

HETEROCYCLES, Vol. 86, No. 1, 2012, pp. 127 - 132. © 2012 The Japan Institute of Heterocyclic Chemistry
Received, 24th May, 2012, Accepted, 15th June, 2012, Published online, 21st June, 2012
DOI: 10.3987/COM-12-S(N)21

A CONCISE SYNTHESIS OF THE AB-RING FRAGMENT OF (-)-GAMBIEROL

Haruhiko Fuwa,* Kazuaki Hirota, and Makoto Sasaki*

Graduate School of Life Sciences, Tohoku University, 2-1-1 Katahira, Aoba-ku,
Sendai 980-8577, Japan. Email: hfuwa@bios.tohoku.ac.jp;
masasaki@bios.tohoku.ac.jp

Abstract – We describe herein a concise synthesis of the AB-ring fragment of gambierol, wherein silver(I) trifluoromethanesulfonate-catalyzed 6-*endo* cyclization of a hydroxy ynone was exploited for the formation of the A-ring.

This paper is dedicated to Professor Ei-ichi Negishi on the occasion of his 77th birthday.

(-)-Gambierol (**1**, Figure 1) is a structurally complex marine polycyclic ether natural product that was isolated from the cultured cells of the ciguatera causative dinoflagellate *Gambierdiscus toxicus* by Satake and co-workers.¹ The entire structure of **1** was elucidated by extensive 2D-NMR analysis and application of a chiral anisotropic reagent. Satake et al. have reported that gambierol exhibits potent lethal toxicity against mice with a minimal lethal dose value of 50 µg/kg (ip) and that the neurological symptoms caused in mice resemble those shown by ciguatoxins, suggesting the possible role of **1** in ciguatera seafood poisoning. Unfortunately, detailed biological studies on **1** had been precluded for almost a decade, due to the extreme natural scarcity of **1**. Motivated by the structural complexity as well as the biological aspects, we have successfully completed the first total synthesis of **1** and enabled material supply for extensive biological investigations.^{2,3} Consequently, we have identified that **1** inhibits voltage-gated potassium channels (Kv channels) in mice taste cells in low nanomolar concentrations,⁴ while **1** acts as a weak partial agonist of voltage-gated sodium channels in human neuroblastoma cells.⁵ Snyders and co-workers have shown that **1** selectively inhibits the Kv1 and Kv3 subfamilies and that **1** binds to the previously undescribed binding site present between the S5 and S6 segments of Kv3.1 channels.⁶ In addition, we have recently found that the EFGH-ring domain of **1** plays a crucial role in inhibiting voltage-gated potassium currents and that, in an *in vitro* model of Alzheimer's disease, **1** and its truncated analogues are able to lower the amyloid β (Aβ) and

hyperphosphorylated tau levels with a possible implication of Kv channel inhibition.⁷ Undoubtedly, **1** represents an intriguing molecular probe for the functional analysis of Kv channels as well as for understanding downstream events of Kv channel inhibition.

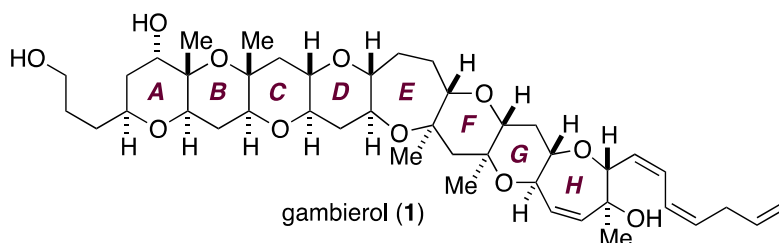
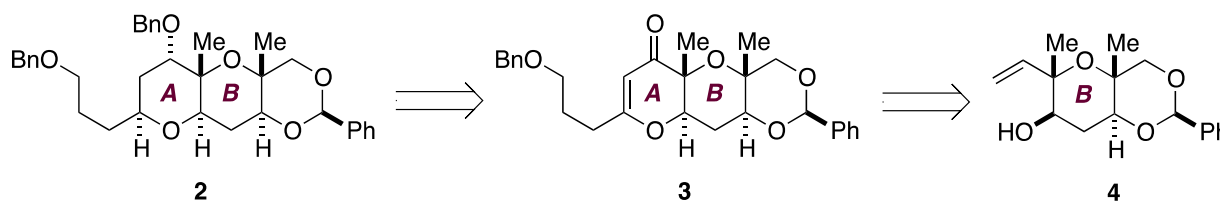


Figure 1. Structure of (-)-gambierol (**1**)

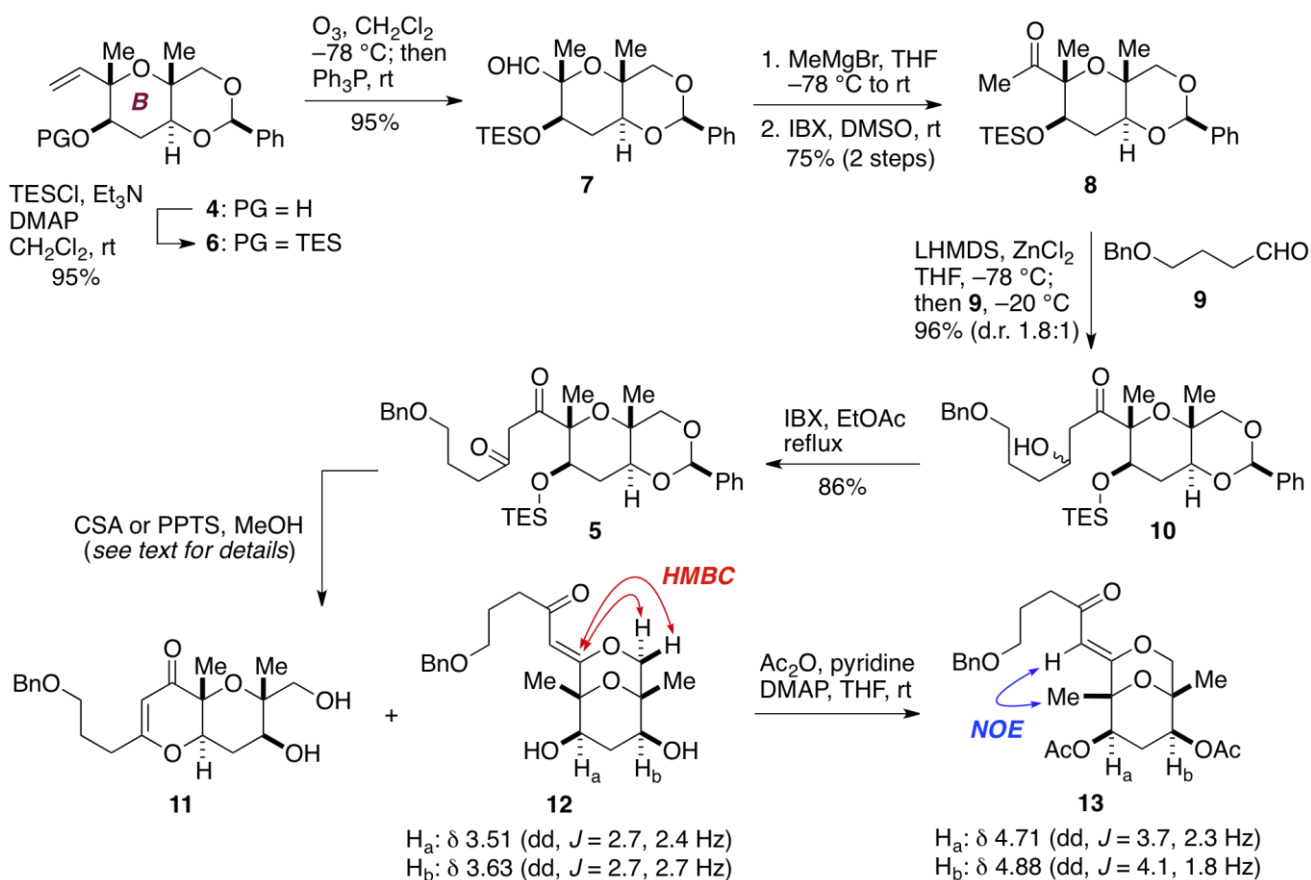
Toward an even more practical synthesis of **1**, we describe herein a concise synthesis of the AB-ring fragment of **1**. We envisioned that the AB-ring fragment **2** would be readily accessible from dihydropyrone **3**, which in turn would be obtainable from the known tetrahydropyran **4**⁸ via standard chemistry (Scheme 1).



Scheme 1. Synthesis plan toward the AB-ring fragment **2**

Our initial synthetic approach toward **2** involved cyclodehydration of 1,3-diketone **5** (Scheme 2).⁹ Protection of **4** with triethylsilyl chloride (TESCl) and Et₃N in the presence of a catalytic amount of 4-(dimethylamino)pyridine (DMAP) gave TES ether **6** in 95% yield. Ozonolysis of the double bond delivered aldehyde **7** in 95% yield. Methylation with MeMgBr, followed by oxidation of the resultant alcohol with 2-iodoxybenzoic acid (IBX), provided methyl ketone **8** in 75% yield (two steps). Enolization of **8** with lithium hexamethyldisilazide (LHMDS) in the presence of ZnCl₂ followed by addition of 4-benzyloxybutanal (**9**) (THF, -78 to -20 °C) afforded β-hydroxy ketone **10** as a 1.8:1 mixture of diastereomers in 96% combined yield (stereochemistry not determined). Oxidation of **10** was most efficiently performed with IBX in refluxing EtOAc¹⁰ to give 1,3-diketone **5** in 86% yield. Unexpectedly, cyclodehydration of **5** was found to be rather problematic. Treatment of **5** with (+)-10-camphorsulfonic acid (CSA) in methanol under reflux resulted in decomposition of the material, possibly due to the instability of the diketone moiety. Accordingly, **5** was reacted with CSA initially at room temperature to cleave the silyl ether and then at 60 °C to effect cyclodehydration. This modified procedure gave

dihydropyrone **11** with concomitant loss of the benzylidene acetal in 41% yield, along with a mixture of several byproducts. Careful examination of the mixture revealed that the major byproduct was bicycle **12**. Running the reaction with pyridinium *p*-toluenesulfonate (PPTS) as an acid only provided a mixture of **12** and unidentified byproducts. The structure of **12** was assigned on the basis of extensive NMR studies on **12** and its acetate **13** as shown. Upon cleavage of the benzylidene acetal, the steric repulsion existing between the 1,3-diaxial methyl groups of the tetrahydropyran would facilitate ring flipping, hemiacetal formation, and dehydration to produce **12**.

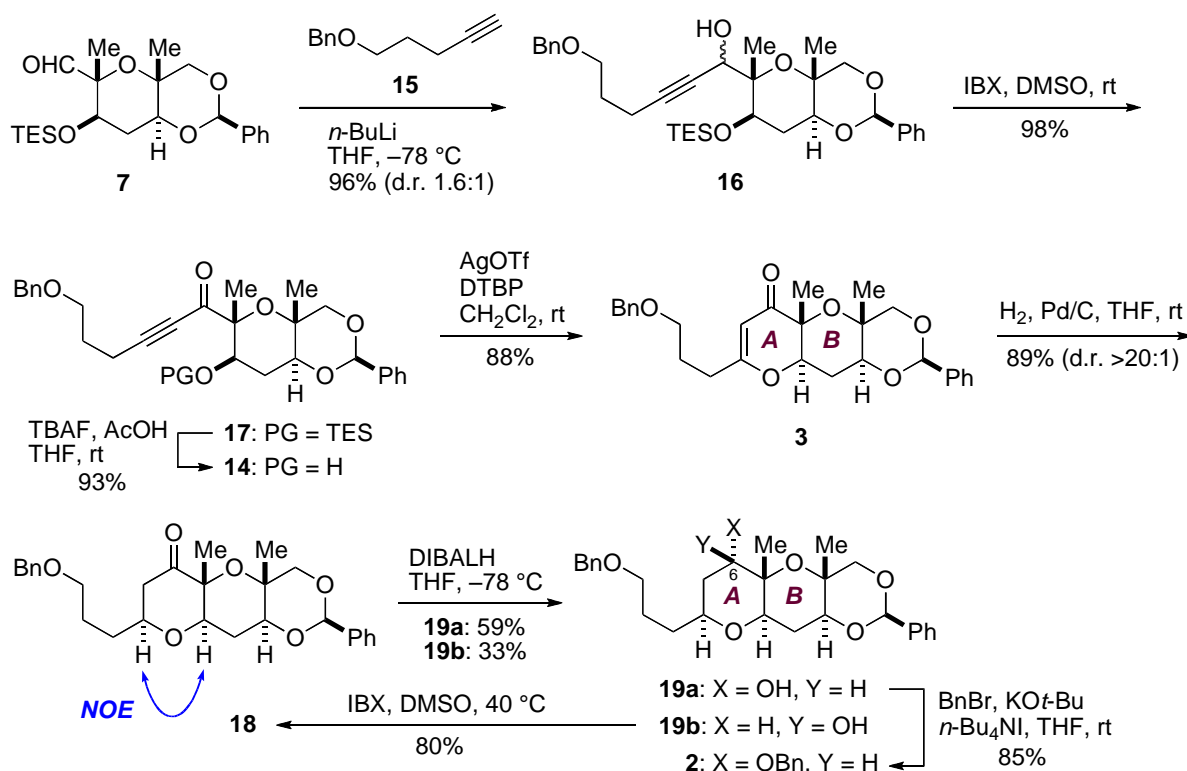


Scheme 2. Initial efforts toward the synthesis of the A-ring

The disappointing outcome of the cyclodehydration of 1,3-diketone **5** led us to investigate 6-*endo* cyclization of hydroxy ynone **14** as the revised approach toward dihydropyrone **3** (Scheme 3). We have previously shown that a variety of substituted dihydropyrones could be accessed based on silver(I) trifluoromethanesulfonate (AgOTf)-catalyzed cyclization of the corresponding hydroxy ynone.^{11,12} Addition of a lithium acetylide generated from 5-benzyloxypentyne (**15**)¹³ to aldehyde **7** gave propargylic alcohol **16** as a 1.6:1 mixture of diastereomers in 96% combined yield. Without separation, this material was oxidized with IBX to deliver ynone **17** in 98% yield. Removal of the TES group was effected by exposure to tetra-*n*-butylammonium fluoride (TBAF) buffered with acetic acid (93%). The resultant

hydroxy ynone **14** was cyclized smoothly upon treatment with AgOTf in the presence of 2,6-di-*tert*-butylpyridine (DTBP),¹⁴ giving rise to dihydropyrone **3** in 88% yield.

Having established the reliable route to **3**, our focus was shifted toward elaboration of the AB-ring fragment **2** (Scheme 3). Hydrogenation of **3** proceeded in a stereoselective manner to give ketone **18** in 89% yield as a single stereoisomer. The newly generated stereogenic center was established through an NOE experiment as shown. Stereoselective reduction of **18** to establish the C6 stereogenic center turned out to be a challenging task. Even bulky reducing agents such as diisobutylaluminum hydride (DIBALH) or L-selectride[®] provided the desired alcohol **19a** with moderate diastereoselectivity (**19a:19b** = 1.4–1.8:1), while reduction with NaBH₄ or LiAlH₄ resulted in preferential formation of the undesired alcohol **19b** (**19a:19b** = ca. 1:6). However, we were fortunate to find that these diastereomeric alcohols were readily separable by flash chromatography on silica gel, and the undesired **19b** could be recycled by oxidation with IBX (80%).¹⁵ Finally, benzylation of **19a** afforded bis(benzyl) ether **2** in 85% yield, successfully intercepting our previous synthesis of **1**.



Scheme 3. Synthesis of the AB-ring fragment **2** via 6-*endo* cyclization of hydroxy ynone **14**

In conclusion, we have devised a concise synthesis of the AB-ring fragment **2** of (–)-gambierol, by exploiting AgOTf-catalyzed 6-*endo* cyclization of hydroxy ynone **14**. The present synthesis allows an efficient access to **2** from the B-ring tetrahydropyran **4** in nine steps, which compares favorably with our previous synthesis (17 steps). Further studies on the chemistry and biology of (–)-gambierol and its synthetic analogues are currently underway and will be reported shortly.

ACKNOWLEDGEMENTS

This work was financially supported in part by Grants-in-Aid for Scientific Research on Innovative Areas “Chemical Biology Using Bioactive Natural Products as Specific Ligands” (Nos. 23102016 and 24102517) from the Ministry of Education, Culture, Sports, Science and Technology (MEXT).

REFERENCES AND NOTES

1. M. Satake, M. Murata, and T. Yasumoto, *J. Am. Chem. Soc.*, 1993, **115**, 361; A. Morohashi, M. Satake, and T. Yasumoto, *Tetrahedron Lett.*, 1999, **40**, 97.
2. H. Fuwa, M. Sasaki, M. Satake, and K. Tachibana, *Org. Lett.*, 2002, **4**, 2981; H. Fuwa, N. Kainuma, K. Tachibana, and M. Sasaki, *J. Am. Chem. Soc.*, 2002, **124**, 14983.
3. I. Kadota, H. Takamura, K. Sato, A. Ohno, K. Matsuda, and Y. Yamamoto, *J. Am. Chem. Soc.*, 2003, **125**, 46; I. Kadota, H. Takamura, K. Sato, A. Ohno, K. Matsuda, M. Satake, and Y. Yamamoto, *J. Am. Chem. Soc.*, 2003, **125**, 11893; H. W. B. Johnson, U. Majumder, and J. D. Rainier, *J. Am. Chem. Soc.*, 2005, **127**, 848; U. Majumder, J. M. Cox, H. W. B. Johnson, and J. D. Rainier, *Chem. Eur. J.*, 2006, **12**, 1736; H. W. B. Johnson, U. Majumder, and J. D. Rainier, *Chem. Eur. J.*, 2006, **12**, 1747; H. Furuta, Y. Hasegawa, and Y. Mori, *Org. Lett.*, 2009, **11**, 4382; H. Furuta, Y. Hasegawa, M. Hase, and Y. Mori, *Chem. Eur. J.*, 2010, **16**, 7586.
4. V. Ghiaroni, M. Sasaki, H. Fuwa, G. P. Rossini, G. Scalera, T. Yasumoto, P. Pietra, and A. Bigiani, *Toxicol. Sci.*, 2005, **85**, 657; V. Ghiaroni, H. Fuwa, M. Inoue, M. Sasaki, K. Miyazaki, M. Hirama, T. Yasumoto, G. P. Rossini, G. Scalera, and A. Bigiani, *Chem. Senses*, 2006, **31**, 673.
5. M. C. Louzao, E. Cagide, M. R. Vieytes, M. Sasaki, H. Fuwa, T. Yasumoto, and L. M. Botana, *Cell. Physiol. Biochem.*, 2006, **17**, 257.
6. E. Cuypers, Y. Abdel-Mottaleb, I. Kopljar, J. D. Rainier, A. L. Raes, D. J. Snyders, and J. J. Tytgat, *Toxicol.*, 2008, **51**, 974; I. Kopljar, A. J. Labro, E. Cuypers, H. W. B. Johnson, J. D. Rainier, J. Tytgat, and D. J. Snyders, *Proc. Natl. Acad. Sci. U. S. A.*, 2009, **106**, 9896.
7. E. Alonso, H. Fuwa, C. Vale, Y. Suga, T. Goto, Y. Konno, M. Sasaki, F. M. LaFerla, M. R. Vieytes, L. Giménez-Llort, and L. M. Botana, *J. Am. Chem. Soc.*, 2012, **134**, 7467.
8. K. C. Nicolaou, D. A. Nugiel, E. Couladouros, and C.-K. Hwang, *Tetrahedron*, 1990, **46**, 4517.
9. K. Tsubone, K. Hashizume, H. Fuwa, and M. Sasaki, *Tetrahedron*, 2011, **67**, 6600.
10. J. D. More and N. S. Finney, *Org. Lett.*, 2002, **4**, 3001.
11. H. Fuwa, S. Matsukida, and M. Sasaki, *Synlett*, 2010, 1239; H. Fuwa, K. Mizunuma, S. Matsukida, and M. Sasaki, *Tetrahedron*, 2011, **67**, 4995.
12. For other reports on the AgOTf-mediated 6-*endo* cyclization of hydroxy ynones, see: S.-L. Shi, M. Kanai, and M. Shibasaki, *Angew. Chem. Int. Ed.*, 2012, **51**, 3932; C. R. Reddy and B. Srikanth, *Synlett*,

- [2010, 1536](#); K. C. Nicolaou, M. O. Frederick, A. C. B. Burtoloso, R. M. Denton, F. Rivas, K. P. Cole, R. J. Aversa, R. Gibe, T. Umezawa, and T. Suzuki, *J. Am. Chem. Soc.*, **2008**, **130**, 7466; C. Wang and C. J. Forsyth, *Org. Lett.*, **2006**, **8**, 2997.
13. Alkyne **15** was prepared from aldehyde **9** in two steps ((i) CBr₄, Ph₃P, CH₂Cl₂, 0 °C; (ii) *n*-BuLi, THF, -78 °C, 96% yield for the two steps).
 14. Addition of DTBP was found to be essential for obtaining **3** in excellent yield.
 15. All attempts at inverting the C6 stereogenic center by means of Mitsunobu reaction were unfruitful (*p*-NO₂C₆H₄CO₂H, Ph₃P, diethyl azodicarboxylate, toluene or THF, reflux or *p*-MeOC₆H₄CO₂H, *n*-Bu₃P, *N,N,N',N'*-tetramethyl azodicarboxamide,¹⁶ benzene, reflux).
 16. T. Tsunoda, J. Otsuka, Y. Yamamiya, and S. Itô, *Chem. Lett.*, **1994**, 539.