

SILICON-MEDIATED ISOQUINOLINE ALKALOID SYNTHESIS: A NEW ROUTE
TO THE DIBENZOPYRROCOLINE ALKALOID (±)-CRYPTAUSTOLINE

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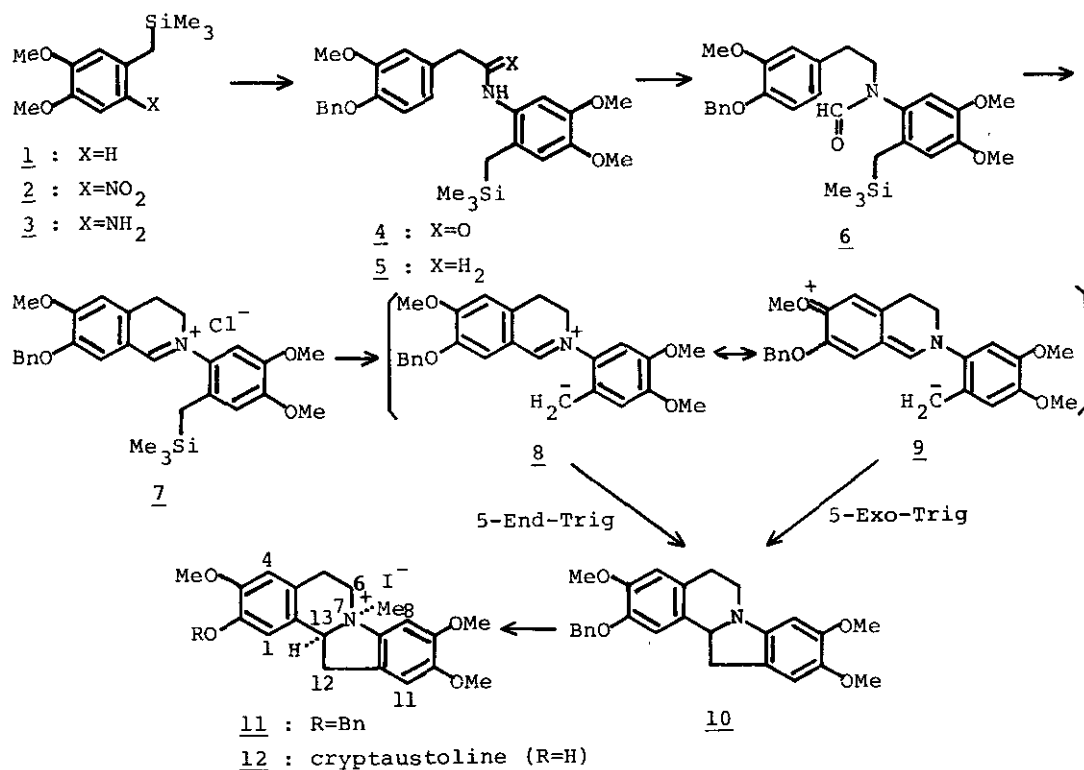
Abstract—A fundamentally new synthesis of the benzopyrrocoline alkaloid (±)-cryptaustoline iodide (12) has been devised by employing the silicon mediated ring closure through a formal 5-Endo-Trigonal process.

Recently, we developed a novel route to the protoberberine type isoquinoline alkaloids via intramolecular nucleophilic addition of a benzylsilane derivative to a carbon-nitrogen double bond through a favorable 6-Endo-Trigonal process.^{1,2} We report here an extension of this method for the construction of a dibenzopyrrocoline framework intending to develop a new route to the dibenzopyrrocoline alkaloids.³ Although some synthetic routes to the benzopyrrocoline alkaloids have been already developed by several workers,⁴ the methodologies employed were fundamentally based on the same principle involving the bond construction between 7-nitrogen and aromatic 7a-position at the key step. The present method involves the bond formation between 12 and 13 carbons at the key step employing the silicon-mediated method.

The key intermediate (7) was prepared from the known 4,5-dimethoxybenzylsilane¹ (1) via a 6 step-sequence of reactions. Nitration of 1 with cupric nitrate in acetic anhydride⁵ gave the 6-nitroderivative (2) in 43% yield. Hydrogenation (H₂, 10% Pd-C, EtOH) of 2 followed by condensation with 4-benzyloxy-3-methoxyphenyl-acetyl chloride yielded the anilide (4) (93% overall) via the amine (3). Reduction of 4 with a 3:1 mixture of lithium aluminum hydride and aluminum chloride gave the secondary amine (5) which was sequentially treated with acetic formic anhydride to give the formamide (6) (89% overall). Treatment of 6 with

phosphorus oxychloride in boiling benzene underwent cyclization to give the desired isoquinolinium base (7) in quantitative yield.

Treatment of 7 with two equivalent of tetra-*n*-butylammonium fluoride⁶ in boiling tetrahydrofuran allowed smooth ring closure to give rise to the known dibenzopyrrocoline⁷ (10) in 60% yield. Employing the established three step sequence of reactions⁴ 10 was converted into the dibenzopyrrocoline alkaloid, ($\pm$)-cryptaustoline iodide⁷ (12) in 44% overall yield. Two diastereomeric products may be formed at the quaternization stage with regard to the 7-nitrogen center, however, a single diastereomer with natural *cis* configuration⁸ was selectively obtained as has been confirmed.^{4c} Although the key cyclization seem to proceed via the betaine (8) formally through a disfavorable 5-Endo-Trigonal process, the observed facile ring closure may be the result of contribution from the quinonoid structure (9) which permits the favored 5-Exo-Trigonal mode of cyclization.⁹



REFERENCES AND NOTES

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2. More recently a similar silicon mediated reaction was also reported, see: G. Dai-Ho, A. J. Y. Lan, and P. S. Mariano, Tetrahedron Lett., **26**, 5867 (1985).
3. Pertinent reviews: (a) M. Shamma, "The Isoquinoline Alkaloids", Academic Press, New York, 1972, pp. 169-176; (b) M. Shamma and J. L. Monniot, "Isoquinoline Alkaloids Research 1972-1977", Plenum Press, New York, 1978, pp. 113-116.
4. (a) T. Kametani and K. Ogasawara, J. Chem. Soc. C, 2208 (1967); (b) F. Benington and R. D. Morin, J. Org. Chem., **32**, 1050 (1967); (c) T. Kametani, K. Fukumoto, and T. Nakano, J. Heterocyclic Chem., **9**, 1363 (1972); (d) I. W. Elliott, Jr., J. Org. Chem., **47**, 5398 (1982).
5. F. A. Carey and R. M. Giuliano, J. Org. Chem., **46**, 1366 (1981).
6. Interestingly, the cyclization did not take place with cesium fluoride under the same conditions which brought the 6-Endo-Trigonal cyclization.¹
7. Identical in all respects with an authentic material obtained by the benzyne route.^{4a} All new isolable compounds described showed satisfactory spectral (IR, ¹H-NMR, Mass) and analytical (combustion and high resolution Ms) data.
8. The configuration of the alkaloids has not yet been determined to date, see: S. Takano, S. Satoh, Y. Oshima, and K. Ogasawara, accompanied paper.
9. (a) J. E. Baldwin, J. Chem. Soc., Chem. Commun., 734 (1976); (b) J. E. Baldwin, J. Cutting, W. Dupont, L. I. Kruse, L. Silberman, and R. C. Thomas, ibid., 736 (1976).

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