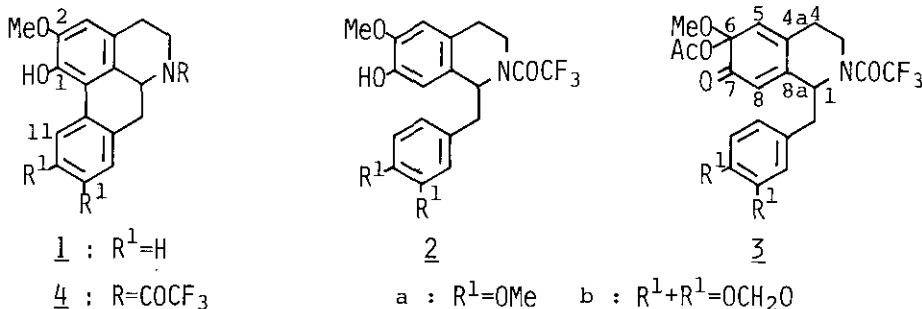


A NOVEL SYNTHESIS OF (\pm)-NORAPORPHINE ALKALOIDS, (\pm)-WILSONIRINE
AND (\pm)-NORDOMESTICINE[#]

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Abstract— (\pm)-Wilsonirine (**1a**) and (\pm)-nordomesticine (**1b**) were synthesized in moderate yields on acid treatment, followed by alkaline hydrolysis of *o*-quinol acetates (*o*-QAs) (**3**) readily obtained by lead tetraacetate oxidation of (\pm)-*N*-trifluoroacetyltetrahydroisoquinolin-7-ols (**2**).

Noraporphine alkaloids have been expected to be useful key compounds for synthesis of biologically active compounds¹. However, only a few papers² have been published on their synthesis so far. In continuation of our studies³ on application of lead tetraacetate [Pb(OAc)₄] oxidation to synthesis of isoquinoline alkaloids, we found that (\pm)-*N*-trifluoroacetyltetrahydroisoquinolin-7-ols (**2**) were readily oxidized with the oxidant in CH₂Cl₂ to give rise to the corresponding (\pm)-*N*-trifluoroacetyl-6-acetyl-6-methoxy-7-oxo- $\Delta^{4a,5,8,8a}$ -hexahydroisoquinolines [*o*-quinol acetates (*o*-QAs)] (**3**), which could be converted into (\pm)-noraporphines. We now wish to report a novel synthesis of (\pm)-wilsonirine (**1a**)⁴ and (\pm)-nordomesticine (**1b**)⁵.



[#] Dedicated to Professor G. Stork on the occasion of his sixty-fifth birthday.

A typical example is as follows. $\text{Pb}(\text{OAc})_4$ (127 mg) was added in one portion to an ice-cold, stirred solution of $\underline{2a}^{2b}$ (mp 149-151°C) (100 mg) in CH_2Cl_2 (1 ml) and stirring was continued at 0-5°C for 0.5 h. A careful work-up of the reaction mixture gave quantitatively a 1:1.2 diastereoisomeric mixture of $\underline{3a}$ (oil) [IR⁶ ν : 1735, 1680 cm^{-1} ; ¹H-nmr⁶ δ : 2.10, 2.12 (1:1.2) (3H, each s, 6-OCOMe), 3.42, 3.43 (1.2:1) (3H, each s, 6-OMe)]. CF_3COOH (5 ml) was added dropwise to an ice-cold, stirred solution of crude $\underline{3a}$ in CH_2Cl_2 (5 ml) and the mixture was stirred at room temperature for 1 h. Usual work-up of the reaction mixture gave quantitatively a solid, which was purified on preparative thin-layer chromatography (SiO_2 ; developing solvent; CHCl_3) to give $\underline{4a}^6$ (61 mg, 61%), mp 190-191°C (CH_2Cl_2 -n-hexane) (lit.^{2b}, 196.5-197°C). ¹H-Nmr spectra datum of $\underline{4a}$ was identical with that noted in literature^{2b}.

From the above results, it was proved that $\text{Pb}(\text{OAc})_4$ oxidation of $\underline{2a}$ gave o-QA ($\underline{3a}$), acid treatment of which afforded (\pm)-N-trifluoroacetylwilsonirine ($\underline{4a}$) in a moderate yield.

Similarly, $\underline{2b}$ (mp 137-138°C) gave o-QA ($\underline{3b}$)⁶, which was converted into $\underline{4b}^6$ in 21% overall yield.

Alkaline hydrolysis (aq. K_2CO_3 -MeOH, reflux, 2 h) of $\underline{4a}$ and $\underline{4b}$ afforded $\underline{1a}^6$ (93%), mp 210-213°C (CH_2Cl_2) (lit.⁴, 211-213°C) and $\underline{1b}^6$ (98%), mp 205-207°C (CH_2Cl_2 -n-hexane), respectively.

Thus, a novel synthesis of (\pm)-noraporphine alkaloids was accomplished via o-QAs ($\underline{3}$) of (\pm)-N-trifluoroacetyltetrahydroisoquinolin-7-ols ($\underline{2}$). Further development of the present method is in progress.

ACKNOWLEDGEMENTS

The authors are indebted to Dr. T. Moroe of Takasago Perfumery Co., Ltd. for his kind supply of vanillin and piperonal. Thanks are also due to Professor M. Ishikawa and Dr. A. Sugimoto of Tokyo Medical and Dental University for high resolution ms spectral measurements, to Sankyo Co., Ltd. for elemental analysis, and to Misses N. Sawabe and N. Yamatani of this Faculty for ¹H-nmr and ms spectral measurements.

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6. IR spectra were taken with a Hitachi 260-10 spectrophotometer in CHCl_3 solution. ^1H -Nmr spectra were measured with a JEOL-JNM-FX-100(100MHz) instrument in CDCl_3 solution using TMS as internal standard. 3a; δ : 2.10, 2.12 (1:1.2) (3H, each s, 6-OCOMe), 3.42, 3.43 (1.2:1) (3H, each s, 6-OMe), 3.82, 3.83, 3.84 (9H, each s, 3xOMe), 5.15 (1H, dd, $J=4.3, 8.5$ Hz, 1-H), 5.14, 5.99 (2H, each br s, 2xolefinic H), 6.60-6.70 (3H, m, 3xAr-H). 4a; δ : 3.94 (3H, s, OMe), 3.96 (3H, s, 2xOMe), 5.08 (1H, dd, $J=5.4, 12.9$ Hz, 6a-H), 6.60, 6.80 (2H, each s, 2xAr-H), 8.15 (1H, s, 11-H). 1a; δ : 3.95 (9H, s, 3xOMe), 6.58, 6.77 (2H, each s, 2xAr-H), 8.10 (1H, s, 11-H). 3b (oil); ν : 1730, 1680 cm^{-1} ; δ : 2.10, 2.12 (1:1.2) (3H, s, 6-OCOMe), 3.42 (3H, s, 6-OMe), 4.96-5.24 (1H, m, 1-H), 5.46, 5.57 (1:1.2) (1H, each s, olefinic H), 5.90 (2H, s, OCH_2O), 5.96-6.08 (1H, m, olefinic H), 6.44-6.78 (3H, m, 3xAr-H). 4b; δ : 3.97 (3H, s, 2-OMe), 5.05 (1H, dd, $J=5.7, 12.8$ Hz, 6a-H), 6.00 (2H, s, OCH_2O), 6.60, 6.78 (2H, each s, 2xAr-H), 8.04 (1H, s, 11-H). 1b; δ : 3.95 (3H, s, 2-OMe), 5.98 (2H, s, OCH_2O), 6.57, 6.76 (2H, each s, 2xAr-H), 8.00 (1H, s, 11-H).

Received, 29th May, 1986