threo) (9)

$NaBH_4$ (20 mg) was added to a solution of **7** (140 mg) in MeOH (5 ml), and the mixture was stirred at -30–40°C for 2 h. After addition of CH_3COOH (4 drops), the reaction mixture was concentrated in vacuo, and the residue was dissolved in $CHCl_3$. Work-up gave an oil, which was purified by prep. TLC (Al_2O_3 , $CHCl_3$ /hexane=4/1, v/v) to yield **8** (96 mg, 68%), Rf 0.30 and **9** (13.4 mg, 10%), Rf 0.30.

The erythro-Isomer (8): A colorless oil. IR ν_{max} cm^{-1} : 3528 ($\epsilon=53$) (OH) ($c=0.0011$ mol/l, CCl_4). 1H NMR (C_6D_6) δ : 7.80–6.82 (9H, m, aromatic H's), 5.38 (1H, t, J 3 Hz, 1-H),⁶ 4.79 (2H, s, OCH_2O), 4.05 (1H, d, J 2.5 Hz, 3-H), 3.48 (1H, dd, J 3 and 2.5 Hz, 2-H), 3.06 (3H, s, OMe), 2.49 (1H, d, J 3 Hz, 1-OH).⁷

MS Calcd for $C_{17}H_{18}O_4$: M, 286.121. Found m/z: M^+ , 286.121.

The threo-Isomer (9): A colorless oil. IR ν_{max} cm^{-1} : 3652 ($\epsilon=38$), 3572 ($\epsilon=30$), 3548 ($\epsilon=29$) (OH) ($c=0.00085$ mol/l, CCl_4). 1H NMR (C_6D_6) δ : 7.74–6.62 (9H, m, aromatic H's), 5.14 (1H, t, J 4 Hz, 1-H),⁶ 4.79 (2H, s, OCH_2O), 4.02 (1H, d, J 2.5 Hz, 3-H), 3.25 (1H, dd, J 4 and 2.5 Hz, 2-H), 3.06 (3H, s, OMe), 2.53 (1H, d, J 4 Hz, 1-OH).⁷ MS Calcd for $C_{17}H_{18}O_4$: M, 286.121. Found m/z: M^+ , 286.121.

(2R*,3S*,4R*)-Flavan-3,4-diol (4)

(1) $NaBH_4$ (25 mg) was added to a solution of **2** (37.0 mg) in MeOH (5 ml), and the mixture was stirred at room temperature for 2 h. Work-up of the reaction

mixture gave an oil, which was crystallized from ether/hexane to yield 4 (24.0 mg, 64%) as colorless needles of mp 139–142°C (lit.³; mp 145°C). Additional 4 (7.4 mg, 20%) was obtained from the mother liquor by prep. TLC (silica gel, acetone/ $C_6H_6=1/3$, v/v). IR $\nu_{\max}^{KBr}$ cm^{-1} : 3370 (OH). 1H NMR ($CDCl_3$) δ : 7.59–6.83 (9H, m, aromatic H's), 4.80 (1H, dd, J 8.5 and 5.5 Hz, 4-H),⁶ 4.76 (1H, d, J 10 Hz, 2-H), 3.80 (1H, ddd, J 10, 8.5 and 3 Hz, 3-H),⁶ 3.03 (1H, d, J 5.5 Hz, 4-OH),⁷ 2.45 (1H, d, J 3 Hz, 3-OH).⁷ MS Calcd for $C_{15}H_{14}O_3$: M, 242.094. Found m/z: M^+ , 242.094.

(2) 10% HCl/MeOH (2 ml) was added to a solution of 9 (35.0 mg) in MeOH (0.4 ml), and the mixture was refluxed for 30 min. Work-up of the reaction mixture gave 4 (21.0 mg, 71%).

(2R*,3S*,4S*)-Flavan-3,4-diol (5)

This compound was prepared from 8 in 71% yield by the procedure employed for the synthesis of 4 from 9 using CF_3COOH instead of 10% HCl/MeOH. Colorless needles of mp 159.5–163°C ($CHCl_3$) (lit.³; mp 160°C). IR $\nu_{\max}^{KBr}$ cm^{-1} : 3375, 3280 (OH). 1H NMR ($CDCl_3$) δ : 7.25–6.88 (9H, m, aromatic H's), 5.04 (1H, d, J 9 Hz, 2-H), 4.72 (1H, d, J 3.5 Hz, 4-H), 4.04 (1H, dd, J 9 and 3.5 Hz, 3-H), 2.35 (2H, s, 3- and 4-OH's).⁷ MS Calcd for $C_{15}H_{14}O_3$: M, 242.094. Found m/z: M^+ , 242.094.

3,4,2',4',6'-Pentakis(methoxymethoxy)chalcone (12)

This compound was prepared from the acetophenone (10)⁸ and the benzaldehyde (11)⁹ in 86% yield by the procedure described in lit.⁵ A colorless oil. IR $\nu_{\max}^{film}$ cm^{-1} : 1650 (CO). 1H NMR ($CDCl_3$) δ : 7.42 (1H, s, 2-H), 7.37 (1H, d, J 17 Hz, β -H), 7.35 (2H, s, 5- and 6-H's), 6.91 (1H, d, J 17 Hz, α -H), 6.63 (2H, s, 3'- and 5'-H's), 5.28, 5.27, 5.13 (2) (10H, each s, OCH_2O 's), 3.52, 3.42 (4) (15H, each s, OMe's). MS Calcd for $C_{25}H_{32}O_{11}$: M, 508.194. Found m/z: M^+ , 508.194.

(α S*, β R*)- α,β -Epoxy-3,4,2',4',6'-pentakis(methoxymethoxy)chalcone (13)

This compound was prepared from 12 in 92% yield by the same procedure as employed for the synthesis of 7 from 6. A colorless oil. IR $\nu_{\max}^{film}$ cm^{-1} : 1700 (CO). 1H NMR (C_6D_6) δ : 7.24–6.80 (3H, m, 2-, 5- and 6-H's), 6.79 (2H, s, 3'- and 5'-H's), 4.17, 3.96 (each 1H, d, J 2 Hz, α - and β -H's), 4.96, 4.92, 4.89, 4.87 (2) (10H, each s, OCH_2O 's), 3.22 (1), 3.19 and 3.16 (4) (15H, each s, OMe's). MS Calcd for $C_{25}H_{32}O_{12}$: M, 524.189. Found m/z: M^+ , 524.189.

(2R*,3R*)-Taxifolin (14)

This compound was prepared from 13 in 92% yield by the procedure employed for the synthesis of 2 from 7 using 10% HCl/MeOH instead of CF₃COOH. Colorless plates of mp 232–234°C (H₂O) (lit.¹⁰; mp 234–236°C). IR $\nu_{\text{max}}^{\text{KBr}}$ cm⁻¹: 3300 (OH), 1640 (CO). ¹H NMR (MeOD) δ : 7.10 (1H, s, 2'-H), 6.94 (2H, s, 5'- and 6'-H's), 6.02, 5.97 (each 1H, d, J 2.5 Hz, 6- and 8-H's), 4.99 (1H, d, J 12 Hz, 2-H), 4.54 (1H, d, J 12 Hz, 3-H). MS Calcd for C₁₅H₁₂O₇: M, 304.058. Found m/z: M⁺, 304.059.

The Tetramethyl Ether (15): Colorless needles of mp 167–169.5°C (EtOH) (lit.¹¹; mp 169–170°C). IR $\nu_{\text{max}}^{\text{KBr}}$ cm⁻¹: 3450, 3340 (OH), 1670 (CO). ¹H NMR (CDCl₃) δ : 7.08 (1H, dd, J 7 and 2 Hz, 6'-H), 7.05 (1H, d, J 2 Hz, 2'-H), 6.91 (1H, d, J 7 Hz, 5'-H), 6.77 (2H, s, 6- and 8-H's), 4.94 (1H, d, J 12 Hz, 2-H), 4.41 (1H, d, J 12 Hz, 3-H), 3.91, 3.89, 3.87, 3.78 (each 3H, s, OMe's), 3-OH signal was not observed. MS Calcd for C₁₉H₂₀O₇: M, 360.121. Found m/z: M⁺, 360.120.

(2R*,3S*,4R*)-5,7,3',4'-Tetramethoxyflavan-3,4-diol (16) and The 4S*-Isomer (17)

NaBH₄ (10 mg) was added to a solution of 15 (96.0 mg) in MeOH (5 ml), and the mixture was stirred at -30°C for 20 h. Work-up of the reaction mixture gave an oil, which was purified by prep. TLC (silica gel, acetone/C₆H₆=2/3, v/v) to yield 16 (13.8 mg, 14%), Rf 0.55 and 17 (64.8 mg, 67%), Rf 0.62.

The 2R*,3S*,4R*-Isomer (16): Colorless crystals of mp 166.5–168°C (EtOH). IR $\nu_{\text{max}}^{\text{KBr}}$ cm⁻¹: 3510, 3460 (OH). ¹H NMR (CDCl₃) δ : 7.08 (1H, dd, J 9 and 2 Hz, 6'-H), 7.02 (1H, d, J 2 Hz, 2'-H), 6.89 (1H, d, J 9 Hz, 5'-H), 6.16, 6.07 (each 1H, d, J 2.5 Hz, 6- and 8-H's), 4.99 (1H, dd, J 7 and 1 Hz, 4-H),⁶ 4.68 (1H, d, J 10 Hz, 2-H), 4.08 (1H, ddd, J 10, 7 and 2.5 Hz, 3-H),⁶ 3.89, 3.86 (3), 3.68 (12H, each s, OMe's), 2.33 (1H, d, J 2.5 Hz, 3-OH),⁷ 4-OH signal was not observed. MS Calcd for C₁₉H₂₂O₇: M, 362.137. Found m/z: M⁺, 362.135.

The 2R*,3S*,4S*-Isomer (17): Colorless crystals of mp 209–212°C (EtOH). IR $\nu_{\text{max}}^{\text{KBr}}$ cm⁻¹: 3590, 3420, 3270 (OH). ¹H NMR (CDCl₃) δ : 7.06 (1H, dd, J 9 and 2 Hz, 6'-H), 6.99 (1H, d, J 2 Hz, 2'-H), 6.89 (1H, d, J 9 Hz, 5'-H), 6.09 (2H, s, 6- and 8-H's), 4.99 (1H, d, J 3.5 Hz, 4-H), 4.88 (1H, d, J 10 Hz, 2-H), 3.82 (1H, m, 3-H), 3.88, 3.85, 3.84, 3.74 (each 3H, s, OMe's), 2.67 (2H, s, 3- and 4-OH's).⁷ MS Calcd for C₁₉H₂₂O₇: M, 362.137. Found m/z: M⁺, 362.138.

REFERENCES AND NOTES

1. Part XIII: Y. Harigaya, S. Takamatsu, H. Yamaguchi, and M. Onda, Chem. Pharm. Bull., 1982, 30, 1244.
2. J. Algar and J. P. Flynn, Proc. Roy. Irish Acad., 1934, 42B, 1; M. Murakami and T. Irie, Proc. Imp. Acad. Jpn., 1935, 11, 229.
3. R. Rognár, M. Rákosi, H. Fletcher, E. M. Philbin, and T. S. Wheeler, Tetrahedron, 1963, 19, 391.
4. A. Gaudemer, 'Stereochemistry', Vol. 1, ed. by H. B. Kagan, Georg Thieme Publishers, Stuttgart, 1977, p. 77.
5. W. Feuerstein, St. and V. Kostaneck, Chem. Ber., 1898, 31, 710.
6. On addition of deuterium oxide, these splittings changed to those corresponding to disappearance of the hydroxyl protons.
7. On addition of deuterium oxide, these signals disappeared.
8. E. A. Sherif, A. Islam, and M. Krishnamurti, Indian J. Chem., 1982, 21B, 478.
9. T. Oyamada and H. Baba, Bull. Chem. Soc. Jpn., 1966, 39, 507.
10. H. Aft, J. Org. Chem., 1961, 26, 1958.
11. H. L. Hergert, P. Coad, and A. V. Logan, J. Org. Chem., 1956, 21, 304.

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