HETEROCYCLES, Vol. 106, No. 9, 2023, pp. 1443 - 1477. © 2023 The Japan Institute of Heterocyclic Chemistry
Received, 22nd April, 2023, Accepted, 7th June, 2023, Published online, 14th June, 2023
DOI: 10.3987/REV-23-1009

RECENT TRENDS IN THE SYNTHESIS OF ENAMINONES

Al-Shimaa Badran,[a] Najla A. Alshaye,[b] Zeinab Hussain,[a]* Aya Ahmed,[a] and Magdy A. Ibrahim[a]

[a]Department of Chemistry, Faculty of Education, Ain Shams University, Roxy, Heliopolis 11757, Cairo, Egypt
[b]Department of Chemistry, College of Science, Princess Nourah bint Abdulrahman University, P.O. Box 84428, Riyadh 11671, Saudi Arabia: naalshaye@pnu.edu.sa
*E-mail: zeinab.foda@edu.asu.edu.eg

Abstract – Enaminones are very useful intermediates for building different categories of heterocyclic compounds. Formation of enaminone derivatives was achieved using a variety of compounds such as alcohols, aldehydes, active methyl ketones, active methylene ketones, *β*-alkoxy ketones, *β*-diketones, acetoacetamides, Grignard reagents, diazocarbonyl compounds, acetylene derivatives, nitrile derivatives, cyclic compounds, acyl chloride and other enaminones. The present review covers the methods developed for the synthesis of enaminone derivatives.

CONTENTS

1. INTRODUCTION

The open-chain enaminones are active substrates containing functional group N–C=C–C=O as a part of a flexible chain. They are typical push-pull ethylenes in which the amine groups push electrons and the carbonyl groups pull electron densities. The presence of π-electron delocalization as well as *α*,*β*-unsaturated ketone are the two specific characteristic of enaminones which make them reactive.

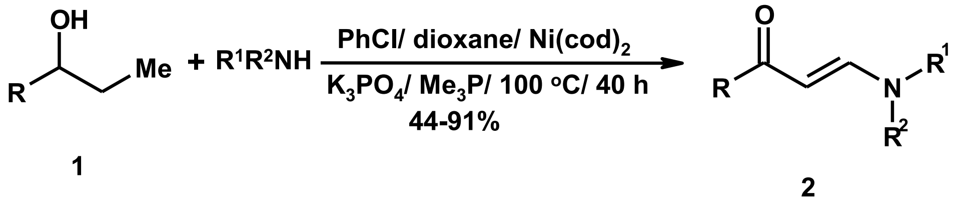
Enaminone compounds possess two electrophilic centers and three nucleophilic centers. So, enaminones are useful in variable scientific branches including synthetic intermediates in organic synthesis,[1,2] in pharmaceutical development[3] and in heterocyclic synthesis.[4,5] Also, they have been used as biological active agents such as antibacterial,[6] ant-inflammatory,[7] antioxidant, antimicrobial,[8] anticonvulsant,[9] antitumor [10] and anticancer agents.[11]

Numerous techniques have been developed for construction of enaminones using a diversity of substrates. The present review aims to collect the several methods used for the synthesis of enaminone derivatives starting from 2010 till now (April 2023).

2. SYNTHESIS OF ENAMINONES

2.1. From alcohols

Ueno *et al.*[12] reported that heating a mixture of α-substituted propanols **1** and amines, in 1,4-dioxane and chlorobenzene in the presence of potassium phosphate (4.0 equiv), Ni(cod)$_2$ (cod= cycloocta-1,4-diene) and trimethylphosphine, gave enaminones **2** (Scheme 1).



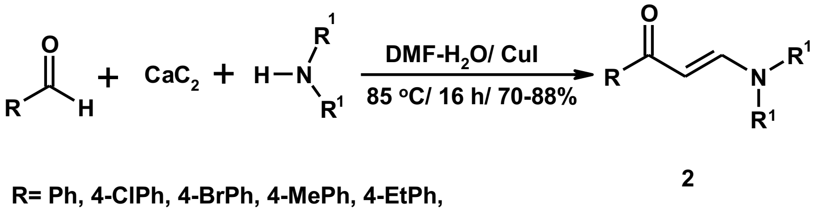
R= Et, *i*-Bu, *c*-hex, Ph, 4-ClPh, 4-F_3CPh, 4-MePh

R^1R^2NH= $HNBu_2$, H_2NBz, $HNBz_2$, piperidinyl, morpholinyl, *N*-Boc-piperazinyl, 1,2,3,4-tetrahydroisoquinolinyl

Scheme 1

2.2. From aldehydes

Various enaminones **2** were synthesized through three-component coupling reaction of aryl aldehyde, amines and calcium carbide, in aqueous dimethylformamide (DMF) containing copper(I) iodide as a catalyst (Scheme 2).[13]

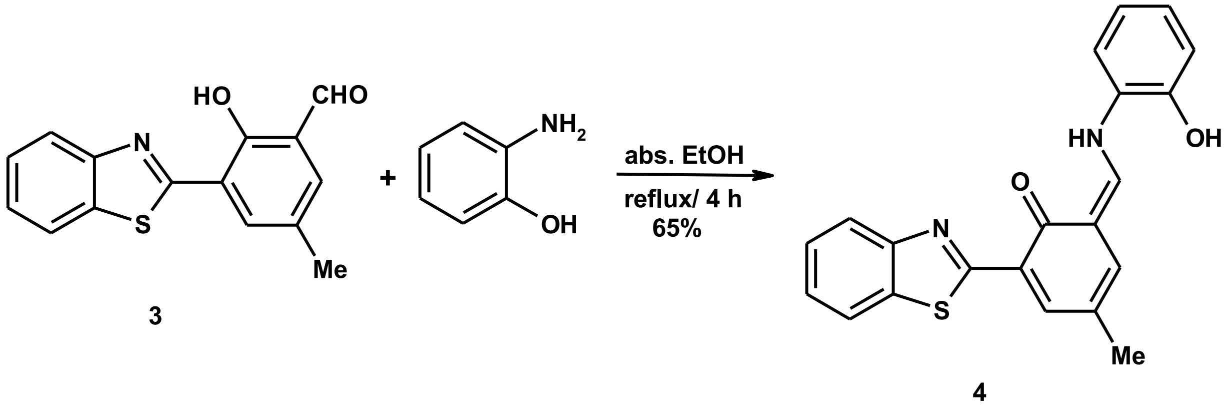


R= Ph, 4-ClPh, 4-BrPh, 4-MePh, 4-EtPh, 4-MeOPh, 4-NO_2Ph, 3,5-$(MeO)_2$Ph, 4-CF_3Ph, 4-CNPh, 4-PhPh

R^1= *i*-Pr, *c*-hex

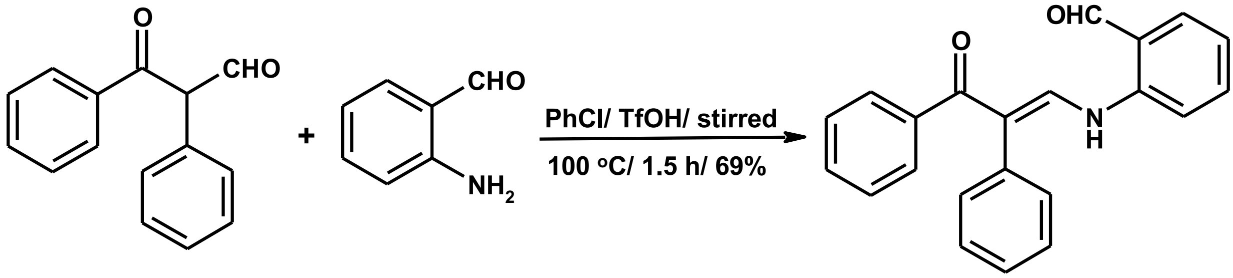
Scheme 2

2-(1,3-Benzothiazol-2-yl)-6-{[(2-hydroxyphenyl)amino]methylidene}-4-methylcyclohexa-2,4-dien-1-one (**4**) was prepared from nucleophilic addition of 2-aminophenol to 3-(benzo[*d*]thiazol-2-yl)-2-hydroxy-5-methylbenzaldehyde (**3**), in boiling ethanol (Scheme 3).[14]



Scheme 3

Luo *et al.*[15] demonstrated that 2-((*E*)-3-oxo-2,3-diphenylprop-1-enylamino)benzaldehyde (**6**) was formed, in good yield, from condensation reaction of 3-oxo-2,3-diphenylpropionaldehyde (**5**) with 2-aminobenzaldehyde, in chlorobenzene in the presence of triflic acid (TfOH) (Scheme 4).



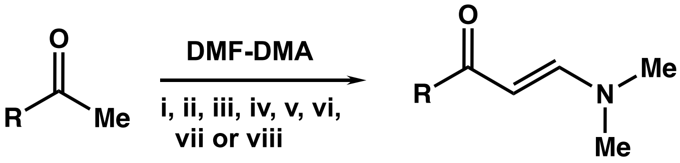
5

6

Scheme 4

2.3. From active methyl ketones

Acetyl derivatives represent one of the most common sources for formation of enaminones. Thus, treatment of acetyl derivatives **7** with dimethylformamide dimethyl acetal (DMF-DMA), under different reaction conditions, afforded (3-(dimethylamino)-1-substituted)prop-2-en-1-one **8** as shown in Scheme 5. Table 1 refers to the types of substituents (R) and their references.



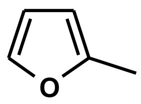
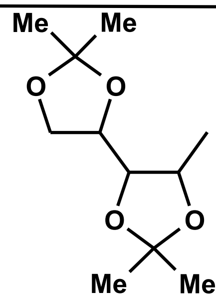
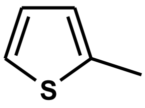
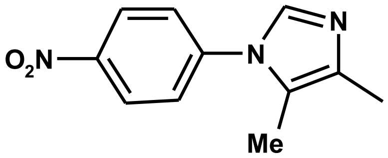
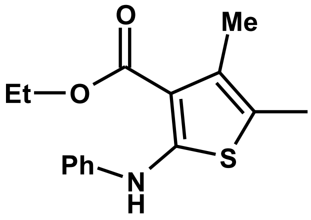
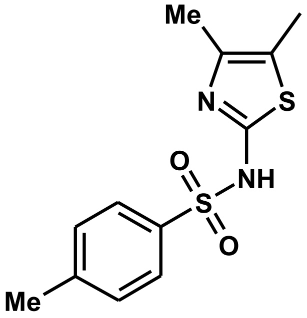
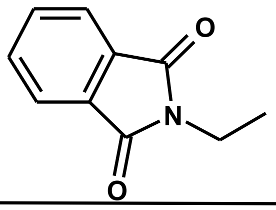
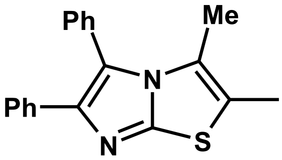
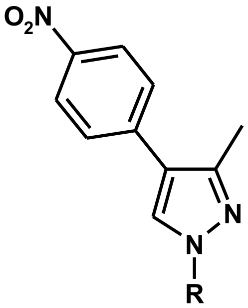
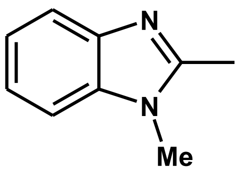
7

8

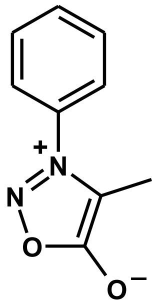
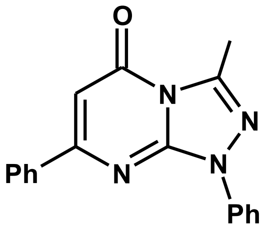
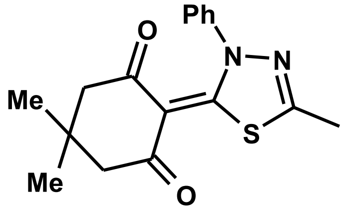
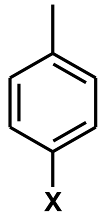
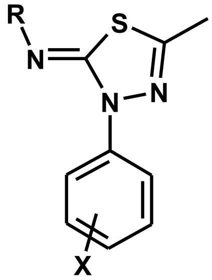
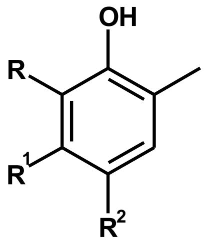
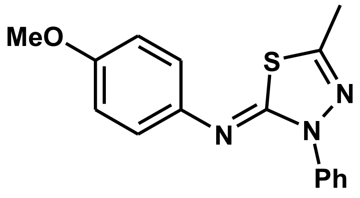
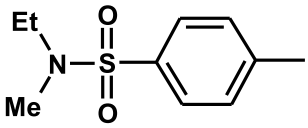
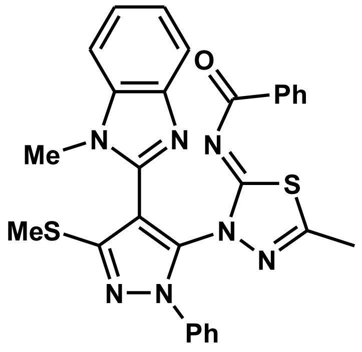
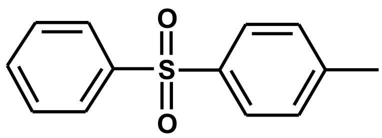
i) EtOH/ TEA/ reflux/ 6 h/ 58%[34]
ii) DMF/ reflux/ 0.5-3 h/ 82-97%[35]
iii) benzene/ reflux/ 5 h/ 77%[27]
iv) toluene/ reflux/ 5-24 h/ 46-90%[26,29,31,33,38]
v) fusion/ 85-120 °C/ 0.5-20 h/ 75-86%[17,19,21,24,40]
vi) dioxane/ reflux/ 5-6 h/ 90-93%[25,30,32,36]
vii) xylene/ reflux/ 2-7 h/ 58-93%[20,22,23,28,37,39]
viii) MW/ 120-180 °C/ 5-45 min/ 77-95%[1,2,16,18,33]

Scheme 5

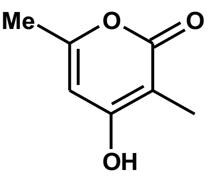
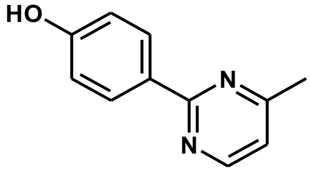
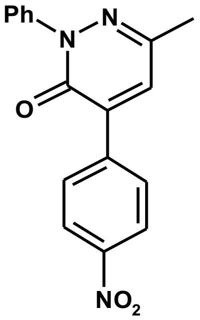
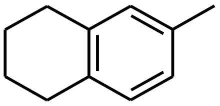
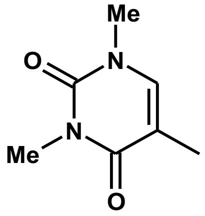
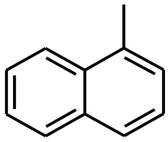
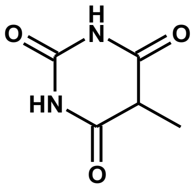
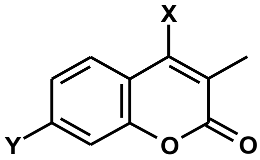
Table 1:

R	Ref	R	Ref
	[16]		[20]
	[1]		[21]
	[17]		[22]
	[18]		[23]
 R= Ph, 4-MePh	[19]		[24]

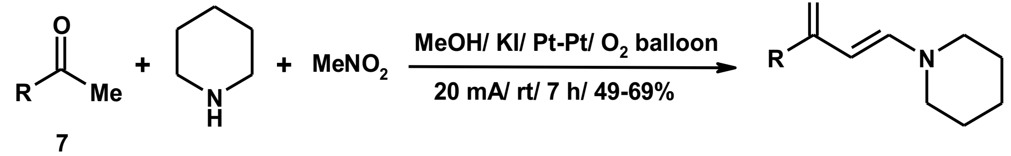
Continued Table 1

	[21]		[29]
	[25]	 X= Cl, NO_2, MeO, CO_2Me	[1,16,21,30]
 X= H, 2-Cl, 4-Cl, 4-Br, 4-NO_2 2-Me, 4-Me, 4-MeO R= 4-MeOPh, PhOC	[26,27]	 R= H, I, Br R^1= H, Me R^2= H, I, Br, Cl, Me	[31]
	[27]		[32]
	[28]		[33]

Continued Table 1

	[34,35]		[37]
	[36]		[38]
	[35]		[2]
	[35]	 X= H, OH Y= H, OH	[35,39,40]

Xu *et al.*[41] showed that enaminones **9** were obtained from electro-synthesis of acetyl derivatives **7**, piperidine and nitromethane. This method proceeded at room temperature to give enaminones **9** in one step, utilizing nitromethane as the carbon source (Scheme 6).



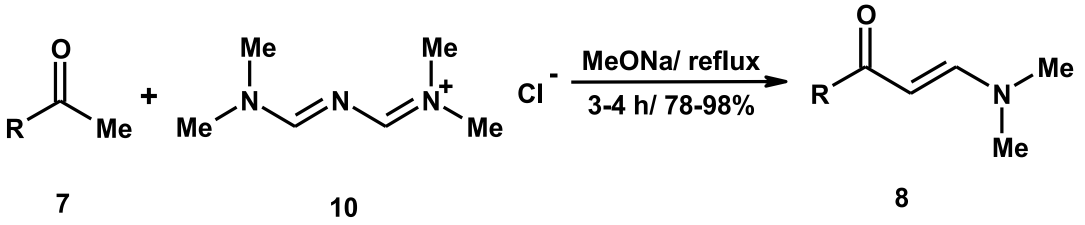
7

9

R= Ph, 2-FPh, 4-FPh, 4-ClPh,
4-BrPh, 4-IPh, 4-F_3CPh, 2-MePh,
3-MePh, 4-MePh, 3-MeOPh, 4-MeOPh

Scheme 6

Enaminones **8** have been prepared by Gupton method through reaction of acetyl derivatives **7** with Gold's reagent [3-(dimethylamino)-2-azaprop-2-en-1-ylidene]dimethylammonium chloride (**10**), in a good yields as shown in Scheme 7.[42,43]

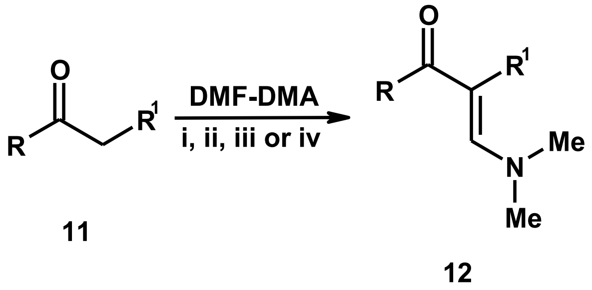


R= 2-pyrrolyl, Ph, 4-BrPh, 4-ClPh,
4-NO_2Ph, 4-MeOPh, 2-naph,
2-(3-methylthiazolo[3,2-*a*]benzimidazolyl

Scheme 7

2.4. From active methylene ketones

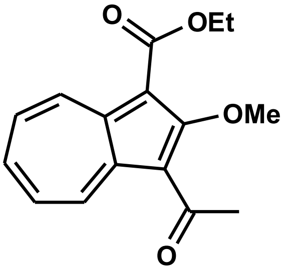
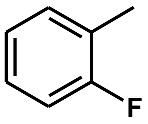
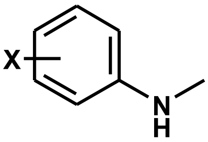
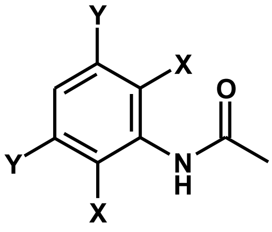
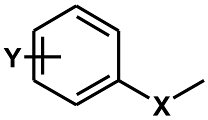
Reaction of active methylene ketones **11** with DMF-DMA under various conditions [boiling in DMF or xylene, fusion or under microwave irradiations] donated enaminones **12**, in good to excellent yields (Scheme 8). Table 2 indicates to types of substituent R, R^1 and their references.



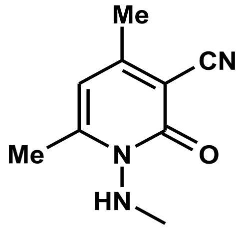
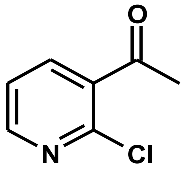
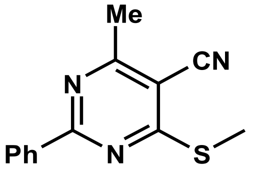
i) DMF/ 80-100 °C/ 0.5-3 h/ 82-97%[45,46,51]
ii) xylene/ reflux/ 4-5 h/ 66-87%[48,50,52]
iii) fusion/ 90-120 °C/ 0.5-24 h/ 55-98%[47,49]
iv) MW/ 5 min/ 90%[44]

Scheme 8

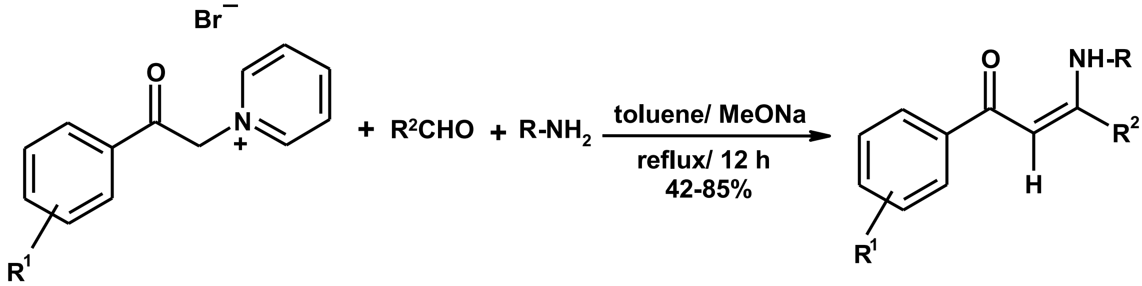
Table 2:

R	R^1	Ref
Me	CO_2Et	[44]
	CN	[45]
	CO_2Et	[46]
 X= Ph, 4-ClPh, 4-MePh, 2-MeOPh, 2,6-$(MeO)_2$Ph, 1-naph, 2-naph	Me, Ph, 4-FPh	[47]
 X=Y= H, Me	CN, COPh	[48]
 X = O, S Y= H, 4-Cl, 3-MeO, 2,5-$(MeO)_2$, 3,4-$(MeO)_2$	Me, 2-BrPh, 3-MePh, 4-MePh, 3-MeOPh, 4-MeOPh, 3-Br-4-MeOPh	[49]

Continued Table 2

	CN	[50]
	CO_2Et	[51]
	Ph	[52]

Enaminones **14** were prepared *via* multicomponent reaction of phenacylpyridinium bromides **13**, primary amines, and phenylglyoxals or ethyl glyoxalate, in moderate to good yields (Scheme 9).[53]



13

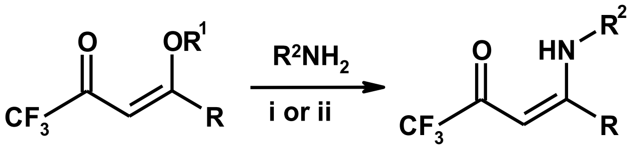
14

$R = 4\text{-ClPh}, 3\text{-MePh}, 4\text{-MePh}, 3,4\text{-(Me)}_2\text{Ph}$
$R^1 = H, 4\text{-Me}, 4\text{-Br}, 4\text{-Cl}$
$R^2 = CO_2Et, PhCO, 4\text{-ClPhCO}, 4\text{-MePhCO}$

Scheme 9

2.5. From *β*-alkoxy ketones

Trifluoromethylated *β*-aminovinyl ketones **16** were prepared, in good to excellent yields, by reaction of (*E*)-4-alkoxy-1,1,1-trifluorobut-3-en-2-one **15** with various amines (slight excess), in anhydrous acetonitrile (Scheme 10). Table 3 illustrates the types of substituent R, R^1, R^2 and their references.



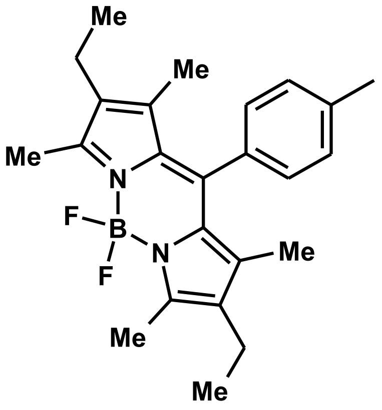
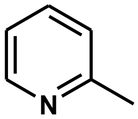
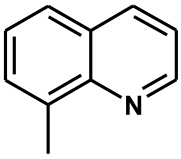
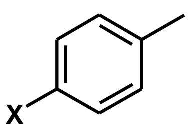
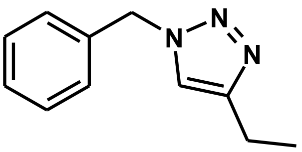
15 **16**

i) MeCN/ stirred/ rt/ 5 h/ 72-94%[54,55,57]

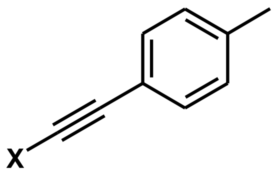
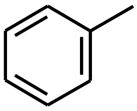
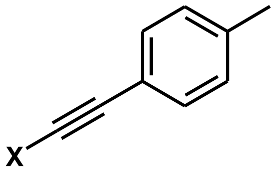
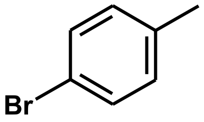
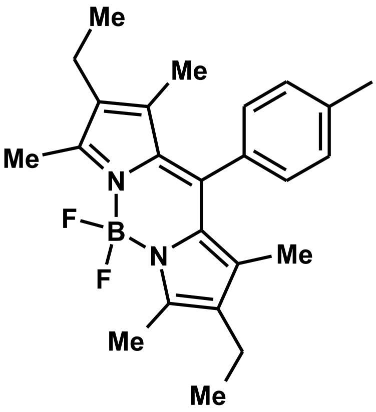
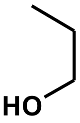
ii) MeCN/ TEA/ stirred/ rt/ 24 h/ 44-80%[56]

Scheme 10

Table 3:

R	R^1	R^2	Ref
H	Et		[54]
H	Et		[55]
H	Et		[55]
 X= H, Br, NO$_2$	Me		[56]

Continued Table 3

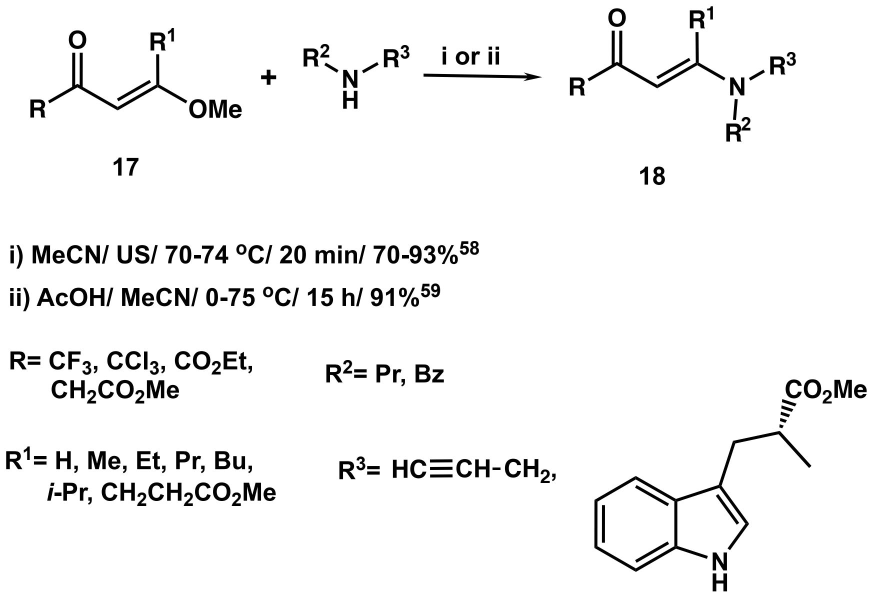
 X= TMS, Ph	Me		[57]
 X= TMS, Ph	Me		[57]
	Me		[54]

Reaction of *β*-methoxyvinyl ketones **17** with secondary amines, under different conditions [thermal or ultrasound irradiation (US), afforded enaminones **18**, in 70-93% yields as illustrated in Scheme 11.[58,59]

2.6. From *β*-diketones

Reacting acetylacetone (**19**) with ammonium salts in the presence of catalytic amount of 1-butyl-3-methylimidazolium hydrogen sulphate [bmim]HSO_4 under solvent free conditions led to 4-aminopent-3-en-2-one (**20**) (Scheme 12).[60]

Enaminones **21** were prepared, in quantitative yields, by the reaction of primary amines with *β*-diketones **19** under different conditions (Scheme 13). The substituents and their references are depicted in Table 4.



17

18

i) MeCN/ US/ 70-74 °C/ 20 min/ 70-93%[58]

ii) AcOH/ MeCN/ 0-75 °C/ 15 h/ 91%[59]

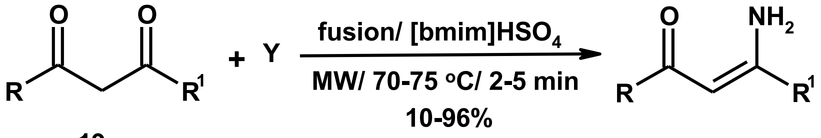
R= CF_3, CCl_3, CO_2Et, CH_2CO_2Me

R^2= Pr, Bz

R^1= H, Me, Et, Pr, Bu, *i*-Pr, $CH_2CH_2CO_2Me$

R^3= HC≡CH-CH_2,

Scheme 11



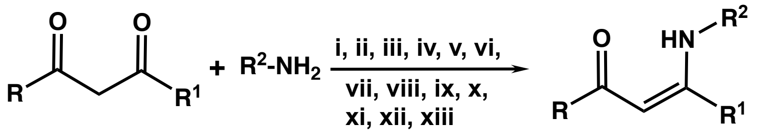
19

20

R= R^1= Me

Y= $(NH_4)_2CO_3$, $(CO_2NH_4)_2$, NH_4OAc, NH_4Cl

Scheme 12



19

21

i) toluene/ *p*-TsOH/ reflux/ 10-48 h/ 40-52%[69]

ii) $CHCl_3$/ AcOH/ 80 °C/ 1 h/ 40-58%[70]

iii) THF/ CuCl/ rt/ 12-48 h/ 75-98%[61]

iv) MeCN/ 40% phosphotungstic acid/ TiO_2/ rt/ 1-30 min/ 65-98%[65]

v) EtOH/ stirred/ rt/ 1 h/ 74-89%[63]

vi) MeOH/ HCO_2H/ 85 °C/ 4 h/ 53-98%[67]

vii) EtOH/ AcOH/ reflux/ 3 h/ 45%[63]

viii) fusion/ $ZrOCl_2{\cdot}8H_2O$/ stirred/ rt/ 10-300 min/ 81-95%[64]

ix) fusion/ [VO(acac)$_2$]/ stirred/ rt/ 15-60 min/ 86-91%[68]

x) fusion/ $Fe(HSO_4)_3{\cdot}SiO_2$/ rt/ 7-25 min/ 81-95%[66]

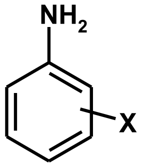
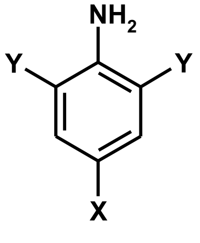
xi) fusion/ 40% phosphotungstic acid/ TiO_2/ rt/ 1-13 min/ 90-98%[65]

xii) fusion/ [bmim]H_2SO_4/ MW/ 70-75 °C/ 2-3 min/ 70-98%[60]

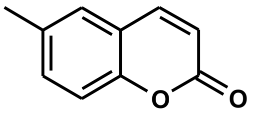
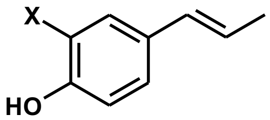
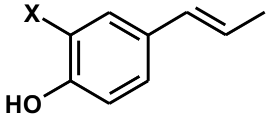
xiii) fusion/ $Zn(X)_2$/ ultraviolet/ 1h/ 65-99% X= Gly, Pro[62]

Scheme 13

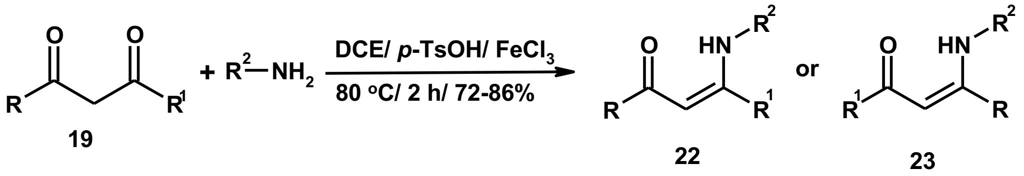
Table 4:

R	R^1	R^2	Ref
Me	Me	Bu, Ph, 2-ClPh, 3-ClPh, 4-ClPh, 4-BrPh, 3HOPh, 2-MePh, 4-MePh, 2-MeOPh, 4-MeOPh, Bz, 1-naph	[61]
Me	Me	Me, *i*-Pr, Pe, hex, 4-ClPh 4-MePh, Bz, 1-naph	[60]
Me	Me	*c*-hex, Ph, Bz, 4-MePh 4-MeOPh	[62]
Me	Me	 X= 2-CO_2H, 4-CO_2H	[63]
Me, Ph	Me	Bn, Ph, 2-MePh , 4-ClPh, 4-MePh, 4-MeOPh, 4-EtOPh	[64]
Me, Ph	Me	$(CH_2)_2NH_2$, *n*-Bu, Ph, 4-ClPh 4-NO_2Ph, 4-MePh, 4-MeOPh, Bz, PhCO	[65]
Me, Ph	Me	Ph, 4-ClPh, 4-NO_2Ph, 4-MePh, 4-MeOPh,	[66]
Me, Ph	Me	 X= H, Me, OMe, Cl, NO_2, CN Y= H, Me, *i*-Pr	[67]
Ph	Me	2-MeOPh, 2,6-$(Me)_2$Ph, Bz	[68]

Continued Table 4

Me, Et, Ph	Me, Et		[69]
 X= H, OMe	 X= H, OMe	*c*-hex, *n*-Pr, *n*-Bu, *i*-Bu, *sec*-Bu, allyl	[70]

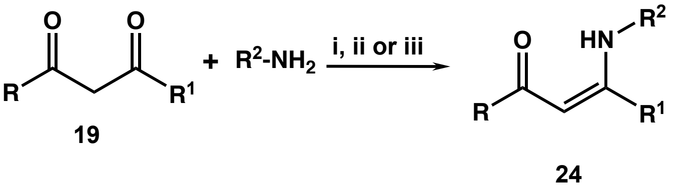
On the other hand, treatment of 1,3-dicarbonyl compounds **19** with some primary amines using $FeCl_3$/*p*-TsOH as a catalyst gave *β*-*N*-substituted enaminones **22** or **23** (Scheme 14).[71]



R= Me
R^1= CF_3, Ph
R^2= hex, 2-py, 3-MePh, 4-MePh, 4-MeOPh, Bz, 1-naph

Scheme 14

Also, acyclic enaminones **24** were prepared, in good to excellent yields, from the condensation reaction of various diketones **19** with different amines, using copper, silver or copper oxide nanoparticles (NPs) as catalyst (Scheme 15).[72-74]

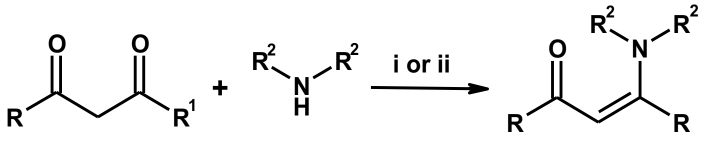


i) MeOH/ NPs Ag/ 60 °C/ 1-8 h/ 70-93%[72]
II) MeOH/ NPs Cu/ 45-50 °C/ 1.15-3.20 h/ 85-94%[73]
iii) fusion/ CuO NPs/ 60 °C/ 9 h/ 70-98%[74]

R= Me, Ph
R^1= Me
R^2= Me, Et, C_2H_2, *i*-Pr, Bu, *c*-hex, 2-pyrrolyl, Ph, 4-BrPh, 4-NO_2Ph, 4-MePh, 4-MeOPh, 2,4-F_2Ph, Bz

Scheme 15

Similarly, 4-*N*,*N*-dialkylaminopent-3-en-2-ones **25** were obtained from reaction of acetylacetone (**19**) with secondary amines under different reaction conditions, in good yields (Scheme 16).[75]



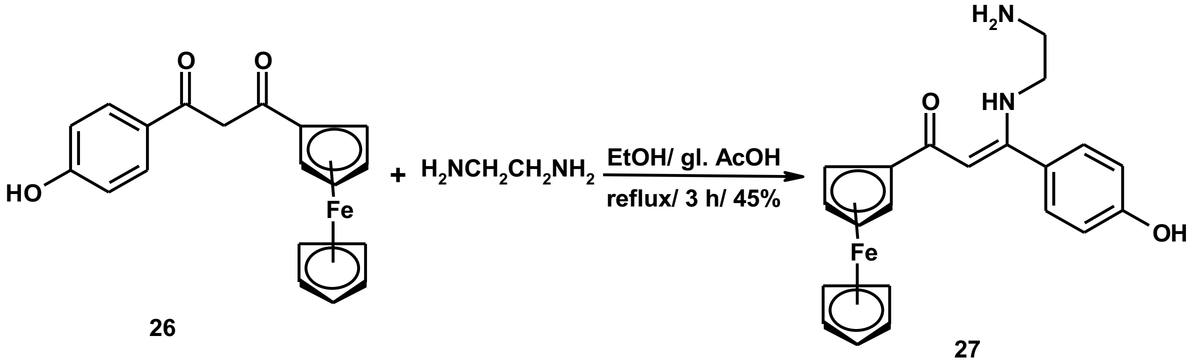
19 **25**

i= ice bath/ 15 min/ 66-68% R^2= Et R=R^1= Me

ii= benzene/ *p*-TsOH/ 85 °C/ 5 h/ 64-67% R^2= Pr, *c*-hex

Scheme 16

Celedon *et al.*[76] reported that formation of *O*,*N*,*N*-tridentate metalloligand enaminone **27**, in 45% yield, was achieved from organometallic *β*-diketone **26** with ethylenediamine, in ethanol containing drops of glacial acetic acid (Scheme 17).

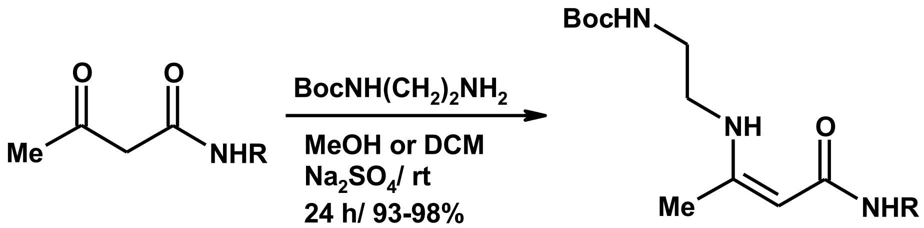


26 **27**

Scheme 17

2.7. From acetoacetamides

Boc-monoprotected ethylenediamines (Boc = *tert*-butyloxycarbonyl) was treated with acetoacetamides **28** to provide *β*-enaminoamides **29**, in quantitative yields (Scheme 18).[77]



28

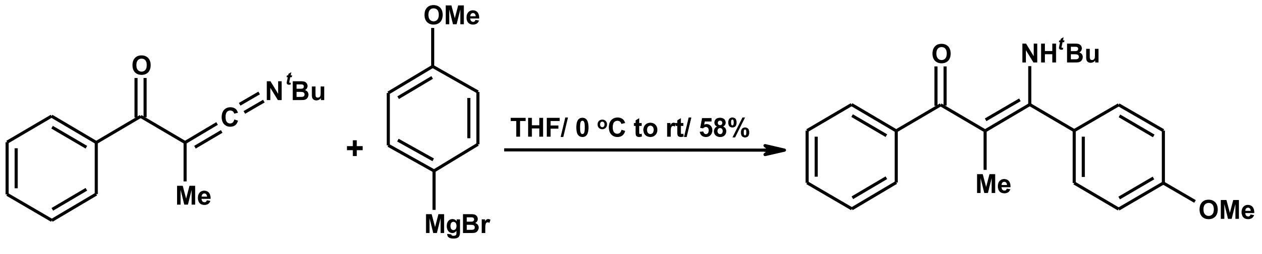
R= H, Ph, $(CH_2)_2Ph$

29

Scheme 18

2.8. From Grignard reagents

Mamboury *et al.*[78] presented that treatment of α-oxo-ketenimine **30** with 4-methoxyphenylmagnesium bromide (Grignard reagent) (**31**) provided 3-(*t*-butyl-amino)-3-(4-methoxyphenyl)-2-methyl-1-phenylprop-2-en-1-one (**32**), in 58% yield, *via* chemoselective nucleophilic addition to the ketenimine function (Scheme 19).

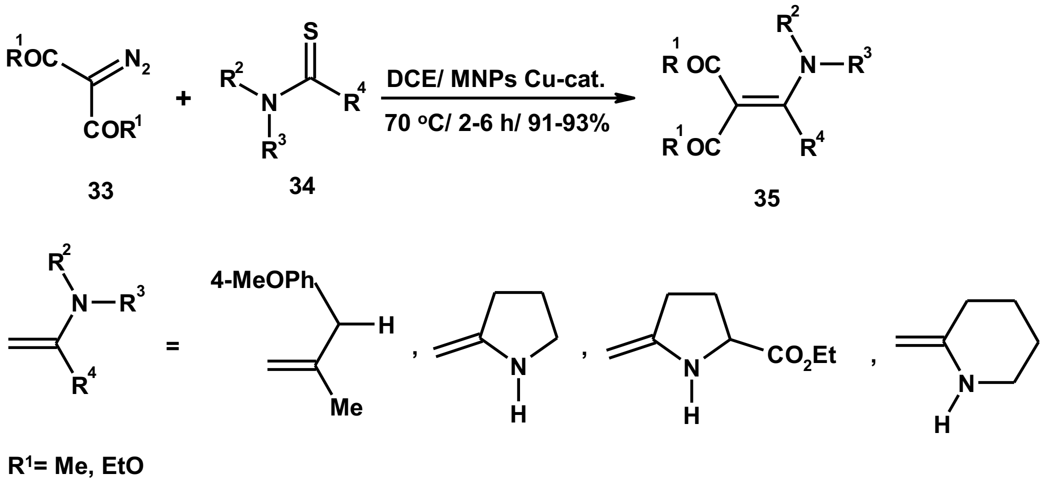


30 **31** **32**

Scheme 19

2.9. From diazocarbonyl compounds

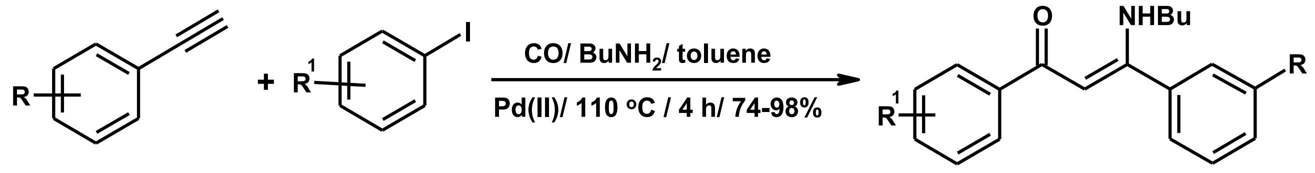
Mohammadi *et al.*[79] presented that coupling of α-diazocarbonyl compounds **33** and thioamides **34**, using a magnetic nanoparticle (MNPs)-Cu as a catalyst, in dry dichloroethane at 70 °C, gave enaminones **35** (Scheme 20).



Scheme 20

2.10. From acetylene derivatives

On the other hand, alkynes represent another way to synthesize enaminones. Carbonylative Sonogashira coupling of terminal alkynes **36** with iodobenzenes **37** is accompanied by Michael addition of butylamine in a one-pot method, to generate enaminones **38**. This reaction occurred in the presence of *N*,*N* bidentate palladium(II) complexes as catalyst (Scheme 21).[80]

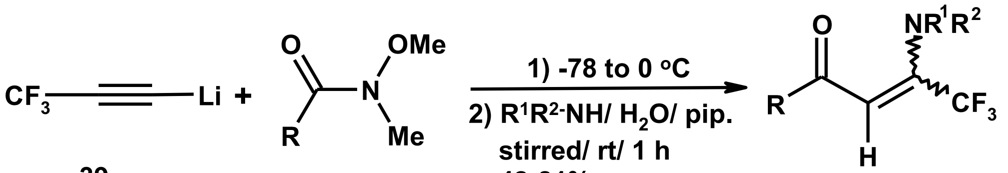


36 **37** **38**

R= H, 4-F, 4-Me, 4-Et, 4-^{t}Bu,
R^1= H, 4-Me

Scheme 21

Trifluoromethyl enaminones **41** were prepared, in good yields, from the reaction of trifluoropropynyllithium (**39**) with *N*-methoxy-*N*-methylamide **40** at -78-0 °C, followed by quenching with a variety of amines, in the presence of H_2O (Scheme 22).[81]

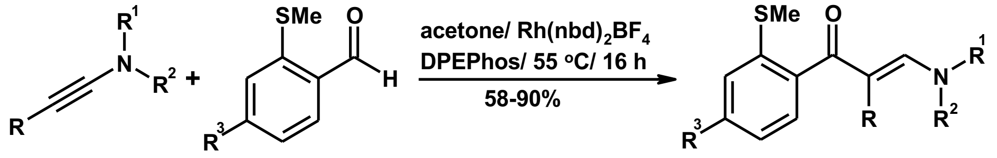


39 **40** **41**

R= Ph, 4-ClPh, 4-MePh
R^1, R^2= *c*-hex, $(Me)_2$, (Bn, Me), (H, Bn),
$((Me)_2CH)_2$, $((Me)_3Si)_2$, (H, Ph)

Scheme 22

Enaminones **44** were generated, in high yields, from the reaction of ynamide derivatives **42** and aldehydes **43**, in the presence of rhodium *bis*[(2-diphenylphosphino)phenyl] ether [DPEPhos] and [*bis*(norbornadiene)rhodium(I) tetrafluoroborate] [$Rh(nbd)_2BF_4$] as depicted in Scheme 23.[82]

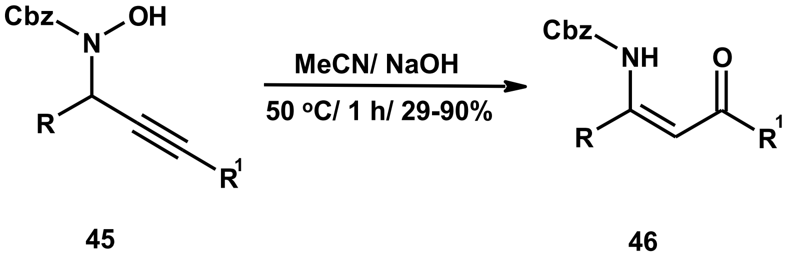


42 **43** **44**

R= 3-thienyl, *n*-hex, Ph, 4-CF_3Ph, 4-MeOPh
R^1= Ts, Cbz, Boc, Ms
R^2= Me, Ph
R^3= H, Br, F_3C, MeS, MeO

Scheme 23

Gayon *et al.*[83] demonstrated that (*N*-carboxybenzyl) Cbz-protected enaminones **46** were obtained from catalytic isomerization of readily available propargylic hydroxylamines **45**, in acetonitrile containing NaOH for 1 h, in 29-90% yields (Scheme 24).

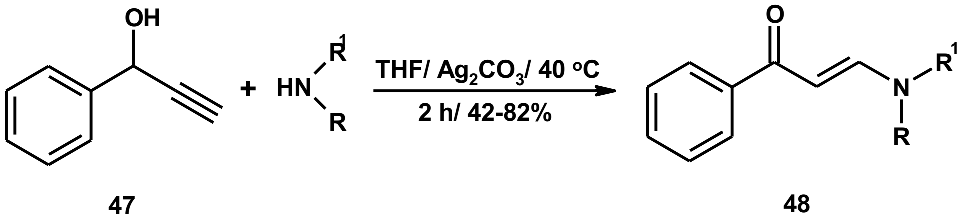


R= thienyl, Ph, 2-FPh, 3-BrPh, 4-FPh, 4-ClPh, 4-BrPh, 4-MePh, 3-MePh, 3-MeOPh, 4-PhPh, 1-naph, 2-naph

R1= Me, *i*-Pr, *n*-Bu, *t*-Bu

Scheme 24

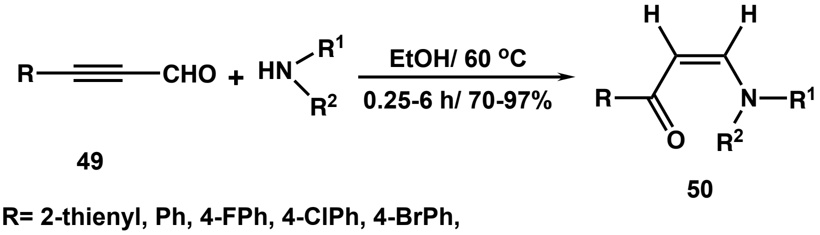
Enaminones **48** were prepared, in good yields, from treatment of propargyl alcohols **47** with various acyclic amines, under mild reaction conditions in the presence of Ag_2CO_3 (Scheme 25).[84]



R, R1= H, Me, Et, Ph, Bz

Scheme 25

A convenient one-pot strategy for the regioselective synthesis of enaminones **50** have been developed *via* the condensation reaction of propiolaldehydes **49** with some acyclic and cyclic amines, in boiling ethanol (Scheme 26).[85]

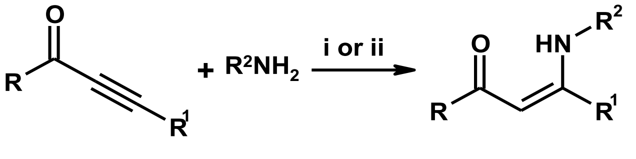


R= 2-thienyl, Ph, 4-FPh, 4-ClPh, 4-BrPh, 2-ClPh, 4-MePh, 4-EtPh, 4-MeOPh,

R1= R2 = Et, Pr, *i*-Pr, *n*-Bu, *n*-Pr, *n*-hex, *c*-hex, *n*-octanyl

Scheme 26

Enaminone derivatives **21** were obtained through addition reactions of primary amines with several α-ketoalkynes **51**, under various reaction conditions (a nickel homogeneous aqueous catalytic system formed from Ni(CN)2/CO/NaOH/KCN or boiling methanol) (Scheme 27). Table 5 refers to substituents R, R^1, R^2 with their reference.



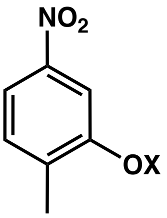
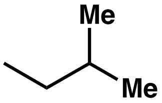
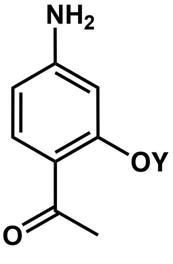
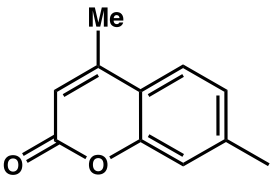
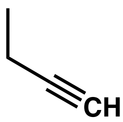
51 **21**

i) EtOH/ Ni(CN)$_2$/KCN/ NaOH/ CO/ H$_2$O/ rt/ 24 h/ 40-60%[87]

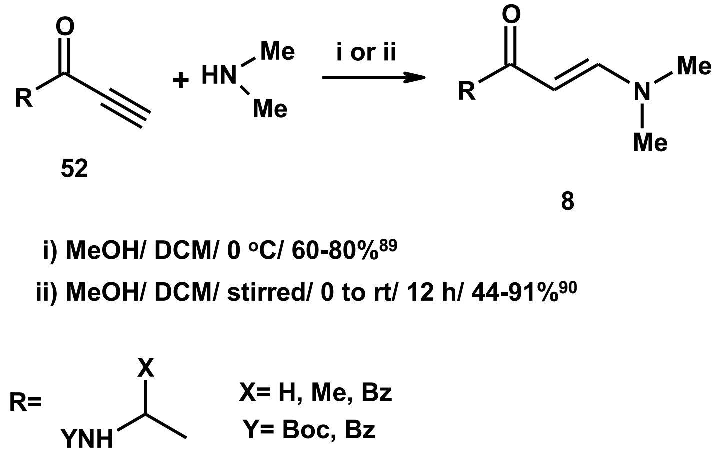
ii) MeOH/ 65-120 °C/ 3-6 h/ 13-98%[86,88]

Scheme 27

Table 5:

R	R^1	R^2	Ref
 X= Me, *i*-Pr, Bn		 Y= Me, *i*-Pr, Bn	[86]
3-MePh, 4-MePh 4-EtPh, 4-BrPh, 3,5-(MeO)$_2$Ph	Fc [Ferrocenyl]		[87]
Me, Ph, 4-ClPh, 4-MePh, 4-MeOPh	Bu, 3-thienyl, Ph 4-MePh, 4-MeOPh		[88]

Enaminones **8** were prepared, in a good yields, from addition of dimethylamine to ynone derivatives **52**, under different reaction conditions (Scheme 28).[89,90]

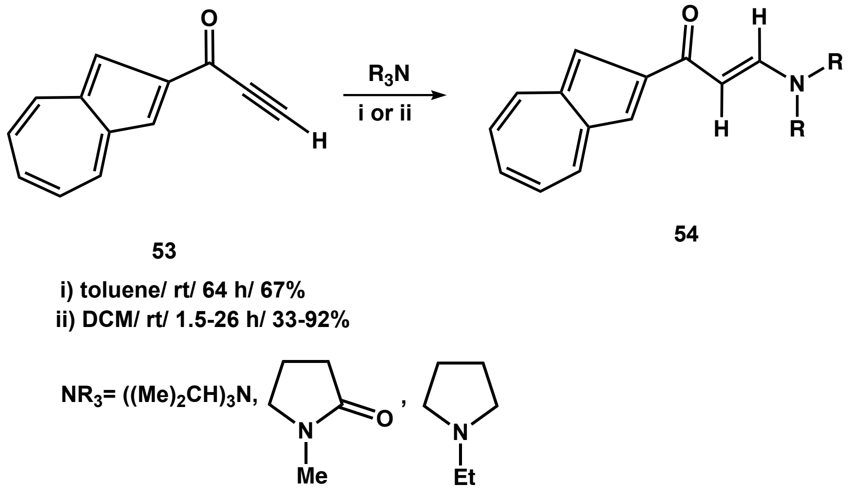


i) MeOH/ DCM/ 0 °C/ 60-80%[89]
ii) MeOH/ DCM/ stirred/ 0 to rt/ 12 h/ 44-91%[90]

X= H, Me, Bz
Y= Boc, Bz

Scheme 28

Treatment of 1-(azulen-2-yl)prop-2-yn-1-one (**53**) with tertiary amines, under various reaction conditions, resulted in 1-((azulen-2-yl)prop-2-en-1-ones **54**, in high yield (Scheme 29).[91]



i) toluene/ rt/ 64 h/ 67%
ii) DCM/ rt/ 1.5-26 h/ 33-92%

NR_3= $((Me)_2CH)_3N$,

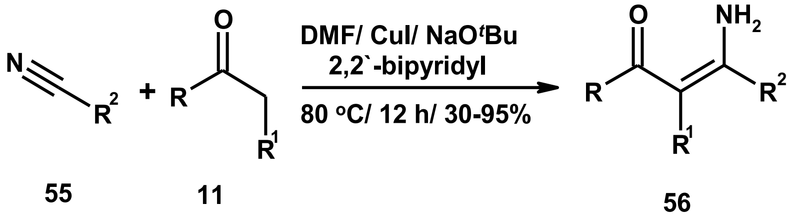
Scheme 29

2.11. From nitrile derivatives

Aldol-type addition of active methylene ketones **11** to aromatic nitriles **55** proceeded, in the presence of a simple copper catalyst system (CuI–2,2-bipyridyl–NaO*ᵗ*Bu), produced enaminones **56**, in satisfactory to excellent yields (Scheme 30).[92]

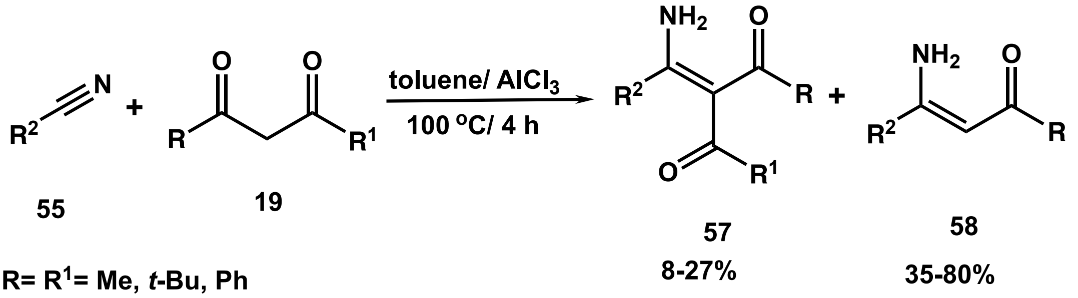
Various enaminone derivatives **57** and **58** were prepared from $AlCl_3$ catalyzed reaction of *β*-diketones **19** with nitriles **55**, in boiling toluene (Scheme 31).[93]

One-pot Ugi/nucleophilic addition reaction of isocyanides **59**, aromatic aldehydes, propiolic acid (**60**) and primary amines in the presence of secondary amines such as piperidine, morpholine and *N*-hydroxyethylpiperazine were carried out at room temperature to produce polyfunctionalized enaminones **61** as shown in Scheme 32.[94]



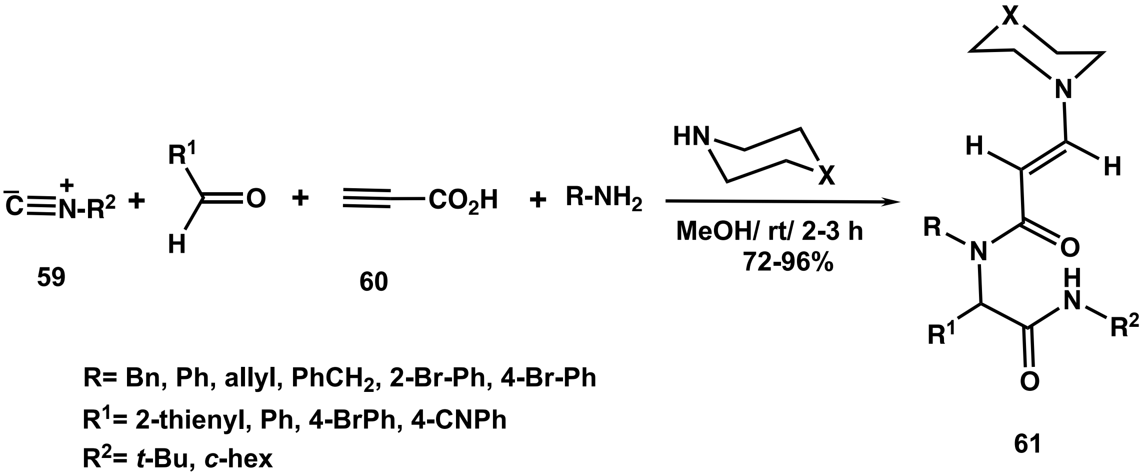
R= Pr, 2-thienyl, Ph, 2-MePh, 4-MePh, 4-MeOPh, 4-(Me)$_2$NPh, 4-BuNHPh, 4-FPh, 4-ClPh
R^1= H, Me
R^2= Ph, 4-ClPh, 4-CF$_3$Ph, 4-MePh, 4-MeOPh, 2-Py, 2-ClPy

Scheme 30



R= R^1= Me, *t*-Bu, Ph
R^2= Me, Br(CH$_2$)$_2$, 2-furyl, Ph, 2-ClPh, 2-MeOPh, 4-NO$_2$Ph, 2-NO$_2$Ph, Bz, 4-HOBz, PhCH=CH

Scheme 31

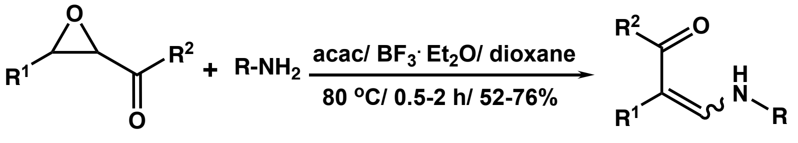


R= Bn, Ph, allyl, PhCH$_2$, 2-Br-Ph, 4-Br-Ph
R^1= 2-thienyl, Ph, 4-BrPh, 4-CNPh
R^2= *t*-Bu, *c*-hex
X= O, CH$_2$, NCH$_2$CH$_2$OH

Scheme 32

2.12. From cyclic compounds

Domino reactions of different oxirane derivatives **62** with substituted arylamines, in the presence of acetylacetone (acac) and $BF_3{\cdot}Et_2O$, in 1,4-dioxane, afforded enaminones **63**, in moderate yields (Scheme 33).[95]

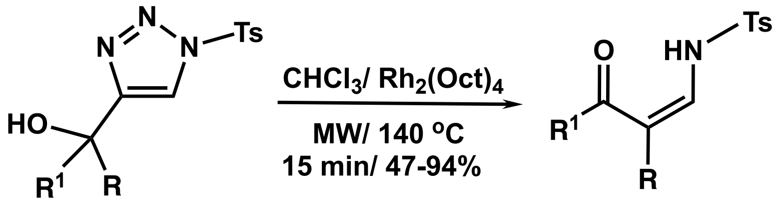


62 **63**

R= Ph, 2-IPh, 3-BrPh, 4-ClPh, 4-BrPh,
4-MeOPh, 3,4-Cl_2Ph, 1-naph
R^1= Ph, 4-BrPh
R^2= Ph, 4-MeOPh

Scheme 33

Enaminones **65** were synthesized by rhodium(II) octanoate, $Rh_2(Oct)_4$, catalyzed denitrogenative rearrangement reaction of 1-(*N*-sulfonyl-1,2,3-triazol-4-yl)alkanols **64**, in $CHCl_3$ at 140 °C under microwave irradiation for 15 min. (Scheme 34).[96]

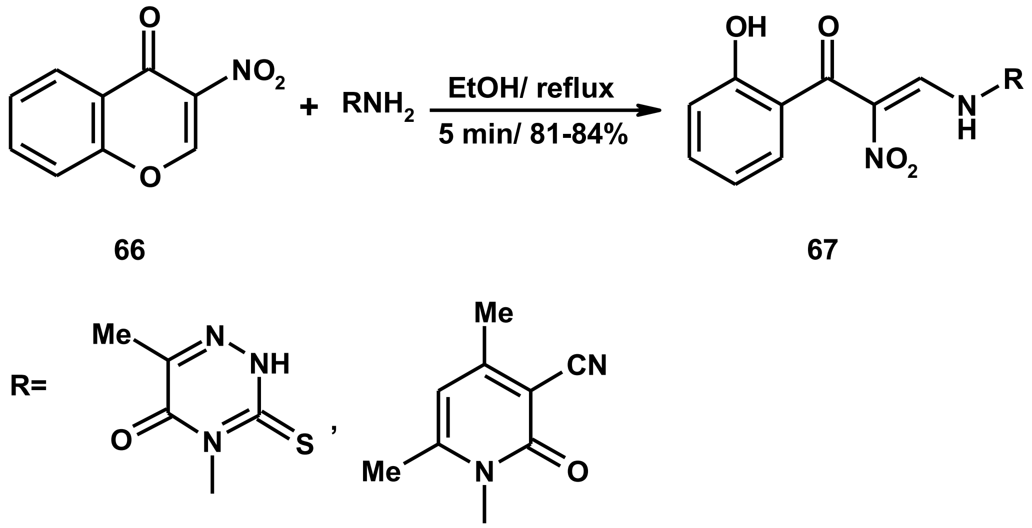


64 **65**

R= H, Me, Ph
R^1= Me, *i*-Pr, *t*-Bu, Ph

Scheme 34

On the other hand, ring opening of 3-nitrochromone (**66**) with some heterocyclic amines, in boiling ethanol, furnished (2*Z*)-1-(2-hydroxyphenyl)-3-(substitutedamino)-2-nitroprop-2-en-1-ones **67** (Scheme 35).[97]



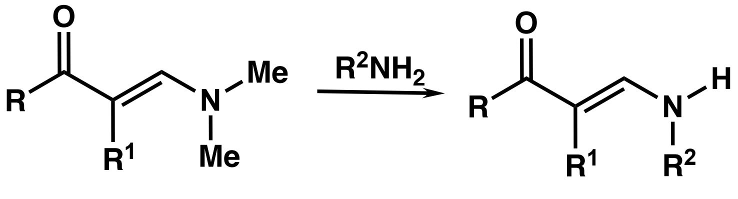
66

67

Scheme 35

2.13. From other enaminones

In addition to the above methods developed for the synthesise of enaminones, another method to synthesize enaminone derivatives are the reaction of enaminone itself with different primary and secondary amines. Thus, a variety of enaminones **68** were efficiently synthesized from reaction of enaminone **12** with a diversity of primary amines, under various reaction conditions (Scheme 36). Table 6 shows the types of substituent R, R^1, R^2 and their references.



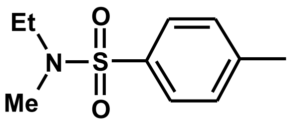
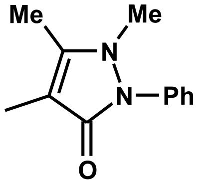
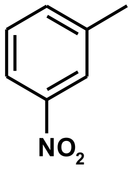
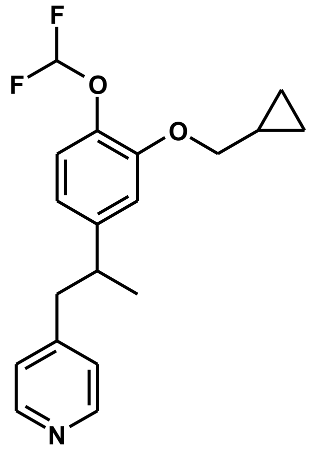
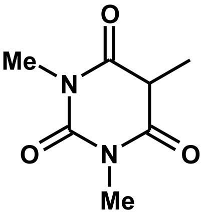
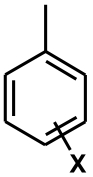
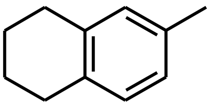
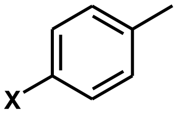
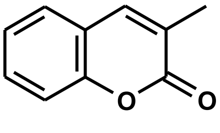
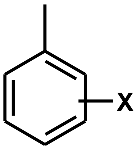
12

68

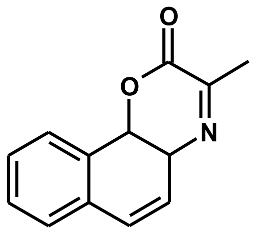
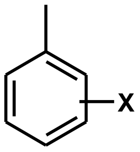
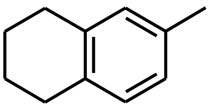
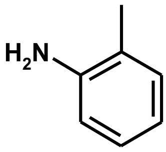
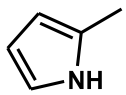
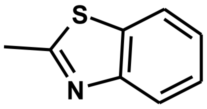
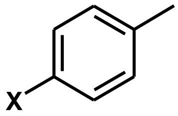
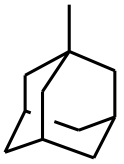
i) EtOH/ reflux/ 4-8 h/ 66-83%[38,103]
ii) EtOH/ AcOH/ reflux/ 3 h/ 66%[4]
iii) AcOH/ reflux/ 5 h/ 56-90%[39,102]
iv) AcOH/ stirred/ rt/ 3 h/ 68-78%[100,101]
v) [PyH]Cl/ EtOH/ 110 °C/ 20 min/ 74%[103]
vi) EtOH/ ultrasound/ 70 °C/ 2 h/ 83%[103]
vii) glycerol/ Lanthanum / chitosan/ 100 °C/ 2-5 h/ 62-72%[99]
viii) glycerol/ Lanthanum / chitosan/ MW/ 100 °C/ 5-10 min/ 90-96%[99]
ix) EtOH/ MW/ 15-25 min/ 71-92%[104]

Scheme 36

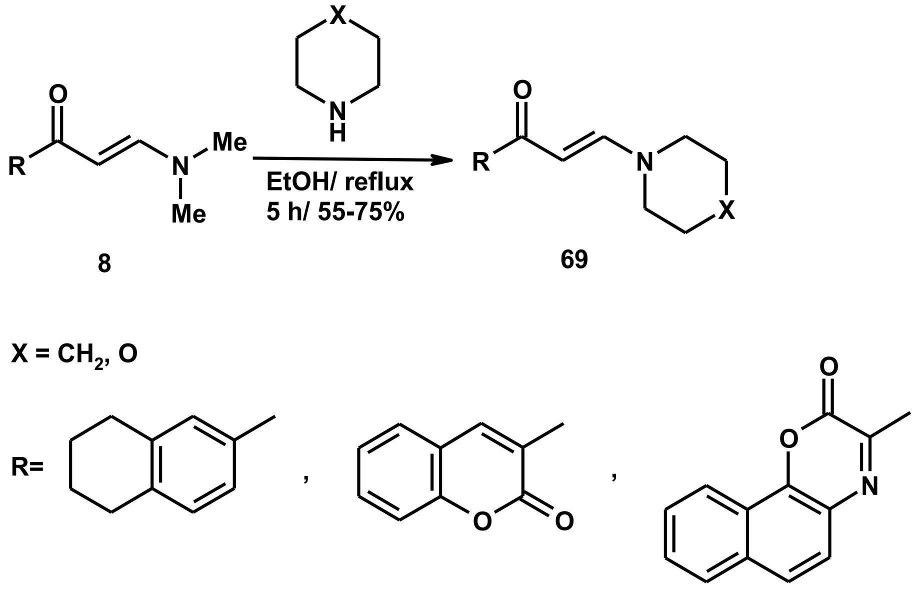
Table 6:

R	R^1	R^2	Ref
	H		[4]
	H		[98]
	H	 X= H, 4-Cl, 4-OH, 4-NO_2, 2-NO_2, 4-Me, 4-OMe	[99]
	H	 X= H, F, Cl, Br, Me, MeO, NO_2	[100,101]
	H	 X= H, 2-CO_2H	[39]

Continued Table 6

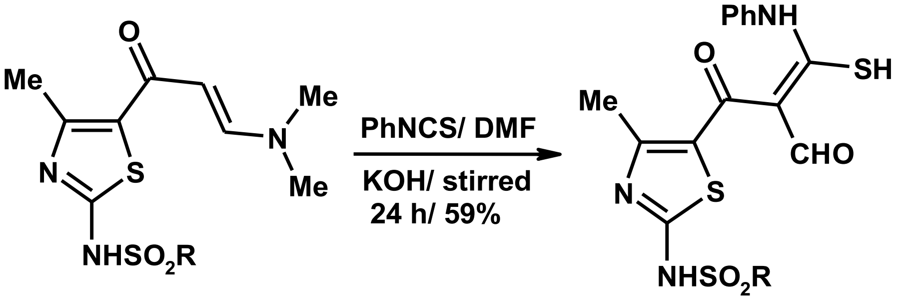
	H	 X= 2-NO$_2$, 4-NO$_2$	[102]
	H		[38]
	CN		[103]
 X= H, Cl, NO$_2$, Me, OMe	H		[104]

Also, enaminone **8** reacted with some secondary amines namely piperidine and morpholine, in refluxing ethanol, to yield 1-substituted-3-(piperidinyl) and/or (morpholinyl)prop-2-en-1-ones **69**, in 55-75% yields (Scheme 37).[39,100,102]



Scheme 37

When 3-(dimethylamino)-1-(4-methyl-2-(tosylamino)thiazol-5-yl)prop-2-en-1-one (**70**) was treated with phenyl isothiocyanate, in the presence of potassium hydroxide, 3-mercapto-2-[4-methyl-2-(tosylamino)thiazole-5-carbonyl]-3-phenylaminopropenal was obtained (**71**) (Scheme 38).[22]



70

R= 4-MePh

71

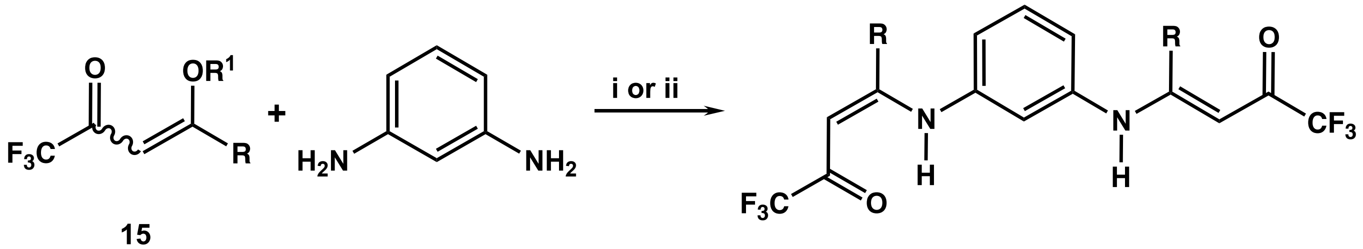
Scheme 38

3. SYNTHESIS OF *BIS*-ENAMINONES

In addition to the reported methods for the synthesis of enaminones, some *bis*-enaminones were also prepared as collected below.

3.1. From *β*-alkoxy ketones

N,N'-Bis{1-[aryl(heteroaryl)]-4,4,4-trifluoro-3-oxo-1-buten-1-yl}-1,3-phenylenediamines **72** was isolated, in 47-91% yields, by reaction of 4-alkoxy-4-(aryl/heteroaryl)-1,1,1-trifluoroalk-3-en-2-ones **15** with 1,3-phenylenediamine, at a molar ratio 2:1, under different reaction conditions (Scheme 39).[105]



15

72

i) EtOH/ stirred/ 40 °C/ 2 h/ 62-91%

ii) H_2O/ $CHCl_3$/ stirred/ 60 °C/ 4 h

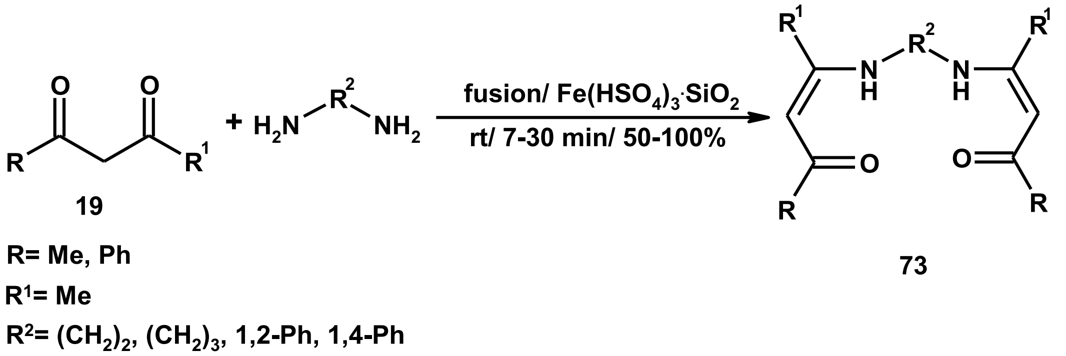
R= H, 2-furyl, Ph, 4-FPh, 4-ClPh, 4-BrPh,
4-NO_2Ph, 4-MePh, 4-MeOPh

R^1= Me, Et

Scheme 39

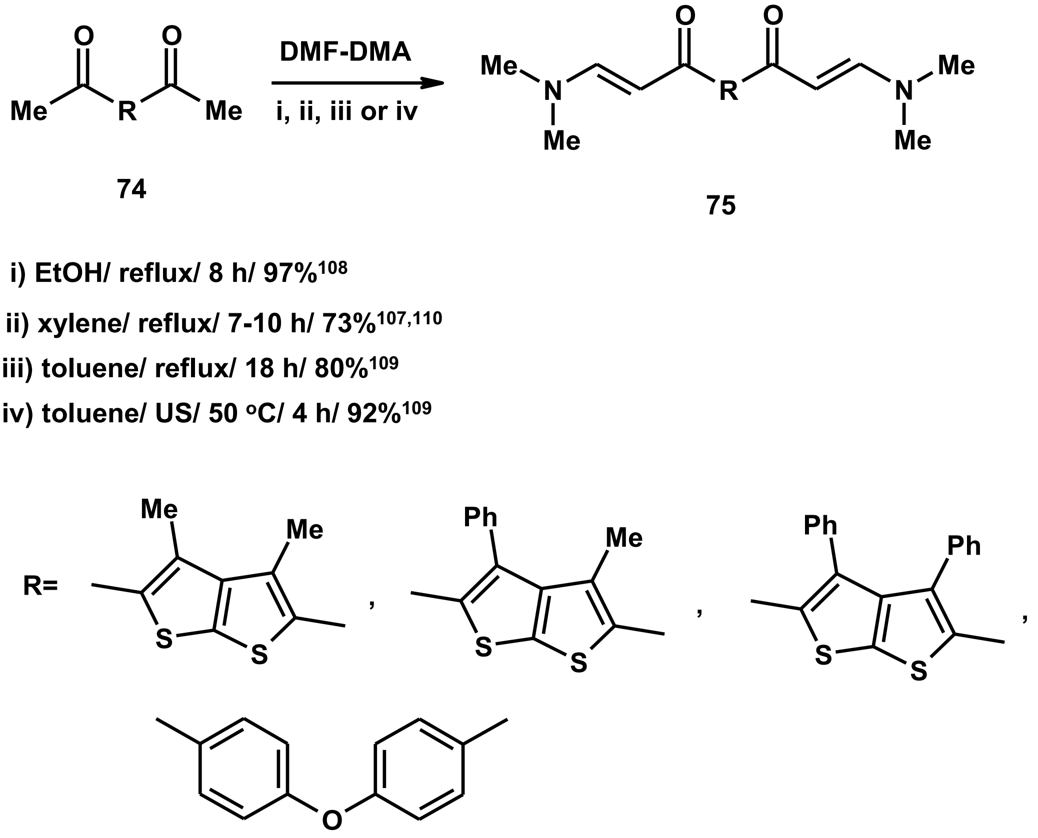
3.2. From *β*-diketones

Condensation of diamines with dicarbonyl compounds **19** under solvent-free conditions, in the presence of silica ferric hydrogen sulfate [$Fe(HSO_4)_3 \cdot SiO_2$] as a catalyst, afforded the corresponding *bis*-enaminones **73**, in good to excellent yields (Scheme 40).[106]



Scheme 40

Reaction of diethanone derivatives **74** with DMF-DMA, under different reaction conditions (in refluxing solvent or under ultrasonic irradiations), gave the corresponding *bis*-enaminone derivatives **75** (Scheme 41).[107-110]

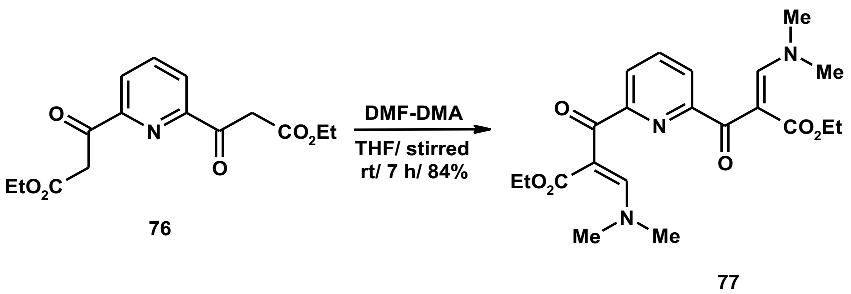


i) EtOH/ reflux/ 8 h/ 97%[108]
ii) xylene/ reflux/ 7-10 h/ 73%[107,110]
iii) toluene/ reflux/ 18 h/ 80%[109]
iv) toluene/ US/ 50 °C/ 4 h/ 92%[109]

Scheme 41

3.3. From *β*-ketoesters

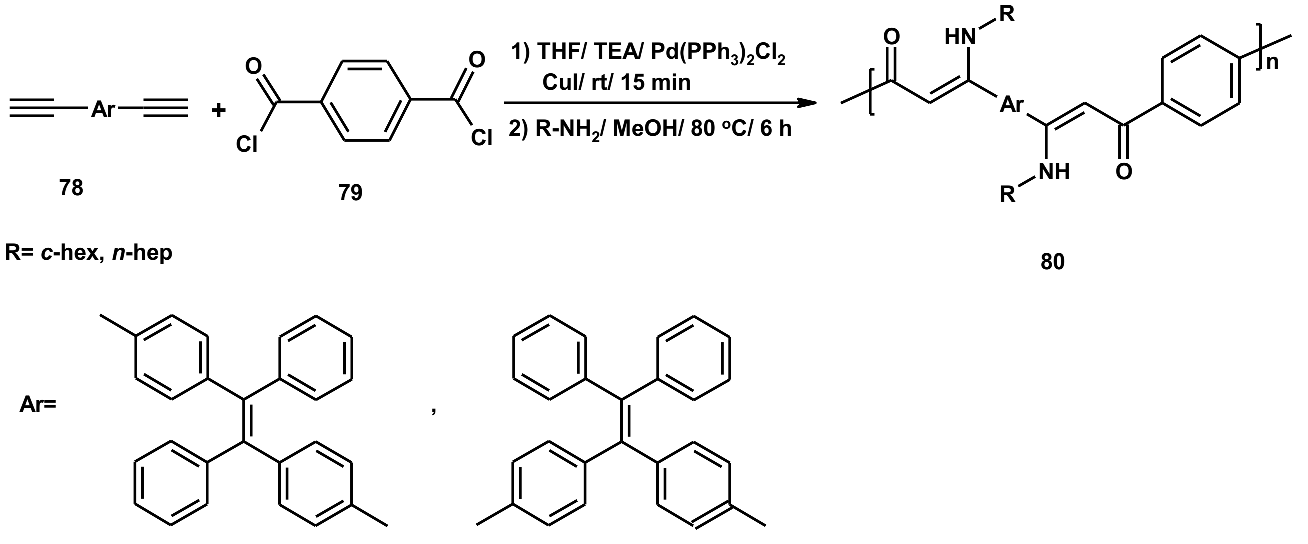
Pyridine-2,6-*bis*[ethyl 2-(*N*,*N*-dimethylamino)methylene-3-oxopropanoate] (**77**) was prepared by the reaction of pyridine-2,6-*bis*(ethyl 3-oxopropanoate) (**76**) with 1,1-dimethoxytrimethylamine, in dry THF (Scheme 42).[111]



Scheme 42

3.4. From acyl chloride

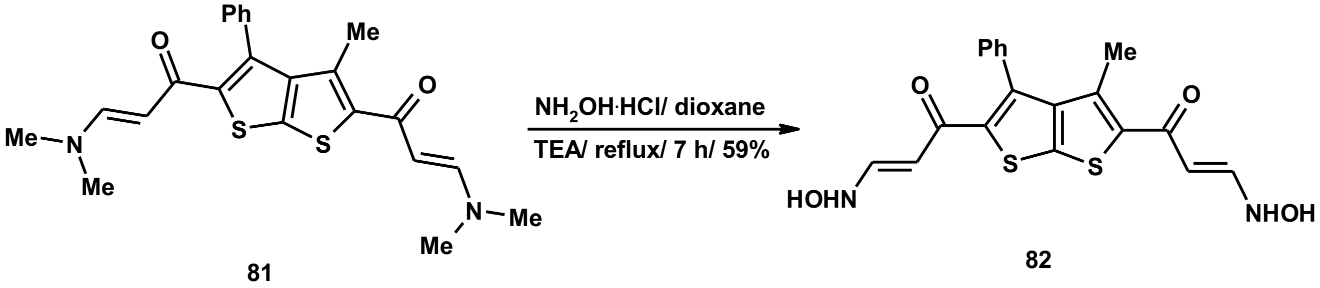
The sequential polymerizations of tetraphenylethene (TPE)-containing diynes **78**, terephthaloyl dichloride (**79**) and primary amines furnished conjugated polymers **80**. This polymerization was carried out in THF under nitrogen in the presence of $Pd(PPh_3)_2Cl_2$, CuI and TEA as depicted in Scheme 43.[112]



Scheme 43

3.5. From other enaminones

1,1'-(3-Methyl-4-phenylthieno[2,3-*b*]thiophene-2,5-diyl)*bis*(3-(hydroxyamino)-prop-2-en-1-one) (**82**) were prepared by reaction of enaminone **81** with hydroxylamine hydrocholoride, in refluxing dioxane (Scheme 44).[107]



Scheme 44

4. CONCLUSION

In conclusion, a variety of enaminones and *bis*-enaminones linked a numerous substituents were efficiently obtained starting from a diversity of substrates and reagents. The methods developed to synthesize these enaminones were collected according to the synthons employed.

5. REFERENCES

1. C. Kouklovsky and G. Vincent, 'Comprehensive Organic Synthesis II: Reduction of N=N, N–N, N–O, and O–O Bonds', Vol. 8. ed. by P. Knochel and G. A. Molander, Elsevier, Amsterdam, 2014, pp. 493-534.
2. H. M. E. Hassaneen, *Molecules*, 2011, **16**, 609.
3. P. Subramammiam, C. Ramasubbu, S. Athiramu, S. Arumugam, and M. Alagumuthu*, Arch. Pharm. Chem. Life Sci.*, 2019, **352**, 1800244.
4. S. A. Al-Harbi and M. S. Bashandy, *Heterocycles*, 2015, **91**, 1905.
5. K. M. Al-Zaydi, R. A. Mekheimer, S. M. Mousally, R. M. Borik, and M. H. Elnagdi, *Arab. J. Chem.*, 2017, **10**, S2697.
6. I. J. Amaye, R. D. Haywood, E. M. Mandzo, J. J. Wirick, and P. L. Jackson-Ayotunde, *Tetrahedron*, 2021, **83**, 131984.
7. R. Kumar, N. Saha, P. Purohit, S. K. Garg, K. Seth, V. S. Meena, S. Dubey, K. Dave, R. Goyal, S. S. Sharma, U. C. Banerjee, and A. K. Chakraborti, *Eur. J. Med. Chem.*, 2019, **182**, 111601.
8. S. Akhramez, A. Oumessaoud, A. Hibot, S. Talbi, S. Hamri, E. M. Ketatni, H. Ouchetto, A. Hafid, H. Ayad, N. El Abbadi, C. Zanane, H. Latrache, M. Khouili, and M. D. Pujol, *Arab. J. Chem.*, 2022, **15**, 103527.
9. I. J. Amaye, T. Harper, and P. L. Jackson-Ayotunde, *J. Fluorine Chem.*, 2021, **251**, 109886.
10. H. A. El-Ghamry, K. Yamauchi, K. Sakai, and T. A. Farghaly, *Inorganica Chim. Acta*, 2021, **516**, 120117.
11. P. Subramamiam, C. Ramasubbu, S. Athiramu, S. Arumugam, and M. Alagumuthu, *Arch. Pharm.*, 2019, **352**, 1800244.

12. S. Ueno, K. Usui, and R. Kuwano, *Synlett*, 2011, 1303.
13. D. Yu, Y. N. Sum, A. C. C. Ean, M. P. Chin, and Y. Zhang, *Angew. Chem. Int. Ed.*, 2013, **52**, 5125.
14. L. Tang, Y. Zou, K. Zhong, and Y. A. Bian, *RSC Adv.*, 2016, **6**, 48351.
15. W. Luo, Q. Mu, W. Qiu, T. Liu, F. Yang, X. Liu, and J. Tang, *Tetrahedron*, 2011, **67**, 7090.
16. B. Prek, J. Bezenšek, M. Kasunič, U. Grošelj, J. Svete, and B. Stanovnik, *Tetrahedron*, 2014, **70**, 2359.
17. Y. M. Mabkhot, N. A. Kaal, S. Alterary, S. S. Al-Showiman, A. Barakat, H. A. Ghabbour, and W. Frey, *Molecules*, 2015, **20**, 8712.
18. S. M. Al-Mousawi, M. A. El-Apasery, and M. H. Elnagdi, *Molecules*, 2010, **15**, 58.
19. S. M. Riyadh, *Molecules*, 2011, **16**, 1834.
20. S. Kantevari and S. R. Putapatri, *Synlett*, 2010, 2251.
21. S. Shetty, A. Asma, and B. Kalluraya, *Indian J. Chem.*, 2016, **55B**, 501.
22. K. M. Dawood and S. M. Sayed, *J. Heterocycl. Chem.*, 2016, **53**, 1950.
23. S. M. Gomha and H. A. Abdel-Aziz, *Heterocycles*, 2012, **85**, 2291.
24. R. Determann, J. Dreher, K. Baumann, L. Preu, P. G. Jones, F. Totzke, C. Schächtele, M. H. G. Kubbutat, and C. Kunick, *Eur. J. Med. Chem.*, 2012, **53**, 254.
25. H. M. Hassaneen, A. S. Shawali, A. Tayseer T. A. Abdallah, and F. M. Saleh, *ARKIVOC*, 2014, **vi**, 156.
26. T. A. Farghaly, M. A. Abdallah, and Z. A. Muhammad, *Molecules*, 2011, **16**, 10420.
27. M. M. Abdellall, *Orient. J. Chem.*, 2014, **30**, 1099.
28. M. A. Raslan and O. A. Omran, *J. Heterocycl. Chem.*, 2016, **53**, 1121.
29. A. S. Shawali, S. M. Sherif, M. A. A. Darwish, and M. M. El-merzabani, *Arch. Pharm. Res.*, 2010, **33**, 55.
30. T. Sun, J. Xu, M. Ji, and P. Wang, *J. Chem. Res.*, 2016, **40**, 511.
31. B. V. Kendre, M. G. Landge, W. N. Jadhav, and S. R. Bhusare, *Chin. Chem. Lett.*, 2013, **24**, 325.
32. H. A. Abdel-Aziz, T. S. Saleh, and H. S. A. El-Zahabi, *Arch. Pharm. Chem. Life Sci.*, 2010, **343**, 24.
33. M. F. Mady, T. S. Saleh, A. A. El-Kateb, N. M. Abd El-Rahman, and S. I. Abd El-Moez, *Res. Chem. Intermed.*, 2015, **42**, 753.
34. O. Bouaziz, F. Abboub, N. Bayou-Khier, M. Fodili, P. Hoffmann, and M. Amari, *Compt. Rend. Chim.*, 2012, **15**, 774.
35. Z. N. Siddiqui and F. Farooq, *J. Mol. Catal. A: Chem.*, 2012, **363**, 451.
36. H. M. Ibrahim and H. Behbehani, *Molecules*, 2014, **19**, 2637.
37. S. Islam and S. M. Firestine, *J. Heterocycl. Chem.*, 2013, **50**, 1043.
38. N. A. Hamdy, W. M. El-Senousy, and I. M. I. Fakhra, *J. Heterocycl. Chem.*, 2013, **50**, 337.
39. I. H. El-Azab, L. M. Break, and Z. A. A. El-Zahrani, *Orient. J. Chem.*, 2016, **32**, 2435.

40. F. M. A. Al-Talbawy, M. H. Abdel-Kader, E. S. S. Darwish, and M. H. El-nagdi, *Asian J. Chem*., 2016, **28**, 2303.

41. K. Xu, Z. Zhang, P. Qian, Z. Zha, and Z. Wang, *Chem. Commun*., 2015, **51**, 11108.

42. T. S. Saleh, M. A. Al-Omar, and H. A. Abdel-Aziz, *Lett. Org. Chem*., 2010, **7**, 483.

43. H. A. Abdel-Aziz, T. S. Saleh, and H. S. A. El-Zahabi, *Arch. Pharm. Chem. Life Sci*., 2010, **343**, 24.

44. Z. Kibou, N. Cheikh, D. Villemin, and N. Choukchou-Braham, *J. Mater. Environ. Sci*., 2016, **7**, 3061.

45. D. L. Wang, Q. T. Cui, S. S. Feng, and J. Y. Yu, *Heterocycles*, 2012, **85**, 697.

46. R. A. Bunce, E. J. Lee, and M. T. Grant, *J. Heterocycl. Chem*., 2011, **48**, 620.

47. A. V. Jerezano, E. M. Labarrios F. E. Jiménez, M. D. C. Cruz, D. C. Pazos, R. U. Gutiérrez, F. Delgado, and J. Tamariz, *ARKIVOC*, 2014, **iii**, 18.

48. F. M. Abdelrazek, M. S. Farghaly, and H. E. Abdelrahman, *J. Heterocycl. Chem*., 2015, **52**, 163.

49. E. Labarrios, A. Jerezano, F. Jiménez, M. D. C. Cruz, F. Delgado, L. G. Zepeda, and J. Tamariz, *J. Heterocycl. Chem*., 2014, **51**, 954.

50. R. E. Khidre, I. A. M. Radini, and D. A. Ibrahim, *ARKIVOC*, 2016, **v**, 301.

51. R. A. Bunce and B. Nammalwar, *J. Heterocycl. Chem*., 2012, **49**, 658.

52. Y. W. Ho and M. C. Suen, *J. Chem*., 2013, **2013**, 765243.

53. M. Ghandi and A. H. Jameà, *Tetrahedron Lett*., 2011, **52**, 4005.

54. N. Chopin, M. Médebielle, O. Maury, G. Novitchi, and G. Pilet, *Eur. J. Inorg. Chem*., 2014, 6185.

55. N. Chopin, M. Médebielle, and G. Pilet, *J. Fluorine Chem.*, 2013, **155**, 89.

56. N. Chopin, S. Decamps, A. Gouger, M. Médebielle, S. Picot, A.S. Bienvenu, and G. Pilet, *J. Fluorine Chem*., 2011, **132**, 850.

57. N. Chopin, G. Pilet, Y. Morita, and M. Médebielle, *J. Fluorine Chem*., 2014, **167**, 211.

58. M, A. P. Martins, M. Rossatto, L. D. T. Prola, L. Pizzuti, D. N. Moreira, P. T. Campos, C. P. Frizzo, N. Zanatta, and H. G. Bonacorso, *Ultrason. Sonochem*., 2012, **19**, 227.

59. R. V. Edwankar, C. R. Edwankar, O. A. Namjoshi, J. R. Deschamps, and J. M. Cook, *J. Nat. Prod*., 2012, **75**, 181.

60. A. Singh, N. Gupta, M. L. Sharma, and J. Singh, *Indian J. Chem*., 2014, **53B**, 900.

61. Q. Yiping, W. Longxiang, Z. Xiaoyu, F. Xiujuan, and B. Ming, *Chin. J. Org. Chem*., 2013, **33**, 860.

62. C. R. Winck, M. P. Darbem, R. S. Gomes, A. W. Rinaldi, and N. L. C. Domingues, *Tetrahedron Lett*., 2014, **55**, 4123.

63. R. Murugavel, M. P. Singh, and M. Nethaji, *J. Chem. Crystallogr*., 2012, **42**, 12.

64. Z. H. Zhang, T. S. Li, and J. J. Li, *Catal. Commun*., 2007, **8**, 1615.

65. E. Rafiee, H. Mahdavi, S. Eavani, M. Joshaghani, and F. Shiri, *Appl. Catal., A*, 2009, **352**, 202.

66. H. Eshghi, S. M. Seyedi, E. Safaei, M. Vakili, A. Farhadipour, and M. Bayat-Mokhtari, *J. Mol. Catal.*

A: Chem., 2012, **363**, 430.

67. S. A Patil, P. A. Medina, D. Gonzalez-Flores, J. K. Vohs. S. Dever, L. W. Pineda, M. L. Montero, and B. D. Fahlman, *Synth. Commun.*, 2013, **43**, 2349.
68. R. A. Laskar, N. A. Begum, M. H. Mir, S. Ali, and A. T. Khan, *Tetrahedron Lett.*, 2013, **54**, 436.
69. H. Doušová, P. Šimůnek, N. Almonasy, and Z. Růžičková, *J. Organomet. Chem.*, 2015, **802**, 60.
70. R. D. Vreese, C. Grootaert, S. D'hoore, A. Theppawong, S. V. Damme, M. V. Bogaert, J. V. Camp, and M. D'hooghe, *Eur. J. Med. Chem.*, 2016, **123**, 727.
71. W. Liu, C. Chen, and Q. Zhang, *J. Chem. Res.*, 2012, **36**, 175.
72. K. D. Bhatte, P. J. Tambade, K. P. Dhake, and B. M. Bhanage, *Catal. Commun.*, 2010, **11**, 1233.
73. M. Kidwai, S. Bhardwaj, N. K. Mishra, V. Bansal, A. Kumar, and S. Mozumdar, *Catal. Commun.*, 2009, **10**, 1514.
74. A. L. Gajengi, T. Sasaki, and B. M. Bhanage, *RSC Adv.*, 2016, **6**, 101800.
75. T. Mahmud, R. Rehman, A. Gulzar, A. Khalid, J. Anwar, U. Shafique, W. Zaman, and M. Salman, *Arab. J. Chem.*, 2010, **3**, 219.
76. S. Celedon, M. Fuentealba, T. Roisnel, J. R. Hamon, D. Carrillo, and C. Manzur, *Inorganica Chim. Acta*, 2012, **390**, 184.
77. P. Angelov, *Synlett*, 2010, 1273.
78. M. Mamboury, Q. Wang, and J. Zhu, *Chem. Eur. J.*, 2017, **23**, 12744.
79. L. Mohammadi, M. A. Zolfigol, M. Ebrahiminia, K. P. Roberts, S. Ansari, T. Azadbakht, and S. R. Hussaini, *Catal. Commun.*, 2017, **102**, 44.
80. L. Shi, L. Xue, R. Lang, C. Xia, and F. Li, *ChemCatChem*, 2014, **6**, 2560.
81. I. H. Jeong, S. L. Jeon, Y. K. Min, and B. T. Kim, *Tetrahedron Lett.*, 2002, **43**, 7171.
82. R. N. Straker, M. K. Majhail, and M. C. Willis, *Chem. Sci.*, 2017, **8**, 7963.
83. E. Gayon, M. Szymczyk, H. Gérard, E. Vrancken, and J. M. Campagne, *J. Org. Chem.*, 2012, **77**, 9205.
84. M. Li, D. Fang, F. Geng, and X. Dai, *Tetrahedron Lett.*, 2017, **58**, 4747.
85. W. Shi, S. Sun, M. Wu, B. Catano, W. Li, J. Wang, H. Guo, and Y. Xing, *Tetrahedron Lett.*, 2015, **56**, 468.
86. M. J. Adler and A. D. Hamilton, *J. Org. Chem.*, 2011, **76**, 7040.
87. I. Arellano, P. Sharma, L. Rubio-Perez, A. Cabrera, N. Rosas, and A. Toscano, *J. Organomet. Chem.*, 2012, **700**, 29.
88. S. Karabiyikoglu, Y. Kelgokmen, and M. Zora, *Tetrahedron*, 2015, **71**, 4324.
89. L. Šenica, U. Grošelj, M. Kasunič, D. Kočar, B. Stanovnik, and J. Svete, *Eur. J. Org. Chem.*, 2014, 3067.
90. B. Prek, J. Bezenšek, and B. Stanovnik, *Tetrahedron*, 2017, **73**, 5260.

91. J. I. Yamaguchi and S. Sugiyama, *Tetrahedron Lett.*, 2016, **57**, 4514.
92. X. Yu, L. Wang, X. Feng, M. Bao, and Y. Yamamoto, *Chem. Commun.*, 2013, **49**, 2885.
93. X. Cheng, S. Pei, C. Xue, K. Cao, L. Hai, and Y. Wu, *RSC Adv.*, 2014, **4**, 63897.
94. M. Bararjanian, S. Balalaie, F. Rominger, B. Movassagh, and H. R. Bijanzadeh, *Mol. Divers.*, 2011, **15**, 583.
95. Z. Yang, B. Jiang, W. J. Hao, P. Zhou, S. J. Tu, and G. Li, *Chem. Commun.*, 2015, **51**, 1267.
96. T. Miura, Y. Funakoshi, M. Morimoto, T. Biyajima, and M. Murakami, *J. Am. Chem. Soc.*, 2012, **134**, 17440.
97. M. A. Ibrahim and Y. A. Alnamer, *J. Heterocycl. Chem.*, 2019, **56**, 2341.
98. C. M. Park and D. J. Jeon, *Org. Biomol. Chem.*, 2012, **10**, 2613.
99. S. Tarannum and Z. N. Siddiqui, *Appl. Organomet. Chem.*, 2016, **30**, 473.
100. A. M. Gamal-Eldeen, N. A. Hamdy, H. A. Abdel-Aziz, E. A. El-Hussieny, and I. M. I. Fakhr, *Eur. J. Med. Chem.*, 2014, **77**, 323.
101. N. A. Hamdy and W. M. El-Senousy, *Acta Pol. Pharm.*, 2013, **70**, 99.
102. I. H. El Azab and K. M. Khaled, *Russ. J. Bioorg. Chem.*, 2015, **41**, 421.
103. K. M. Al-Zaydi, R. A. Mekheimer, S. M. Mousally, R. M. Borik, and M. H. Elnagdi, *Arab. J. Chem.*, 2017, **10**, S2697.
104. U. Kalita, S. Kaping, R. Nongkynrih, M. Sunn, I. Boiss, L. I. Singha, and J. N. Vishwakarma, *Med. Chem. Res.*, 2015, **24**, 32.
105. H. G. Bonacorso, R. Andrighetto, N. Krüger, M. A. P. Martins, and N. Zanatta, *J. Braz. Chem. Soc.*, 2011, **22**, 1426.
106. H. Eshghi, E. Safaei, S. M. Seyedi, and T. Eshghi, *Compt. Rend. Chim.*, 2014, **17**, 1088.
107. Y. N. Mabkhot, A. Barakat, and S. Alshahrani, *J. Mol. Struct.*, 2012, **1027**, 15.
108. Y. N. Mabkhot, A. M. Al-Majid, and A. S. Alamary, *Molecules*, 2011, **16**, 7706.
109. A. E. M. Mekky and A. S. Al-Bogami, *J. Heterocycl. Chem.*, 2016, **53**, 595.
110. Y. N. Mabkhot, F. D. Aldawsari, S. S. Al-Showiman, A. Barakat, S. M. Soliman, M. I. Choudhary, S. Yousuf, M. S. Mubarak, and T. B. Hadda, *Chem. Cent. J.*, 2015, **9**, 24.
111. K. A. Ali, M. A. Elsayed, and A. M. Farag, *Heterocycles*, 2012, **85**, 1913.
112. H. Deng, R. Hu, A. C. S. Leung, E. Zhao, J. W. Y. Lam, and B, Z. Tang, *Polym. Chem.*, 2015, **6**, 4436.

Al shimaa Badran Abdel Monem was born in Cairo, Egypt, in 1987. She received his B.Sc. degree in chemistry (2008) and her M.Sc. degree in organic chemistry (2013). Also, she received her Ph.D. degree (2016) from the organic division, Department of Chemistry, Faculty of Education, Ain Shams University. In 2021 she was promoted to assistant professor in organic chemistry. Her research is focused on the synthesis of new heterocyclic compounds containing chromone, γ-pyrone and quinolin-2(1*H*)-one moiety.

Najla Alshaye was born in Riyadh, Saudi Arabia in 1975. She received her BSc degree in chemistry (1997) from the College of Science and Education, Riyadh, Saudi Arabia. She also received her MSc (2008) and PhD (2012) from the Department of Chemistry of the Faculty of Science and Engineering at the University of Hull, Hull, United Kingdom. Her research focuses on stereochemistry and synthesis of heterocyclic compounds. Since 2013 she has been working as an associate Professor at Princesses Nourah Bint Abdulrahman University, Riyadh, Saudi Arabia.

Zeinab Hussain Foda Khedr was born in Cairo, Egypt, in 1995. She received his B.Sc. degree in chemistry (2017) Department of Chemistry, Faculty of Education, Ain Shams University. Her research is focuse on synthesis of new heterocyclic compounds containing pyrimidine nucleus.

Aya Ahmed Hamed Shahat was born in Cairo, Egypt, in 1994. She received his B.Sc. degree in chemistry (2017) Department of Chemistry and her M.Sc. degree in organic chemistry (2021), Faculty of Education, Ain Shams University. Her research is focuse on synthesis of new heterocyclic compounds containing chromone nucleus.

Magdy Ahmed Mohamed Ibrahim was born in Cairo, Egypt, in 1977. He received his B.Sc. degree in physics and chemistry (1999) and his M.Sc. degree in organic chemistry (2004). Also, he received his Ph.D. degree (2007) from the organic division, Department of Chemistry, Faculty of Education, Ain Shams University. He won the award for the best research article in heterocyclic chemistry field at the Egyptian universities and research centers in 2010. In 2012 he was promoted to assistant professor in organic chemistry. In 2017 he was promoted to professor in organic chemistry. His research is focused on the synthesis of new heterocyclic compounds containing chromone, pyridine, pyrimidine, quinolin-2(1*H*)-one and 1,2,4-triazine. He published more than hundred scientific papers in the field of organic and theoretical chemistry. In addition, He has published about 130 scientific papers in international scientific journals. He was included in world's top 2% scientific list Stanford university in 2021 and 2022.