

ASYMMETRIC SYNTHESIS OF 4-ACETOXY-3-HYDROXYETHYL-
AZETIDIN-2-ONE, A KEY INTERMEDIATE FOR THE PREPARATION OF PENEM AND
CARBAPENEM ANTIBIOTICS

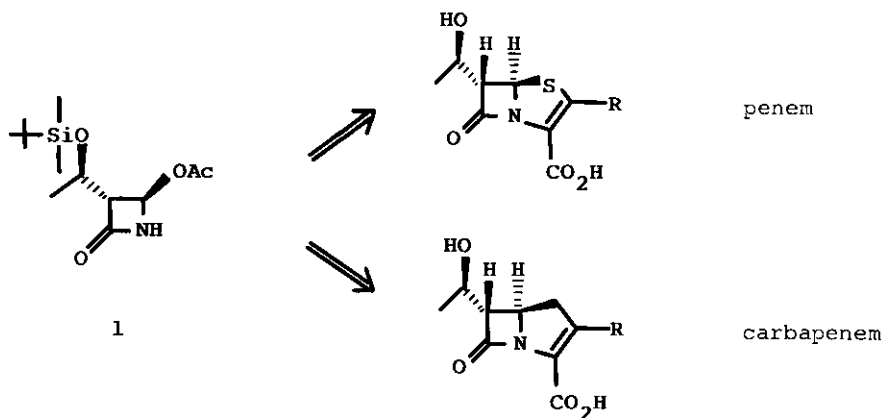
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Dedicated to Professor Gilbert Stork on the occasion of his
65th birthday.

Abstract — Asymmetric synthesis of 4-acetoxy-3-hydroxyethyl-
azetidin-2-one with desired stereochemistry has been achieved
by employing [3+2]dipolar cycloaddition of a chiral nitron
with benzyl crotonate as a key reaction.

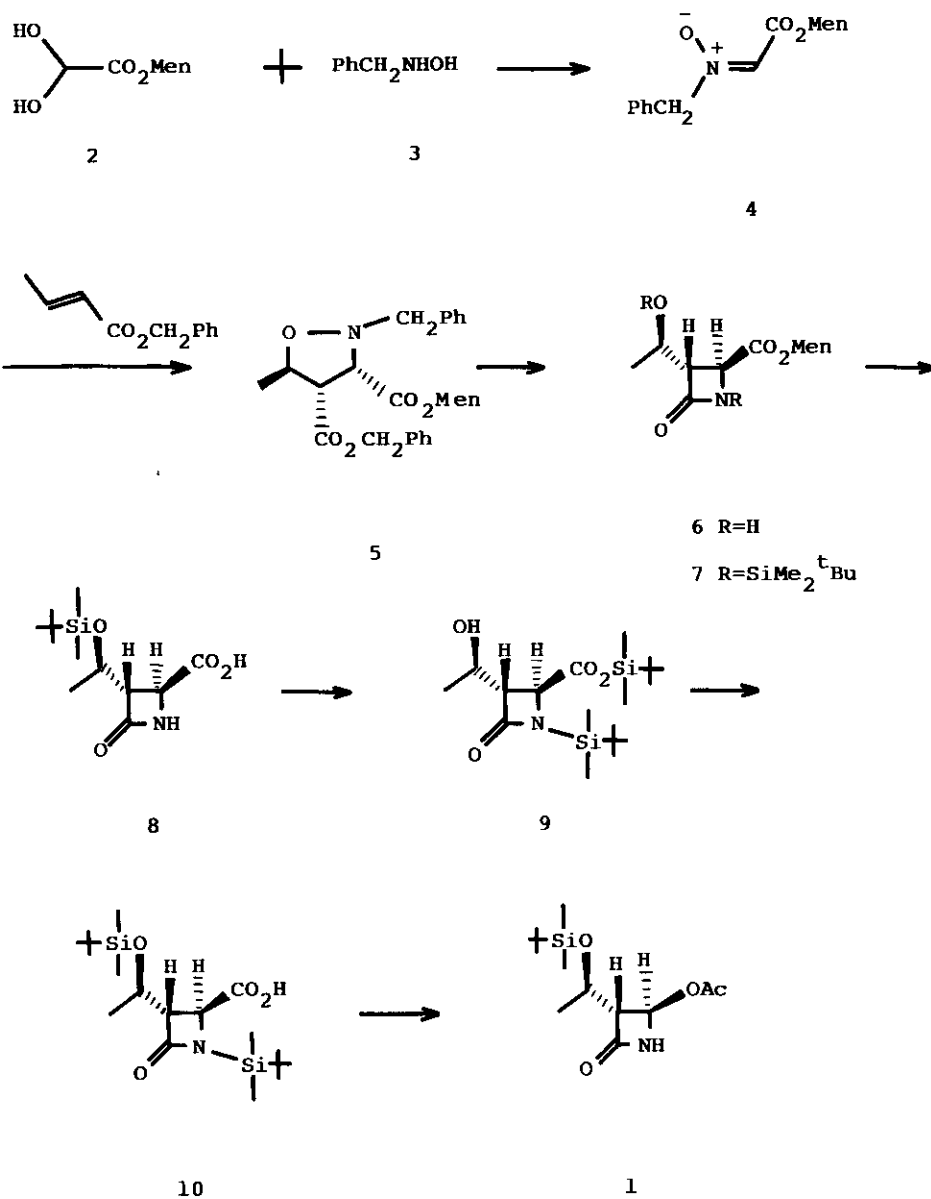
4-Acetoxy-3-hydroxyethylazetidin-2-one derivative (1) was recognized to be a versatile starting material in the synthesis of non-classical β -lactam antibiotics because of its successful conversion into penem¹ and carbapenem² derivatives, including thienamycin. Therefore a number of enantiospecific synthesis of 1 have appeared by elaboration of a chiral natural source such as 6-aminopenicillanic acid,³ (+)-aspartic acid,⁴ and β -hydroxybutyrate⁵ and by an asymmetric [2+2]cyclo-



Scheme 1

addition reaction.⁶

We have recently demonstrated⁷ the enantioselective synthesis of a carbapenem antibiotic, thienamycin, in which [3+2]dipolar cycloaddition of a chiral nitronone has played a key role to control the stereochemistry with the desired chirality. We here report a novel chiral synthesis of 4-acetoxy-3-hydroxyethylazetidin-2-one derivative (1) by further application of the above synthetic strategy employing an asymmetric [3+2]dipolar cycloaddition reaction as a key step. Condensation of L-(-)-menthyl glyoxalate hydrate (2)⁸ with benzylhydroxylamine (3) in refluxing benzene afforded the nitronone (4), whose treatment with benzyl crotonate brought about [3+2]cycloaddition reaction to furnish the desired isoxazolidine (5)⁹ as a major product in 30 % yield, together with the other stereoisomers. Catalytic reduction of 5 over platinum oxide in methanol under a current of hydrogen gave the amino acid, which without purification was treated with N,N-dicyclohexylcarbodiimide in acetonitrile to afford the azetidinone (6)¹⁰ in 39 % yield. The stereochemistry of 6 was determined based on its nmr spectrum to have 3,4-trans relationship, though the absolute configuration could not be determined at this stage. After silylation of the hydroxy and amide groups of 6 with tert-butyldimethylchlorosilane, the silyl derivative (7) was subjected to hydrolysis with 1N-NaOH to provide the acid (8)¹¹ in 64 % yield from 6. Treatment of the resulting acid (8) with 2 eq. molar amount of tert-butyldimethylchlorosilane in N,N-dimethylformamide in the presence of triethylamine yielded the silyl ester (9) in 94 % yield. Obviously the migration of the silyl group occurred during the silylation reaction. The silyl ester (9) was interestingly converted into the acid (10) by treatment with acetic acid in tetrahydrofuran in 82 % yield. Finally the acid (10) was transformed to the desired 4-acetoxy derivative (1), mp 105 °C (lit.,^{3a} 104 - 106 °C), $[\alpha]_D^{25} +49.31^\circ$ (c=0.073, CHCl₃) [lit.,^{3a} +48.8° (CHCl₃)], by oxidative acetoxylation with lead tetraacetate in 70 % yield. Since the physicochemical properties of 1 were identical with those reported, its stereochemistry was determined to have 3R and 4R configuration. Thus, the chiral synthesis of 1 was achieved by employing [3+2]dipolar cycloaddition of a chiral nitronone with benzyl crotonate as a key step. Furthermore, since the acid (8) was already converted^{3b} to thienamycin, this synthesis constitutes a formal synthesis of (+)-thienamycin.



Scheme 2

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9. mp 95 °C; $[\alpha]_D^{25} -76.42^\circ$ (c=1.086, CHCl₃); IR (CHCl₃) 1740 cm⁻¹; NMR (CDCl₃) δ 1.35 (3H, d, J = 6.1 Hz, 5-Me), 3.16 (1H, dd, J = 8.3 and 8.5 Hz, 4-H), 3.75 (1H, d, J = 8.5 Hz, 3-H), 4.41 (1H, dd, J = 6.1 and 8.3 Hz, 5-H), 5.12 (2H, s, CH₂Ar); MS m/z 493 (M⁺).
10. An oil; $[\alpha]_D^{25} -55.48^\circ$ (c=1.77, CHCl₃); IR (CHCl₃) 3400, 1765, 1730 cm⁻¹; NMR (CDCl₃) δ 1.31 (3H, d, J = 6.3 Hz, 1'-Me), 2.75 - 3.27 (1H, ddd, J = 1, 2.8 and 5.6 Hz, 3-H), 4.29 (1H, d, J = 2.8 Hz, 4-H); MS m/z 298 (M⁺ + 1).
11. mp 134 °C [lit., ^{3b} 134 °C]; $[\alpha]_D^{25} -21.0^\circ$ (c=0.2, CHCl₃); The melting point of this acid was identical with the literature value, ^{3b} however, its optical rotation was not reported.

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