

HETEROCYCLES, Vol. 104, No. 9, 2022, pp. 1559 - 1564. © 2022 The Japan Institute of Heterocyclic Chemistry
 Received, 17th June, 2022, Accepted, 13th July, 2022, Published online, 13th July, 2022
 DOI: 10.3987/COM-22-14698

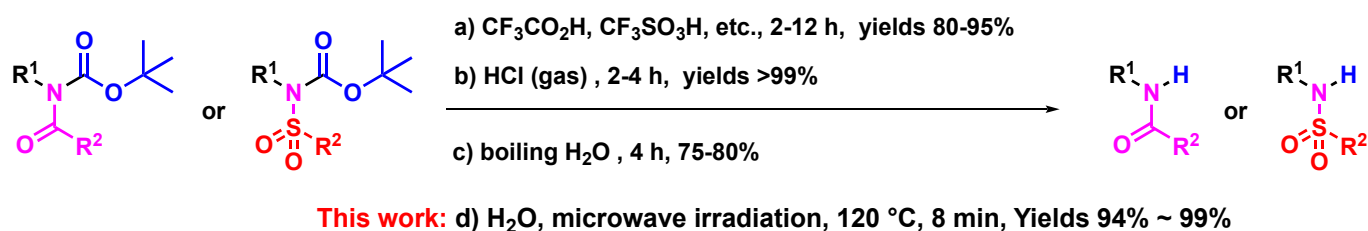
FACILE AND EFFICIENT *N*-BOC DEPROTECTION OF AMIDE AND SULFONAMIDE UNDER MICROWAVE IRRADIATION IN WATER

Zhi-you Huang,^{1*} Wen-sheng Li,² Jia Xu,¹ Bing-rong Xiang,¹
 Min-xin Li,¹ Li-jun Chen,¹ and Ye-she Wang¹

¹College of Urban and Rural Construction, Shaoyang University, Shaoyang, 422000, P.R.China; ²Shaoyang City Sixteen middle School, Shaoyang, 422000, P.R.China. Email Address: huangzhiyou@syxy247.wecom.work (Zhi-you Huang).

Abstract – A facile and efficient microwave-assisted *N*-Boc deprotection of amide and sulfonamide in water was developed. With environmentally friendly, operational-simplicity, good substrate scope and excellent product yields, this methodology is superior to the existing approach. Significantly, the target compound was obtained after filtration and dry.

The *tert*-butyloxycarbonyl (Boc) group is one of the most important protection group in the core fragment amide and sulfonamide.¹⁻⁵ Also, the application of *N*-Boc protection in the organic synthesis was widely due to the mild reaction conditions and high yields.⁶⁻⁸ Herein, the removal of the Boc group remains of vital importance. While, the most general approach to the *N*-Boc deprotection of amide and sulfonamide was under the acid conditions (**Scheme 1, method a**).⁹⁻¹¹ Recently, the alternative deprotection has been developed such as HCl gas and high-temperature water (**Method b, c**).¹³⁻¹⁷ However, the **methods a** and **b** could produce the waste acid (The HCl gas was released from the reaction of H₂SO₄ and NaCl). Additionally, the **method c** offers moderate yields in 4 hours. Furthermore, the organic solvent was inevitably used in the chemical synthesis such as the post-manipulations, purification steps.¹⁸⁻²⁰ Thus, the facile, eco-friendly strategy with excellent yields was strongly desired. In our previous works, the

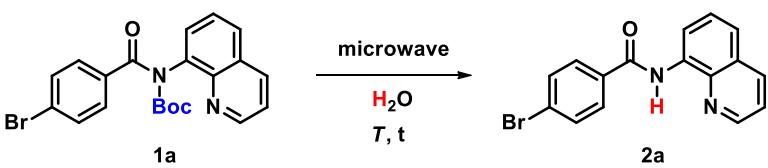


Scheme 1. Selected *N*-Boc deprotection of amide and sulfonamide

microwave irradiation could accelerate the reaction and enhance the yields.²¹⁻²³ Herein, the green solvent water was used to decompose the *N*-Boc in amide and sulfonamide under microwave irradiation with operational-simplicity, environmentally friendly, excellent yields and good substrate scope (**Method d**). Significantly, the target compound was obtained after filtration and dry.

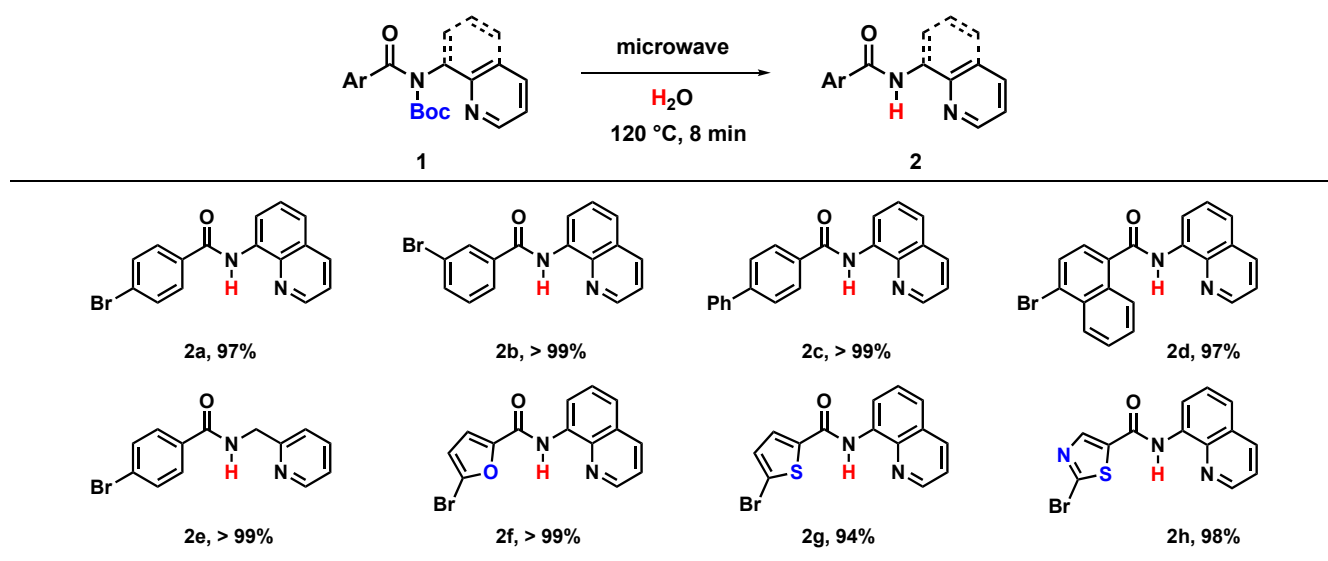
Although the deprotection of *N*-Boc is known, it is still great challenge to find the optimum condition to perform microwave-assisted approach in water. The reaction of **1a** and water was chosen as a benchmark reaction (**Table 1**). Unfortunately, the *N*-Boc deprotection was not performed at 60 °C and 80 °C (**Table 1, entries 1-2**). Then, 52% of **2a** was obtained with a temperature of 100 °C in 10 minutes (**Entry 3**). Therefore, the reaction temperature was enhanced to improve the yields. Thus, the yield of **2a** was increased with an increase of temperature (**Entries 4-6**). Then, the reaction yields of 8 minutes were higher than that of 6 minutes (**Entries 7, 8**). Additionally, the conversion of **1a** was completely at 130 °C in 6 minutes (**Entry 9**). Compared with conventional heating method (**Entry 10**), microwave irradiation could significantly accelerate the reaction and notably improve the yield. As a result, the combination of 0.5 mmol **1a** and 3.0 mL water with a temperature of 120 °C in 8 minutes was fixed (**Entry 7**), as optimal conditions.

Table 1. The optimization microwave-assisted approach in water^a



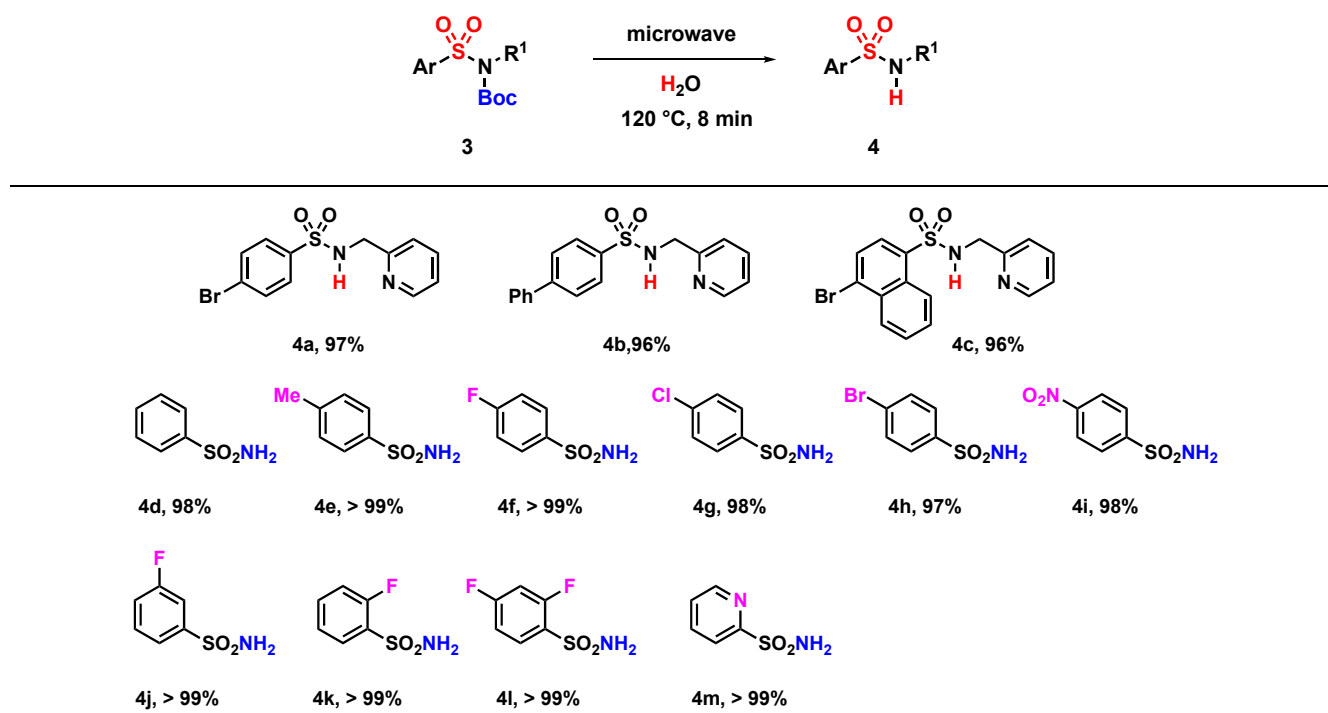
Entry	<i>T</i> / (°C)	<i>t</i> / (min)	Yield (2a) / % ^b
1	60	10	0
2	80	10	0
3	100	10	52
4	110	10	85
5	120	10	97
6	130	10	97
7	120	8	97
8	120	6	93
9	130	6	97
10 ^c	100	180	79

^a Reactions conditions : **1a** 0.5 mmol, **water** 3.0 mL, microwave irradiation; ^b Isolated yield; ^c Conventional heating.

Table 2. Microwave-assisted *N*-Boc deprotection of amide (**1**)^a

^a Reactions conditions : **1** 0.5 mmol, **water** 3.0 mL, microwave irradiation; ^b Isolated yield.

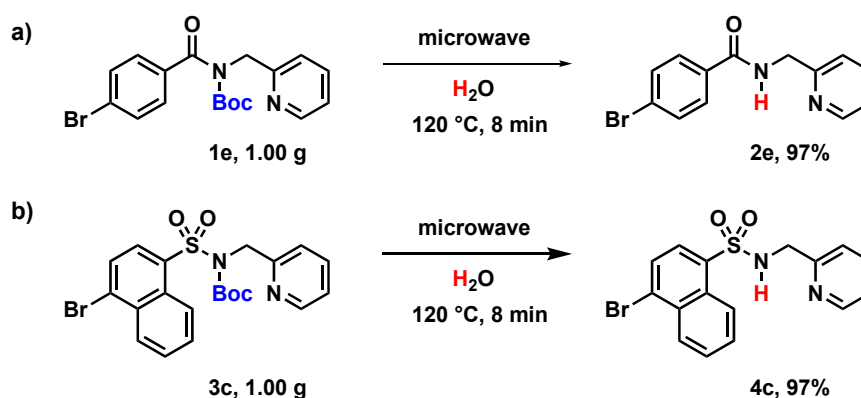
Once the optimized reaction conditions were identified, the limitation and scope of the microwave-assisted approach were examined. Various *N*-Boc aromatic amides were explored, the result was shown in **Table 2**. The deprotection of *N*-Boc benzamide (**2a**, **2b**, **2e**), biphenylamide (**2c**) and naphthylamide (**2d**) always went smoothly and the yields were excellent. In addition, the five-membered heteroaryl carboxamide were also afforded (**2f-2h**) in excellent yields.

Table 3. Microwave-assisted *N*-Boc deprotection of sulfonamide (**3**)^a

^a Reactions conditions : **3** 0.5 mmol, **water** 3.0 mL, microwave irradiation; ^b Isolated yield.

To increase the scope of microwave-assisted method, the *N*-Boc aromatic sulfonamide (**3**) was tested (**Table 3**). The deprotection of **3a-3c** (R^1 was pyridin-2-ylmethyl) in water gave excellent yields. Furthermore, it was obvious that the deprotection of **3d-3m** (R^1 was hydrogen) took place extremely smoothly. Both electron-donating and electron-withdrawing groups, such as 4-methyl (**4e**) and 4-nitro (**4i**), 4-halogen (**4f-4h**) afforded the desired products in excellent yields. More importantly, the group in different substituted position had no effect on the reaction, such as 3-fluoro (**4j**) and 2-fluoro (**4k**). In addition, the *N*-Boc-2,4-difluorobenzenesulfonamide (**3l**) was also investigated and afforded the corresponding product (**4l**). Finally, the deprotection of **3m** could offer the desired compound **4m** in excellent yields.

To fully demonstrate the applicability of this methodology, a scale-up reaction for **1e** and **3c** was carried out with three times. The yields were excellent (**Scheme 2**).



Scheme 2. Gram-scale synthesis

More importantly, the deprotection mechanism was shown in **Scheme 3**. Firstly, the water attacked the amid carbonyl group in high temperature under microwave irradiation.^{14,15} Then, **5** could be decomposed into target molecule (**4**) and **6** (**Route A**). Subsequently, **6** was attacked by water and CO_2 , H_2O and tertiary butanol or 2-methylpropene was released. Additionally, **5** could also be decomposed into **7** and tertiary butanol or 2-methylpropene (**Route B**). Subsequently, the water attacked the carboxy group of **7**. Finally, the target molecule **4** was obtained.

4. Y. Hu, X. Yan, Y. Shen, M. Di, and J. Wang, *Ecotoxicol. Environ. Saf.*, 2018, **157**, 150.
5. J. W. Fulp, L. He, S. Toldo, Y. Jiang, A. Boice, C. Guo, X. Li, A. Rolfe, D. Sun, and A. Abbate, *J. Med. Chem.*, 2018, **61**, 5412.
6. K. Takamatsu, K. Hirano, and M. Miura, *Org. Lett.*, 2015, **17**, 4066.
7. E. P. Talbot, T. D. Fernandes, J. M. McKenna, and F. D. Toste, *J. Am. Chem. Soc.*, 2014, **136**, 4101.
8. M. Nishino, K. Hirano, T. Satoh, and M. Miura, *Angew. Chem.*, 2013, **125**, 4553.
9. D. C. Johnson and T. S. Widlanski, *Tetrahedron Lett.*, 2004, **45**, 8483.
10. B. Li, R. Bemish, R. A. Buzon, C. K. F. Chiu, S. T. Colgan, W. Kissel, T. Le, K. R. Leeman, L. Newell, and J. Roth, *Tetrahedron Lett.*, 2003, **44**, 8113.
11. D. S. Coffffey, M. K. N. Hawk, S. W. Pedersen, S. J. Ghera, P. G. Marler, P. N. Dodson, and M. L. Lytle, *Org. Process Res. Dev.*, 2004, **8**, 945.
12. P. Strazzolini, N. Misuri, and P. Polese, *Tetrahedron Lett.*, 2005, **46**, 2075.
13. R. H. Verschuere, P. Gilles, S. V. Mileghem, and W. M. De Borggraeve, *Org. Biomol. Chem.*, 2021, **19**, 5782.
14. G. Wang, C. Li, J. Li, and X. Jia, *Tetrahedron Lett.*, 2009, **50**, 1438.
15. J. Wang, Y. L. Liang, and J. Qu, *Chem. Commun.*, 2009, 5144.
16. A. R. Bogdan, M. Charaschanya, A. W. Dombrowski, Y. Wang, and S. W. Djuric, *Org. Lett.*, 2016, **18**, 1732.
17. B. Li, R. Li, P. Dorffff, J. C. McWilliams, R. M. Guinn, S. M. Guinness, L. Han, K. Wang, and S. Yu, *J. Org. Chem.*, 2019, **84**, 4846.
18. G. Pavan Kumar, D. Rambabu, M. V. Basaveswara Rao, and M. Pal, *J. Chem.*, 2013, **2013**, Article ID 916960.
19. S. S. Bhawal, R. A. Patil, and D. W. Armstrong, *RSC Adv.*, 2015, **5**, 95854.
20. D. Margetić and M. Đud, *Int. J. Org. Chem.*, 2017, **7**, 140.
21. Z. Y. Huang, M. Liu, Y. J. Mao, Y. D. Chen, Y. P. Wang, and C. Liu, *Tetrahedron Lett.*, 2019, **60**, 626.
22. Z. Y. Huang, J. F. Yang, K. Song, Q. Chen, S. L. Zhou, G. F. Hao, and G. F. Yang, *J. Org. Chem.*, 2016, **81**, 9647.
23. Z. Y. Huang, J. F. Yang, Q. Chen, R. J. Cao, W. Huang, G. F. Hao, and G. F. Yang, *RSC Adv.*, 2015, **5**, 75182.