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## TOTAL SYNTHESIS OF MARINE SESQUITERPENOID SINULARIANIN B AND 8-EPI-SINULARIANIN B

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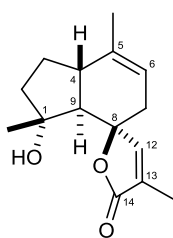
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Dedicated to Professor Dr. Isao Kuwajima on his 77<sup>th</sup> birthday

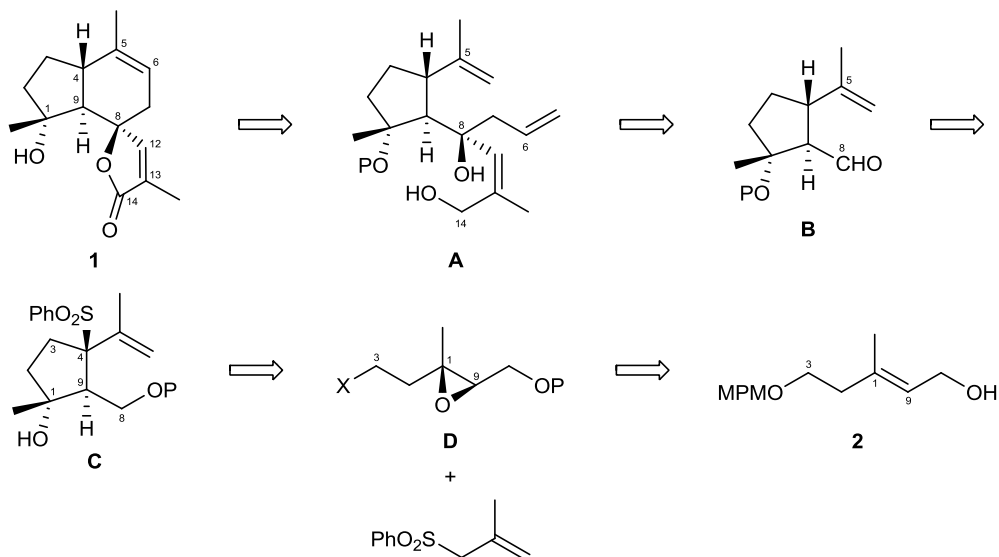
**Abstract** – The enantioselective total synthesis of marine sesquiterpenoid sinularianin B, isolated from the Formosan soft coral *Sinularia* sp., has been accomplished in 16 steps based on a highly concise synthetic strategy. The synthesis features two highly stereoselective sulfone-mediated reactions, including a key tandem intermolecular–intramolecular alkylation developed in our laboratory, and a palladium-catalyzed desulfonylation of an allyl sulfone through  $\pi$ -allyl palladium complex formation.

### INTRODUCTION

Sinularianin B (**1**) is a structurally unique valerenane sesquiterpenoid first isolated in 2006 from extracts of the Formosan soft coral *Sinularia* sp. by Sheu and co-workers (Figure 1).<sup>1</sup> The structure of this unique tricyclic sesquiterpenoid was determined by detailed NMR spectroscopic analysis to comprise a highly strained *trans*-5,6-fused ring appended to  $\gamma$ -spirolactone. Although the overall relative configuration of sinularianin B was assigned on the basis of spectral analysis, the absolute configuration had not been determined. Furthermore, sesquiterpenoids possessing a valerenane skeleton have been isolated mostly from terrestrial plants, e.g. *Valeriana officinalis*,<sup>2</sup> although one report has described the isolation of valerenane sesquiterpenoid from soft coral.<sup>3</sup> Recently, we were first to report on the asymmetric total synthesis of sinularianin B (**1**) using our developed method involving sulfone-mediated highly stereoselective tandem intermolecular–intramolecular alkylation.<sup>4</sup> In this paper, we wish to report on the detailed synthesis of sinularianin B and 8-*epi*-sinularianin B.

Figure 1. Structure of sinularianin B (**1**)

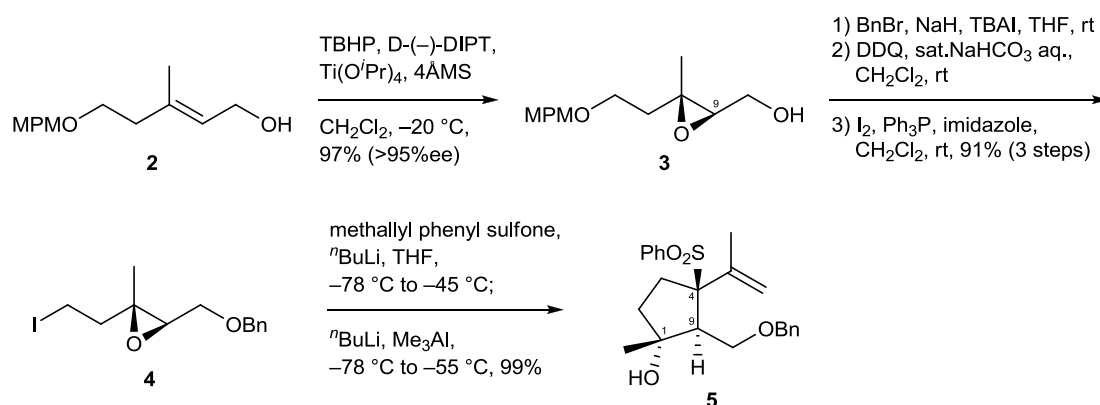
Our initial retrosynthetic analysis for the synthesis of sinularianin B (**1**) is outlined in Figure 2. We envisioned that the *trans*-5,6-fused ring moiety should be constructed in the latter stages due to potential isomerization of the double bond between C-5 and C-6 under both acidic and basic conditions. Therefore, we set triene **A** for further elaboration toward **1**. Triene **A** was expected to be generated from aldehyde **B** through stepwise double alkylation at C-8. Aldehyde **B** could be generated from sulfonylcyclopentane **C** by stereoselective palladium-mediated hydrogenolysis and oxidation. Key intermediate sulfonylcyclopentane **C** would be constructed by our developed tandem intermolecular–intramolecular alkylation of epoxide **D** and methallyl phenyl sulfone.<sup>5</sup> Optically active epoxide **D** could be derived from known allylic alcohol **2**.<sup>6</sup>

Figure 2. Retrosynthetic analysis of sinularianin B (**1**)

## RESULTS AND DISCUSSION

The first stage involved construction of the sulfonylcyclopentane possessing a tertiary hydroxy group at C-1 (sinularianin B numbering). The asymmetric synthesis of sulfonylcyclopentane **5** from epoxyiodide **4** and methallyl phenyl sulfone-mediated tandem intermolecular–intramolecular alkylation is shown in Scheme 1. According to the known three-step procedure, 4-hydroxybutan-2-one was elaborated to allylic

alcohol **2**,<sup>6</sup> which was then converted to epoxyalcohol **3** in 97% yield using Sharpless asymmetric epoxidation under standard conditions.<sup>7</sup> The optical purity of epoxyalcohol **3** was determined as >95% ee, which was estimated from the <sup>1</sup>H NMR spectra in the presence of chiral shift reagent “Chirabite-AR”.<sup>8</sup> To demonstrate the practical utility of Chirabite-AR, at first, synthetic racemic epoxyalcohol *rac*-**3** was tested. Prior to the comparison between synthetic (–)-epoxyalcohol **3** and *rac*-**3** with Chirabite-AR, we examined the effect of differing amounts of Chirabite-AR regarding *rac*-**3**, in an effort to determine sufficient signal separations between (+)- and (–)-**3**. Consequently, a mixture of *rac*-**3** with 75 mol% of Chirabite-AR was measured sequentially by 400 MHz <sup>1</sup>H NMR at 25 °C in CDCl<sub>3</sub>. The methine proton at C-9 signal separations were observed between 3.32 to 3.14 ppm, and good enantiomeric discrimination was achieved for (+)- and (–)-**3**. NMR analysis of (–)-epoxyalcohol **3** under the same conditions indicated that separated signals exhibited >39/1 ratio in numerical integration value. Therefore, the optical purity of (–)-**3** was determined as >95 % ee. Epoxyiodide **4**, required for the key tandem intermolecular–intramolecular alkylation necessary for the construction of sulfonylcyclopentane **5**, was obtained by protection of the primary hydroxy group in chiral epoxyalcohol **3** with benzyl bromide, sodium hydride, and TBAI, selective deprotection with DDQ, and iodination of the primary hydroxy group with I<sub>2</sub>, Ph<sub>3</sub>P and imidazole in 91% yield for the three steps and in multigram quantities. The sulfonylcarbanion prepared from methallyl phenyl sulfone (1.30 equiv.) and <sup>n</sup>BuLi (1.25 equiv.) was treated with epoxyiodide **4** at –45 °C. Following confirmation of the disappearance of the starting material by TLC, <sup>n</sup>BuLi (2.00 equiv.) and Me<sub>3</sub>Al (1.50 equiv.) were added sequentially at –78 °C, and allowing the reaction mixture to warm to –55 °C cleanly promoted 5-*endo* cyclization to give sulfonylcyclopentane **5** as the sole product in 99% yield.



Scheme 1

The relative configuration of sulfonylcyclopentane **5** was determined by NOESY analysis (Figure 3). The *trans*-configuration of the isopropenyl and benzyloxymethyl groups in sulfonylcyclopentane **5** was determined by NOE correlation between the  $\beta$ -methylene proton at C-3 and the  $\text{sp}^2$ -methine proton at the *ortho* position of the phenylsulfonyl group, the  $\alpha$ -methylene proton at C-3 and the methine proton at C-9,

the methylene protons at C-8 and the methyl protons at C-10, and the methine proton at C-9 and the methyl protons at C-11.

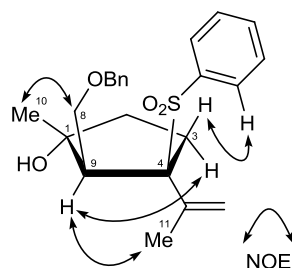


Figure 3. Selected NOE correlations of sulfonylcyclopentane **5**

The complete diastereoselectivity associated with the tandem intermolecular–intramolecular alkylation was attributed to the stability of the transition state in the intramolecular alkylation from the sulfonylcarbanion intermediate. The carbanion of methallyl phenyl sulfone reacted with epoxyiodide **4** to form epoxysulfone **6**, which is rapidly converted into corresponding sulfonylcarbanion **6a**. Furthermore, activation of the epoxide that followed the addition of  $\text{Me}_3\text{Al}$  as a Lewis acid, shifted the equilibrium between sulfonylcarbanions **6b** and **6c**. In the equilibrium state, intramolecular cyclization mostly proceeded from sulfonylcarbanion **6c**, as a result of transition state destabilization due to steric hindrance between the phenylsulfonyl group and the benzyloxymethyl group in sulfonylcarbanion **6b**. Consequently, sulfonylcyclopentane **5** was generated by acidic workup of alkoxyaluminum anionic species **6d**.

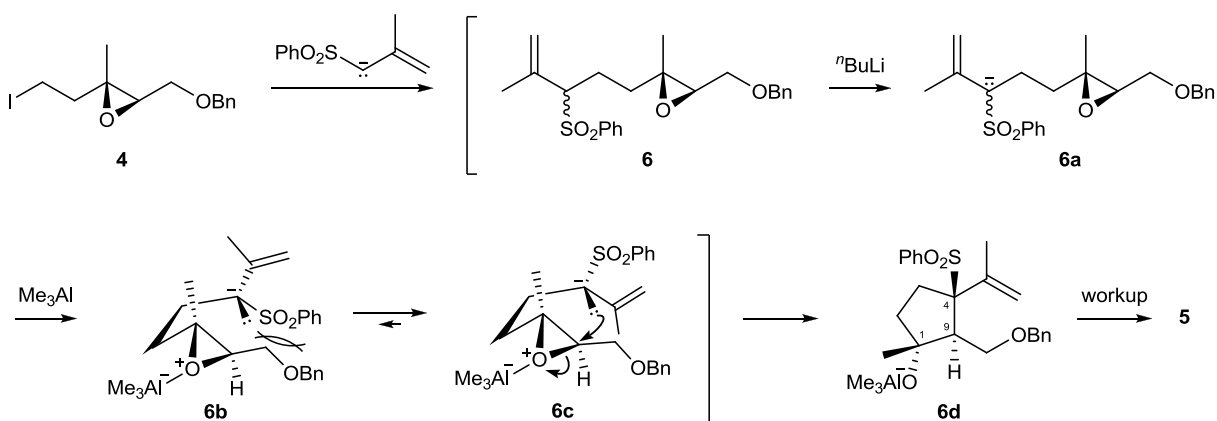
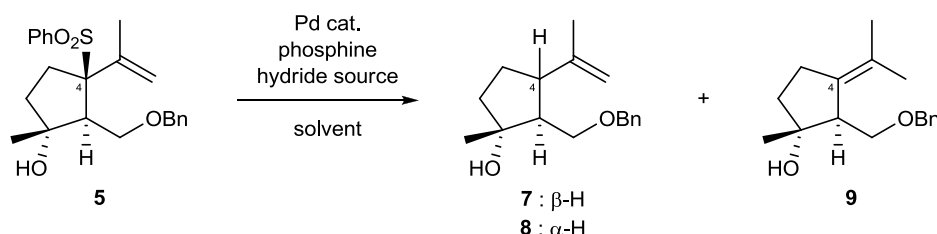


Figure 4. A plausible mechanism for the formation of sulfonylcyclopentane **5**

We next focused on removal of the phenylsulfonyl group in sulfonylcyclopentane **5**. We conceived a hydrogenolysis approach *via* a  $\pi$ -allyl palladium complex (Table 1). Initially, we applied  $\text{NaBH}_4$ <sup>9</sup> or  $\text{LiBHET}_3$ <sup>10</sup> as a standard hydride source for the palladium-mediated hydrogenolysis, however, the desired cyclopentane **7** was not generated, and isomerization proceeded to the corresponding internal olefin **9** (entries 1 and 2). A solution to this problem was found by using *in situ*-generated ammonium formate as the hydride source,<sup>11</sup> where treatment of sulfonylcyclopentane **5** with triethylamine (8.0 equiv.), formic acid

(8.0 equiv.), and PdCl<sub>2</sub>(dppp) (0.005 equiv.) in THF at rt facilitated regioselective hydrogenolysis and provided cyclopentane **7** in 16% yield and 4-epimer **8** in 18% yield with low diastereoselectivity (entry 3). After some optimization, optimal reaction conditions included a reagent combination comprising Pd<sub>2</sub>(dba)<sub>3</sub>·CHCl<sub>3</sub> (0.08 equiv.), <sup>n</sup>Bu<sub>3</sub>P (0.20 equiv.), triethylamine (8.0 equiv.), and formic acid (8.0 equiv.) in 1,4-dioxane (0.05 M) under reflux, and these conditions generated cyclopentane **7** in 91% yield with high stereoselectivity at C-4 (>20 : 1 d.r.) (entry 8).

Table 1. Palladium catalyst screening in the desulfonylation of sulfonylcyclopentane **5**



entry	Pd catalyst	equiv.	phosphine	hydride source	solvent	temperature	yield (%)		
							<b>7</b>	<b>8</b>	<b>9</b>
1	PdCl <sub>2</sub> (dppp)	0.005	-	NaBH <sub>4</sub>	THF	rt	0	0	44
2	PdCl <sub>2</sub> (dppp)	0.005	-	LiBHEt <sub>3</sub>	THF	rt	0	0	98
3	PdCl <sub>2</sub> (dppp)	0.005	-	Et <sub>3</sub> N, HCO <sub>2</sub> H	THF	rt	16	18	0
4	Pd <sub>2</sub> (dba) <sub>3</sub> ·CHCl <sub>3</sub>	0.08	<sup>n</sup> Bu <sub>3</sub> P	Et <sub>3</sub> N, HCO <sub>2</sub> H	THF	40 °C	57	38	0
5	Pd <sub>2</sub> (dba) <sub>3</sub> ·CHCl <sub>3</sub>	0.08	<sup>n</sup> Bu <sub>3</sub> P	Et <sub>3</sub> N, HCO <sub>2</sub> H	THF	reflux	73	18	0
6	Pd <sub>2</sub> (dba) <sub>3</sub> ·CHCl <sub>3</sub>	0.08	<sup>n</sup> Bu <sub>3</sub> P	Et <sub>3</sub> N, HCO <sub>2</sub> H	1,4-dioxane	40 °C	78	21	0
7	Pd <sub>2</sub> (dba) <sub>3</sub> ·CHCl <sub>3</sub>	0.08	<sup>n</sup> Bu <sub>3</sub> P	Et <sub>3</sub> N, HCO <sub>2</sub> H	1,4-dioxane	80 °C	83	13	0
8	Pd <sub>2</sub> (dba) <sub>3</sub> ·CHCl <sub>3</sub>	0.08	<sup>n</sup> Bu <sub>3</sub> P	Et <sub>3</sub> N, HCO <sub>2</sub> H	1,4-dioxane	reflux	91	4	0

The relative configurations of cyclopentane **7** were determined by NOE correlation between the methine proton at C-4 and the methyl protons at C-10, the methylene protons at C-8 and the methyl protons at C-10, and the methine proton at C-9 and the methyl protons at C-11 (Figure 5). Similarly, the relative configurations of 4-epimer **8** were determined by NOE correlation between the methine proton at C-4 and the methine proton at C-9, the methylene protons at C-8 and the methyl protons at C-10, and the methylene protons at C-8 and the methyl protons at C-11.

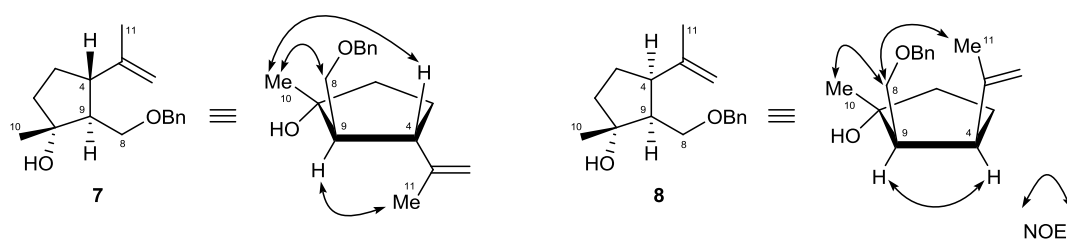


Figure 5. Selected NOE correlations of cyclopentanes **7** and **8**

In order to account for the observed improvement in the reaction outcome, a possible mechanism of the regio- and stereoselective hydrogenolysis has been outlined in Figure 6. In palladium-catalyzed hydrogenolysis, it is well-established that the initial step in  $\pi$ -allyl palladium complex formation involves inversion of stereochemistry. The resulting  $\pi$ -allyl palladium complex **10a** rapidly rearranges to the  $\sigma$ -allyl palladium formate **10b**, and subsequently the single bond rotation, which was caused by the non-bonding interactions between the oxygen atom and the cationic palladium species, generated **10c**. Subsequent migration of the hydride from the  $\sigma$ -allyl palladium formate on the more substituted end of the allylic system afforded cyclopentane **7**.

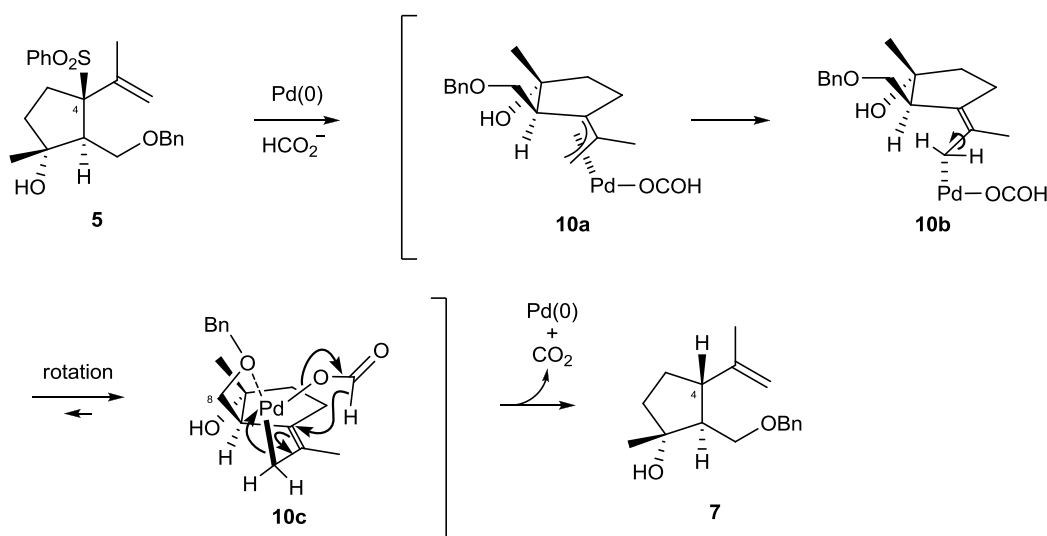
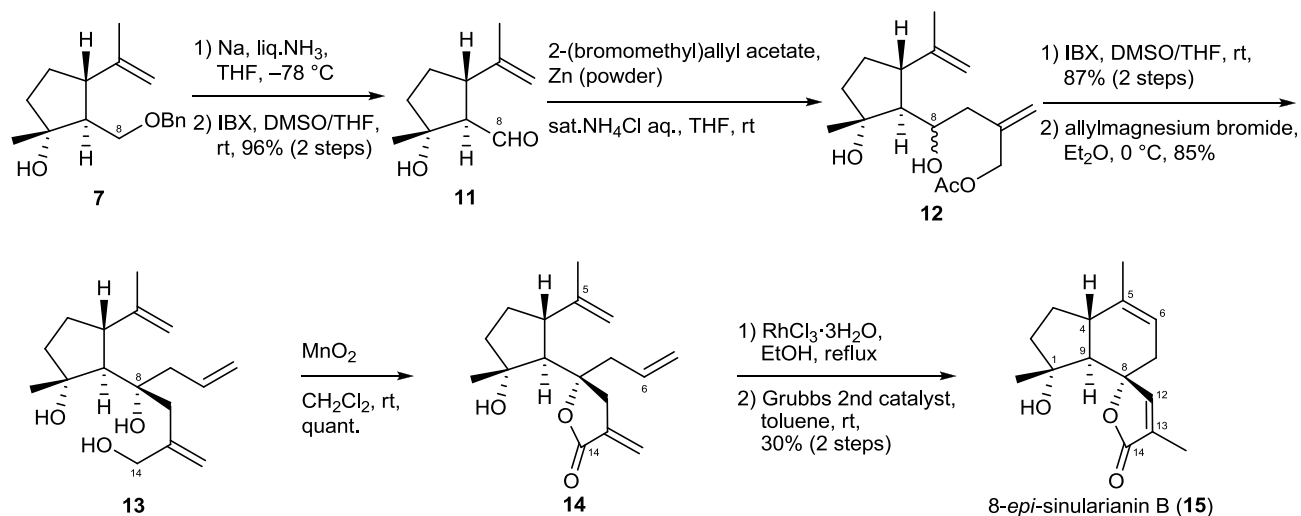


Figure 6. A plausible mechanism for desulfonylation following decarboxylation

As shown in Scheme 2, removal of the benzyl group in cyclopentane **7** with Na/liq.NH<sub>3</sub> did not affect the terminal olefin and proceeded smoothly to give the primary hydroxy group. Subsequent treatment of the derived diol with IBX<sup>12</sup> in DMSO/THF oxidized the primary alcohol to corresponding aldehyde **11** in 96% yield for the two steps. Next, according to the synthetic plan, we introduced two different allylic groups at C-8. Thus, zinc-mediated Barbier–Grignard type allylation<sup>13</sup> of aldehyde **11** with 2-(bromomethyl)allyl acetate in saturated NH<sub>4</sub>Cl aqueous/THF furnished corresponding diol **12** as diastereomeric mixtures. Oxidation of diol **12** with IBX in DMSO/THF afforded the  $\beta,\gamma$ -unsaturated ketone in 87% yield for the two steps. To the resulting ketone was introduced an allyl group using allyl magnesium bromide, giving triene **13** in 85% yield as a single diastereomer. Although it was not possible to determine the configuration of the tertiary hydroxy group at C-8 in triene **13** at this stage, the presence of an opposite configuration to that of the natural sinularianin B was determined by NOESY analysis at a later step. Oxidation of the allylic hydroxy group in stereochemically-undefined triene **13** with manganese dioxide<sup>14</sup> led to  $\alpha$ -methylene- $\gamma$ -lactone **14** in quantitative yield. Isomerization with rhodium chloride trihydrate,<sup>15</sup> and ring-closing metathesis (RCM) of the resulting  $\alpha,\beta$ -unsaturated lactone gave tricyclic compound **15**

in 30% yield for the two steps.



Scheme 2

Since the <sup>1</sup>H- and <sup>13</sup>C NMR data of synthetic **15** and the natural product did not match,<sup>1</sup> the aforementioned determination of the relative configuration at C-8 was performed at this stage as shown in Figure 7. The relative configurations of **15** were determined by NOE correlation between the methine proton at C-4 and the methyl protons at C-10, the methine proton at C-4 and the olefin proton at C-12, the β-methylene proton at C-7 and the olefin proton at C-12, the α-methylene proton at C-7 and the methine proton at C-9, and the methyl protons at C-10 and the olefin proton at C-12. Therefore, synthetic compound **15** was determined to be 8-*epi*-sinularianin B.

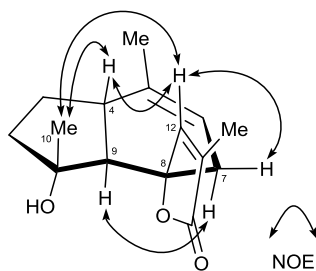
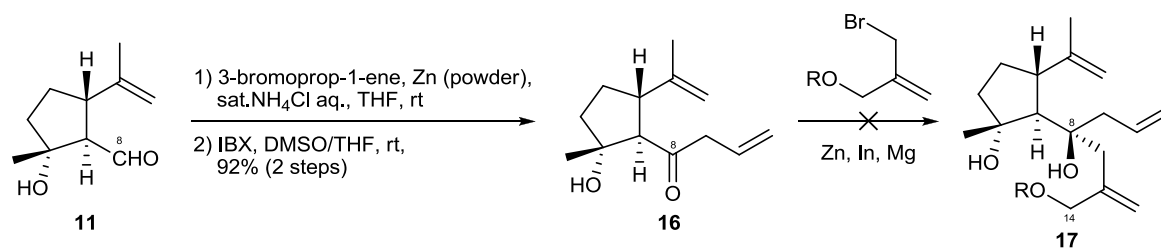


Figure 7. Selected NOE correlations of 8-*epi*-sinularianin B (**15**)

Given the above results, a second synthetic attempt was performed whereby the order of introducing the side chain was altered. Zinc-mediated Barbier–Grignard type allylation<sup>13</sup> of aldehyde **11** with 3-bromoprop-1-ene in saturated NH<sub>4</sub>Cl aqueous/THF furnished the corresponding diol as diastereomeric mixtures (Scheme 3). Subsequent treatment of the derived diol with IBX<sup>12</sup> in DMSO/THF oxidized the secondary alcohol to β,γ-unsaturated ketone **16** in 92% yield for the two steps. Although metal-mediated allylation of β,γ-unsaturated ketone **16** with various hydroxy group protected allylic bromides (R = Bn, MPM, THP, TBS) was negligible, β,γ-unsaturated ketone **16** underwent decomposition.



Scheme 3

On the basis of this unsatisfactory result, an alternative strategy had to be implemented (Figure 8). We envisaged that the  $\gamma$ -spirolactone moiety of sinularianin B (**1**) could be assembled from precursor **E**. Construction of the cyclohexene ring of **E** from the *O*-silylated cyanohydrin **F** would involve sequential RCM, conversion of the nitrile to the aldehyde, and deprotection. *O*-Silylated cyanohydrin **F** could be generated from aldehyde **11** by cyanohydrin formation and stereoselective alkylation with allyl halide at C-8.

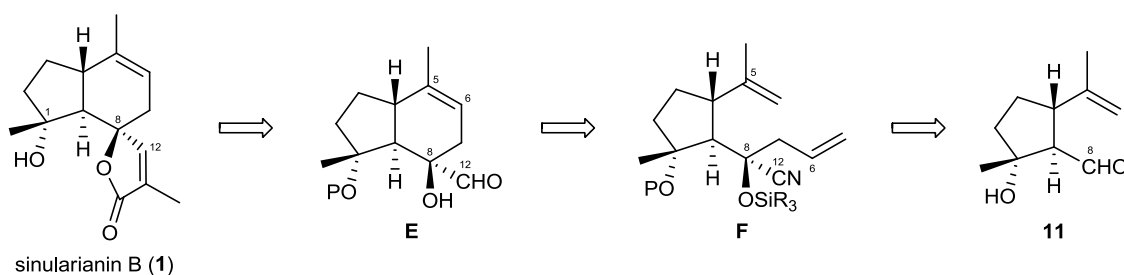
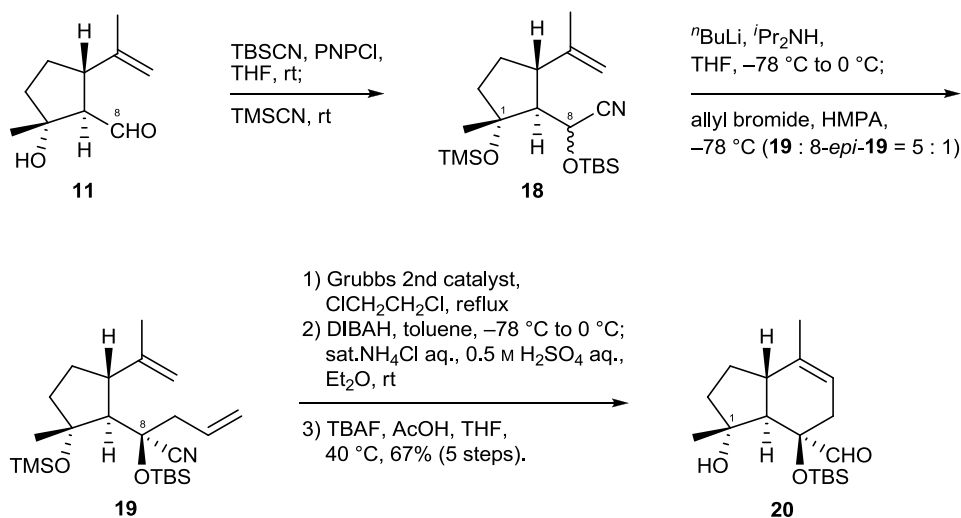


Figure 8. Modified retrosynthetic analysis

As shown in Scheme 4, for the pivotal construction of the *trans*-5,6-fused ring, aldehyde **11** was treated with TBSCN in the presence of a catalytic amount of bis(triphenylphosphoranylidene)ammonium chloride (PNPCL).<sup>16</sup> Following confirmation of the disappearance of **11** by TLC, TMSCN was added, which promoted silylation of the bulky tertiary hydroxy group at C-1 to afford  $\alpha$ -siloxynitrile **18** as the diastereomeric mixture at C-8. Initially, allylation of crude  $\alpha$ -siloxynitrile **18** with allyl bromide was negligible given the steric bulkiness of the nucleophile. However, following extensive optimization, it was found that generation of the carbanion of  $\alpha$ -siloxynitrile **18** with LDA in THF at 0 °C followed by cooling to -78 °C and addition of allyl bromide premixed with HMPA gave  $\alpha$ -siloxynitrile **18** as an inseparable diastereomeric mixture at C-8 and acceptable selectivity (5 : 1). After considerable optimization, RCM of a mixture comprising  $\alpha$ -siloxynitrile **18** and C-8 epimer cleanly generated the *trans*-5,6-fused ring using Grubbs 2nd generation catalyst,<sup>17</sup> with high catalyst loading (10 mol%). Reduction of the cyano group with DIBAH led to the  $\alpha$ -siloxyaldehyde. The reaction proceeded in good yields, based on careful monitoring of the weak acidic workup conditions and a minimal excess of DIBAH. Interestingly, degradation of the C-8 epimer occurred concurrently with DIBAH reduction. Deprotection with TBAF/AcOH afforded

$\alpha$ -siloxyaldehyde **20** as a sole product in 67% yield for the five steps from **11**. The desired *R* configuration at C-8 of  $\alpha$ -siloxyaldehyde **20** was determined by single-crystal X-ray diffraction (Figure 9).



Scheme 4

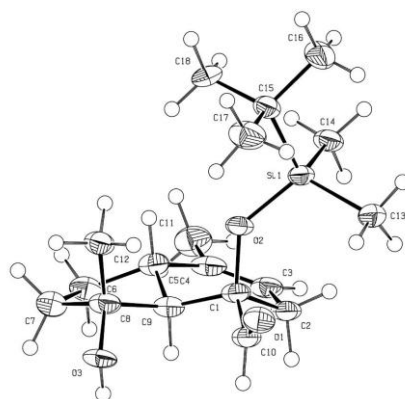
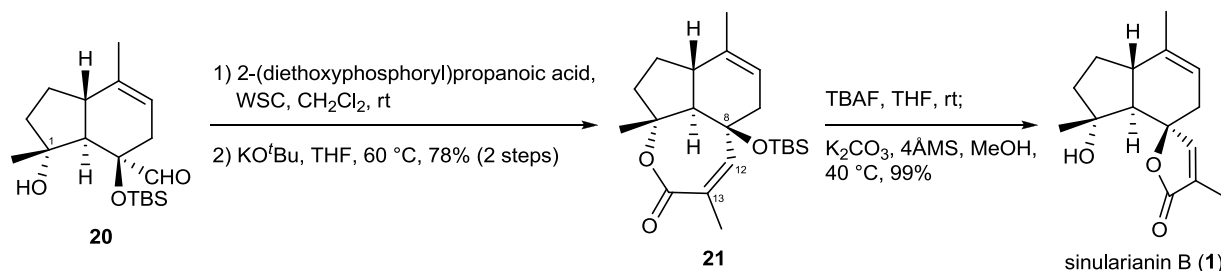


Figure 9. ORTEP drawing of  $\alpha$ -siloxyaldehyde **20**

The final critical step, involving assembly of sinularianin B (**1**), was executed *via* the  $\alpha,\beta$ -unsaturated seven-membered lactone.  $\alpha$ -Siloxyaldehyde **20** was coupled with 2-(diethoxyphosphoryl)propanoic acid using carbodiimide<sup>18</sup> to give the phosphate as shown in Scheme 5. The resultant phosphate was utilized in an intramolecular Horner-Wadsworth-Emmons reaction<sup>19</sup> to give  $\alpha,\beta$ -unsaturated seven-membered lactone **21** in 78% yield for the two steps from **20**. The last transformation was best achieved by one-pot desilylation/methanolysis/re-lactonization methodologies, and where it was imperative to remove the TBS group from lactone **21** with TBAF to generate  $\gamma$ -hydroxylactone. Following confirmation of the disappearance of lactone **21** by TLC, treatment with K<sub>2</sub>CO<sub>3</sub> in MeOH at 40 °C for 2 days cleanly promoted methanolysis and rapid re-lactonization to give sinularianin B (**1**) as the sole product in 99% yield. The <sup>1</sup>H and <sup>13</sup>C NMR spectroscopic data of synthetic **1** matched those reported for sinularianin B.

Comparison of the specific optical rotation data  $\{[\alpha]_D -111 (c 1.55, \text{CHCl}_3); \text{lit.}^1 [\alpha]_D -27 (c 1.04, \text{CHCl}_3)\}$  enabled us to establish the absolute configuration of natural sinularianin B as being identical to that of synthetic product **1**. Despite the discrepancy in magnitude, the match of all other data has confirmed that the constitution and stereochemistry of sinularianin B (**1**) assigned by previously determined spectroscopic analyses are correct.



Scheme 5

In conclusion, a highly stereoselective total synthesis of sinularianin B (**1**) was accomplished by a concise strategy. The procedure comprises 16 steps (longest linear sequence) with 39.5% yield starting from known allylic alcohol **2**. Furthermore, this work has unambiguously established the absolute configuration. Our synthetic strategy can be highlighted by the use of sulfone-mediated tandem intermolecular–intramolecular alkylation for construction of the cyclopentane core, palladium-catalyzed hydrogenolysis of allyl sulfone through  $\pi$ -allyl palladium complex formation, ring-closing metathesis to build the highly strained *trans*-5,6-fused ring, and an intramolecular Horner-Wadsworth-Emmons reaction for conversion of the seven-membered lactone into  $\gamma$ -spirolactone by an apparent acyl rearrangement.

## EXPERIMENTAL

All reagents (Aldrich, Kanto, TCI and Wako) and solvents were of commercial quality and were used as received. Reactions were monitored by thin layer chromatography on glass plates coated with a fluorescent indicator with a 254 nm excitation wavelength (Merck Merck-5554-7). Flash column chromatography was performed using Kanto Chemical Silica Gel 60N (spherical, natural) 40–50  $\mu\text{m}$ . Melting points (mp) were measured using the Yanaco melting point apparatus MP-S3 and are uncorrected. Optical rotations were measured with a JASCO P-1030 polarimeter. IR spectra were recorded with a JASCO FT-IR/620 spectrometer. UV spectra were recorded using a SHIMADZU UV-1200 spectrophotometer. Single crystal X-ray diffraction was recorded using a MacScience Co., Ltd DIP 2020 Image Plate.  $^1\text{H}$ - and  $^{13}\text{C}$  NMR spectra were recorded on a Bruker AVANCE III HD Nanobay or Bruker Biospin AV-600 spectrometer. Chemical shifts are given on the  $\delta$  (ppm) scale using tetramethylsilane (TMS) as the internal standard (s, singlet; d, doublet; t, triplet; q, quartet; quint., quintet; m, multiplet; br, broad). High resolution ESIMS

(HRESIMS) spectra were obtained using a Micromass LCT spectrometer. Elemental analysis data were obtained using an Elementar Vario EL.

**((2R,3R)-3-(2-(4-Methoxybenzyloxy)ethyl)-3-methyloxiran-2-yl)methanol (3):** To a cold ( $-20\text{ }^{\circ}\text{C}$ ) suspension of  $4\text{ \AA}$  molecular sieves (8.44 g) in  $\text{CH}_2\text{Cl}_2$  (38.0 mL) were added D-( $-$ )-DIPT (1.50 mL, 7.13 mmol),  $\text{Ti}(\text{O}^i\text{Pr})_4$  (1.10 mL, 3.73 mmol) and TBHP (5.55 M solution in  $\text{CH}_2\text{Cl}_2$ , 38.6 mL, 214 mmol). After stirring for 30 min at the same temperature, a solution of allylic alcohol **2** (16.9 g, 71.5 mmol) in  $\text{CH}_2\text{Cl}_2$  (200 mL) was added over 10 h. After stirring at  $-20\text{ }^{\circ}\text{C}$  for 2 h, NaOH (30% solution in brine, 3.25 mL) was added. The mixture was diluted with  $\text{Et}_2\text{O}$ , warmed to room temperature, and stirred for 10 min.  $\text{MgSO}_4$  (2.90 g) and Celite (0.35 g) were then added, and after stirring for 15 min the mixture was passed through a pad of Celite and then concentrated *in vacuo*. The residue was purified with flash column chromatography on silica gel (hexane/AcOEt = 1:1) to give epoxyalcohol **3** (17.5 g, 97% yield) as a colorless oil.  $R_f = 0.20$  (hexane/AcOEt 1:1);  $[\alpha]_D^{25} +0.29$  ( $c$  1.06 in  $\text{CHCl}_3$ );  $^1\text{H NMR}$  (400 MHz,  $\text{CDCl}_3$ ):  $\delta = 7.24$  (d,  $J = 8.6$  Hz, 2H), 6.87 (d,  $J = 8.6$  Hz, 2H), 4.43 (d,  $J = 14.7$  Hz, 1H), 4.40 (d,  $J = 14.7$  Hz, 1H), 3.80 (m, 1H), 3.79 (s, 3H), 3.65 (m, 1H), 3.57–3.51 (m, 2H), 3.02 (dd,  $J = 4.4, 6.5$  Hz, 1H), 2.15 (brs, 1H), 1.94 (ddd,  $J = 5.9, 6.0, 14.3$  Hz, 1H), 1.78 (ddd,  $J = 6.6, 7.1, 14.3$  Hz, 1H), 1.30 (s, 3H);  $^{13}\text{C NMR}$  (100 MHz,  $\text{CDCl}_3$ ):  $\delta = 159.1$  (s), 130.2 (s), 129.2 (d) $\times 2$ , 113.8 (d) $\times 2$ , 72.6 (t), 66.0 (t), 62.8 (d), 61.2 (t), 59.6 (s), 55.2 (q), 38.2 (t), 17.2 (q); IR (neat):  $\tilde{\nu} = 3441, 2932\text{ cm}^{-1}$ ; HRMS (ESI):  $m/z$  calcd for  $\text{C}_{14}\text{H}_{20}\text{O}_4 + \text{Na}^+$ : 275.1259 [ $M + \text{Na}^+$ ]; found: 275.1247; Anal. Calcd for  $\text{C}_{14}\text{H}_{20}\text{O}_4$ : C 66.65, H 7.99; found: C 66.64, H 8.14.

**Determination of optical purity of synthetic ( $-$ )-epoxyalcohol 3:** Before comparison between synthetic ( $-$ )-epoxyalcohol **3** and synthetic racemic epoxyalcohol *rac*-**3** with Chirabite-AR, we examined the effect of differing amounts of Chirabite-AR regarding *rac*-**3**, to determine sufficient signal separations between (+)- and ( $-$ )-**3**. Consequently, a mixture of *rac*-**3** with 75 mol% of Chirabite-AR was measured sequentially by 400 MHz  $^1\text{H NMR}$  at  $25\text{ }^{\circ}\text{C}$  in  $\text{CDCl}_3$ , the methine proton at C-9 signal separations were observed between 3.32 to 3.14 ppm, and good enantiomeric discrimination was achieved for (+)- and ( $-$ )-**3**. NMR analysis of ( $-$ )-epoxyalcohol **3** under the same conditions as used to obtain the results indicated that separated signals exhibited  $>39/1$  ratio in numerical integration value. Therefore, the optical purity of synthetic ( $-$ )-**3** was determined as  $>95\%$  ee.

**(2R,3R)-3-Benzyloxymethyl-2-(2-iodoethyl)-2-methyloxirane (4):** To a stirring solution of epoxyalcohol **3** (16.5 g, 65.4 mmol) in THF (109 mL) were added NaH (55%, 5.70 g, 432 mmol), BnBr (11.7 mL, 98.5 mmol) and TBAI (2.40 g, 6.50 mmol) at  $0\text{ }^{\circ}\text{C}$  and then allowed to warm to room temperature. After stirring for 7 h, MeOH (10.0 mL) was slowly added at  $0\text{ }^{\circ}\text{C}$ . The mixture was then allowed to warm to room temperature. After stirring for 1 h, the mixture was diluted with  $\text{Et}_2\text{O}$ , washed with saturated aqueous  $\text{NaHCO}_3$  solution,  $\text{H}_2\text{O}$  and brine, and then concentrated *in vacuo*. The residue was passed through a pad of

silica gel (hexane/AcOEt = 4:1) and then concentrated *in vacuo* to give a crude product.

To a stirring suspension of the crude product in CH<sub>2</sub>Cl<sub>2</sub>/saturated aqueous NaHCO<sub>3</sub> solution (10:1, 127 mL) were added DDQ (22.3 g, 98.2 mmol) over 10 min at room temperature. After stirring for 30 min, the reaction mixture was diluted with Et<sub>2</sub>O, washed with saturated aqueous NH<sub>4</sub>Cl solution, H<sub>2</sub>O and brine, dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>, and then concentrated *in vacuo*. The residue was passed through a pad of silica gel (hexane/AcOEt = 2:1) and then concentrated *in vacuo* to give a crude product.

To a cold (0 °C) solution of the crude product in CH<sub>2</sub>Cl<sub>2</sub> (211 mL) were added Ph<sub>3</sub>P (19.9 g, 75.9 mmol), imidazole (6.47 g, 95.0 mmol). After stirring for 5 min at the same temperature, I<sub>2</sub> (19.4 g, 76.4 mmol) was slowly added. The mixture was then allowed to warm to room temperature. After stirring for 10 min, the solvent was removed *in vacuo*. The residue was passed through a pad of silica gel (hexane/Et<sub>2</sub>O = 6:1) and then concentrated *in vacuo*. The residue was purified with flash column chromatography on silica gel (hexane/AcOEt = 9:1) to give epoxyiodide **4** (19.8 g, 91% yield for 3 steps) as a colorless oil. *R*<sub>f</sub> = 0.60 (hexane/AcOEt 2:1); [ $\alpha$ ]<sub>D</sub><sup>25</sup> +7.19 (*c* 1.53 in CHCl<sub>3</sub>); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  = 7.37–7.30 (m, 5H), 4.64 (d, *J* = 11.9 Hz, 1H), 4.55 (d, *J* = 11.9 Hz, 1H), 3.75 (dd, *J* = 4.2, 11.3 Hz, 1H), 3.55 (dd, *J* = 6.2, 11.3 Hz, 1H), 3.20 (ddd, *J* = 5.3, 8.6, 9.2 Hz, 1H), 3.15 (ddd, *J* = 1.9, 7.7, 8.6 Hz, 1H), 3.10 (dd, *J* = 4.2, 6.2 Hz, 1H), 2.24 (ddd, *J* = 5.3, 8.6, 14.3 Hz, 1H), 2.04 (ddd, *J* = 7.7, 9.2, 14.3 Hz, 1H), 1.26 (s, 3H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta$  = 137.8 (s), 128.4 (d)×2, 127.7 (d), 127.7 (d)×2, 73.2 (t), 68.5 (t), 61.1 (d), 60.1 (s), 42.2 (t), 16.2 (q), -1.2 (t); IR (neat):  $\tilde{\nu}$  = 2925, 2857 cm<sup>-1</sup>; HRMS (ESI): *m/z* calcd for C<sub>13</sub>H<sub>17</sub>IO<sub>2</sub>+H<sup>+</sup>: 333.0352 [*M*+H<sup>+</sup>]; found: 333.0335; Anal. Calcd for C<sub>13</sub>H<sub>17</sub>IO<sub>2</sub>: C 47.00, H 5.16; found: C 46.94, H 5.27.

**(1R,2S,3R)-2-Benzylloxymethyl-1-methyl-3-phenylperoxythio-3-prop-1-en-2-ylcyclopentanol (5)**: To a solution of methallyl phenyl sulfone (1.45 g, 7.39 mmol) in THF (24.0 mL) were added <sup>n</sup>BuLi (1.58 M solution in hexane, 4.50 mL, 7.11 mmol) at -78 °C. After stirring for 30 min at the same temperature, a solution of epoxyiodide **4** (1.89 g, 5.69 mmol) in THF (90.0 mL) was added and then allowed to warm to -45 °C over 18 h. After cooling to -78 °C, <sup>n</sup>BuLi (1.58 M solution in hexane, 7.20 mL, 11.4 mmol). After stirring for 15 min at the same temperature, Me<sub>3</sub>Al (1.07 M solution in hexane, 8.00 mL, 8.56 mmol) was introduced and then allowed to warm to -55 °C. After stirring for 2.5 h, the reaction mixture was diluted with Et<sub>2</sub>O, washed with saturated aqueous NH<sub>4</sub>Cl solution, 1.0 M aqueous HCl solution, H<sub>2</sub>O and brine, dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>, and then concentrated *in vacuo*. The residue was purified with flash column chromatography on silica gel (hexane/AcOEt = 2:1) to give cyclopentane **5** (2.26 g, 99% yield) as a white needle-like crystalline solid. *R*<sub>f</sub> = 0.15 (hexane/AcOEt 2:1); mp 108–109 °C (recrystallized from hexane/AcOEt); [ $\alpha$ ]<sub>D</sub><sup>25</sup> -45.7 (*c* 1.16 in CHCl<sub>3</sub>); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  = 7.73–7.72 (m, 2H), 7.59 (m, 1H), 7.47–7.44 (m, 2H), 7.36–7.29 (m, 5H), 4.99 (s, 1H), 4.66 (d, *J* = 11.6 Hz, 1H), 4.61 (s, 1H), 4.59 (d, *J* = 11.6 Hz, 1H), 4.46 (dd, *J* = 4.8, 9.5 Hz, 1H), 4.41 (dd, *J* = 9.5, 11.4 Hz, 1H), 3.07 (brs, 1H), 2.75 (dd, *J*

= 4.8, 11.4 Hz, 1H), 2.41 (dt,  $J = 6.4, 12.9$  Hz, 1H), 1.98 (s, 3H), 1.92–1.81 (m, 2H), 1.71 (dt,  $J = 6.4, 12.9$  Hz, 1H), 1.60 (s, 3H); NOE correlations (H/H): H-3 ( $\delta_{\text{H}}$  2.41)/*ortho*-H ( $\delta_{\text{H}}$  7.73), H-3 ( $\delta_{\text{H}}$  2.41)/H-10 ( $\delta_{\text{H}}$  1.60), H-3 ( $\delta_{\text{H}}$  1.71)/H-9 ( $\delta_{\text{H}}$  2.75), H-8 ( $\delta_{\text{H}}$  4.46)/H-10 ( $\delta_{\text{H}}$  1.60), H-8 ( $\delta_{\text{H}}$  4.41)/H-11 ( $\delta_{\text{H}}$  1.98), H-9 ( $\delta_{\text{H}}$  2.75)/H-11 ( $\delta_{\text{H}}$  1.98);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ ):  $\delta = 142.2$  (s), 137.8 (s), 136.0 (s), 133.5 (d), 130.3 (d) $\times 2$ , 128.4 (d) $\times 2$ , 128.0 (d) $\times 2$ , 127.7 (d), 127.7 (d) $\times 2$ , 118.1 (t), 80.4 (s), 77.5 (s), 73.6 (t), 68.8 (t), 57.9 (d), 38.0 (t), 31.4 (t), 22.9 (q), 20.8 (q); IR (KBr):  $\tilde{\nu} = 3363, 2975, 2931, 1291, 1135$   $\text{cm}^{-1}$ ; HRMS (ESI):  $m/z$  calcd for  $\text{C}_{23}\text{H}_{28}\text{O}_4\text{S}+\text{Na}^+$ : 423.1606 [ $M+\text{Na}^+$ ]; found: 423.1591; Anal. Calcd for  $\text{C}_{23}\text{H}_{27}\text{O}_4\text{S}$ : C 68.97, H 7.05; found: C 69.17, H 7.00.

**(1R,2S,3S)-2-(Benzyloxymethyl)-1-methyl-3-(prop-1-en-2-yl)cyclopentanol (7)** and **(1R,2S,3R)-2-(benzyloxymethyl)-1-methyl-3-(prop-1-en-2-yl)cyclopentanol (8)**:  $n\text{Bu}_3\text{P}$  (0.82 mL, 3.28 mmol) was added to a solution of  $\text{Pd}_2(\text{dba})_3\cdot\text{CHCl}_3$  (1.36 g, 1.31 mmol) in 1,4-dioxane (200 mL) at room temperature and the mixture was stirred for 10 min.  $\text{Et}_3\text{N}$  (18.3 mL, 131 mmol) and  $\text{HCO}_2\text{H}$  (4.95 mL, 131 mmol) were added to the mixture at the same temperature. After stirring for 10 min, the mixture was refluxed. A solution of cyclopentane **5** (6.56 g, 16.4 mmol) in 1,4-dioxane (128 mL) was added to the mixture and the mixture was stirred for 15 min. The reaction mixture was concentrated *in vacuo*. The residue was purified with flash column chromatography on silica gel (hexane/AcOEt = 5:1) to give *trans*-cyclopentane **7** (3.89 g, 91% yield) as a colorless oil and *cis*-cyclopentane **8** (171 mg, 4% yield) as a colorless oil. *trans*-cyclopentane **7**:  $R_{\text{f}} = 0.40$  (hexane/AcOEt 3:1);  $[\alpha]_{\text{D}}^{25} +31.9$  ( $c$  1.23 in  $\text{CHCl}_3$ );  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ ):  $\delta = 7.37$ – $7.29$  (m, 5H), 4.68 (s, 1H), 4.68 (s, 1H), 4.52 (d,  $J = 11.8$  Hz, 1H), 4.45 (d,  $J = 11.8$  Hz, 1H), 3.55 (dd,  $J = 4.0, 9.4$  Hz, 1H), 3.43 (t,  $J = 9.4$  Hz, 1H), 2.85 (brs, 1H), 2.28–2.16 (m, 2H), 1.87–1.59 (m, 4H), 1.72 (s, 3H), 1.24 (s, 3H);  $^1\text{H}$  NMR (400 MHz,  $\text{C}_6\text{D}_6$ ):  $\delta = 7.31$ – $7.16$  (m, 5H), 4.84 (s, 1H), 4.81 (s, 1H), 4.33 (d,  $J = 11.9$  Hz, 1H), 4.26 (d,  $J = 11.9$  Hz, 1H), 3.55 (dd,  $J = 4.4, 9.4$  Hz, 1H), 3.38 (t,  $J = 9.4$  Hz, 1H), 2.72 (brs, 1H), 2.40 (dt,  $J = 4.4, 10.7$  Hz, 1H), 2.24 (dt,  $J = 10.7, 8.7$  Hz, 1H), 2.05 (dt,  $J = 11.0, 7.4$  Hz, 1H), 1.78–1.64 (m, 3H), 1.71 (s, 3H), 1.33 (s, 3H); NOE correlations (H/H): H-4 ( $\delta_{\text{H}}$  2.24)/H-10 ( $\delta_{\text{H}}$  1.33), H-8 ( $\delta_{\text{H}}$  3.55 and 3.38)/H-10 ( $\delta_{\text{H}}$  1.33), H-9 ( $\delta_{\text{H}}$  2.40)/H-11 ( $\delta_{\text{H}}$  1.71);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ ):  $\delta = 146.7$  (s), 138.0 (s), 128.4 (d) $\times 2$ , 127.7 (d), 127.7 (d) $\times 2$ , 110.5 (t), 80.1 (s), 73.5 (t), 70.8 (t), 51.5 (d), 47.6 (d), 39.4 (t), 27.2 (t), 23.5 (q), 18.7 (q);  $^{13}\text{C}$  NMR (100 MHz,  $\text{C}_6\text{D}_6$ ):  $\delta = 147.2$  (s), 138.6 (s), 128.6 (d) $\times 2$ , 128.5 (d), 127.8 (d) $\times 2$ , 110.6 (t), 80.0 (s), 73.5 (t), 70.9 (t), 52.5 (d), 48.4 (d), 40.4 (t), 27.8 (t), 24.1 (q), 18.8 (q); IR (neat):  $\tilde{\nu} = 3446, 2962, 2871, 1645, 1098$   $\text{cm}^{-1}$ ; HRMS (ESI):  $m/z$  calcd for  $\text{C}_{17}\text{H}_{24}\text{O}_2+\text{Na}^+$ : 283.1674 [ $M+\text{Na}^+$ ]; found: 283.1677; Anal. Calcd for  $\text{C}_{17}\text{H}_{24}\text{O}_2$ : C 78.42, H 9.29; found: C 78.26, H 9.03. *cis*-cyclopentane **8**:  $R_{\text{f}} = 0.35$  (hexane/AcOEt 3:1);  $[\alpha]_{\text{D}}^{25} -15.6$  ( $c$  1.27 in  $\text{CHCl}_3$ );  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ ):  $\delta = 7.35$ – $7.27$  (m, 5H), 4.85 (s, 1H), 4.74 (s, 1H), 4.40 (s, 2H), 3.40 (dd,  $J = 3.4, 9.8$  Hz, 1H), 3.26 (dd,  $J = 7.7, 9.8$  Hz, 1H), 3.08 (dd,  $J = 8.2, 16.2$  Hz, 1H), 2.15 (dt,  $J = 3.4, 7.7$  Hz, 1H), 1.89–1.73 (m, 4H), 1.76 (s, 3H), 1.62 (m, 1H),

1.41 (s, 3H); NOE correlations (H/H): H-4 ( $\delta_{\text{H}}$  3.08)/H-9 ( $\delta_{\text{H}}$  2.15), H-8 ( $\delta_{\text{H}}$  3.40 and 3.26)/H-10 ( $\delta_{\text{H}}$  1.41), H-8 ( $\delta_{\text{H}}$  3.40 and 3.26)/H-11 ( $\delta_{\text{H}}$  1.76);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ ):  $\delta$  = 145.7 (s), 138.5 (s), 128.3 (d) $\times 2$ , 127.4 (d) $\times 2$ , 127.4 (d), 110.5 (t), 81.9 (s), 73.2 (t), 68.7 (t), 52.7 (d), 47.0 (d), 39.3 (t), 26.3 (t), 25.5 (q), 23.7 (q); IR (neat):  $\tilde{\nu}$  = 3387, 2963, 2936, 2871, 1646, 1070  $\text{cm}^{-1}$ ; HRMS (ESI):  $m/z$  calcd for  $\text{C}_{17}\text{H}_{24}\text{O}_2 + \text{Na}^+$ : 283.1674 [ $M + \text{Na}^+$ ]; found: 283.1669; Anal. Calcd for  $\text{C}_{17}\text{H}_{24}\text{O}_2$ : C 78.42, H 9.29; found: C 78.18, H 9.29.

**(1R,2R,5S)-2-Hydroxy-2-methyl-5-(prop-1-en-2-yl)cyclopentanecarbaldehyde (11)**: A solution of *trans*-cyclopentane **7** (2.63 g, 10.1 mmol) in THF (50.5 mL) was added to a pre-prepared Na (2.53 g, 110 mmol)/liquid ammonia (50.5 mL) at  $-78$  °C. After stirring for 20 min,  $\text{NH}_4\text{Cl}$  (10.1 g, 189 mmol) was added to the mixture and excess  $\text{NH}_3$  was removed by warming. The reaction mixture was diluted with  $\text{Et}_2\text{O}$ , washed with  $\text{H}_2\text{O}$  and brine, dried over anhydrous  $\text{Na}_2\text{SO}_4$ , and then concentrated *in vacuo*. The residue was passed through a pad of silica gel (hexane/ $\text{AcOEt}$  = 3:2) and then concentrated *in vacuo* to give a crude product.

To a solution of IBX (5.66 g, 20.2 mmol) in DMSO (50.5 mL) was added a solution of the above crude product in THF (50.5 mL). After stirring for 2.5 h at room temperature,  $\text{H}_2\text{O}$  was added to the mixture. After diluting with  $\text{Et}_2\text{O}$ , the mixture was filtered through celite, washed with  $\text{H}_2\text{O}$  and brine, dried over anhydrous  $\text{Na}_2\text{SO}_4$ , and then concentrated *in vacuo*. The residue was purified with flash column chromatography on silica gel (hexane/ $\text{AcOEt}$  = 2:1) to give aldehyde **11** (1.64 g, 96% yield for 2 steps) as a colorless oil.  $R_f$  = 0.30 (hexane/ $\text{AcOEt}$  2:1);  $[\alpha]_{\text{D}}^{25}$   $-29.5$  ( $c$  1.27 in  $\text{CHCl}_3$ );  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ ):  $\delta$  = 9.72 (d,  $J$  = 2.8 Hz, 1H), 4.74 (s, 1H), 4.73 (s, 1H), 2.99 (m, 1H), 2.75 (dd,  $J$  = 2.8, 9.8 Hz, 1H), 2.12 (brs, 1H), 1.95–1.67 (m, 4H), 1.72 (s, 3H), 1.34 (s, 3H);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ ):  $\delta$  = 203.7 (d), 145.7 (s), 110.5 (t), 81.3 (s), 65.8 (d), 45.7 (d), 41.9 (t), 28.0 (t), 25.3 (q), 20.3 (q); IR (neat):  $\tilde{\nu}$  = 3416, 2968, 1717, 1652, 1104  $\text{cm}^{-1}$ ; HRMS (ESI):  $m/z$  calcd for  $\text{C}_{10}\text{H}_{16}\text{O}_2 + \text{Na}^+$ : 191.1048 [ $M + \text{Na}^+$ ]; found: 191.1041; Anal. Calcd for  $\text{C}_{10}\text{H}_{16}\text{O}_2$ : C 71.39, H 9.59; found: C 71.49, H 9.50.

**4-((1R,2R,5S)-2-Hydroxy-2-methyl-5-(prop-1-en-2-yl)cyclopentyl)-2-methylene-4-oxobutyl acetate**: To a stirring solution of aldehyde **11** (22.7 mg, 0.135 mmol) in THF (1.4 mL) were added 2-(bromomethyl)allyl acetate (65.2 mg, 0.338 mmol), Zn powder (44.2 mg, 0.675 mmol) and saturated aqueous  $\text{NH}_4\text{Cl}$  solution (2.7 mL) at room temperature. After stirring for 2 h,  $\text{H}_2\text{O}$  (1.4 mL) was slowly added. After stirring for 30 min, the mixture was diluted with  $\text{Et}_2\text{O}$ , passed through a pad of celite ( $\text{Et}_2\text{O}$ ). The resulting solution was washed with  $\text{H}_2\text{O}$  and brine, dried over anhydrous  $\text{Na}_2\text{SO}_4$ , and then concentrated *in vacuo* to give a crude product.

To a solution of IBX (133 mg, 0.405 mmol) in DMSO (1.4 mL) was added a solution of the above crude product in THF (1.4 mL). After stirring for 10 h at room temperature,  $\text{H}_2\text{O}$  was added to the mixture. After diluting with  $\text{Et}_2\text{O}$ , the mixture was filtered through celite, washed with  $\text{H}_2\text{O}$  and brine, dried over

anhydrous  $\text{Na}_2\text{SO}_4$ , and then concentrated *in vacuo*. The residue was purified with flash column chromatography on silica gel (hexane/AcOEt = 3:1) to give ketone (32.9 mg, 87% yield for 2 steps) as a colorless oil.  $R_f = 0.30$  (hexane/AcOEt 3:1);  $[\alpha]_D^{25} -52.3$  ( $c$  1.17 in  $\text{CHCl}_3$ );  $^1\text{H NMR}$  (400 MHz,  $\text{CDCl}_3$ ):  $\delta = 5.24$  (s, 1H), 5.03 (s, 1H), 4.69 (s, 2H), 4.60 (d,  $J = 13.2$  Hz, 1H), 4.52 (d,  $J = 13.2$  Hz, 1H), 3.39 (d,  $J = 16.9$  Hz, 1H), 3.25 (d,  $J = 16.9$  Hz, 1H), 3.08 (d,  $J = 10.6$  Hz, 1H), 2.98 (dd,  $J = 7.8, 16.2$  Hz, 1H), 2.06 (s, 3H), 1.78–1.89 (m, 3H), 1.70 (s, 3H), 1.65 (m, 1H), 1.21 (s, 3H);  $^{13}\text{C NMR}$  (100 MHz,  $\text{CDCl}_3$ ):  $\delta = 208.3, 170.8, 146.4, 137.2, 117.4, 110.2, 81.3, 66.6, 64.9, 49.2, 47.2, 42.2, 27.3, 25.2, 20.9, 20.1$ ; IR (neat):  $\tilde{\nu} = 3481, 3080, 2964, 1737, 1652, 1237$   $\text{cm}^{-1}$ ; HRMS (ESI):  $m/z$  calcd for  $\text{C}_{16}\text{H}_{24}\text{O}_4 + \text{H}^+$ : 281.1753 [ $M + \text{H}^+$ ]; found: 281.1748; Anal. Calcd for  $\text{C}_{16}\text{H}_{24}\text{O}_4$ : C 68.54, H 8.63; found: C 68.32, H 8.61.

**(R)-4-((1R,2R,5S)-2-Hydroxy-2-methyl-5-(prop-1-en-2-yl)cyclopentyl)-2-methylenehept-6-ene-1,4-diol (13)**: To a stirring solution of above ketone (32.9 mg, 0.117 mmol) in  $\text{Et}_2\text{O}$  (2.3 mL) were added allylmagnesium bromide (1.0 M solution in  $\text{Et}_2\text{O}$ , 0.35 mL, 0.35 mmol) at  $0^\circ\text{C}$ . After stirring for 30 min, the mixture was diluted with  $\text{Et}_2\text{O}$ , washed with saturated aqueous  $\text{NH}_4\text{Cl}$  solution,  $\text{H}_2\text{O}$  and brine, dried over anhydrous  $\text{Na}_2\text{SO}_4$ , and then concentrated *in vacuo*. The residue was purified with flash column chromatography on silica gel (hexane/AcOEt = 3:2) to give triol **13** (27.8 mg, 85% yield) as a white scaly crystal. mp  $103\text{--}104^\circ\text{C}$ ;  $R_f = 0.25$  (hexane/AcOEt 3:2);  $[\alpha]_D^{25} +38.4$  ( $c$  1.33 in  $\text{CHCl}_3$ );  $^1\text{H NMR}$  (400 MHz,  $\text{CDCl}_3$ ):  $\delta = 5.96$  (m, 1H), 5.19 (s, 1H), 5.15 (d,  $J = 10.3$  Hz, 1H), 5.05 (d,  $J = 16.2$  Hz, 1H), 4.89 (s, 1H), 4.78 (s, 1H), 4.73 (s, 1H), 4.14 (d,  $J = 12.3$  Hz, 1H), 4.07 (d,  $J = 12.3$  Hz, 1H), 3.92–4.02 (m, 3H), 2.74 (d,  $J = 13.6$  Hz, 1H), 2.67 (m, 1H), 2.45 (m, 1H), 2.34 (d,  $J = 13.6$  Hz, 1H), 2.28 (d,  $J = 11.6$  Hz, 1H), 2.01 (dd,  $J = 8.1, 14.8$  Hz, 1H), 1.51–1.81 (m, 4H), 1.72 (s, 3H), 1.48 (s, 3H);  $^{13}\text{C NMR}$  (100 MHz,  $\text{CDCl}_3$ ):  $\delta = 147.8, 144.8, 134.8, 118.6, 118.4, 112.1, 81.5, 75.6, 67.3, 53.6, 46.1, 43.6, 41.9, 40.8, 26.0, 25.0, 17.4$ ; IR (KBr):  $\tilde{\nu} = 3292, 3075, 2959, 1699, 1653$   $\text{cm}^{-1}$ ; HRMS (ESI):  $m/z$  calcd for  $\text{C}_{17}\text{H}_{28}\text{O}_3 + \text{H}^+$ : 303.1936 [ $M + \text{H}^+$ ]; found: 303.1947; Anal. Calcd for  $\text{C}_{17}\text{H}_{28}\text{O}_3$ : C 72.82, H 10.06; found: C 72.83, H 10.07.

**(R)-5-Allyl-5-((1R,2R,5S)-2-hydroxy-2-methyl-5-(prop-1-en-2-yl)cyclopentyl)-3-methylenedihydrofuran-2(3H)-one (14)**: To a stirring solution of triol **13** (27.8 mg, 0.099 mmol) in  $\text{CH}_2\text{Cl}_2$  (5.0 mL) was added manganese dioxide (556 mg, 6.40 mmol) at room temperature. After stirring for 1 h, the mixture was diluted with  $\text{Et}_2\text{O}$ , passed through a pad of silica gel (hexane/AcOEt = 3:1), and then concentrated *in vacuo*. The residue was purified with flash column chromatography on silica gel (hexane/AcOEt = 3:1) to give  $\alpha$ -methylene-lactone **14** (27.4 mg, quantitative yield) as a colorless oil.  $R_f = 0.30$  (hexane/AcOEt 3:1);  $[\alpha]_D^{25} +3.81$  ( $c$  0.93 in  $\text{CHCl}_3$ ); UV  $\lambda_{\text{max}}$  (MeOH) nm ( $\epsilon$ ): 208 (8900, sh);  $^1\text{H NMR}$  (400 MHz,  $\text{CDCl}_3$ ):  $\delta = 6.23$  (t,  $J = 3.0$  Hz, 1H), 5.74 (ddt,  $J = 9.2, 17.0, 7.0$  Hz, 1H), 5.58 (t,  $J = 2.5$  Hz, 1H), 5.15 (d,  $J = 9.2$  Hz, 1H), 5.12 (d,  $J = 17.0$  Hz, 1H), 4.75 (s, 1H), 4.70 (s, 1H), 3.05 (dt,  $J = 17.6, 3.1$  Hz, 1H), 2.69 (dt,  $J = 17.6, 2.3$  Hz, 1H), 2.45–2.62 (m, 3H), 2.42 (d,  $J = 10.2$  Hz, 1H), 2.17 (brs, 1H), 1.77–1.89 (m, 2H), 1.75 (s, 3H),

1.52–1.68 (m, 2H), 1.33 (s, 3H);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ ):  $\delta$  = 169.8, 148.3, 135.0, 131.5, 122.4, 120.7, 111.8, 86.9, 81.0, 56.2, 46.4, 45.4, 41.9, 34.6, 27.6, 24.9, 18.4; IR (neat):  $\tilde{\nu}$  = 3444, 3075, 2955, 1746, 1649, 1272  $\text{cm}^{-1}$ ; HRMS (ESI):  $m/z$  calcd for  $\text{C}_{17}\text{H}_{24}\text{O}_3+\text{Na}^+$ : 299.1627 [ $M+\text{Na}^+$ ]; found: 299.1623; Anal. Calcd for  $\text{C}_{17}\text{H}_{24}\text{O}_3$ : C 73.88, H 8.75; found: C 73.78, H 8.74.

**(2R,3'R,3a'R,7a'S)-3'-Hydroxy-3',4,7'-trimethyl-1',2',3',3a',5',7a'-hexahydro-5H-spiro[furan-2,4'-inden]-5-one (8-*epi*-sinularianin B, 15)**: To a stirring solution of  $\alpha$ -methylene-lactone **14** (27.4 mg, 0.099 mmol) in EtOH (19.8 mL) was added rhodium chloride trihydrate (26.1 mg, 0.099 mmol) at room temperature and then refluxed for 5 h. The mixture was cooled to room temperature, diluted with EtOH, passed through a pad of silica gel ( $\text{Et}_2\text{O}$ ), and then concentrated *in vacuo* to give a crude product.

To a stirring solution of the above crude  $\alpha,\beta$ -unsaturated lactone in degassed toluene (19.8 mL) was added Grubbs 2nd generation catalyst (17.0 mg, 0.020 mmol) at room temperature. After stirring for 12 h, the reaction mixture was diluted with  $\text{Et}_2\text{O}$  and the residue was passed through a pad of silica gel (hexane/AcOEt = 12:1) and then concentrated *in vacuo*. The residue was purified with flash column chromatography on silica gel (hexane/AcOEt = 2:1) to give 8-*epi*-sinularianin B (**15**) (7.4 mg, 30% yield for 2 steps) as a colorless oil.  $R_f$  = 0.30 (hexane/AcOEt 2:1);  $[\alpha]_{\text{D}}^{25}$   $-52.2$  ( $c$  0.24 in  $\text{CHCl}_3$ ); UV  $\lambda_{\text{max}}$  (MeOH) nm ( $\epsilon$ ): 213 (5400, sh);  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ ):  $\delta$  = 7.19 (d,  $J$  = 1.5 Hz, 1H), 5.23 (s, 1H), 2.81 (m, 1H), 2.46 (d,  $J$  = 13.1 Hz, 1H), 2.33 (m, 1H), 1.89–1.99 (m, 4H), 1.91 (d,  $J$  = 1.5 Hz, 3H), 1.71 (s, 3H), 1.70 (m, 1H), 1.51 (m, 1H), 1.06 (s, 3H);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ ):  $\delta$  = 173.5, 150.8, 136.8, 129.2, 118.4, 88.2, 78.8, 58.8, 45.5, 40.8, 40.6, 27.9, 26.5, 19.8, 10.7; IR (neat):  $\tilde{\nu}$  = 3435, 2963, 2925, 2849, 1747, 1645  $\text{cm}^{-1}$ ; HRMS (ESI):  $m/z$  calcd for  $\text{C}_{15}\text{H}_{20}\text{O}_3+\text{Na}^+$ : 271.1310 [ $M+\text{Na}^+$ ]; found: 271.1317; Anal. Calcd for  $\text{C}_{15}\text{H}_{20}\text{O}_3$ : C 72.55, H 8.12; found: C 72.26, H 8.13.

**1-((1R,2R,5S)-2-Hydroxy-2-methyl-5-(prop-1-en-2-yl)cyclopentyl)but-3-en-1-one (16)**: To a stirring solution of aldehyde **11** (432 mg, 2.57 mmol) in THF (25.7 mL) were added 3-bromoprop-1-ene (777 mg, 6.42 mmol), Zn powder (841 mg, 12.9 mmol) and saturated aqueous  $\text{NH}_4\text{Cl}$  solution (51.4 mL) at room temperature. After stirring for 2 h,  $\text{H}_2\text{O}$  (25.7 mL) was slowly added. After stirring for 30 min, the mixture was diluted with  $\text{Et}_2\text{O}$ , passed through a pad of celite ( $\text{Et}_2\text{O}$ ). The resulting solution was washed with  $\text{H}_2\text{O}$  and brine, dried over anhydrous  $\text{Na}_2\text{SO}_4$ , and then concentrated *in vacuo* to give a crude product.

To a solution of IBX (1.44 g, 5.14 mmol) in DMSO (12.9 mL) was added a solution of the above crude product in THF (12.9 mL). After stirring for 3 h at room temperature,  $\text{H}_2\text{O}$  was added to the mixture. After diluting with  $\text{Et}_2\text{O}$ , the mixture was filtered through celite, washed with  $\text{H}_2\text{O}$  and brine, dried over anhydrous  $\text{Na}_2\text{SO}_4$ , and then concentrated *in vacuo*. The residue was purified with flash column chromatography on silica gel (hexane/AcOEt = 2:1) to give  $\beta,\gamma$ -unsaturated ketone **16** (493 mg, 92% yield for 2 steps) as a colorless oil.  $R_f$  = 0.25 (hexane/AcOEt 2:1);  $[\alpha]_{\text{D}}^{25}$   $-65.5$  ( $c$  1.00 in  $\text{CHCl}_3$ );  $^1\text{H}$  NMR (400

MHz, CDCl<sub>3</sub>):  $\delta$  = 5.92 (dddd,  $J$  = 6.8, 7.0, 10.2, 17.2 Hz, 1H), 5.17 (d,  $J$  = 10.2 Hz, 1H), 5.12 (d,  $J$  = 17.2 Hz, 1H), 4.69 (s, 2H), 3.35 (dd,  $J$  = 6.8, 17.1 Hz, 1H), 3.23 (dd,  $J$  = 7.0, 17.1 Hz, 1H), 3.05 (d,  $J$  = 10.4 Hz, 1H), 2.98 (d,  $J$  = 7.8 Hz, 1H), 1.95 (brs, 1H), 1.70–1.89 (m, 3H), 1.62–1.71 (m, 1H), 1.69 (s, 3H), 1.18 (s, 3H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta$  = 209.4, 146.4, 130.4, 118.8, 110.0, 81.3, 65.2, 49.4, 47.2, 42.3, 27.6, 25.5, 20.4; IR (neat):  $\tilde{\nu}$  = 3446, 3080, 1966, 1699, 1646 cm<sup>-1</sup>; HRMS (ESI):  $m/z$  calcd for C<sub>13</sub>H<sub>20</sub>O<sub>2</sub>+Na<sup>+</sup>: 231.1361 [ $M$ +Na<sup>+</sup>]; found: 231.1370; Anal. Calcd (%) for C<sub>13</sub>H<sub>20</sub>O<sub>2</sub>: C 74.96, H 9.68; found: C 75.23, H 9.80.

**(3R,3aR,4R,7aS)-4-(tert-Butyldimethylsilyloxy)-3-hydroxy-3,7-dimethyl-2,3,3a,4,5,7a-hexahydro-1H-indene-4-carbaldehyde (20)**: To a stirring solution of aldehyde **11** (121 mg, 0.718 mmol) in THF (109 mL) were added TBSCN (122 mg, 0.862 mmol), and PNPCI (41.3 mg, 0.0718 mmol) at room temperature. After stirring for 2 h, TMSCN (0.14 mL, 1.12 mmol) was added at the same temperature. The mixture was concentrated *in vacuo*, and the residue was passed through a pad of silica gel (hexane/AcOEt = 15:1) and then concentrated *in vacuo* to give a crude TMS-protected cyanohydrin **18**.

A solution of the above crude TMS-protected cyanohydrin **18** in THF (4.00 mL) was added to a pre-prepared LDA (0.39 M solution in THF, 4.97 mL, 1.94 mmol) at -78 °C and then allowed to warm to 0 °C. After stirring for 40 min, the mixture was cooled to -78 °C, and then pre-mixed allyl bromide (0.31 mL, 3.59 mmol) and HMPA (0.31 mL, 1.80 mmol) was introduced to the mixture. After stirring for 15 min at the same temperature, the reaction mixture was diluted with Et<sub>2</sub>O, washed with saturated aqueous NH<sub>4</sub>Cl solution, H<sub>2</sub>O and brine, dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>, and then concentrated *in vacuo*. The residue was passed through a pad of silica gel (hexane/AcOEt = 30:1) and then concentrated *in vacuo* to give a crude diene **19**.

To a stirring solution of the above crude diene **19** in degassed 1,2-dichloroethane (144 mL) was added Grubbs 2nd generation catalyst (61.0 mg, 0.0718 mmol) at room temperature, and then refluxed. After stirring for 6 h, cooled to room temperature, and then added DMSO (0.51 mL, 7.18 mmol). After stirring for 12 h, the solvent was removed *in vacuo*, the residue was diluted with Et<sub>2</sub>O and the residue was passed through a pad of silica gel (hexane/AcOEt = 12:1) and then concentrated *in vacuo* to give a crude product. To a stirring solution of the above crude product in toluene (14.4 mL) was added DIBAH (1.02 M solution in hexane, 1.41 mL, 1.44 mmol) at -78 °C and then allowed to warm to 0 °C. After stirring for 30 min, the mixture was cooled to -78 °C, and then Et<sub>2</sub>O (14.4 mL), saturated aqueous NH<sub>4</sub>Cl solution (7.18 mL), and 0.5 M aqueous H<sub>2</sub>SO<sub>4</sub> solution (14.4 mL) were introduced to the mixture. The mixture was then allowed to warm to room temperature and then the reaction mixture was diluted with Et<sub>2</sub>O, washed with saturated aqueous NH<sub>4</sub>Cl solution, H<sub>2</sub>O and brine, dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>, and then concentrated *in vacuo*. The residue was passed through a pad of silica gel (hexane/AcOEt = 15:1) and then concentrated *in vacuo*

to give a crude product.

To a stirring solution of the above crude product in THF (7.18 mL) were added acetic acid (0.82 mL, 14.4 mmol) and TBAF (1.0 M solution in THF, 7.18 mL, 7.18 mmol) at room temperature, and then allowed to warm to 40 °C. After stirring for 5 h, the reaction mixture was diluted with Et<sub>2</sub>O, washed with saturated aqueous NaHCO<sub>3</sub> solution, H<sub>2</sub>O and brine, dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>, and then concentrated *in vacuo*. The residue was purified with flash column chromatography on silica gel (hexane/AcOEt = 4:1) to give  $\alpha$ -siloxyaldehyde **20** (156 mg, 67% yield for 5 steps) as a white needle-like crystalline solid.  $R_f$  = 0.60 (hexane/AcOEt 2:1); mp 77–78 °C (recrystallized from hexane/AcOEt);  $[\alpha]_D^{25}$  +9.70 ( $c$  1.51 in CHCl<sub>3</sub>); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  = 9.77 (s, 1H), 5.21 (s, 1H), 2.58 (m, 1H), 2.43 (m, 1H), 1.99 (d,  $J$  = 12.6 Hz, 1H), 1.96–1.73 (m, 4H), 1.69 (3H, s), 1.45 (m, 1H), 1.30 (s, 3H), 0.85 (s, 9H), 0.18 (s, 3H), 0.02 (s, 3H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta$  = 204.0 (d), 137.4 (s), 116.5 (d), 81.3 (s), 78.4 (s), 58.9 (d), 41.5 (t), 39.8 (d), 35.3 (t), 27.9 (q), 25.8 (q)×3, 25.6 (t), 20.2 (q), 18.6 (s), –2.5 (q), –3.1 (q); IR (KBr):  $\tilde{\nu}$  = 3420, 2956, 2931, 2857, 1733, 1653 cm<sup>-1</sup>; HRMS (ESI):  $m/z$  calcd for C<sub>10</sub>H<sub>16</sub>O<sub>2</sub>+Na<sup>+</sup>: 347.2018 [ $M$ +Na<sup>+</sup>]; found: 347.2023; Anal. Calcd for C<sub>18</sub>H<sub>32</sub>O<sub>3</sub>Si: C 66.62, H 9.94; found: C 66.54, H 9.90.

**(4a*S*,4a<sup>1</sup>*R*,7a*S*,9a*R*)-4a-(*tert*-Butyldimethylsilyloxy)-3,7,9a-trimethyl-4a,5,7a,8,9,9a-hexahydroindeno[1,7-*bc*]oxepin-2(4a<sup>1</sup>*H*)-one (21)**: To a stirring solution of  $\alpha$ -siloxyaldehyde **20** (101 mg, 0.311 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (6.22 mL) were added 2-(diethoxyphosphoryl)propanoic acid (211 mg, 0.933 mmol), and WSC (179 mg, 0.933 mmol) at room temperature. After stirring for 30 min, the mixture was diluted with Et<sub>2</sub>O, washed with saturated aqueous NaHCO<sub>3</sub> solution, H<sub>2</sub>O and brine, and then concentrated *in vacuo*. The residue was passed through a pad of silica gel (hexane/AcOEt = 3:2) and then concentrated *in vacuo* to give a crude product.

To a stirring solution of the above crude product in THF (62.4 mL) was added KO<sup>t</sup>Bu (48.9 mg, 0.435 mmol) at room temperature and then allowed to warm to 60 °C. After stirring for 2 h, the mixture was cooled to room temperature and then the reaction mixture was diluted with Et<sub>2</sub>O, washed with saturated aqueous NH<sub>4</sub>Cl solution, H<sub>2</sub>O and brine, and then concentrated *in vacuo*. The residue was purified with flash column chromatography on silica gel (hexane/AcOEt = 12:1) to give  $\alpha,\beta$ -unsaturated lactone **21** (88.1 mg, 78% yield for 2 steps) as a colorless oil.  $R_f$  = 0.35 (hexane/AcOEt 12:1);  $[\alpha]_D^{25}$  +162 ( $c$  1.29 in CHCl<sub>3</sub>); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  = 6.20 (s, 1H), 5.14 (s, 1H), 2.46 (m, 1H), 2.31 (dd,  $J$  = 11.4, 13.8 Hz, 1H), 2.24 (m, 1H), 2.13 (m, 1H), 2.08 (d,  $J$  = 12.7 Hz, 1H), 2.05 (s, 3H), 1.96 (m, 1H), 1.86 (dt,  $J$  = 14.8, 8.8 Hz, 1H), 1.68 (s, 3H), 1.51 (s, 3H), 1.49 (m, 1H), 0.83 (s, 9H), 0.11 (s, 6H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta$  = 167.9 (s), 140.7 (d), 137.0 (s), 127.7 (s), 117.7 (d), 85.8 (s), 73.1 (s), 58.7 (d), 41.3 (d), 41.1 (t), 40.7 (t), 26.5 (t), 25.7 (q)×3, 24.6 (q), 24.1 (q), 20.1 (q), 18.4 (s), –2.1 (q), –2.4 (q); IR (neat):  $\tilde{\nu}$  = 2957, 2930, 2857, 1698, 1684, 1254 cm<sup>-1</sup>; UV/Vis:  $\lambda_{\max}$  (MeOH)/nm 215sh ( $\epsilon$ /dm<sup>3</sup> mol<sup>-1</sup> cm<sup>-1</sup> 9100); HRMS (ESI):  $m/z$

calcd for  $C_{21}H_{34}O_3Si+H^+$ : 363.2355 [ $M+H^+$ ]; found: 363.2350; Anal. Calcd for  $C_{21}H_{34}O_3Si$ : C 69.56, H 9.45; found: C 69.39, H 9.49.

**Sinularianin B (1)**: To a stirring solution of  $\alpha,\beta$ -unsaturated lactone **21** (71.0 mg, 0.196 mmol) in THF (1.96 mL) was added TBAF (1.0 M solution in THF, 0.49 mL, 0.490 mmol) at room temperature. After stirring for 2 h, a suspension of 4Å molecular sieves (14.0 mg) and  $K_2CO_3$  (271 mg, 1.96 mmol) in MeOH (9.80 mL) was added and the mixture was warmed to 40 °C. After stirring for 36 h, the solvent was removed *in vacuo*. The residue was passed through a pad of silica gel ( $Et_2O$ ) and then concentrated *in vacuo*. The residue was purified with flash column chromatography on silica gel (hexane/AcOEt = 2:1) to give sinularianin B (**1**) (48.0 mg, 99% yield) as a colorless oil.  $R_f$  = 0.15 (hexane/AcOEt 2:1);  $[\alpha]_D^{25}$  -111 ( $c$  1.55 in  $CHCl_3$ );  $^1H$  NMR (500 MHz,  $CDCl_3$ ):  $\delta$  = 7.16 (d,  $J$  = 1.5 Hz, 1H), 5.24 (dd,  $J$  = 1.9, 2.3 Hz, 1H), 2.57 (m, 1H), 2.53 (m, 1H), 1.98 (d,  $J$  = 12.8 Hz, 1H), 1.95–1.75 (m, 4H), 1.93 (d,  $J$  = 1.5 Hz, 3H), 1.71 (s, 3H), 1.47 (brs, 1H), 1.43 (m, 1H), 1.12 (s, 3H);  $^{13}C$  NMR (125 MHz,  $CDCl_3$ ):  $\delta$  = 173.7 (s), 152.2 (d), 137.3 (s), 129.1 (s), 117.2 (d), 85.4 (s), 78.4 (s), 56.6 (d), 41.8 (t), 40.9 (d), 39.7 (t), 26.0 (q), 25.3 (t), 20.2 (q), 10.6 (q); IR (neat):  $\tilde{\nu}$  = 3445, 2962, 1734, 1658  $cm^{-1}$ ; UV/Vis  $\lambda_{max}$  (MeOH)/nm 213sh ( $\epsilon$  / $dm^3 mol^{-1} cm^{-1}$  9600); HRMS (ESI):  $m/z$  calcd for  $C_{15}H_{20}O_3+Na^+$ : 271.1310 [ $M+Na^+$ ]; found: 271.1313; Anal. Calcd for  $C_{15}H_{20}O_3$ : C 72.55, H 8.12; found: C 72.37, H 7.93.

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