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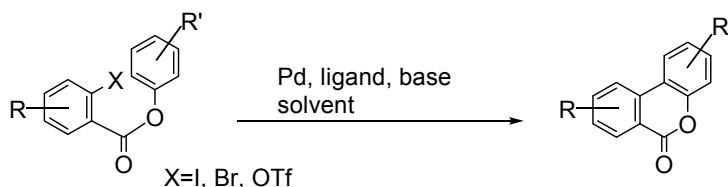
## HIGHLY REGIOSELECTIVE INTRAMOLECULAR BIARYL COUPLING REACTION OF A PHENYL BENZOATE DERIVATIVE FOR THE SYNTHESIS OF GRAPHISLACTONE G

Hitoshi Abe,<sup>a\*</sup> Takuya Matsukihira,<sup>a</sup> Tomoko Fukumoto,<sup>b</sup>  
 Yoshikazu Horino,<sup>a</sup> Yasuo Takeuchi,<sup>b\*</sup> and Takashi Harayama<sup>b</sup>

a) Graduate School of Science and Engineering, University of Toyama, Gofuku,  
 Toyama 930-8555, Japan b) Graduate School of Medicine, Dentistry, and  
 Pharmaceutical Sciences, Okayama University, Okayama 700-8530, Japan  
 abeh@eng.u-toyama.ac.jp; take@pharm.okayama-u.ac.jp

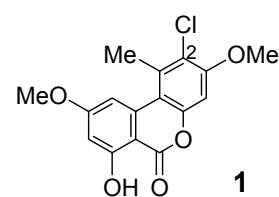
**Abstract** – The 6*H*-dibenzo[*b,d*]pyran-6-one type natural product graphislactone G was synthesized using a palladium-mediated aryl-aryl coupling reaction of the phenyl benzoate derivative. The regioselectivity in the coupling step was also investigated.

A palladium-catalyzed intramolecular aryl-aryl coupling reaction is a useful technique for preparing heterocyclic compounds.<sup>1</sup> For example, phenyl benzoates are good precursors for constructing a 6*H*-dibenzo[*b,d*]pyran-6-one ring system (Scheme 1).<sup>2</sup> Using this strategy, we synthesized several natural products possessing a 6*H*-dibenzo[*b,d*]pyran-6-one skeleton,<sup>3</sup> such as graphislactones A-D<sup>4</sup> and H,<sup>5</sup> and alternariol.<sup>6,7</sup>



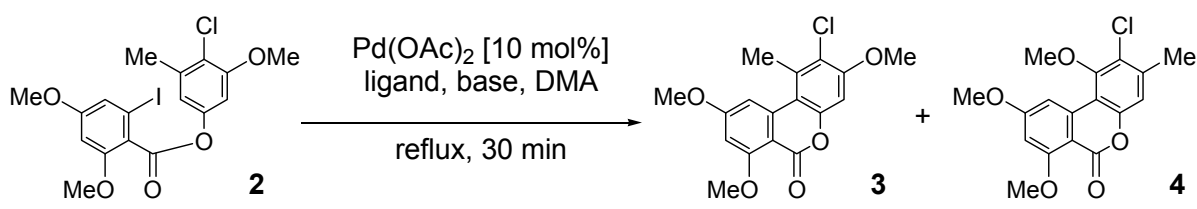
**Scheme 1.** Intramolecular biaryl coupling reaction of phenyl benzoate

Recently, the total synthesis of graphislactone G (**1**), which is a phenolic constituent isolated from *Cepharosporium acremonium* IFB-E007 in 2005,<sup>8</sup> was reported by Cudaj and Podlech.<sup>9</sup> Structurally, compound **1** possesses a highly functionalized 6*H*-dibenzo[*b,d*]pyran-6-one ring system and has a chlorine atom at the 2-position. Biologically, **1** is cytotoxic to the SW1116 cell line.<sup>8</sup> In



the Podlech's synthesis, they employed the Suzuki-Miyaura coupling reaction as the key step.<sup>9</sup> In this study, we planned the synthesis of **1**, using the intramolecular aryl-aryl coupling reaction of a phenyl benzoate derivative. Initially, we prepared the ester (**2**) by simple condensation between the corresponding phenol and benzoic acid derivative as a precursor for the intramolecular biaryl coupling reaction. Several reaction conditions were examined for the palladium-mediated biaryl coupling of **2** in order to construct the 6*H*-dibenzo[*b,d*]pyran-6-one skeleton (Table 1).

**Table 1.** Reactivity and regioselectivity in the palladium-mediated biaryl coupling



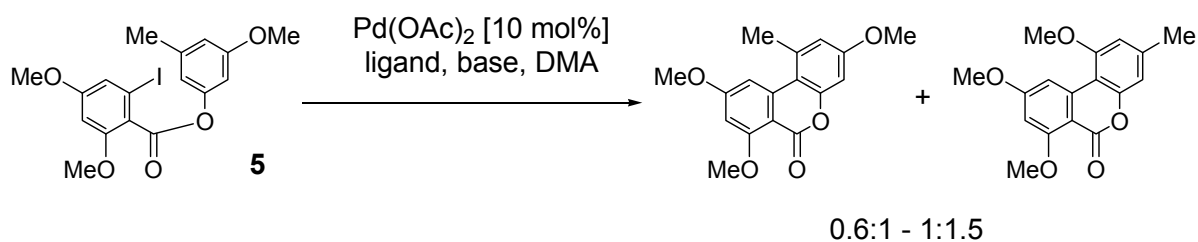
Run	Ligand [mol%]	Base	Yield (%) <b>3 + 4</b>	Ratio <sup>a)</sup> <b>3 : 4</b>
1	P( <i>o</i> -tolyl) <sub>3</sub> [20]	K <sub>2</sub> CO <sub>3</sub>	63	>20 : 1
2	PPh <sub>3</sub> [20]	K <sub>2</sub> CO <sub>3</sub>	68	11 : 1
3	<sup>n</sup> Bu <sub>3</sub> P [20]	K <sub>2</sub> CO <sub>3</sub>	57	7.7 : 1
4	DPPP [10], <sup>n</sup> Bu <sub>3</sub> P [10]	Ag <sub>2</sub> CO <sub>3</sub>	80	1.4 : 1

a) Determined by NMR analysis

In each case, the coupling reaction proceeded smoothly, producing the lactone compounds **3** and **4** in moderate to good yield. The product ratio of **3** and **4** changed dramatically with the reaction conditions. When P(*o*-tolyl)<sub>3</sub> was used as the ligand, extremely high regioselectivity was observed (run 1), generating the product **3**, which was reacted at the position *ortho* to the methyl group. By contrast, the DPPP-<sup>n</sup>Bu<sub>3</sub>P combination system<sup>10</sup> was not effective for regioselectivity (run 4).

These results are very different from our previous report on the synthesis of alternariol, in which the similar coupling reaction of **5** proceeded with the regioselectivity of 0.6:1 to 1.5:1 ratio (Scheme 2).<sup>6</sup> This difference of selectivity indicates that the chlorine atom on the phenoxy group exhibits a powerful directing property in the coupling reaction. The mechanistic aspect of this selectivity is being studied intensively in our laboratory.

Using the major cyclized product **3**, selective demethylation was achieved with Node's protocol (AlCl<sub>3</sub>-NaI/MeCN-CH<sub>2</sub>Cl<sub>2</sub>, 0 °C to rt, 84%),<sup>11</sup> to complete the synthesis of graphis lactone **G** (**1**). The <sup>1</sup>H- and <sup>13</sup>C-NMR data of the product were identical to those reported by Podlech *et al.*<sup>9</sup>



**Scheme 2.** Intramolecular biaryl coupling reaction of **5**

## ACKNOWLEDGEMENTS

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