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ANTHRAQUINONES FROM *CASSIA FISTULA* AND THEIR ANTI-TOBACCO MOSAIC VIRUS ACTIVITY

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Abstract – Two new anthraquinones, 9,11-dihydroxy-2-(hydroxymethyl)-5-methyl-4*H*-naphtho[2,3-*h*]chromene-4,7,12-trione (**1**) and 9,11-dihydroxy-2,5-dimethyl-4*H*-naphtho[2,3-*h*]chromene-4,7,12-trione (**2**), together with five known anthraquinones (**3-7**) were isolated from the stems of *Cassia fistula*. Their structures were elucidated by spectroscopic methods, including extensive 1D- and 2D NMR techniques. Compounds **1-7** were tested for their anti-tobacco mosaic virus activity. The results showed that compound **7** exhibited high anti-TMV activity with inhibition rate of 35.2%. The inhibition rate is higher than that of positive control. The other compounds also showed potential anti-TMV activities with inhibition rates in the range of 15.2–24.8%, respectively.

Cassia fistula L., (Leguminosae) is an ornamental tree with beautiful yellow flowers.¹ In China, it has been used as traditional Chinese medicine by people of Dai nationality, who lived in Xishuangbanna, Yunnan province for treatment of diarrhea, gastritis, ringworm, and fungal skin infections.^{2,3} Previous phytochemical studies of *C. fistula* have shown the presence of anthraquinones,^{4,5} steroids,⁶ chromones,^{7,8} and flavonoids.⁹ Flavonoids possessing anti-tobacco mosaic virus (anti-TMV) have been isolated from *C. fistula* grown in De Hong Prefecture by our group.⁹ Motivated by a search for new bioactive metabolites from local plants, our group investigated the chemical constituents of the stems of *C. fistula* growing in Xishuangbanna Prefecture, which led to the isolation and characterization of two new (**1** and **2**) and five

known (3-7) anthraquinones derivatives. This paper deals with the isolation, structural characterization, and the anti-tobacco mosaic virus (anti-TMV) activities of these compounds.

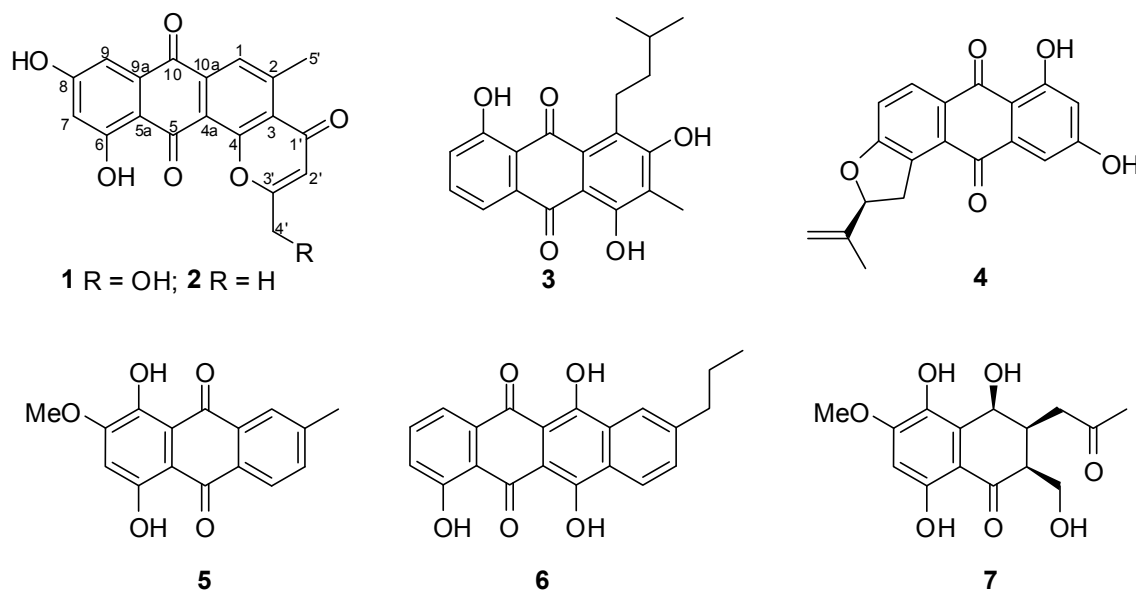


Figure 1. The structures of anthraquinones from *C. fistula*

The air-dried and powdered stems of *C. fistula* (4.8 kg) was extracted with 70% aqueous acetone (4 × 5.0 L) at room temperature and filtered to yield a filtrate, which was successively evaporated under reduced pressure to obtain a crude extract (258 g). This crude extract was subjected repeatedly to column chromatography on Silica gel, Sephadex LH-20, RP-18 and preparative HPLC to afford compounds 1-7. Their structures were shown in **Figure 1**. The ^1H - and ^{13}C NMR data of the compounds 1 and 2 were listed in Table 1. By compared with the literature, the known compounds were identified as lupinacidin A (3),¹⁰ (2*S*)-7,9-dihydroxy-2-(prop-1-en-2-yl)-1,2-dihydroanthra[2,1-*b*]furan-6,11-dione (4),¹¹ austrocortirubin (5),¹² 4,6,11-trihydroxy-9-propyltetracene-5,12-dione (6),¹³ fusarnaphthoquinone B (7).¹⁴

Compound 1 was isolated as a yellow powder. High-resolution ESIMS analysis gave a quasi-molecular ion at m/z 351.0501 $[\text{M}-\text{H}]^-$, consistent with a molecular formula of $\text{C}_{19}\text{H}_{12}\text{O}_7$, which indicated 14 degrees of unsaturation. The UV spectrum of 1 exhibited absorption

bands at 364, 276, 254, and 210 nm, highly suggesting the existence of aromatic chromophore.¹⁵ Strong absorption bands accounting for hydroxy (3392 cm^{-1}), carbonyl (1693 and 1650 cm^{-1}), and aromatic

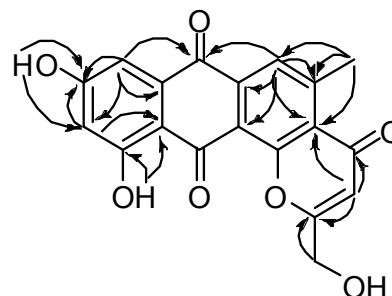


Figure 2. key HMBC (\curvearrowright) correlations of 1.

groups (1605, 1564, and 1483 cm^{-1}) could also be observed in its IR spectrum. The ^1H NMR spectrum of **1** (Table 1) showed the presence of two phenolic hydroxy proton (δ_{H} 11.83 and 12.20), four singlet aromatic protons (δ_{H} 7.62, 7.07, 6.90, and 6.52), and two aliphatic protons contributed by one methyl singlet (δ_{H} 2.05), one *O*-methylene singlet (δ_{H} 4.43). In the ^{13}C NMR spectrum of **1** (Table 1), 14 sp^2 carbon signals, including three oxygenated quaternary sp^2 carbon signals (δ_{C} 155.0, 161.3, and 164.0), and two carbonyl carbon signals (δ_{C} 184.4 and 182.0) were observed, which highly suggested the presence of anthraquinone core.¹⁵ The additional carbons account for the remaining substituents, a hydroxymethyl chromone ring (δ_{C} 182.9 s, 108.2 d, 168.1 s, 62.0 t),¹⁶ and a methyl carbon (δ_{C} 18.8) on the anthraquinone ring. The substituents and their location on the anthraquinone ring were established by analysis of the HMBC spectra of **1** (Figure 2). The HMBC correlations from a methyl singlet (δ_{H} 2.05) to C-1 (δ_{C} 124.0), C-2 (δ_{C} 144.1), and C-3 (δ_{C} 133.0) established the location of a methyl at C-2. HMBC correlations between the hydroxy proton (δ_{H} 12.20) and C-6 (δ_{C} 161.3), C-7 (δ_{C} 109.1), and C-5a (δ_{C} 113.7), as well as those between the other hydroxy proton (δ_{H} 11.83) and C-7 (δ_{C} 109.1), C-8 (δ_{C} 164.0), and C-9 (δ_{C} 111.0), led to the assignment of the phenolic hydroxy groups at C-6 and C-8. Additionally, H-2' (δ_{H} 6.52) showed correlation with the carbon signal of C-3 (δ_{C} 133.0) clearly indicated that the hydroxymethyl chromone ring should be located between C-3 and C-4. On the basis of the above evidence, the structure of **1** was established as shown.

Compounds **2** was also obtained as yellow powder, a molecular formula of $\text{C}_{19}\text{H}_{12}\text{O}_6$ was deduced from the HRESIMS data m/z 335.0562 $[\text{M}-\text{H}]^-$ (calcd m/z 335.0556). The ^1H and ^{13}C NMR spectra of **2** were very similar to those of **1**. The obvious chemical shift differences resulted from the disappearance of a hydroxymethyl signal δ_{C} (62.0 t) and δ_{H} (4.43 s, 2H), and appearance of a methyl group signal δ_{C} (20.0 q) and δ_{H} (2.25 s, 3H) in **2**. This indicated that the hydroxymethyl group in **1** was substituted by a methyl group in **2**. The HMBC correlations of the methyl proton signal (δ_{H} 2.25) with C-2' (δ_{C} 108.8) and C-3' (δ_{C} 165.4) also indicated the position of methyl group at C-3'. Thus, the structure of **2** was established as 9,11-dihydroxy-

Table 1. ^1H and ^{13}C NMR data of compounds **1** and **2** (δ in ppm, in CDCl_3)

No.	Compound 1		Compound 2	
	δ_{C} (m)	δ_{H} (m, J =Hz)	δ_{C} (m)	δ_{H} (m, J =Hz)
1	124.0 d	7.62, s	124.5 d	7.61, s
2	144.1 s		143.8 s	
3	133.0 s		133.4 s	
4	155.0 s		155.5 s	
4a	117.1 s		117.3 s	
5	184.4 s		183.8 s	
5a	113.7 s		113.6 s	
6	161.3 s		161.8 s	
7	109.1 d	6.90, s	109.1 d	6.80, s
8	164.0 s		163.9 s	
9	111.0 d	7.07 s	110.9 d	7.00 s
9a	122.2 s		122.6 s	
10	182.0 s		182.1 s	
10a	125.4 s		125.2 s	
1'	182.9 s		183.0 s	
2'	108.2 d	6.52, s	108.8 d	6.33, s
3'	168.1 s		165.4 s	
4'	62.0 t	4.43, s	20.0 q	2.25, s
5'	18.8 q	2.05, s	18.3 q	2.01, s
Ar-OH-6		12.20, s		12.22, s
Ar-OH-8		11.83, s		11.78, s

2,5-dimethyl-4*H*-naphtho[2,3-*h*]chromene-4,7,12-trione.

Since certain of the anthraquinones exhibit potential antiviral activities,^{17,18} compounds **1-7** were tested for their anti-TMV activities. The inhibitory activities of compounds **1-7** against TMV replication were tested using the half-leaf method.¹⁹ Ningnanmycin, a commercial product for plant disease in China, was used as a positive control. The antiviral inhibition rates of compounds **1-7** at the concentration of 20 μM were listed Table 2. The results showed that compound **7** exhibited high anti-TMV activity with inhibition rate of 35.2%. The inhibition rate is higher than that of positive control. The other compounds also showed potential anti-TMV activities with inhibition rates in the range of 15.2–24.8%, respectively.

EXPERIMENTAL

General. UV spectra were obtained using a Shimadzu UV-2401A spectrophotometer. IR spectra were obtained in KBr disc on a Bio-Rad Wininfrared spectrophotometer. ESI-MS were measured on a VG Auto Spec-3000 MS spectrometer. ¹H-, ¹³C, and 2D NMR spectra were recorded on Bruker

DRX-500 instrument with TMS as internal standard. Column chromatography was performed on silica gel (200-300 mesh), or on silica gel H (10~40 μm , Qingdao Marine Chemical Inc., China). Second separation was performed by an Agilent 1100 HPLC equipped with ZORBAX-C₁₈ (21.2 mm \times 250 mm, 7.0 μm) column and DAD detector.

Plant material. The stems of *Cassia fistula* L., (Leguminosae) were collected in Xishuangbanna Prefecture, Yunnan Province, People's Republic of China, in September 2011. The identification of the plant material was verified by Prof. Yuan. N (Xishuangbanna Botanical Garden). A voucher specimen (YNNI-2010-9-28) has been deposited in our laboratory.

Extraction and Isolation. The air-dried and powdered leaves and stems of *C. fistula* (4.8 kg) were extracted four times with 70% acetone (4 \times 5 L) at room temperature and filtered. The crude extract (258 g) was applied to silica gel (200–300 mesh) column chromatography, eluting with a MeOH-CHCl₃ gradient system (9:1, 8:2, 7:3, 6:4, 5:5), to give five fractions A–E. The further separation of fraction A (9:1, 22.6 g) by silica gel column chromatography, eluted with acetone-CHCl₃ (9:1, 8:2, 7:3, 6:4, 1:1), yielded the mixtures A1–A5. The subfraction A1 (9:1, 5.6 g) was subjected to preparative HPLC (65% MeOH, flow rate 12 mL/min) to give **2** (8.57 mg), **3** (12.4 mg), and **4** (22.6 mg). The further separation of subfraction A2 (8:2, 4.8 g) by silica gel column chromatography, and preparative HPLC (60% MeOH,

Table 2. TMV infection inhibition activities of compounds **1-7**

Compounds	Inhibition rates (%)	Compounds	Inhibition rates (%)
1	24.8 \pm 3.6	5	17.5 \pm 3.1
2	22.4 \pm 2.8	6	21.3 \pm 2.4
3	15.2 \pm 2.4	7	35.2 \pm 3.5
4	18.7 \pm 2.5	ningnamycin	31.2 \pm 3.4

All results are expressed as mean \pm SD; n = 3 for all groups.

flow rate 12 mL/min) to give **1** (8.8 mg), **5** (16.4 mg), and **6** (15.8 mg). The further separation of subfraction A3 (7:3, 3.4 g) by silica gel column chromatography, and preparative HPLC (50% MeOH, flow rate 12 mL/min) to give **7** (14.9 mg).

Anti-TMV Assays. The Anti TMV activities were tested using the half-leaf method, and ningnanmycin,¹⁹ a commercial product for plant disease in China, was used as a positive control.

9,11-Dihydroxy-2-(hydroxymethyl)-5-methyl-4H-naphtho[2,3-h]chromene-4,7,12-trione (1):

C₁₉H₁₂O₇, obtained as yellow powder; UV (MeOH), λ_{\max} (log ϵ) 367 (3.68), 278 (4.20), 255 (3.87), 210 (4.42) nm; IR (KBr) ν_{\max} 3392, 2928, 2876, 1693, 1650, 1605, 1564, 1483, 1418, 1362, 1276, 1158, 1134, 1065, 872, 768 cm⁻¹; ¹H and ¹³C NMR data (CDCl₃, 500 and 125 MHz), Table 1; ESI-MS (negative ion mode) m/z 351[M-H]⁻; HR-ESI-MS (negative ion mode) m/z 351.0501 [M-H]⁻ (calcd 351.0505 for C₁₉H₁₁O₇).

9,11-Dihydroxy-2,5-dimethyl-4H-naphtho[2,3-h]chromene-4,7,12-trione (2): C₁₉H₁₂O₆, obtained as

yellow powder; UV (MeOH), λ_{\max} (log ϵ) 364 (3.71), 276 (4.15), 254 (3.85), 210 (4.38) nm; IR (KBr) ν_{\max} 3395, 2926, 2874, 1695, 1652, 1608, 1559, 1487, 1438, 1369, 1274, 1152, 1128, 1057, 883, 764 cm⁻¹; ¹H and ¹³C NMR data (CDCl₃, 500 and 125 MHz), Table 1; ESI-MS (negative ion mode) m/z 335 [M-H]⁻; HR-ESI-MS (negative ion mode) m/z 335.0562 [M-H]⁻ (calcd 335.0556 for C₁₉H₁₁O₆).

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REFERENCES

1. V. Duraipandiyan and S. Ignacimuthu, *J. Ethnopharmacol.*, 2007, **112**, 590.
2. S. Rajan, D. S. Baburaj, M. Sethuraman, and S. Parimala, *Ethnobotany*, 2001, **6**, 19.
3. J. Ma, L. X. Zhang, and Y. H. Guan, *Chin. J. Ethnomed. Ethnopharm.*, 2004, **5**, 178.
4. K. A. Abo, A. A. Adeyemi, and A. O. Sobowale, *Afr. J. Med. Med. Sci.*, 2001, **30**, 9.
5. S. Aurapa and G. Wandee, *Int. J. Biomed. Pharm. Sci.*, 2009, **3**, 42.
6. P. Sartorelli, S. P. Andrade, M. S. Melhem, F. O. Prado, and A. G. Tempone, *Phytother. Res.*, 2007, **21**, 644.
7. Y. H. Kuo, P. H. Lee, and Y. S. Wein, *J. Nat. Prod.*, 2002, **65**, 1165.

8. Z. Zuraini, C. Yeng, L. L. Yee, Y. L. Lachimanan, N. S. Lai, and S. Sreenivasan, *Molecules*, 2011, **16**, 7583.
9. W. Zhao, X. Y. Zeng, T. Zhang, L. Wang, G. Y. Yang, Y. K. Chen, Q. F. Hu, and M. M. Miao, *Phytochem. Lett.*, 2013, **6**, 179.
10. Y. Igarashi, M. E. Trujillo, E. Martinez-Molina, S. Yanase, S. Miyanaga, T. Obata, H. Sakurai, I. Saiki, T. Fujita, and T. Furumai, *Bioorg. Med. Chem. Lett.*, 2007, **17**, 3702.
11. F. Zhao, S. J. Wang, S. Lin, C. G. Zhu, S. P. Yuan, X. Y. Ding, Z. G. Yue, Y. Yu, B. Liu, X. L. Wu, Q. Hou, and J. G. Shi, *J. Asian Nat. Prod. Res.*, 2011, **13**, 1023.
12. X. K. Xia, H. R. Huang, Z. G. She, C. L. Shao, F. Liu, X. L. Cai, L. L. P. Vrijmoed, and Y. C. Lin, *Magn. Reson. Chem.*, 2007, **45**, 1006.
13. T. D. Sousa, P. C. Jimenez, E. G. Ferreira, E. R. Silveira, R. Braz, O. D. L. Pessoa, L. V. Costa-Lotufo, and O. D. L. Pessoa, *J. Nat. Prod.*, 2012, **75**, 489.
14. K. Trisuwan, N. Khamthong, V. Rukachaisirikul, S. Phongpaichit, S. Preedanon, and J. Sakayaroj, *J. Nat. Prod.*, 2010, **73**, 1507.
15. Y. C. Hu, E. D. Martinez, and J. B. MacMillan, *J. Nat. Prod.*, 2012, **75**, 1759.
16. P. Tuntiwachwuttikul, P. Phansa, Y. Pootaeng-On, and W. C. Taylor, *Chem. Pharm. Bull.*, 2006, **54**, 44.
17. T. Arakawa, H. Yamasaki, K. Ikeda, D. Ejima, T. Naito, and A. Koyama, *Curr. Med. Chem.*, 2009, **16**, 2485.
18. K. A. El Sayed, *Stud. Nat. Prod. Chem.*, 2000, **24(E)**, 473.
19. Q. F. Hu, B. Zhou, J. M. Huang, X. M. Gao, L. D. Shu, G. Y. Yang, and C. T. Che, *J. Nat. Prod.*, 2013, **76**, 292.