

SYNTHESIS OF 3-ARYL-3,4-DIHYDROISOCARBOSTYRILS BY CONDENSATION OF LITHIATED N,N-DIETHYL-O-TOLUAMIDES WITH BENZALDIMINES¹

Robin D. Clark

Institute of Organic Chemistry, Syntex Research, Palo Alto, California 94304, U.S.A.

Abstract - The cyclocondensation of lithiated N,N-diethyl-O-toluamide (1) with benzaldimines (2) at -70°C gave directly 2-alkyl-3-aryl-3,4-dihydroisocarbostyrils (3) in 37-50% yield.

The synthesis of 3-aryl-3,4-dihydroisocarbostyrils (3) using conventional methodology would be expected to involve multistep, and often low-yielding, reaction sequences.^{2,3} Our interest in a direct route to this class of compounds prompted us to investigate the reaction of lithiated N,N-diethyl-O-toluamide (1)⁴ with benzaldimines (2). We now report that this reaction furnishes 2-alkyl-3-aryl-3,4-dihydroisocarbostyrils (3) in one step, albeit in modest yield. The results of a number of these cyclocondensations are given in the Table.

In a typical procedure, a THF solution of 1 equiv. of N,N-diethyl-O-toluamide was added to 1 equiv. of lithium diisopropylamide in THF at -70°C. The resulting dark purple solution of 1 was then treated with a THF solution of imine 2 (1.2 equiv.) and the dark mixture was then stirred at -70°C for 5-20 min.⁵ The reaction mixture was quenched with dilute hydrochloric acid and the products were isolated by ether extraction and purified by silica gel chromatography (3a-d) or crystallization (3e). Although the yields of 3 are modest (37-48%), this is overcome somewhat by the simplicity of the process and by the fact that the crude products are relatively pure since the major by-products are basic materials which remain in the aqueous acidic layer upon workup.

The condensation can also be performed by adding a preformed solution of LDA to a -70°C solution of 1 and 2 in THF. In the case of imine 2c, the yield of 3c was improved somewhat (to 50%) using this modification.⁶

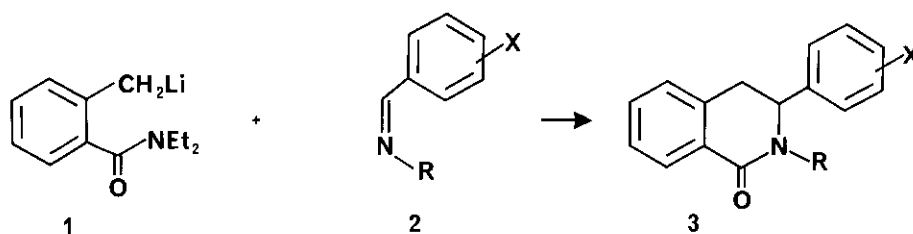


Table: 3,4-Dihydroisocarbostyryls from Condensation of 1 and 2

Imine	R	X	Product	Yield (%) ^{a, b}	M.P. (°C)
2a	cyclohexyl	4-OCH ₃	3a	48	oil
2b	cyclohexyl	2-CH ₃	3b	42	foam
2c	n-butyl	4-OCH ₃	3c	42	oil
2d	CH ₃	4-OCH ₃	3d	37	98-99
2e	6,7-dimethoxy-3,4-dihydro-isoquinoline		3e ^c	43	138-139 ^d

a) Compounds 3a-d were purified by silica gel chromatography (ethyl acetate/hexane). 3e was obtained by crystallization of the crude product from ethyl acetate/hexane.

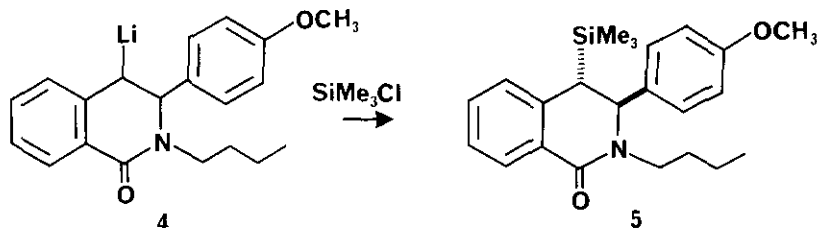
b) Satisfactory IR and NMR spectra and combustion analyses were obtained for 3a-e. Representative spectral data for 3d is given in footnote 18.

c) See structure in text.

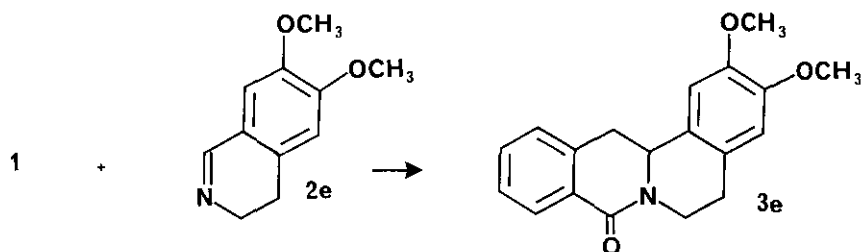
d) Literature, mp 142°C (Reference 8). The NMR spectral data matched those reported in References 9 and 10.

It is important to keep the reaction temperature below -60°C since above this temperature 1 begins to self-condense. In addition, since the lithium dialkylamide base is regenerated in the cyclocondensation process, the product 3 is deprotonated under the reaction conditions to a 4-lithiated dihydroiso-

carbostyril (e.g. 4). This species undergoes a number of side reactions (such as ring opening to a stilbene) if the low temperature is not maintained. In one example, this lithiated dihydroisocarbostyril was trapped with trimethylsilyl chloride to afford 5⁷ in 30% yield.



Reaction of the cyclic imine 6,7-dimethoxy-3,4-dihydroisoquinoline (2e) with 1 gave 2,3-dimethoxy-8-oxoberbine (3e).⁸⁻¹⁰ A related synthesis in which lithiated phthalide was condensed with 2e to give 2,3-dimethoxy-13-hydroxy-8-oxoberbine has been reported.¹¹ Another similar transformation is the condensation of imines with homophthalic anhydrides.^{12,13}



Lithiated *o*-toluic acid,¹⁴ methyl *o*-toluate, and *N*-methyl-*o*-toluamide,¹⁵ were also investigated in condensations with 2. However, none of these species led to production of dihydroisocarbostyrils. In the case of dilithiated *N*-methyl-*o*-toluamide, an adduct was formed which did not cyclize either upon workup or on heating at 150°C. This is in contrast to the reaction of this dilithio species with nitriles in which cyclization to 3-substituted isocarbostyrils occurs readily.¹⁶ 3-Substituted isocarbostyrils have also been prepared by reaction of dilithiated *N*-methyl-*o*-toluamide with *N,N*-dimethylcarboxamides.¹⁷

The results presented herein demonstrate that the cyclocondensation of 1 and 2 affords a direct route to 3-aryl-3,4-dihydroisocarbostyrils (3). In addition, lithiated intermediates such as 4 may serve as useful synthetic intermediates.

ACKNOWLEDGEMENT

We thank Dr. J. Muchowski for his encouragement and invaluable suggestions in regard to this work.

REFERENCES AND NOTES

1. Contribution No. 687 from the Institute of Organic Chemistry.
2. For example, cyclization of phenethyl isocyanates: H. Irie, A. Shiina, T. Fushimi, J. Katakawa, N. Fujii, and H. Yajima, Chem. Lett., 1980, 875.
3. A two step synthesis of substituted 3,4-dihydroisocarbostyrils involving addition of lithiated N-methyl-o-toluamide to carbonyl compounds followed by cyclodehydration was reported: C.L. Mao, I.T. Barnish, and C.R. Hauser, J. Heterocycl. Chem., 1969, 6, 83. However, a subsequent report demonstrated that most of these purported dihydroisocarbostyrils were in fact cyclic imino ethers: D.M. Bailey and C.G. DeGrazia, Tetrahedron Lett., 1970, 11, 633.
4. The condensation of lithiated 6-methoxy-N,N-dimethyl-o-toluamide with aldehydes has been reported: P. Beak and V. Snieckus, Acc. Chem. Res., 1982, 15, 306.
5. Imines 2c-e condense with 1 almost immediately. The cyclohexylimines 2a-b were somewhat slower and in these cases a 20 min reaction time was employed.
6. The use of substantially less than 1 equiv. of LDA was also investigated. However, this did not lead to useful results.
7. 5: oil; IR (film) 1645 cm^{-1} ; NMR (CDCl_3) δ 8.10 (m, 1H, H-8), 7.33-6.70 (m, 3H), 6.76 (AB, 4H), 4.76 (s, 1H, H-3), 3.90 (dt, 1H, NCH_a), 3.60 (s, 3H), 2.93 (dt, 1H, NCH_b), 2.45 (s, 1H, H-4), 1.80-1.00 (m, 4H), 0.92 (t, 3H), 0.05 (s, 9H). Anal. for $\text{C}_{23}\text{H}_{31}\text{NO}_2\text{Si}$: C,H,N.

8. D.W. Brown and S.F. Dyke, Tetrahedron, 1966, 22, 2429.
9. G.R. Lenz, J. Org. Chem., 1974, 39, 2846.
10. I. Ninomiya, T. Naito, and H. Takasugi, J. Chem. Soc. Perkin I, 1975, 1720.
11. R. Marsden and D.B. MacLean, Tetrahedron Lett., 1983, 24, 2063.
12. M.A. Haimova, N.M. Mollov, S.C. Ivanova, A.I. Dimitrova, and V.I. Ognyanov, Tetrahedron, 1977, 33, 331.
13. M. Cushman and L. Cheng, J. Org. Chem., 1978, 43, 286.
14. P.L. Creger, J. Am. Chem. Soc., 1970, 92, 1396.
15. R.L. Vaulx, W.H. Puterbaugh, and C.R. Hauser, J. Org. Chem., 1964, 29, 3514.
16. G.S. Poindexter, J. Org. Chem., 1982, 47, 3787.
17. R.S. Mali, B.K. Kulkarni, and K. Shankaran, Synthesis, 1982, 329.
18. 3d: IR (KBr) 1650 cm^{-1} ; NMR (CDCl_3) δ 8.10 (m, 1H, H-8), 7.33-6.50 (m, 6H), 4.67 (dd, $J = 2.5, 5 \text{ Hz}$, 1H, H-3), 3.78 (s, 3H), 3.67 (s, 3H), 3.60 (dd, $J = 5, 12 \text{ Hz}$, H-4a), 3.05 (s, 3H), 2.93 (dd, $J = 2.5, 12 \text{ Hz}$, H-4b).

Received, 7th January, 1985